

Electrophilically activated nitroalkanes in double annulation of [1,2,4]triazolo[4,3-a]quinolines and 1,3,4-oxadiazole rings

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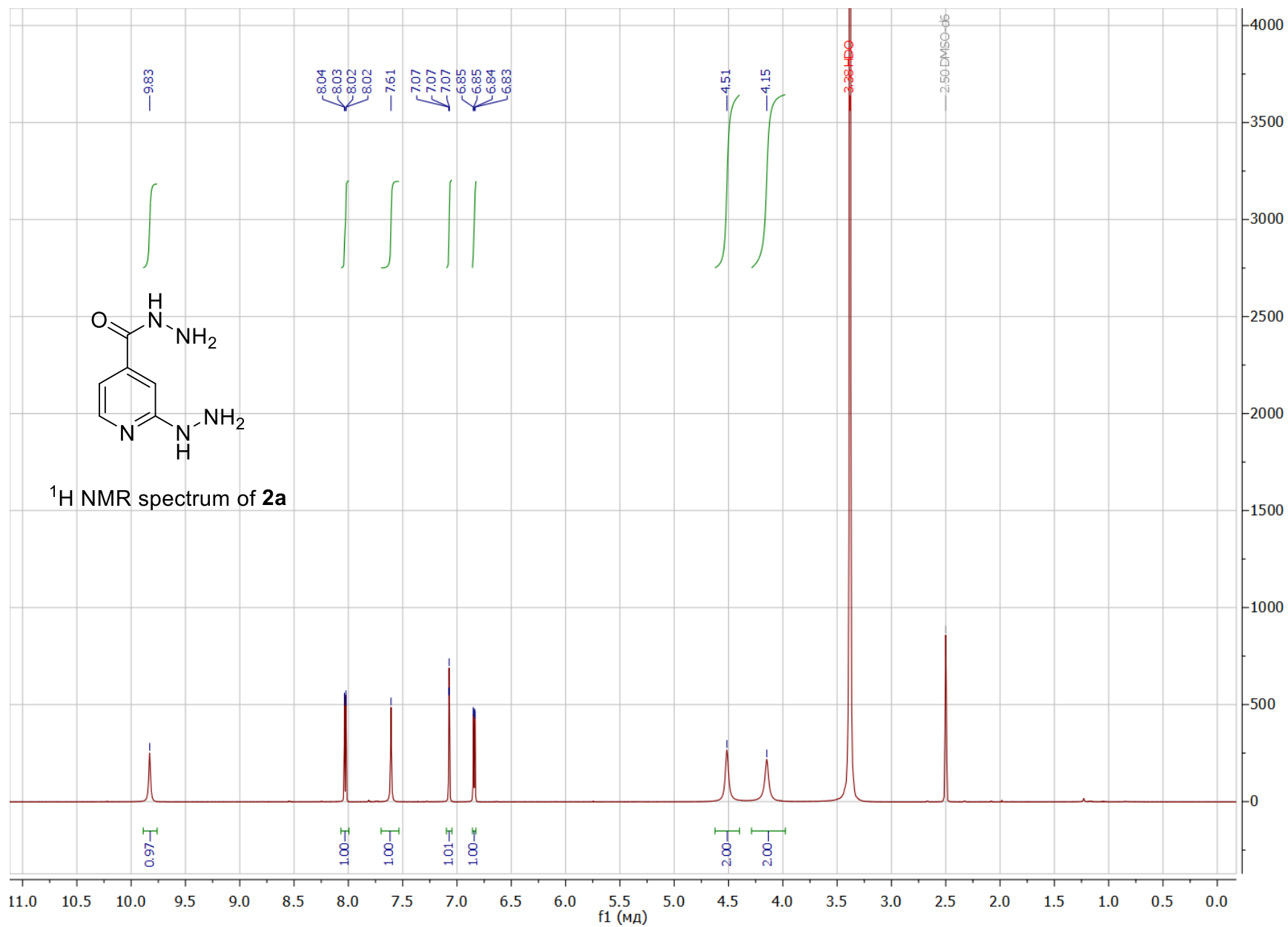
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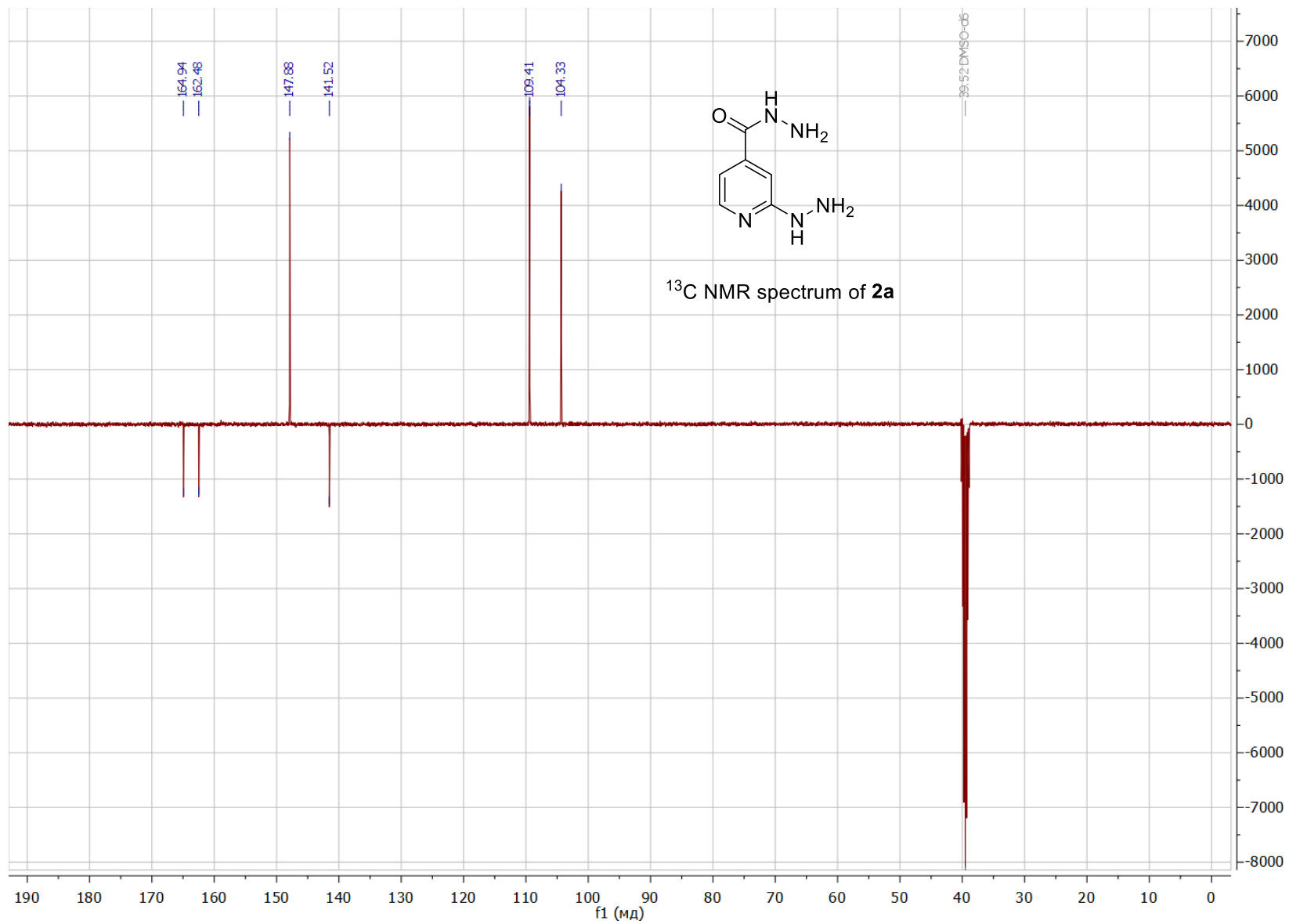
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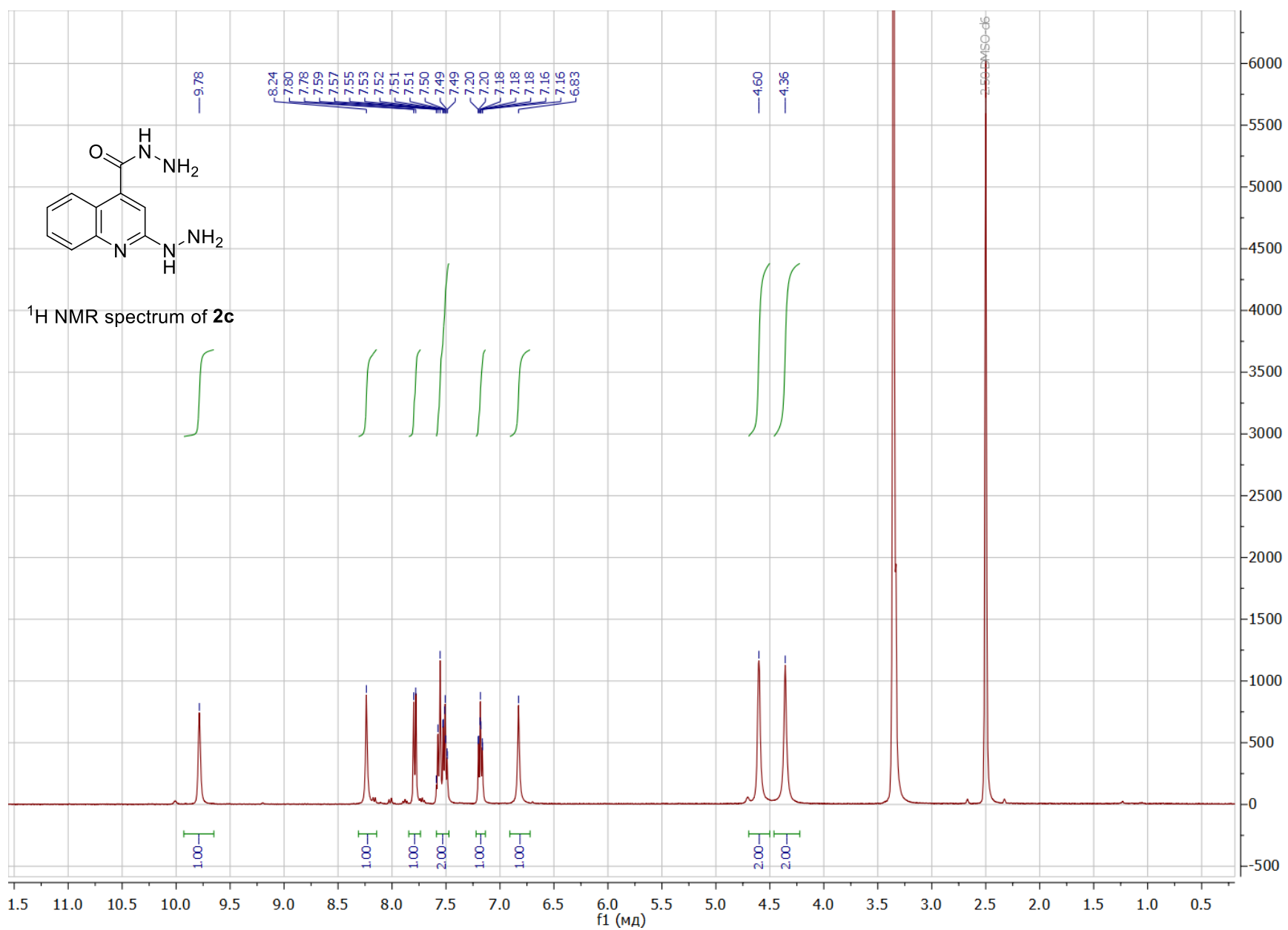
Supporting Information

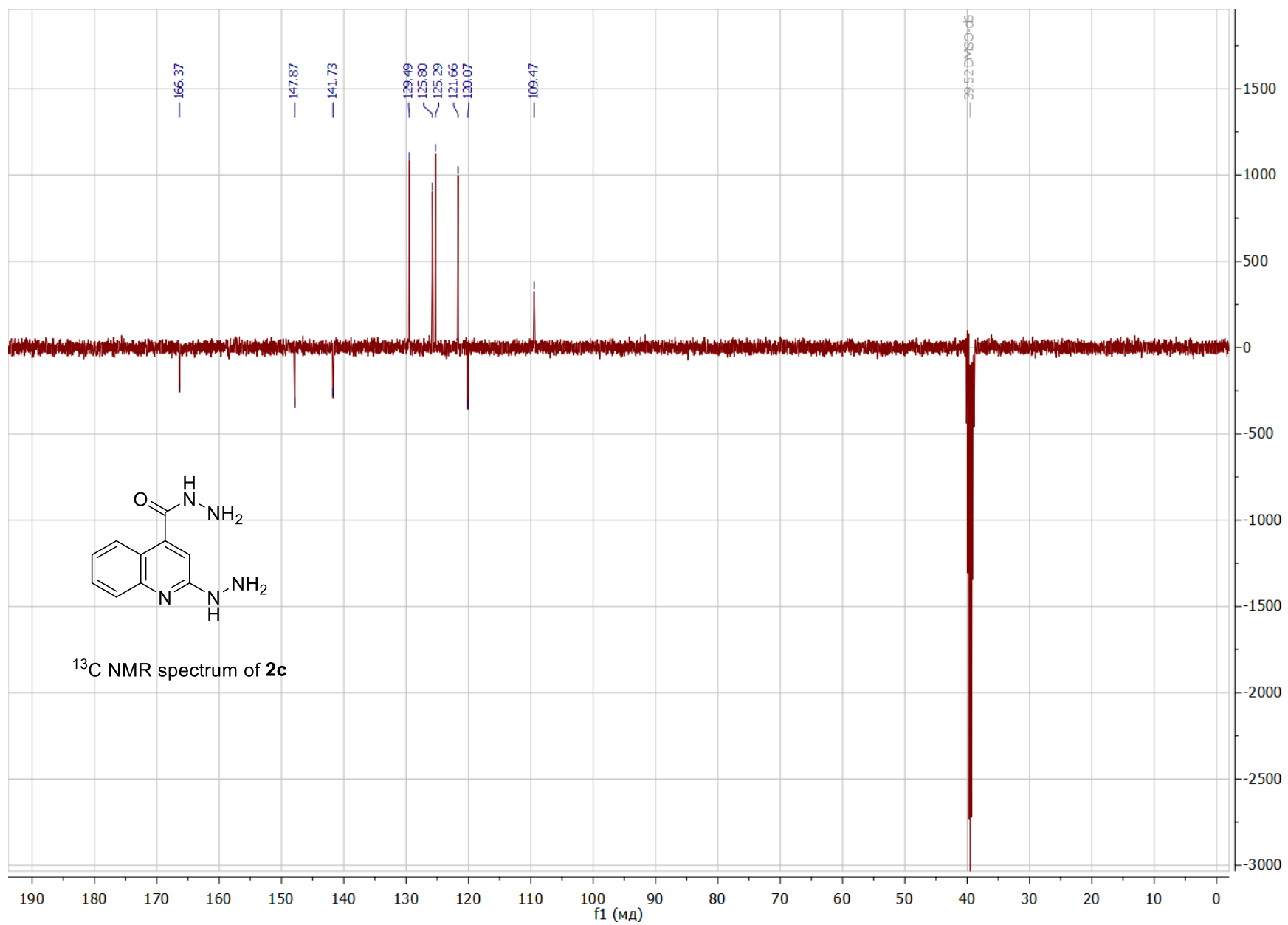
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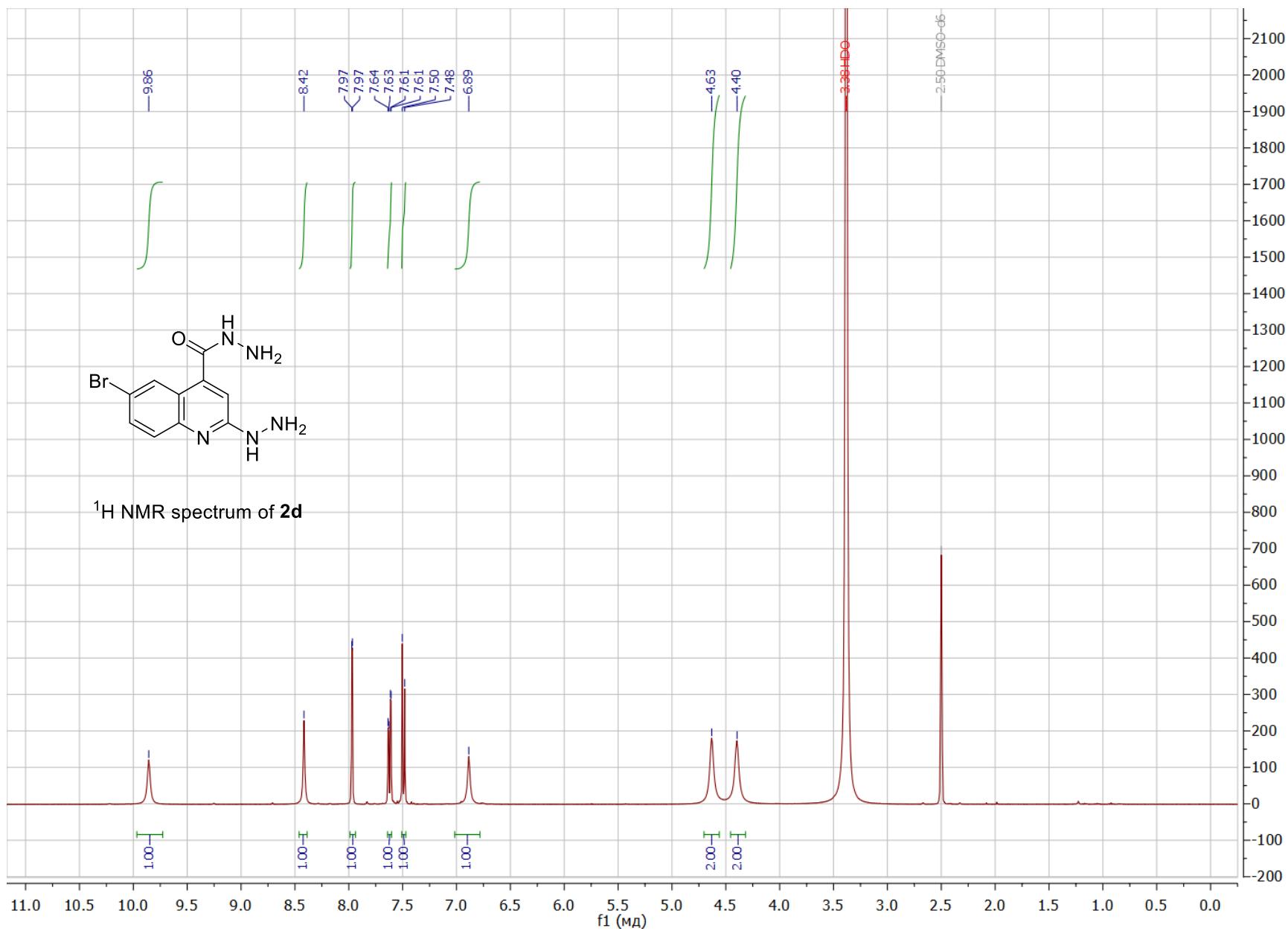
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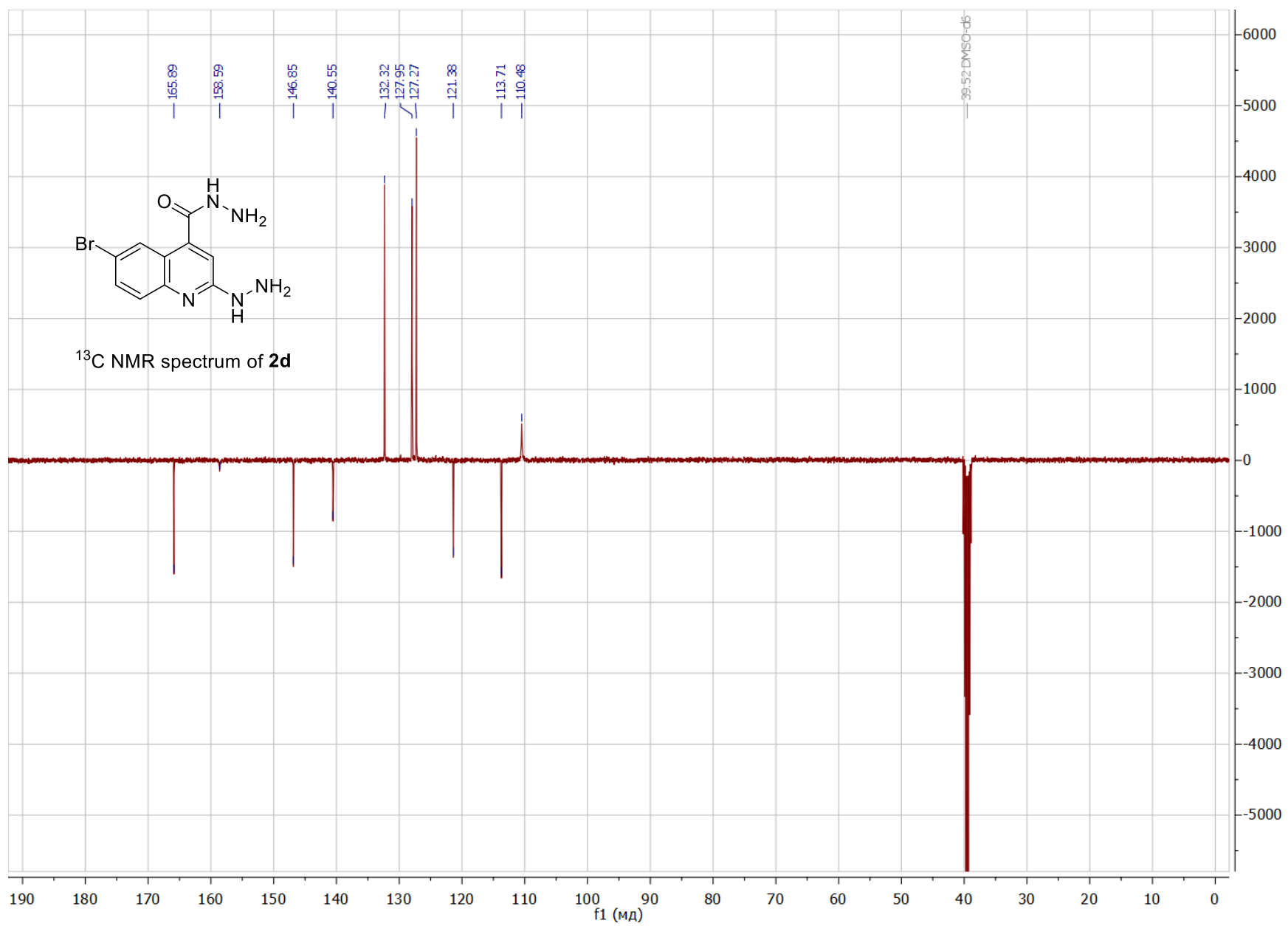


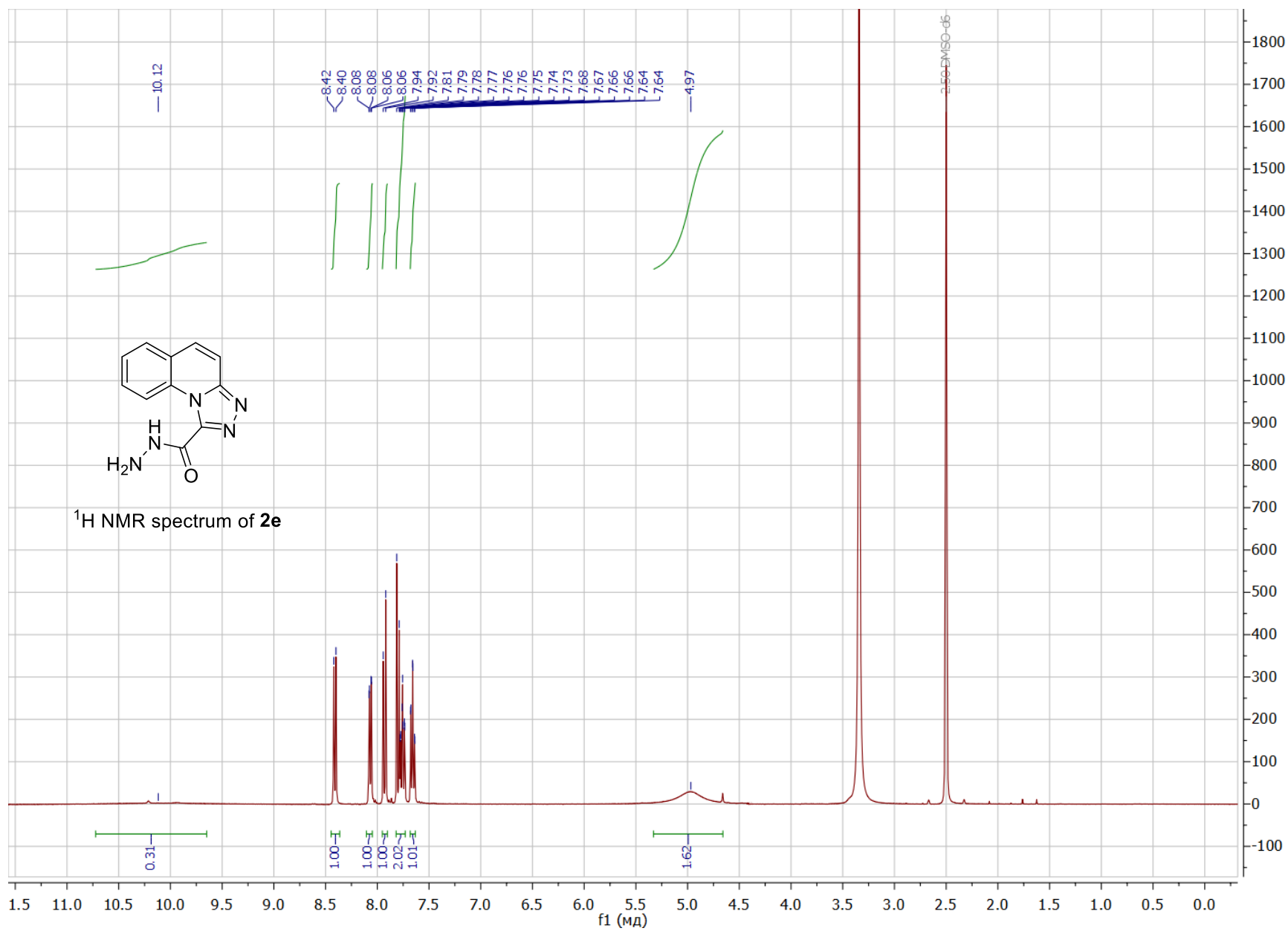


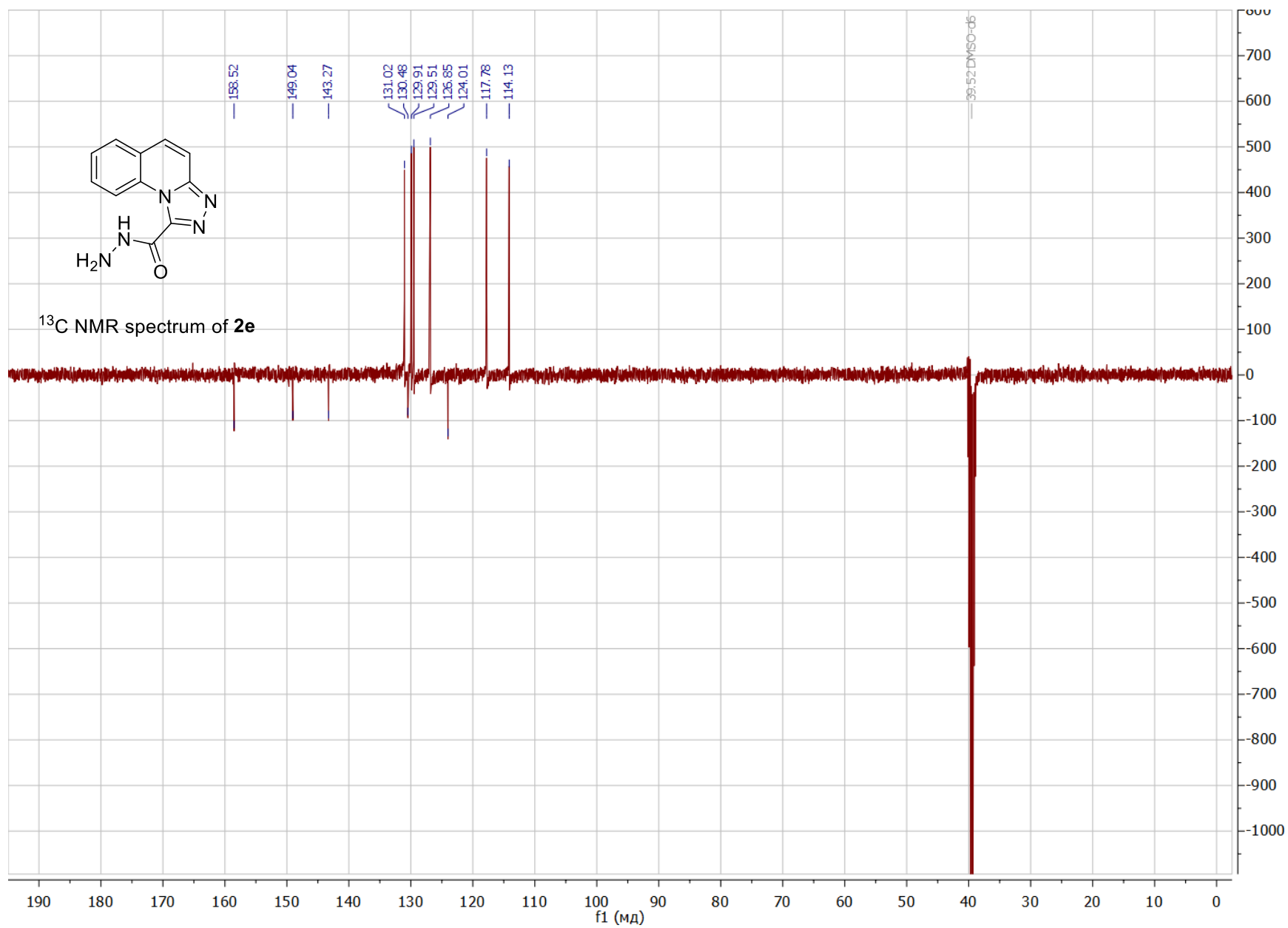


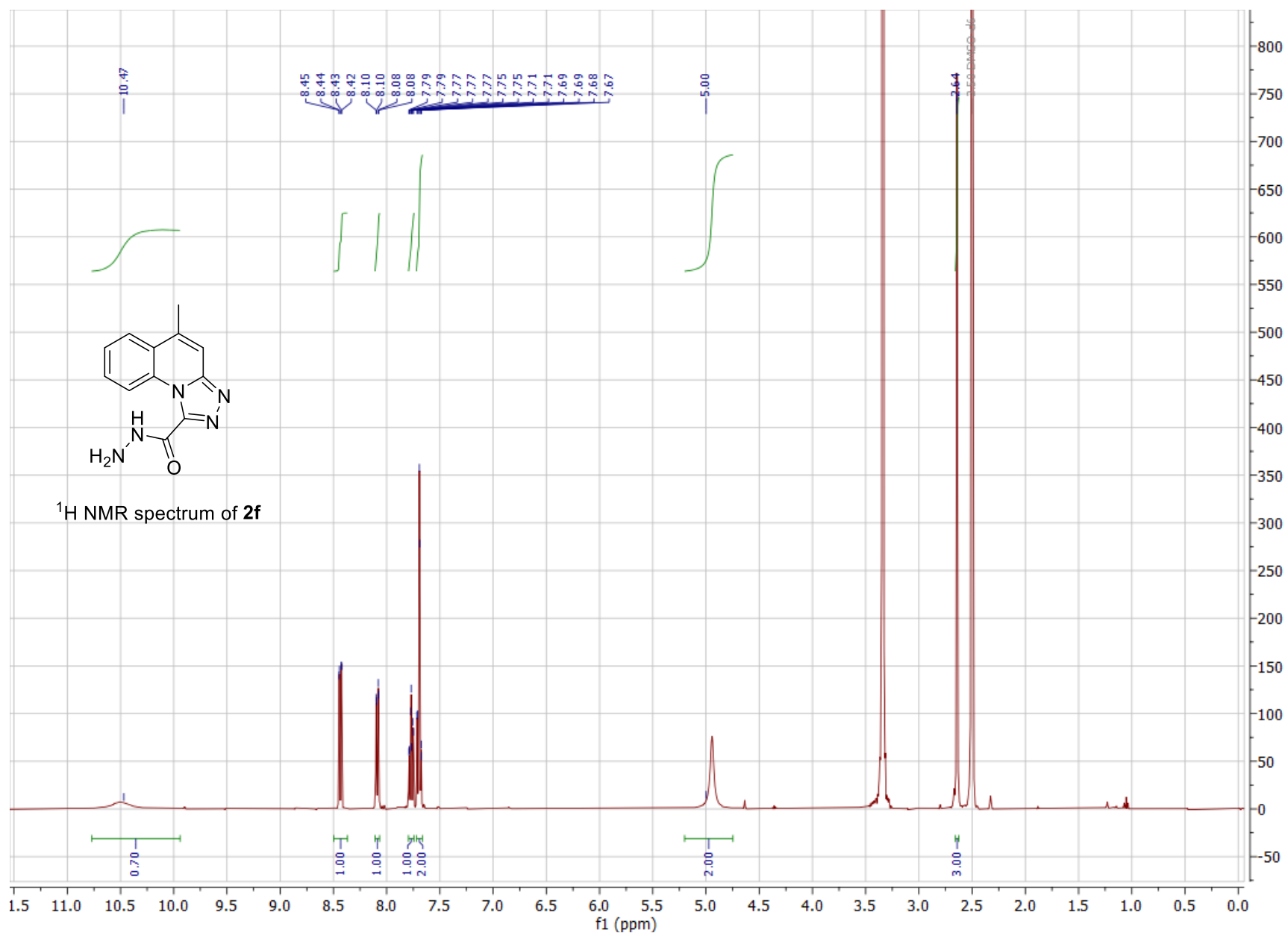


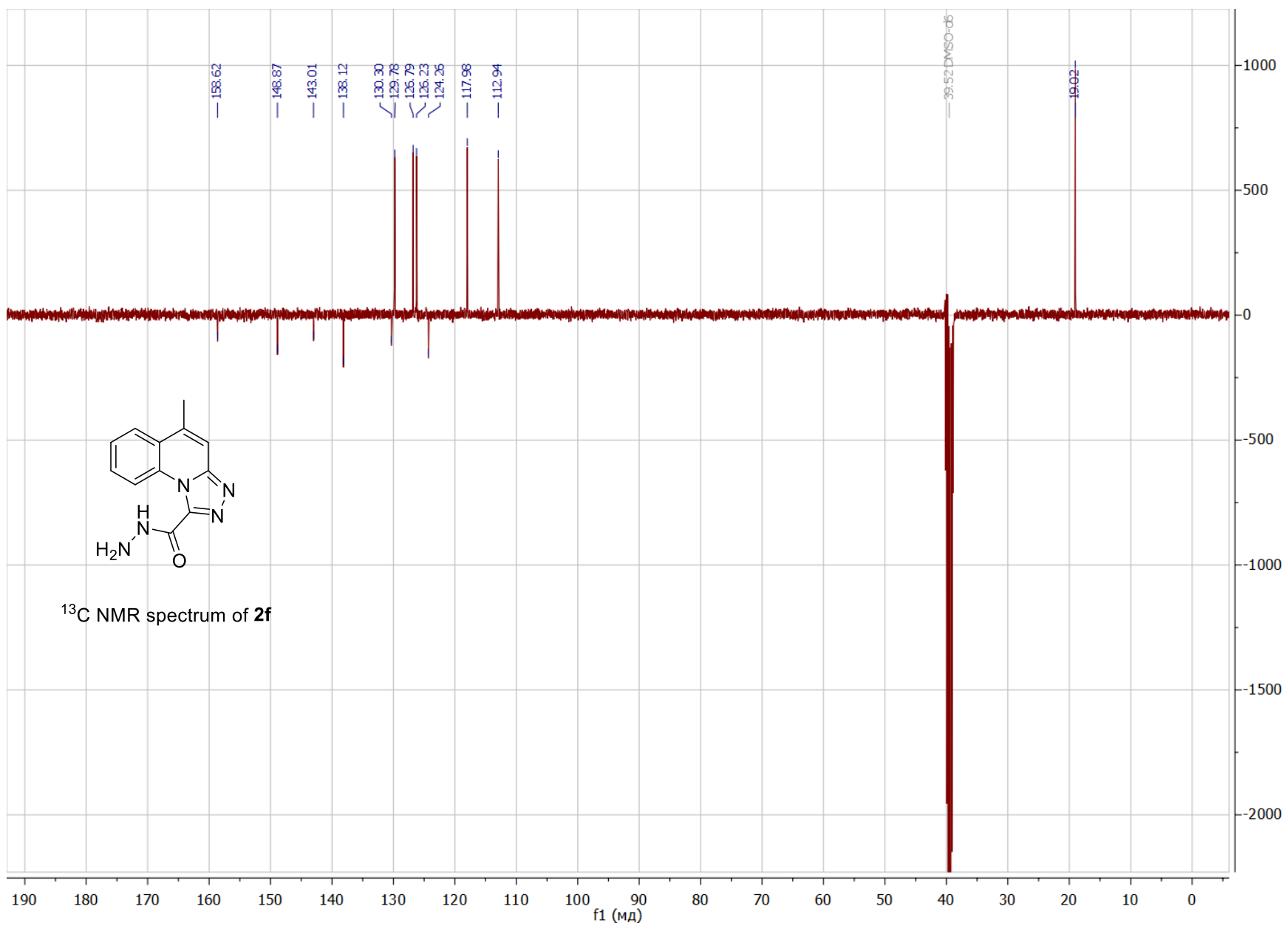










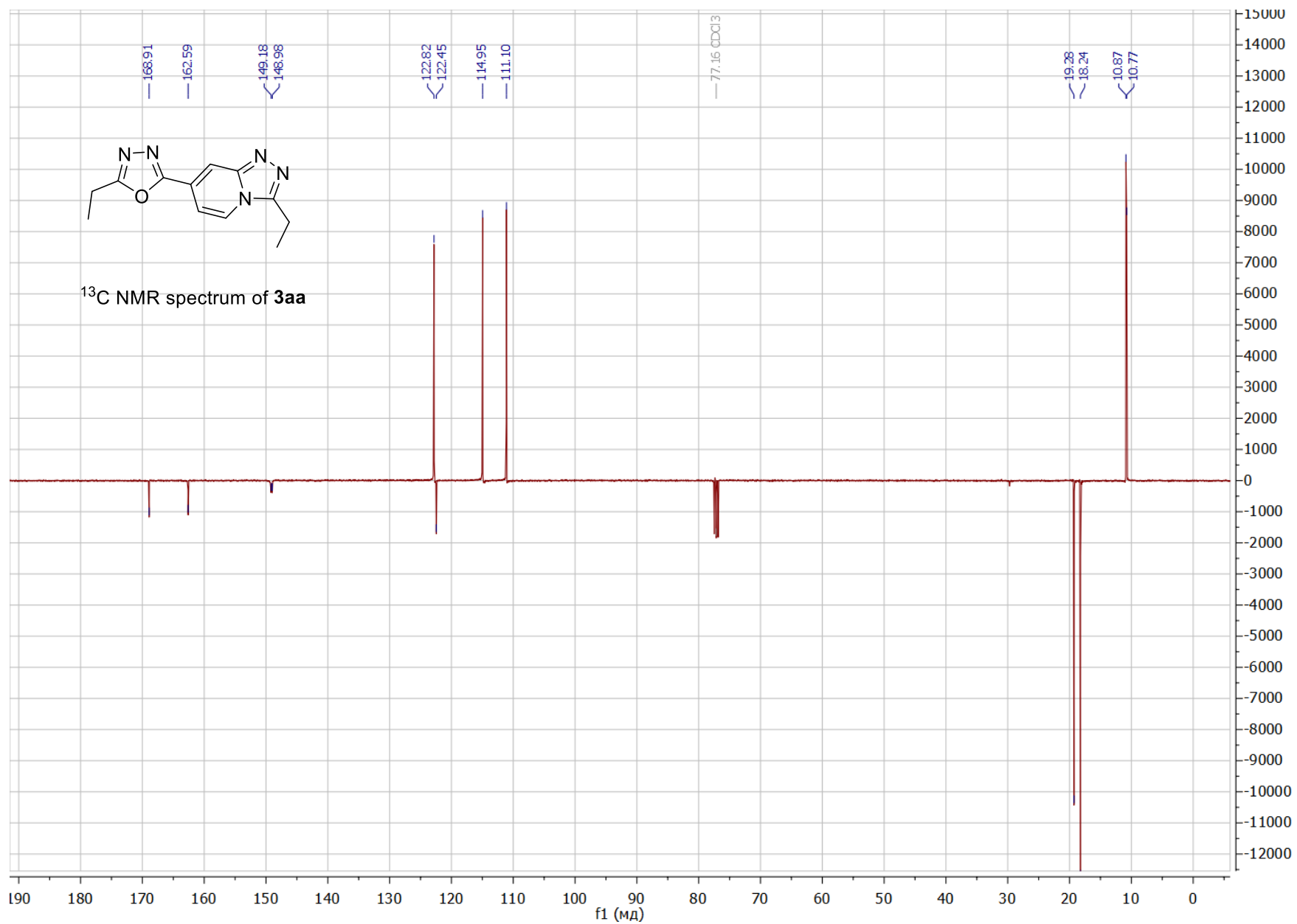


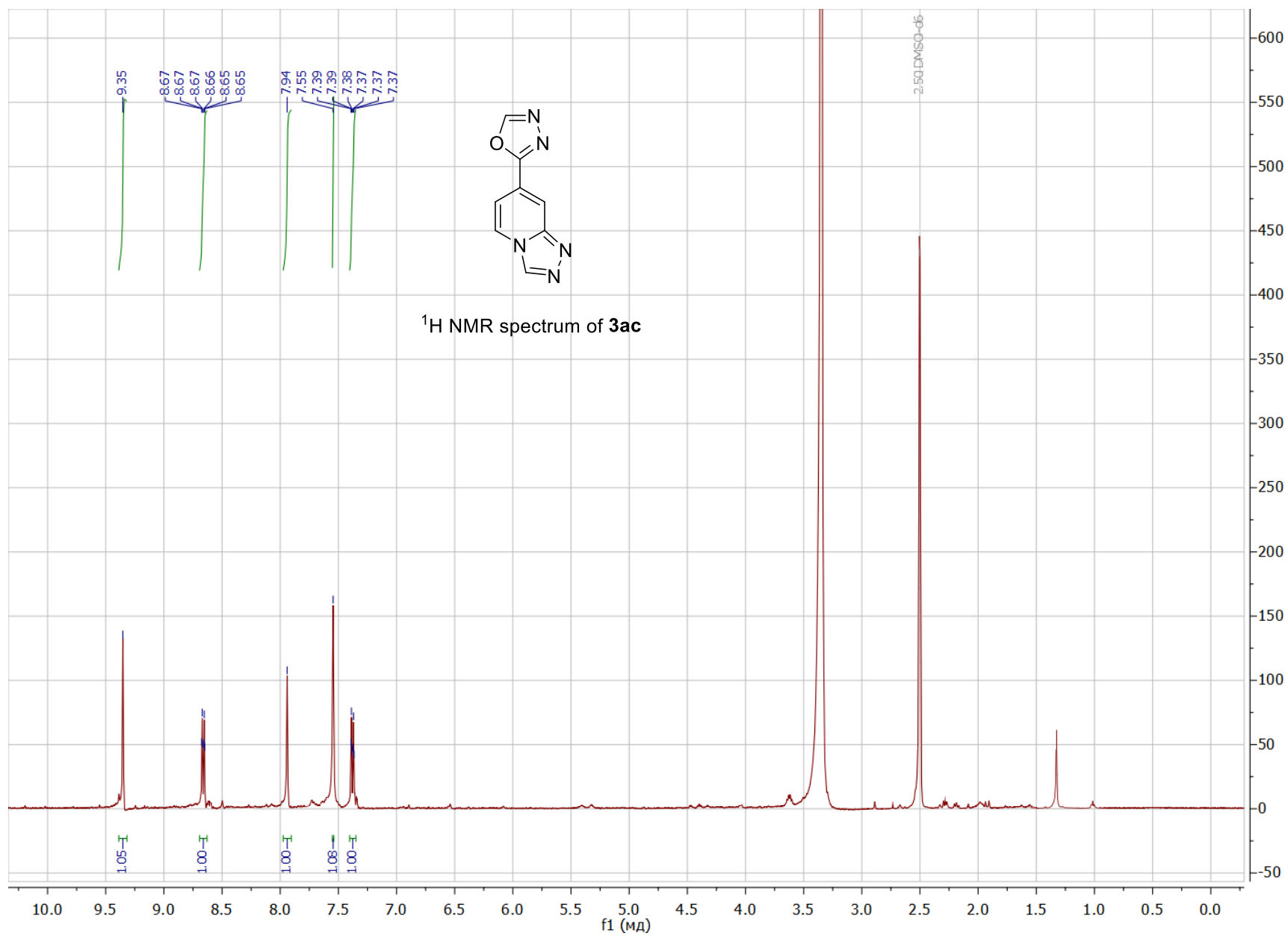
¹H NMR spectrum of 3aa

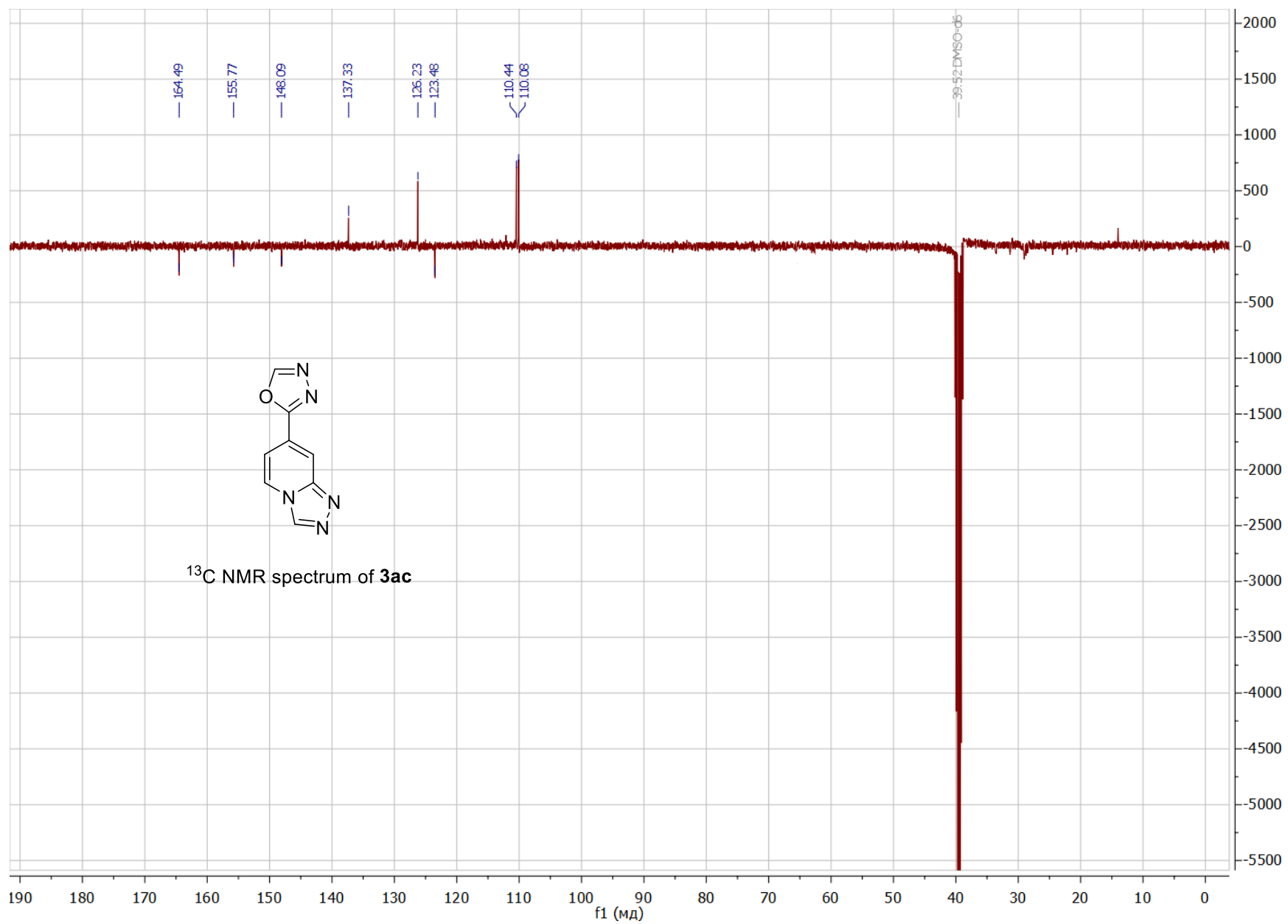
Chemical structure of 3aa: CCc1nc2cc(ccc2n1)C(=O)N=CC

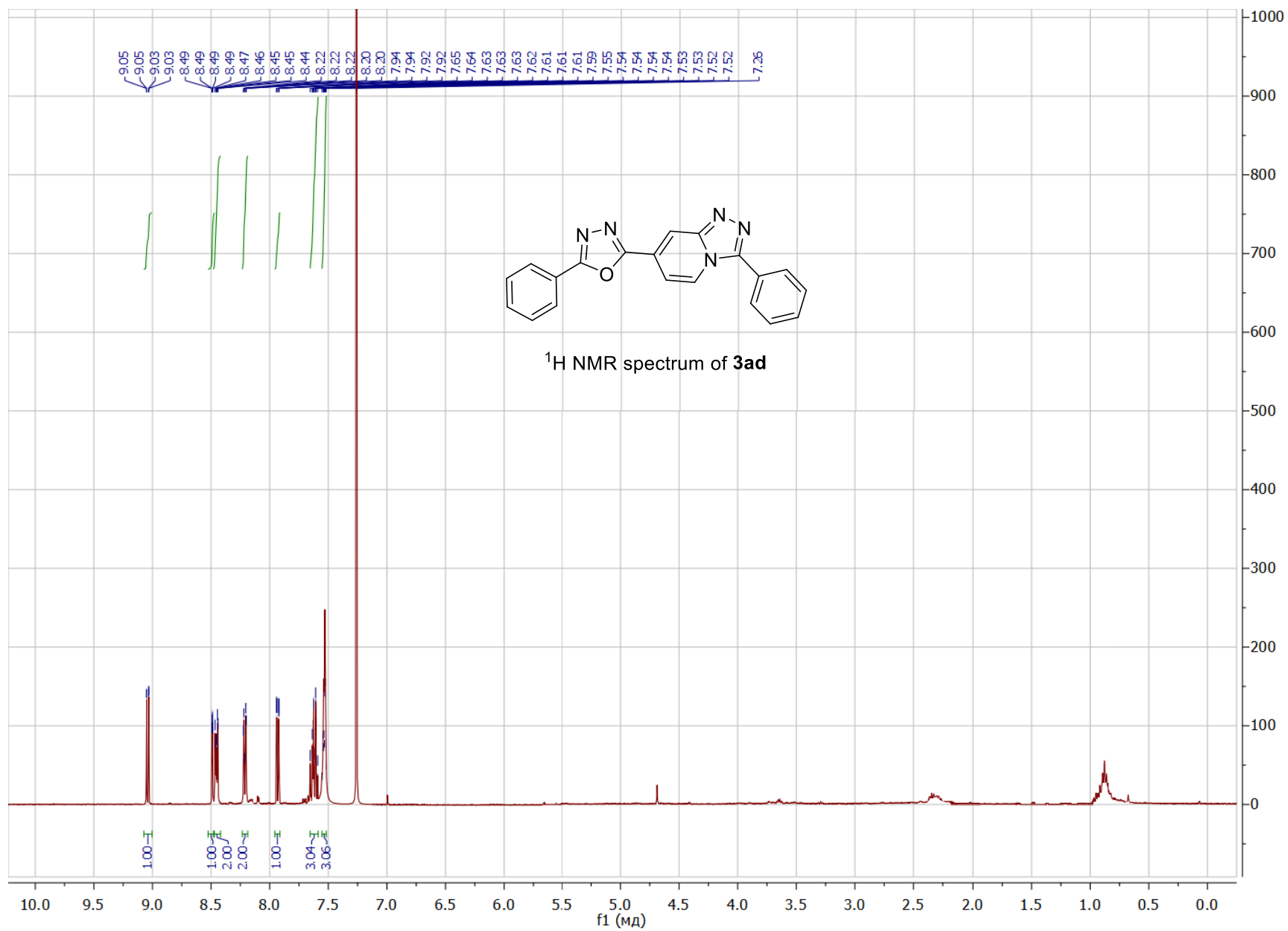
The spectrum displays the following peak regions and integrations:

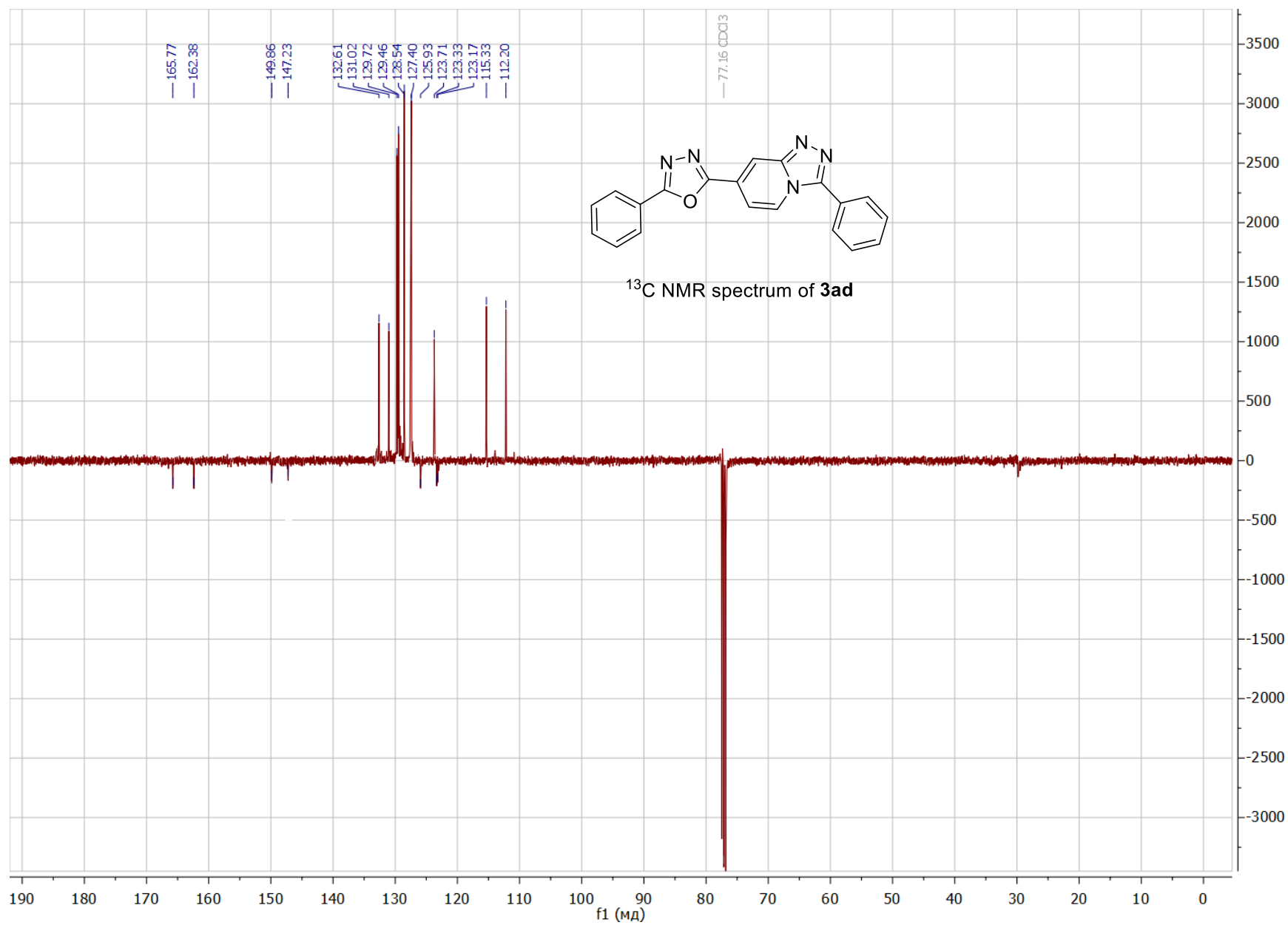
- Aromatic region (7.2-8.3 ppm):
 - 8.24, 8.23, 8.23, 8.23 ppm (integration: 1.00)
 - 8.00, 7.54, 7.53, 7.53, 7.52, 7.52, 7.51 ppm (integration: 1.01)
 - 7.26 ppm (integration: 1.00)
- Aliphatic region (1.4-3.2 ppm):
 - 3.14, 3.14, 3.12, 3.12, 3.10, 3.10, 3.08, 3.08, 2.99, 2.99, 2.97, 2.97, 2.95, 2.95, 2.93, 2.93, 1.52, 1.52, 1.51, 1.51, 1.50, 1.50, 1.49, 1.48, 1.44, 1.44, 1.42, 1.42, 1.40, 1.40 ppm (integration: 2.00, 2.00)
 - 1.48, 1.44, 1.44, 1.42, 1.42, 1.40, 1.40 ppm (integration: 3.00, 3.00)

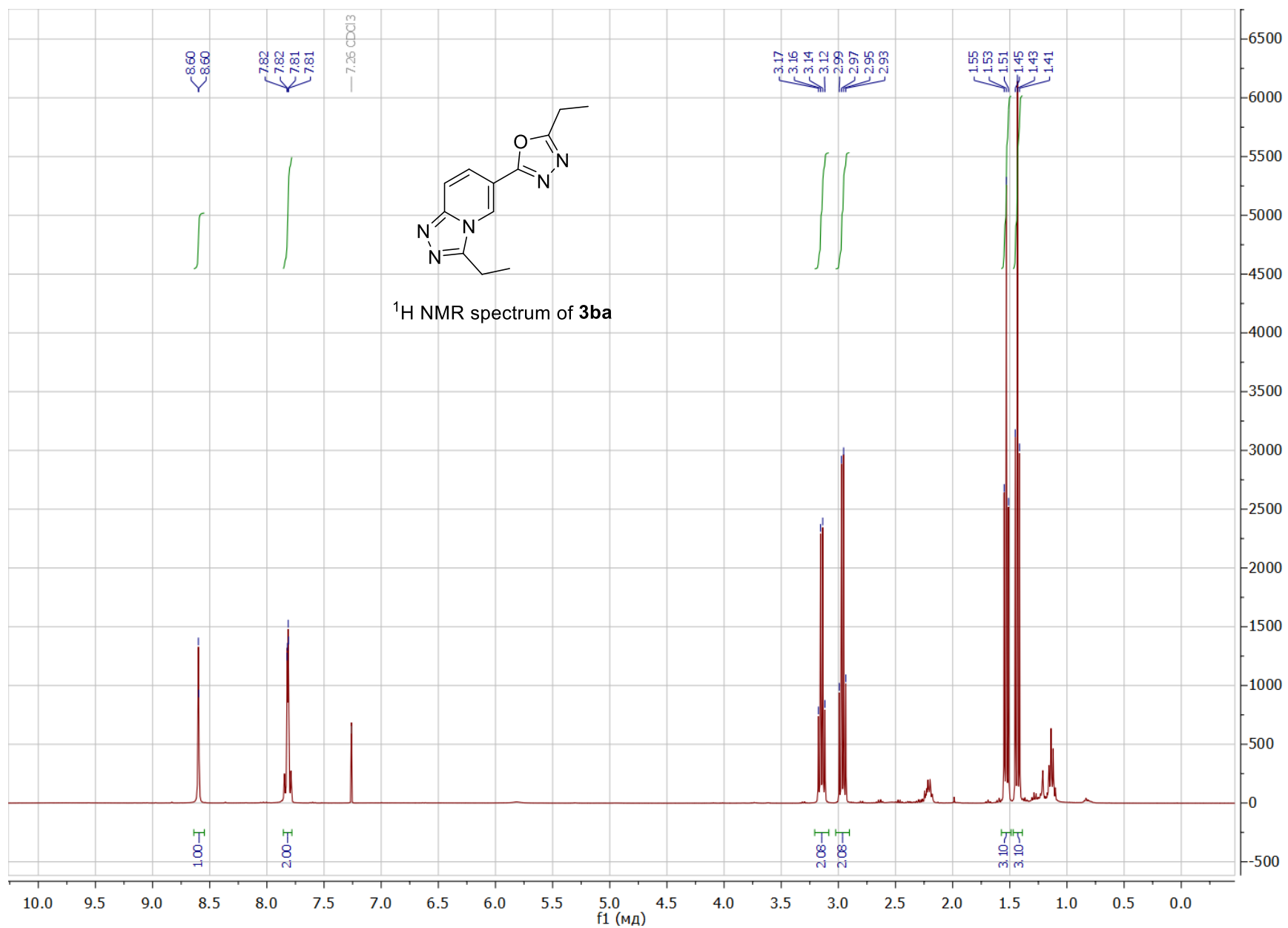


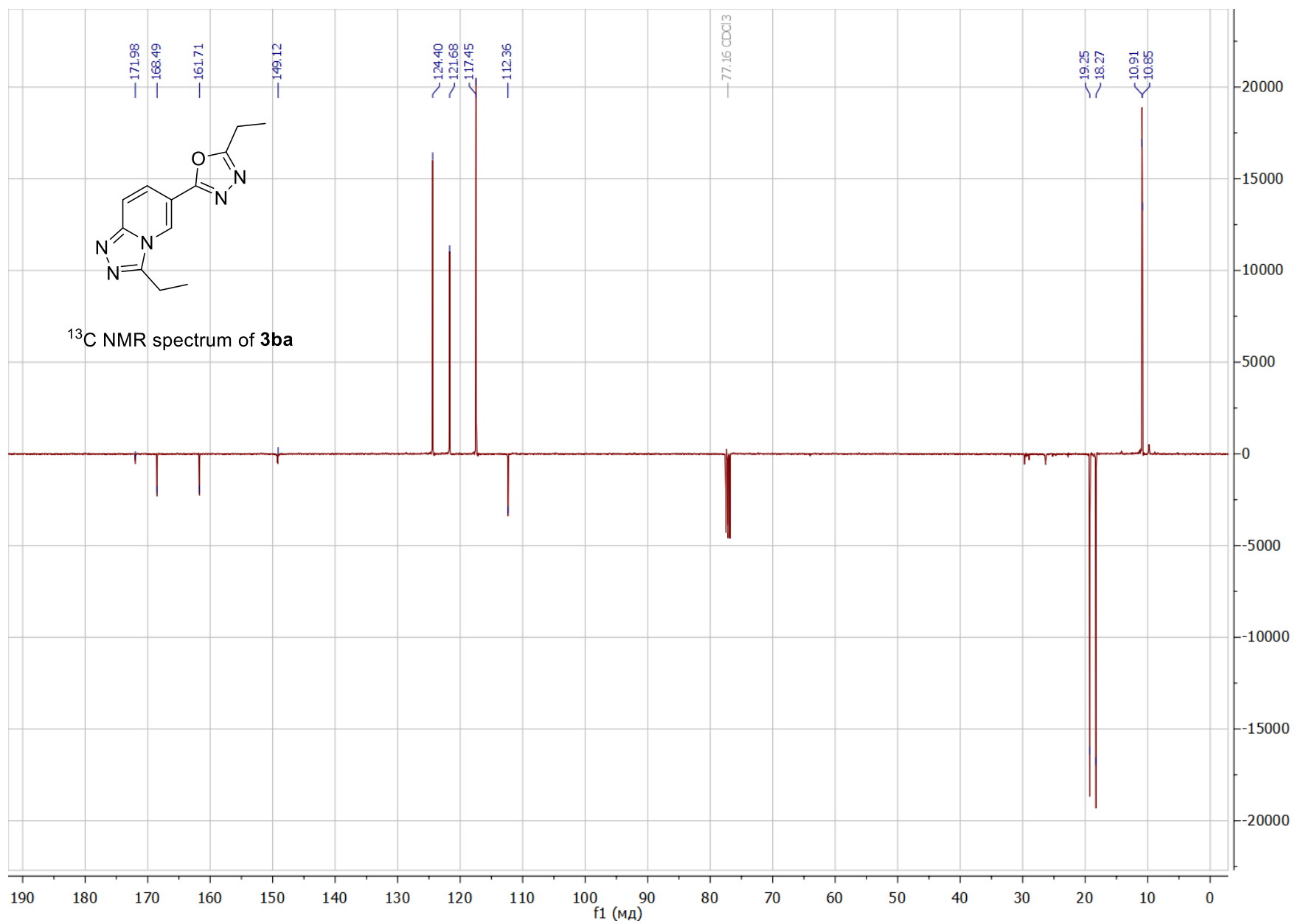


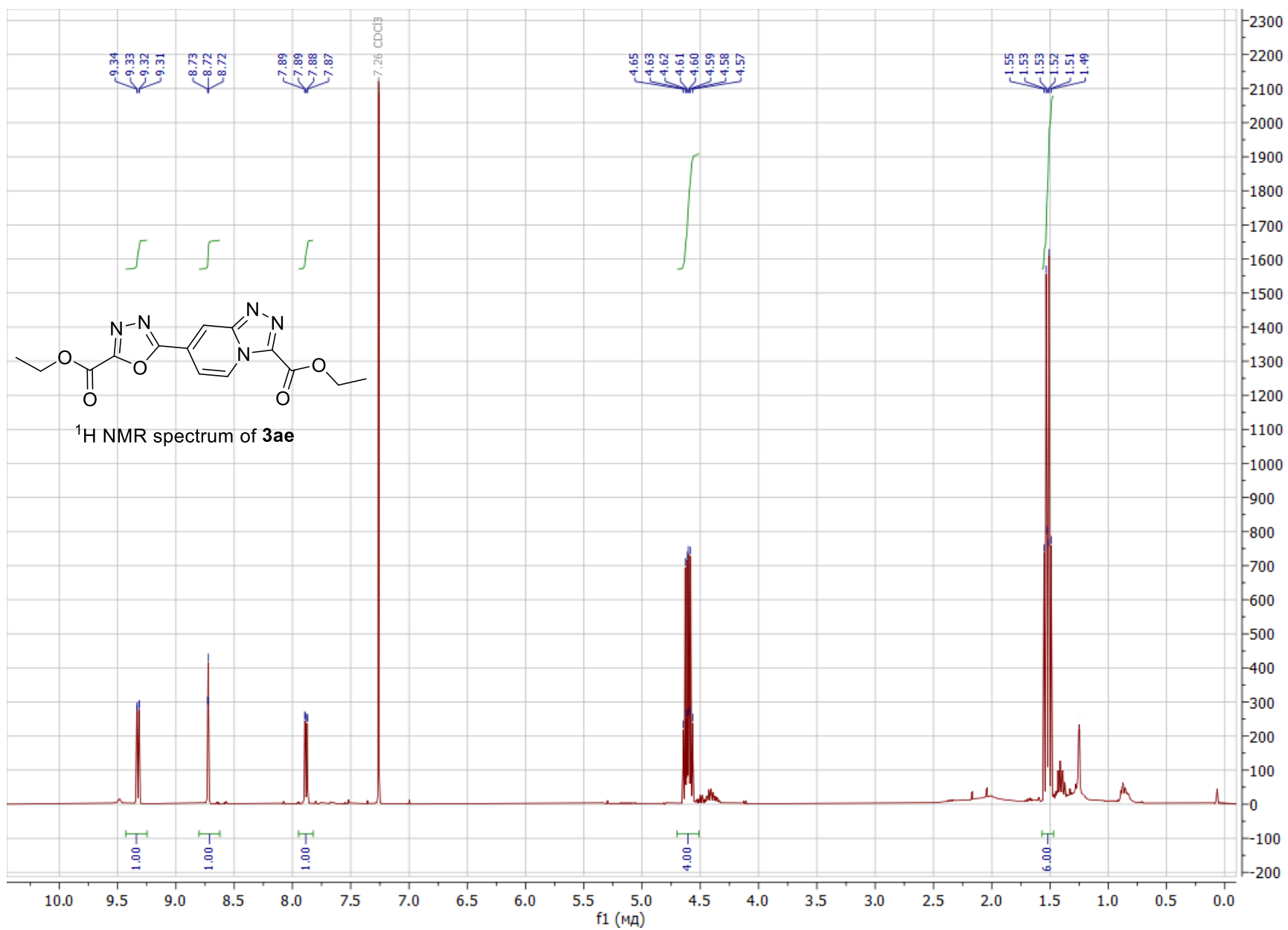


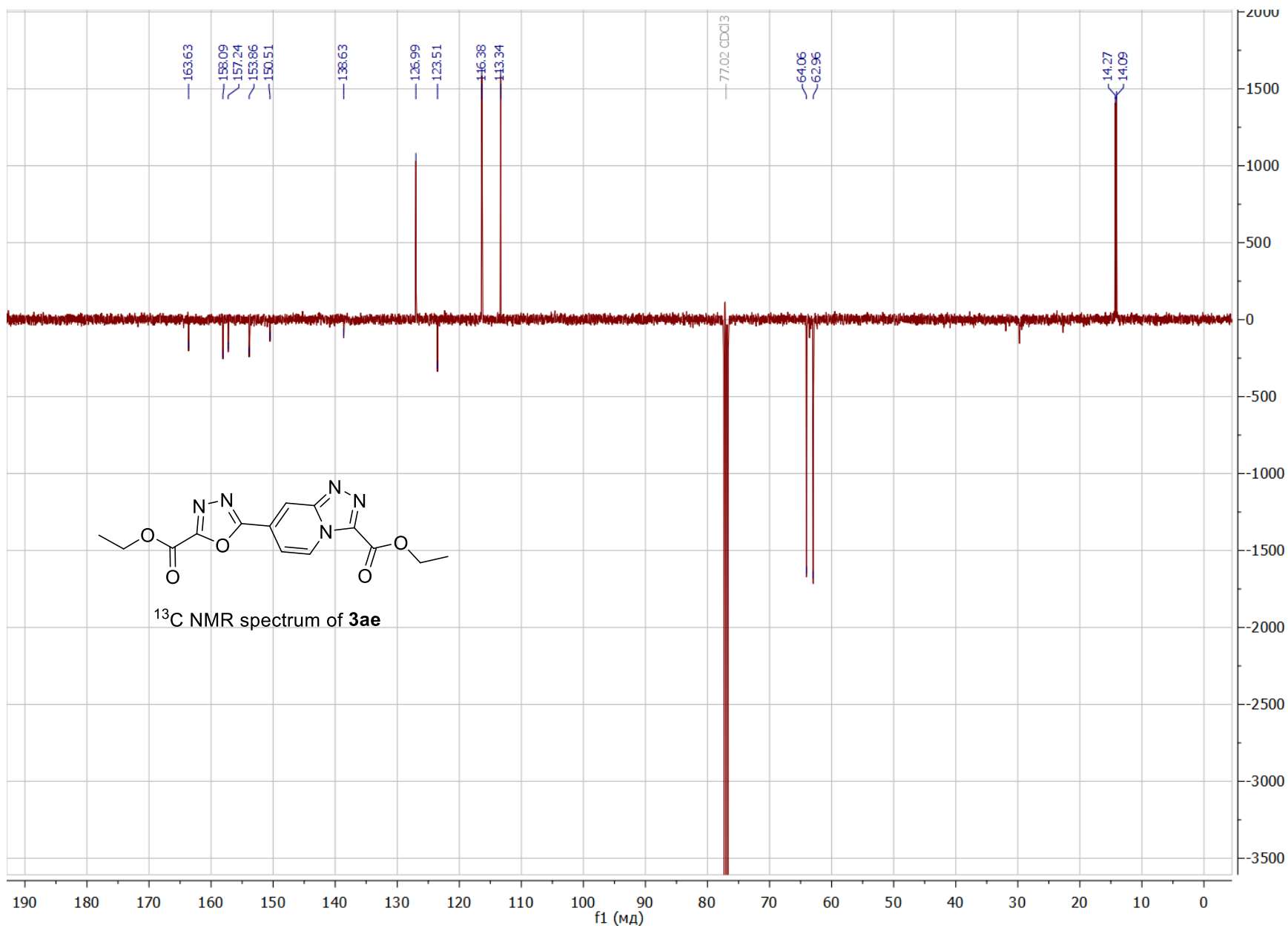




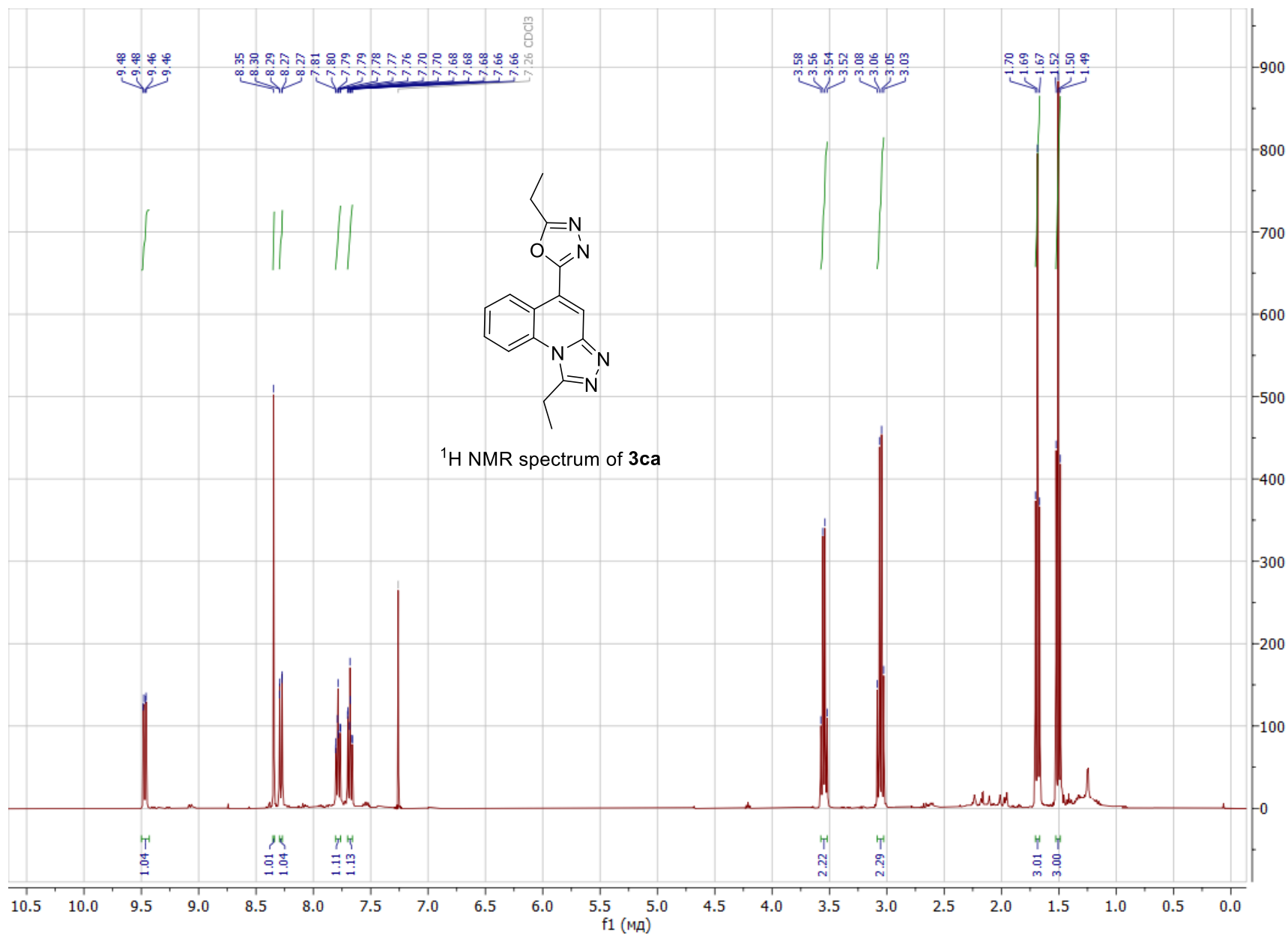


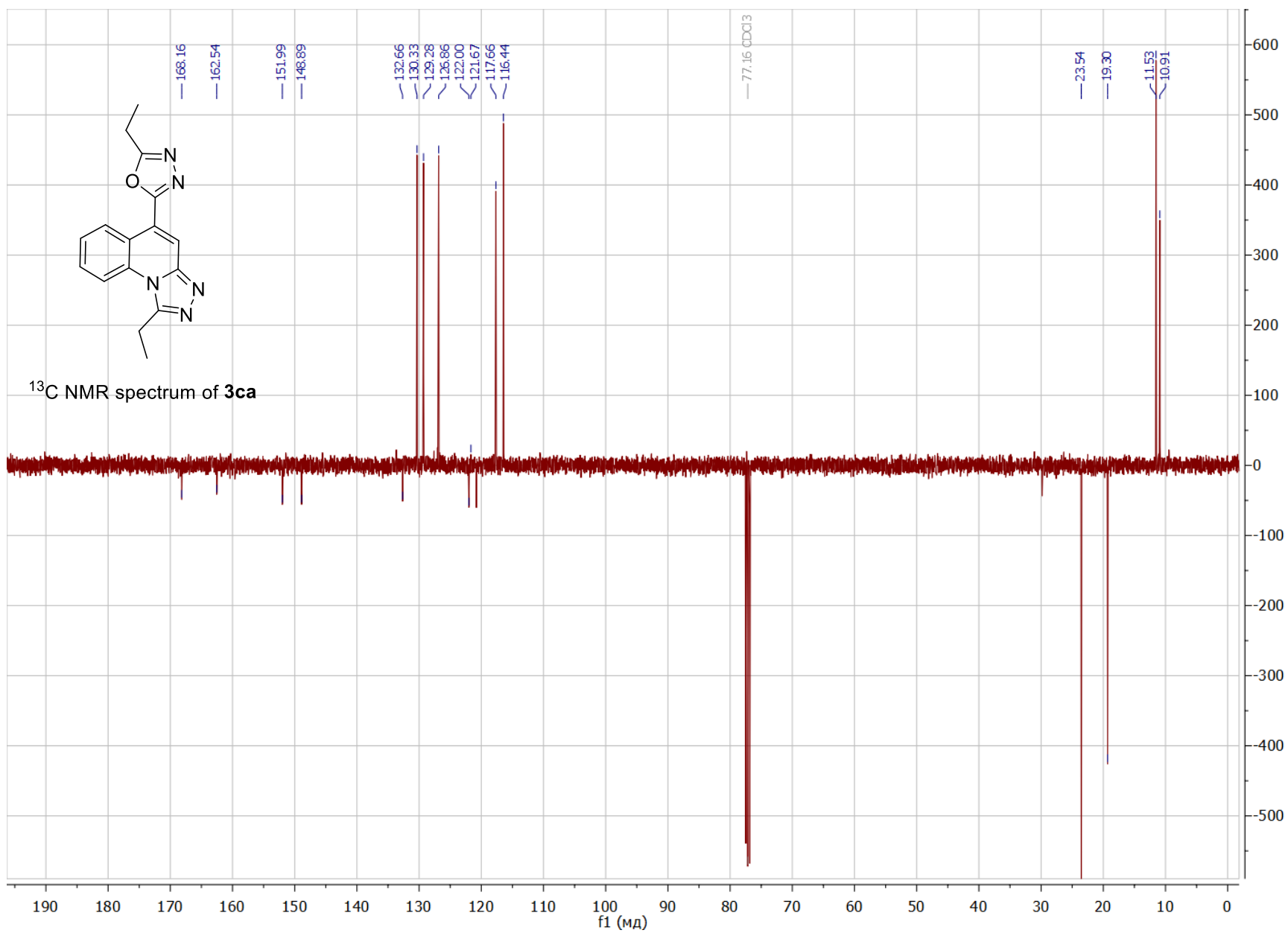


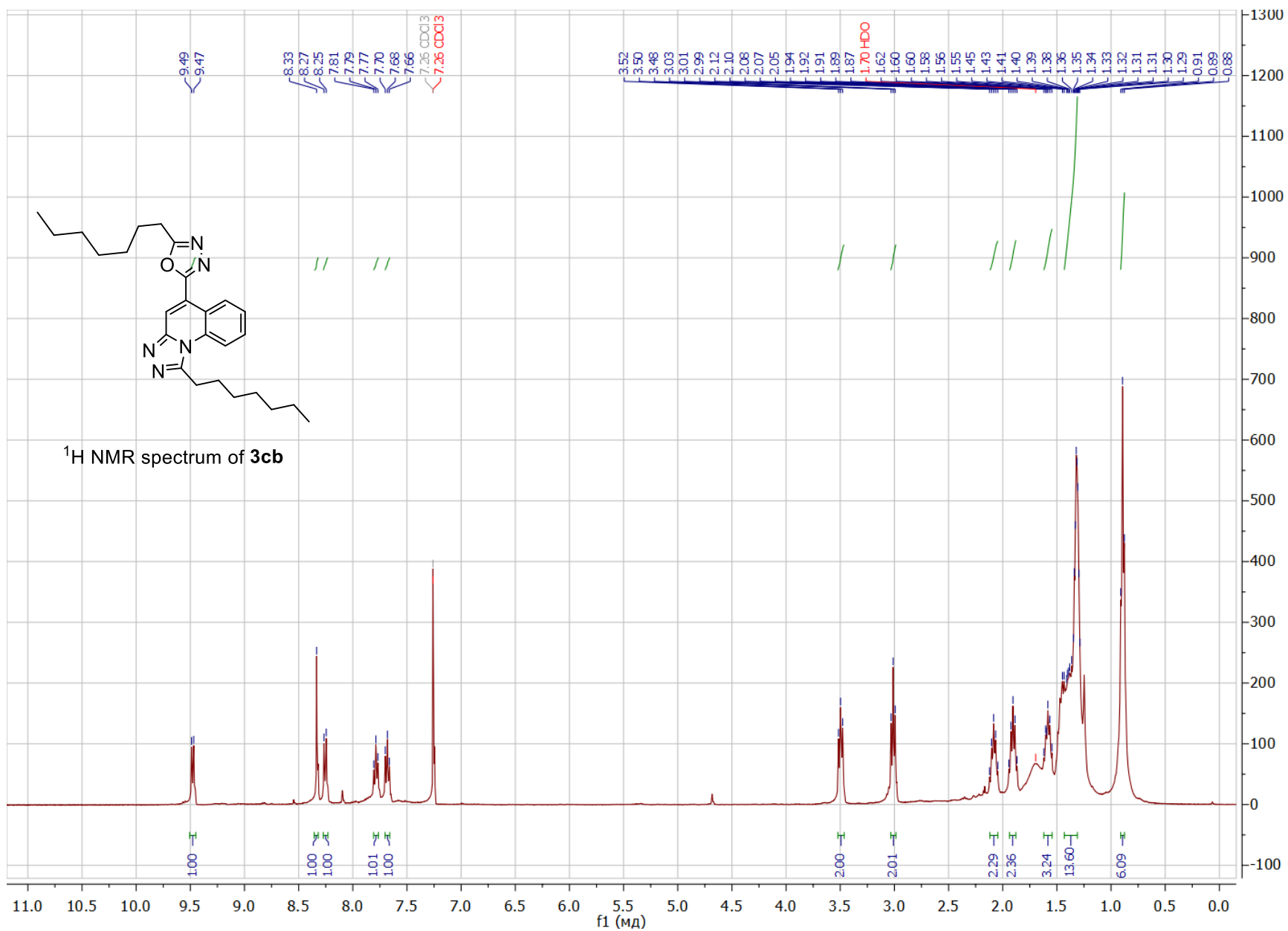


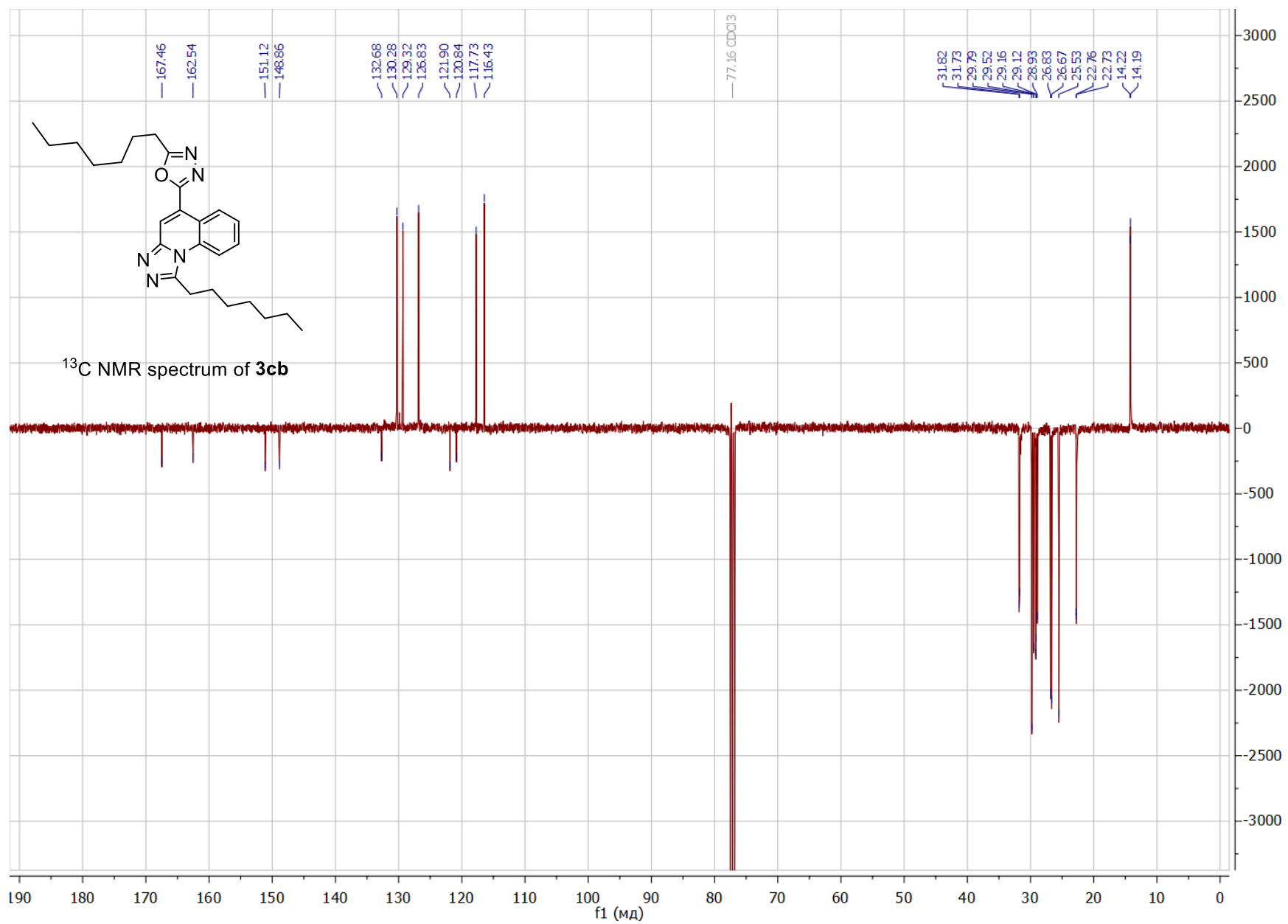


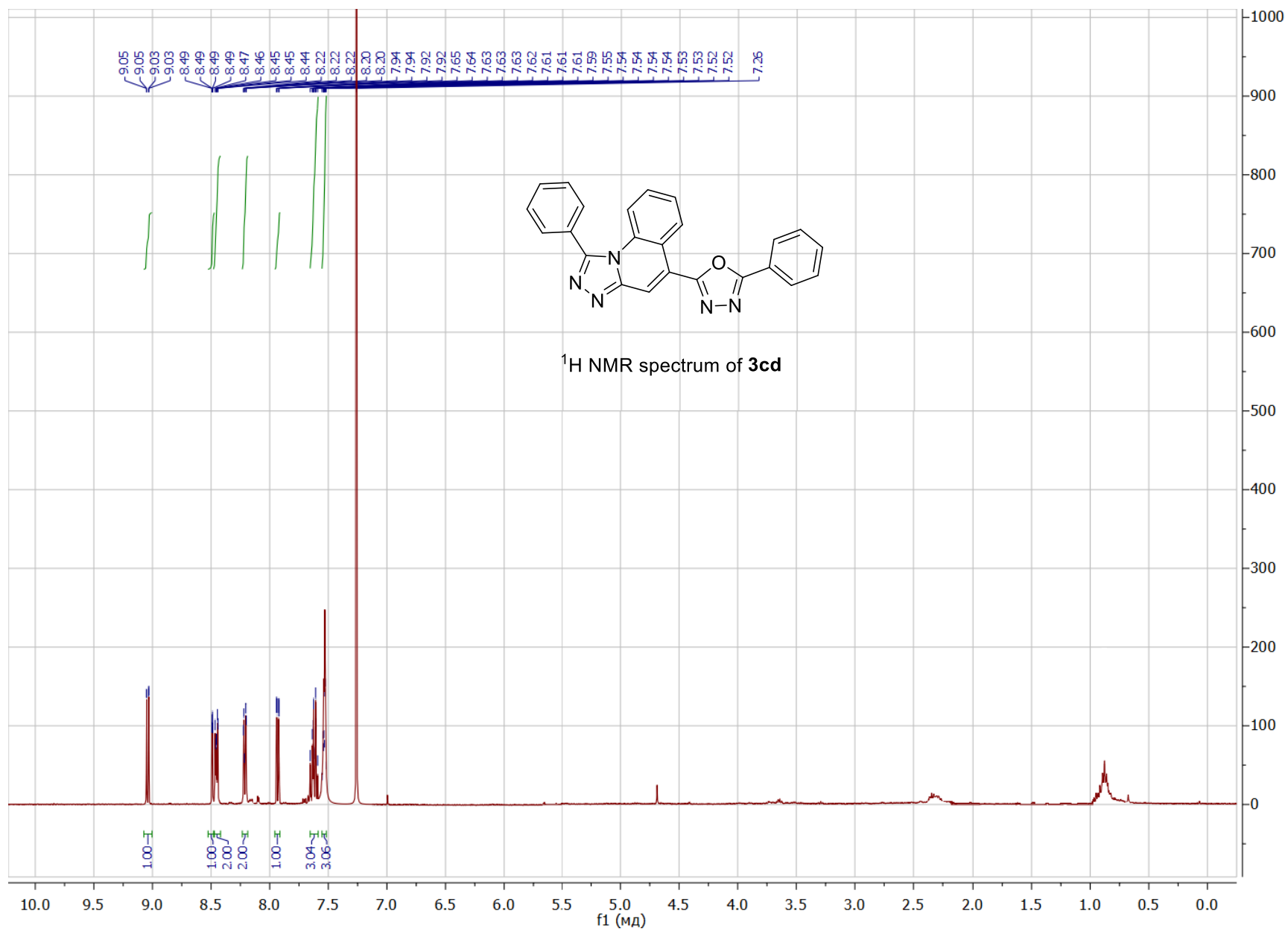
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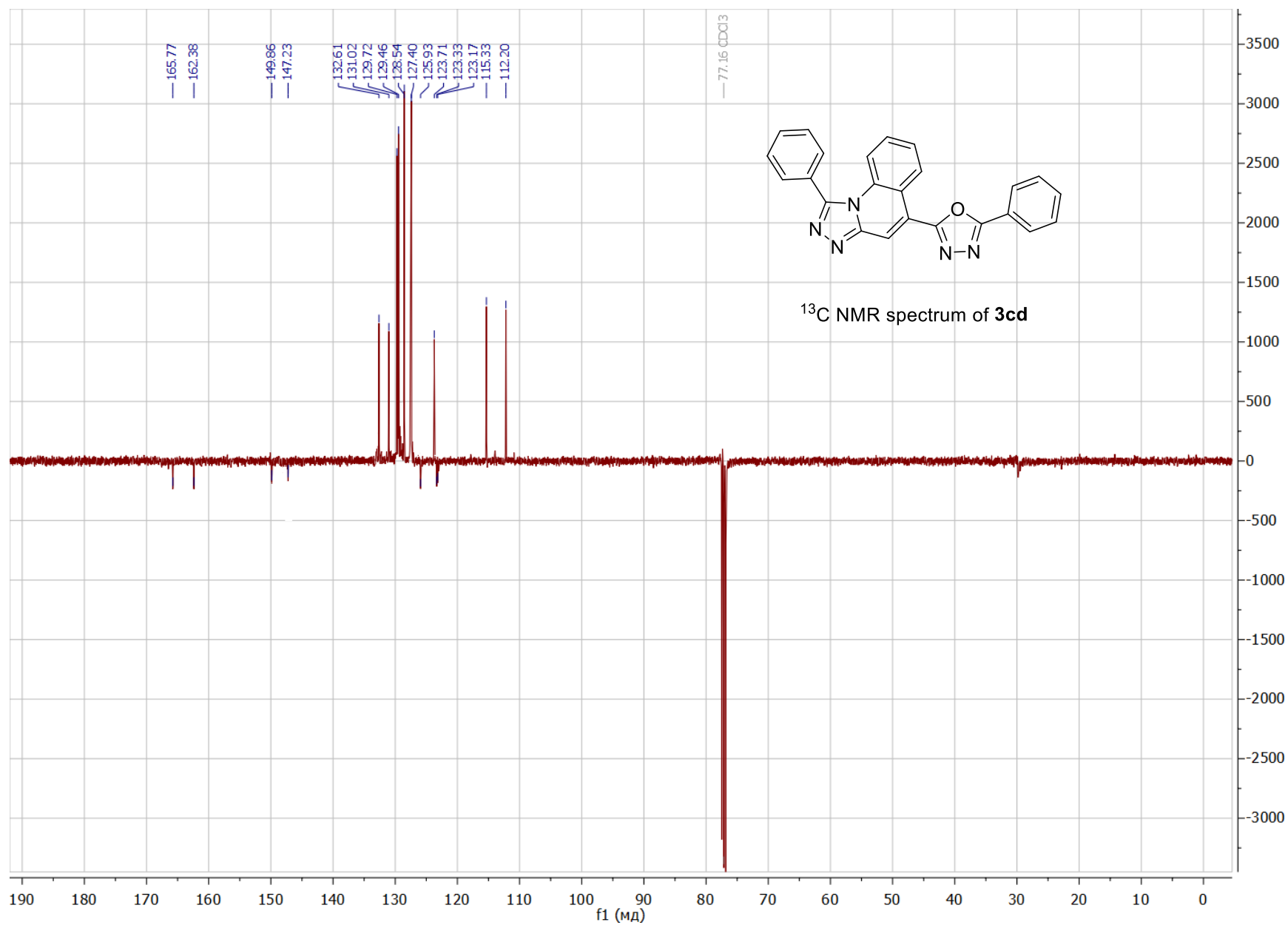


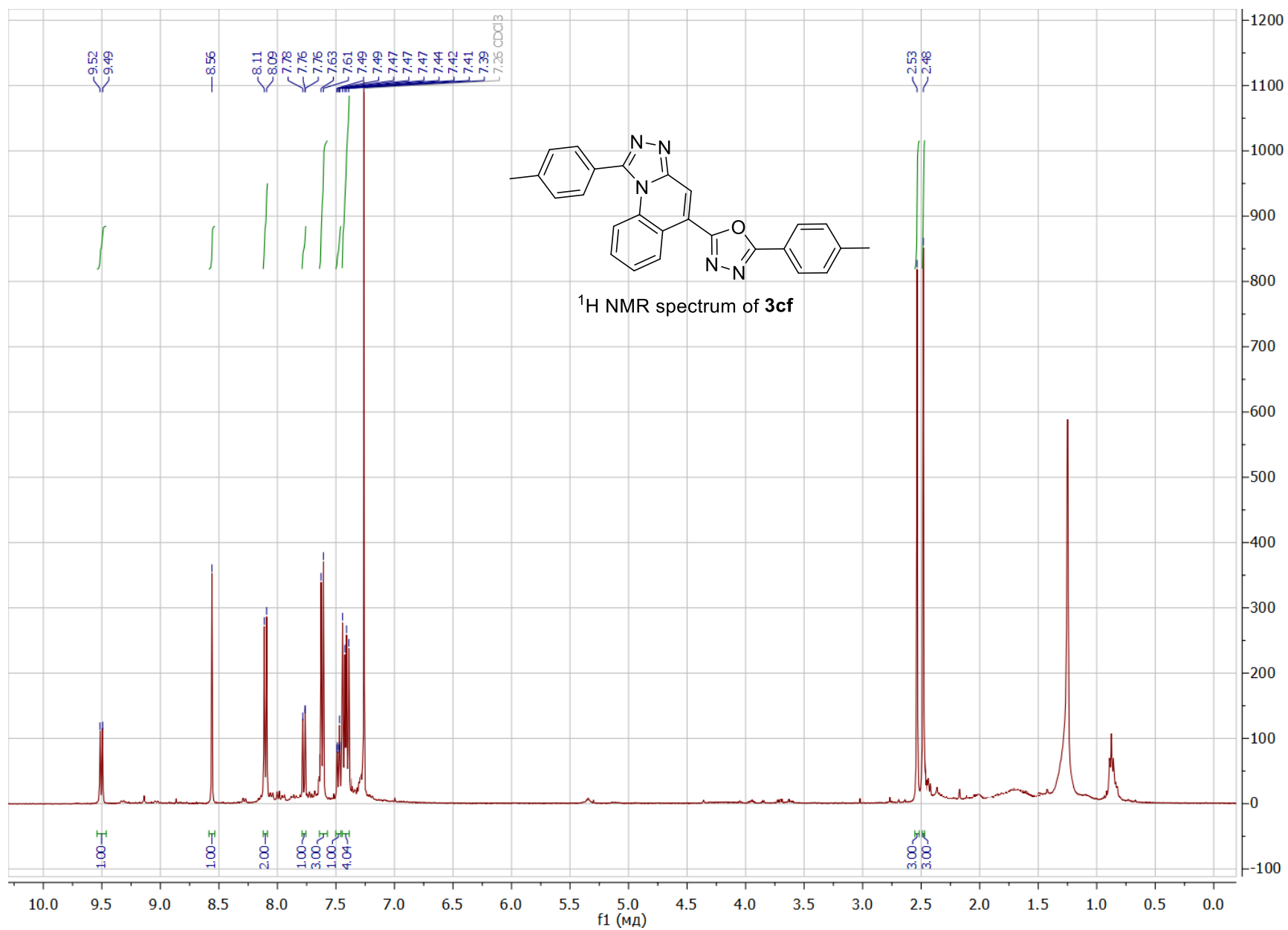


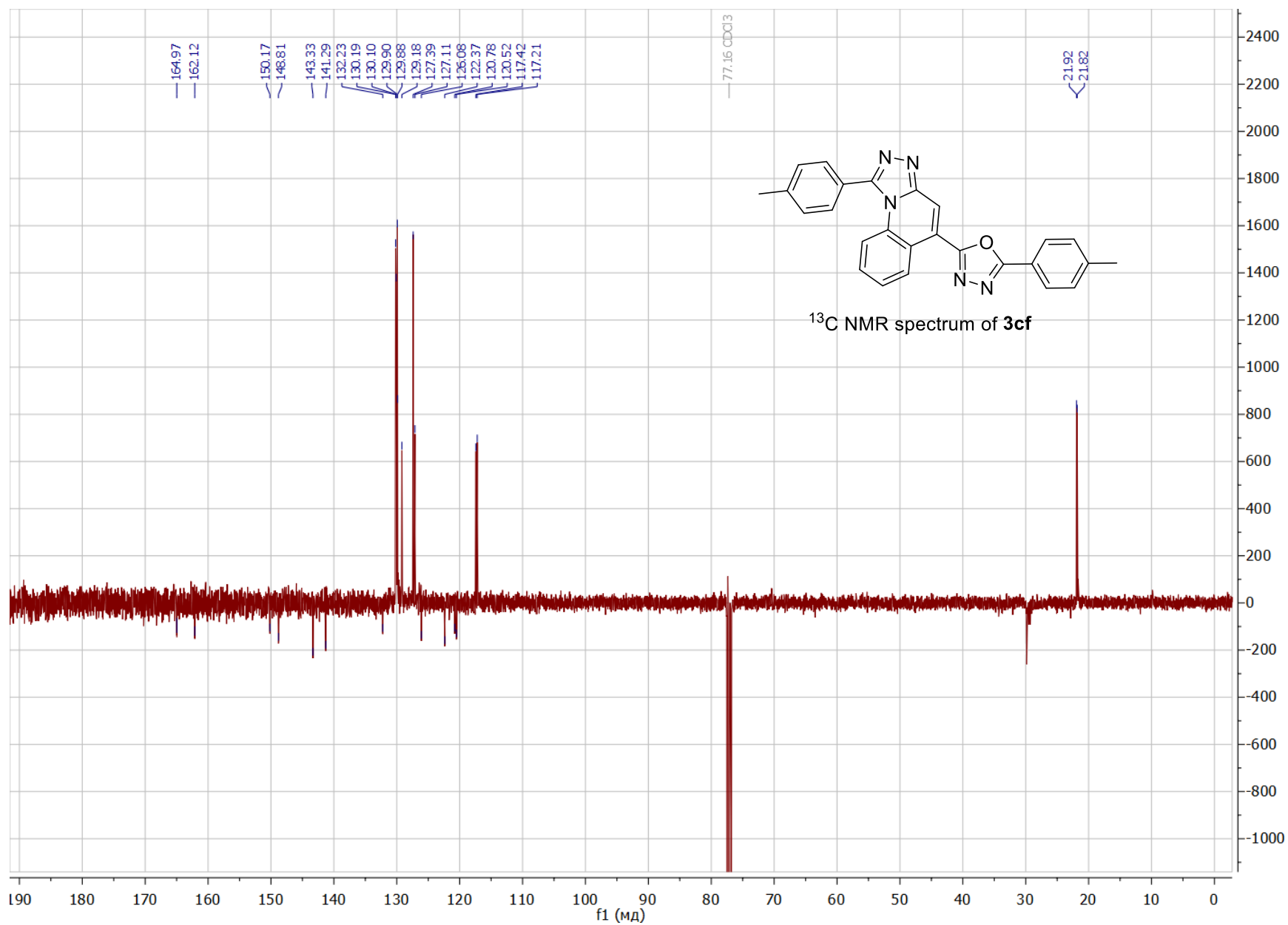


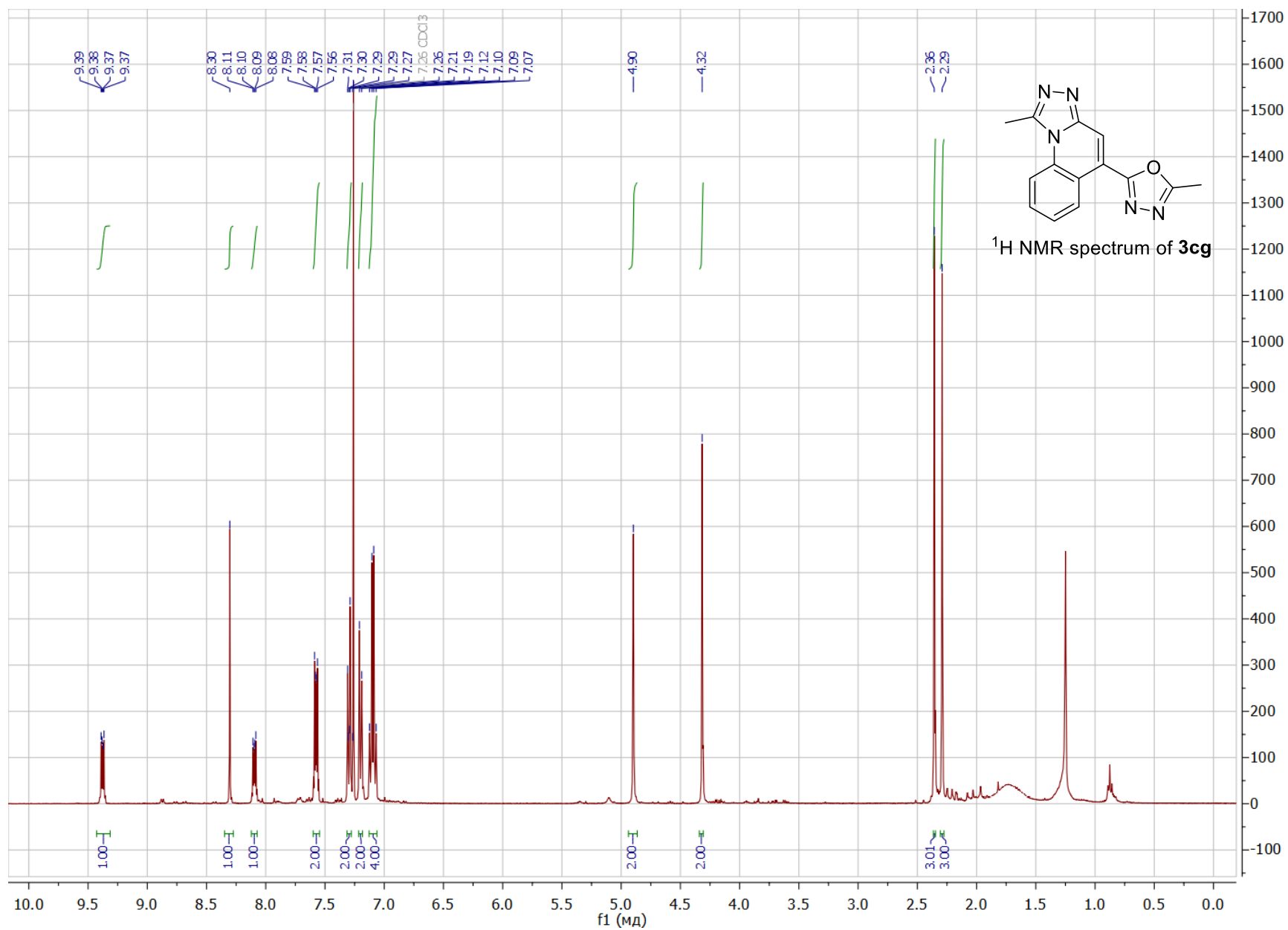


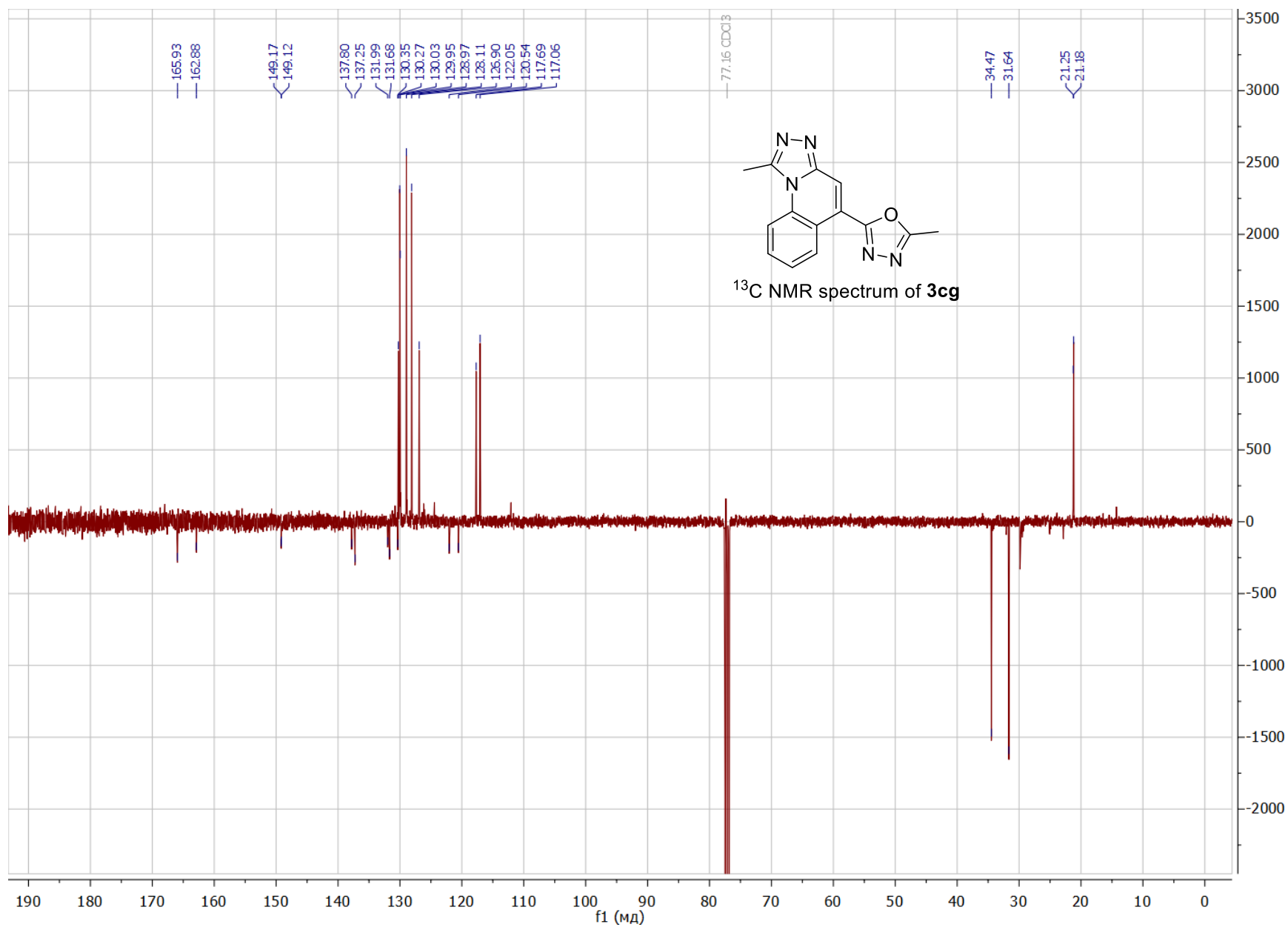


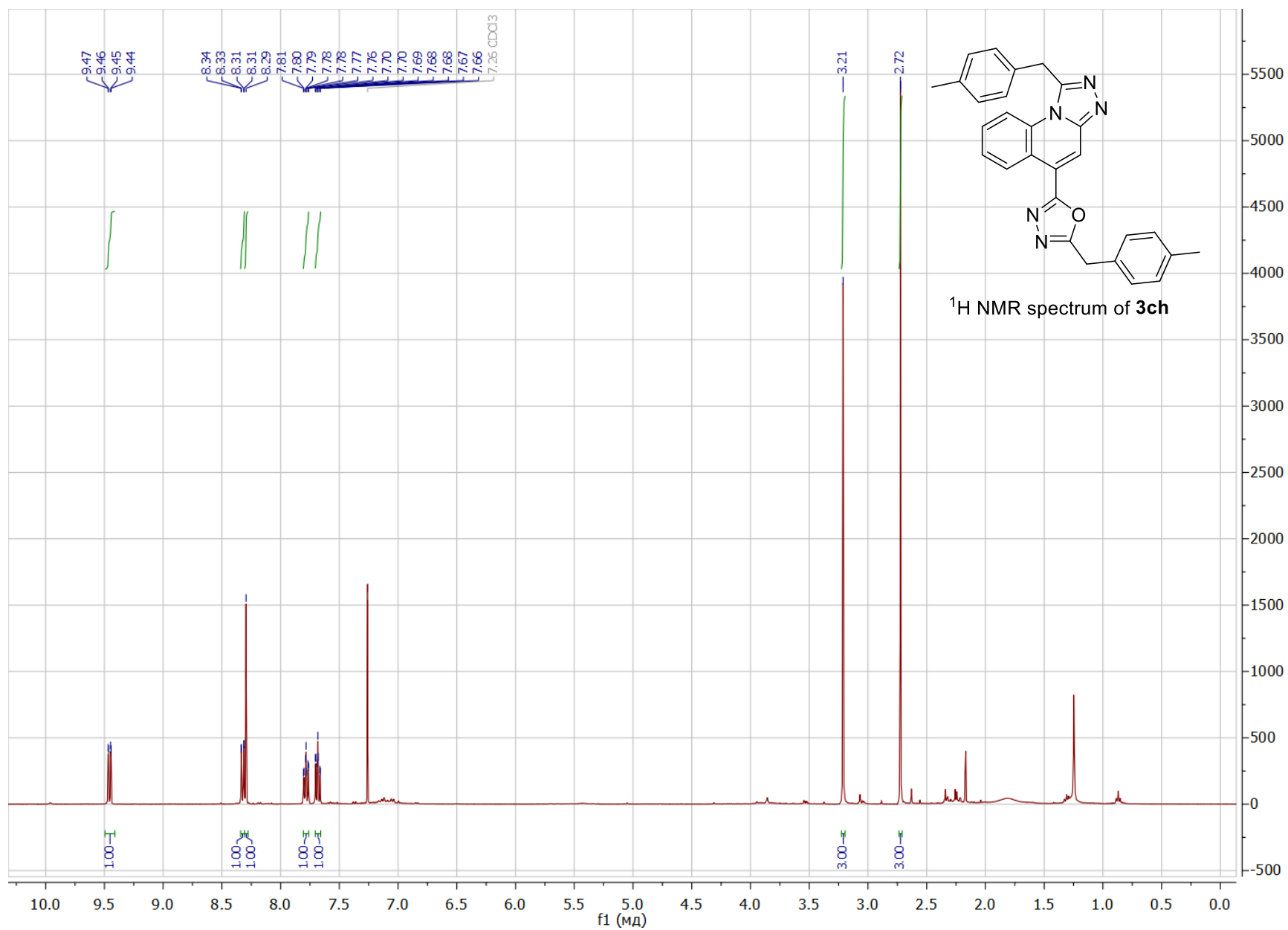


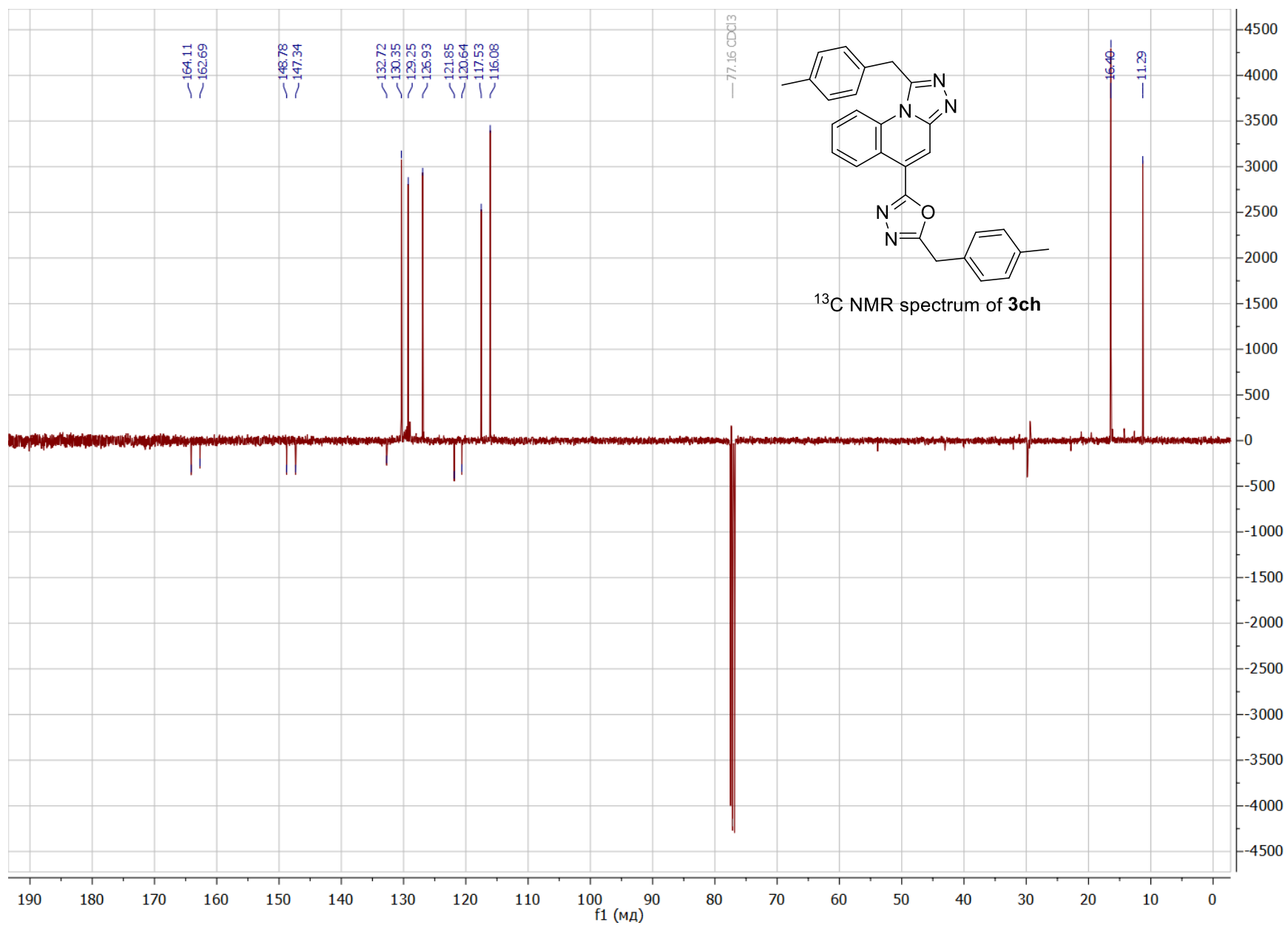


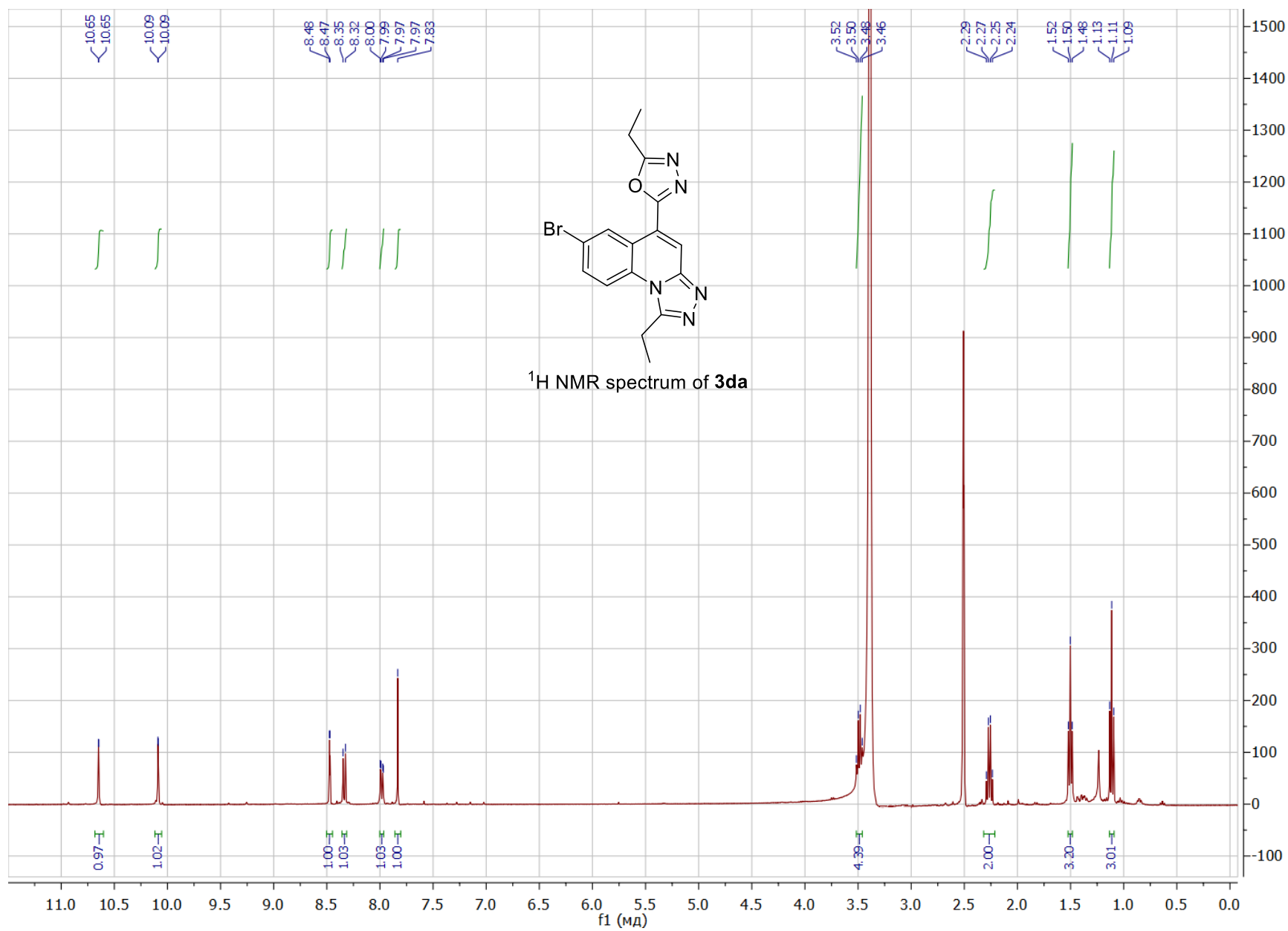


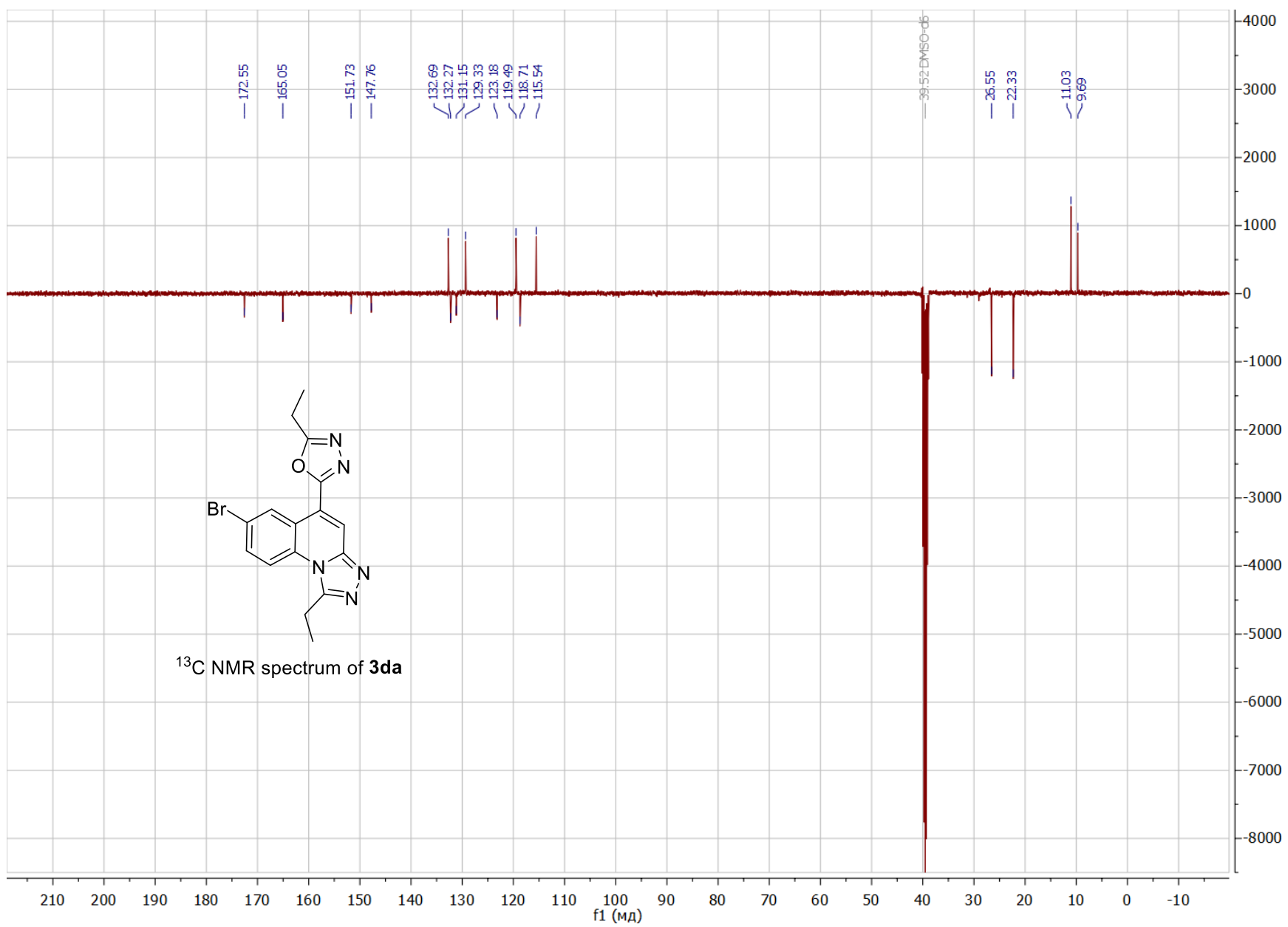


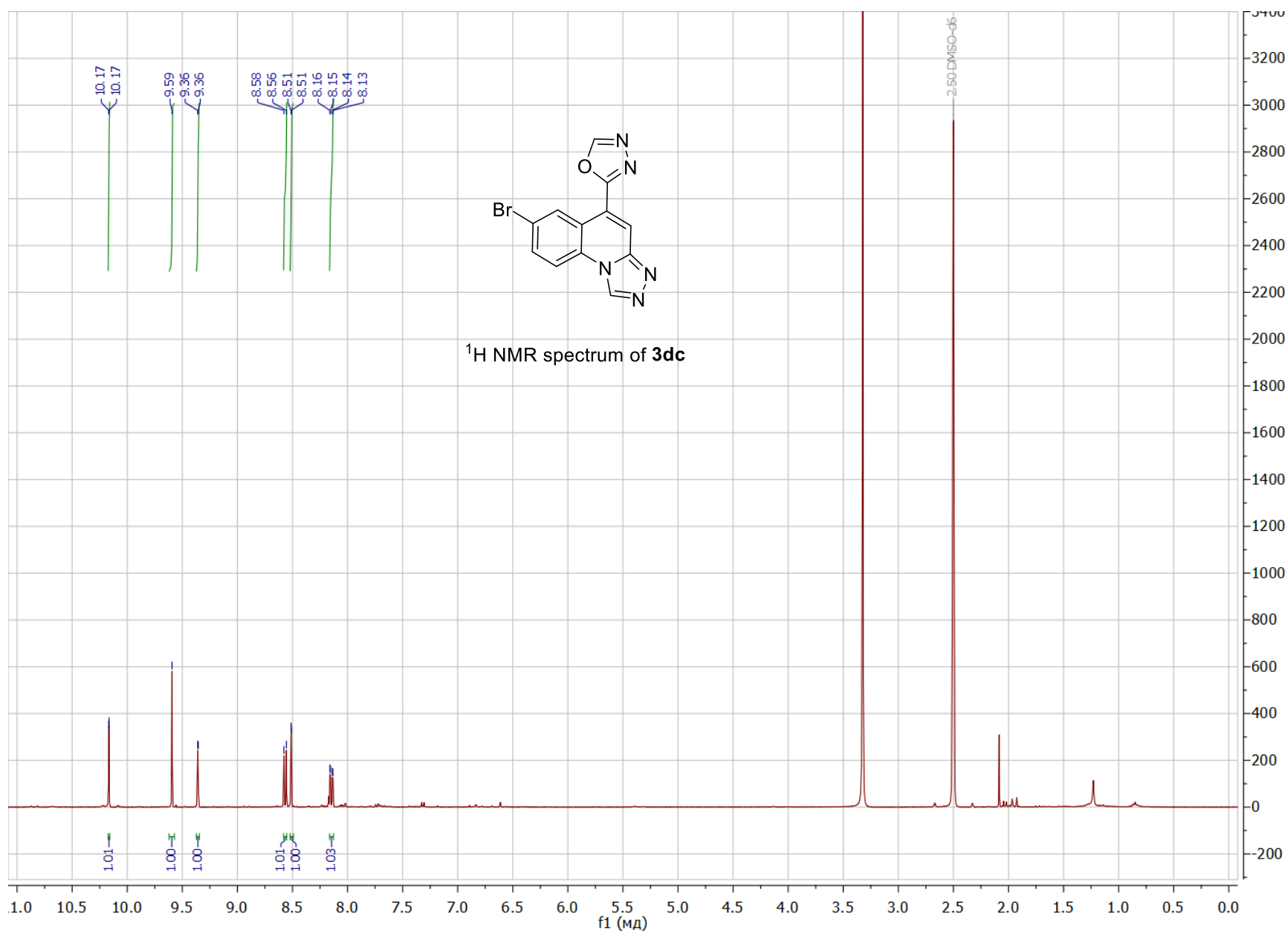


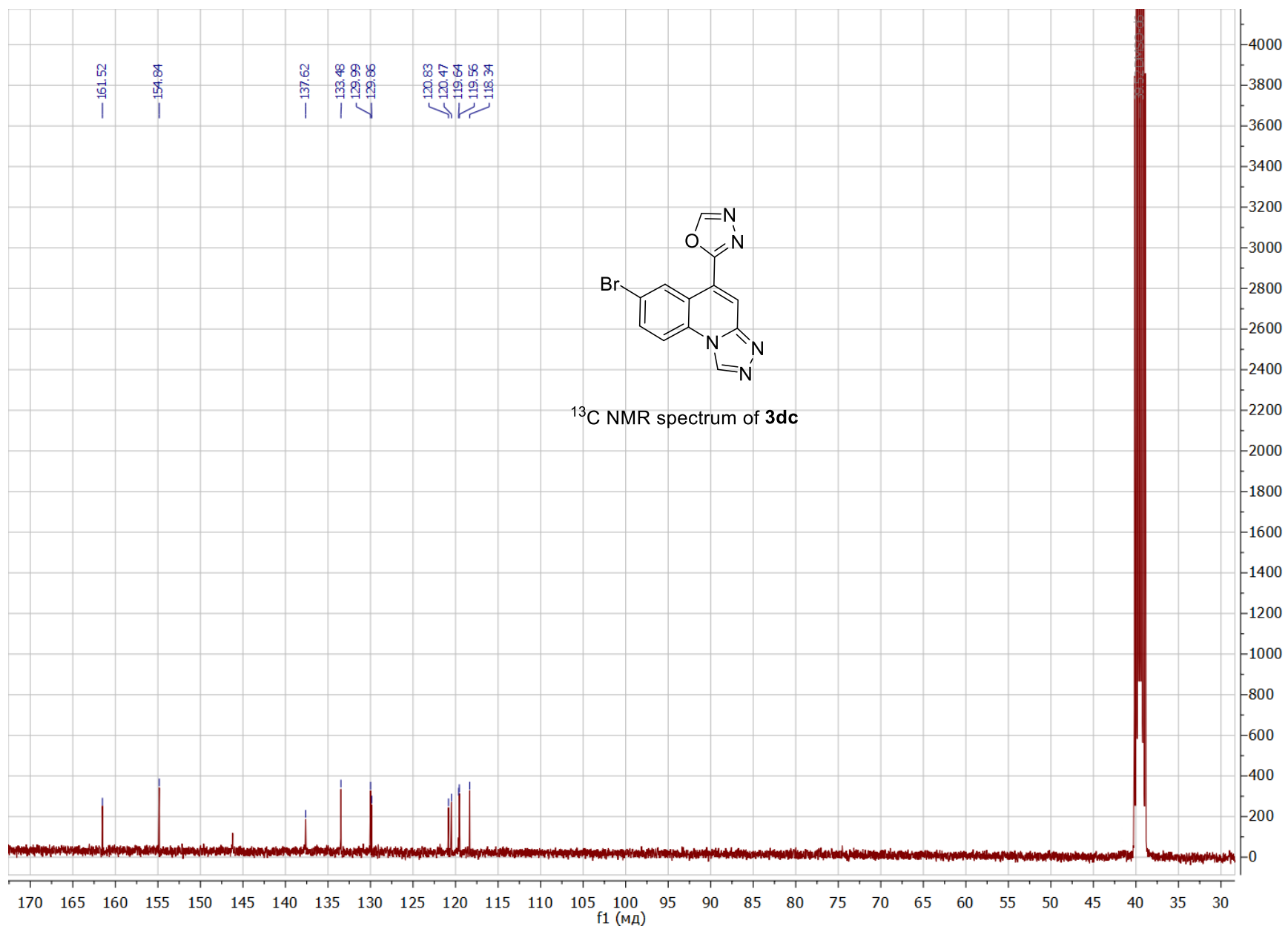


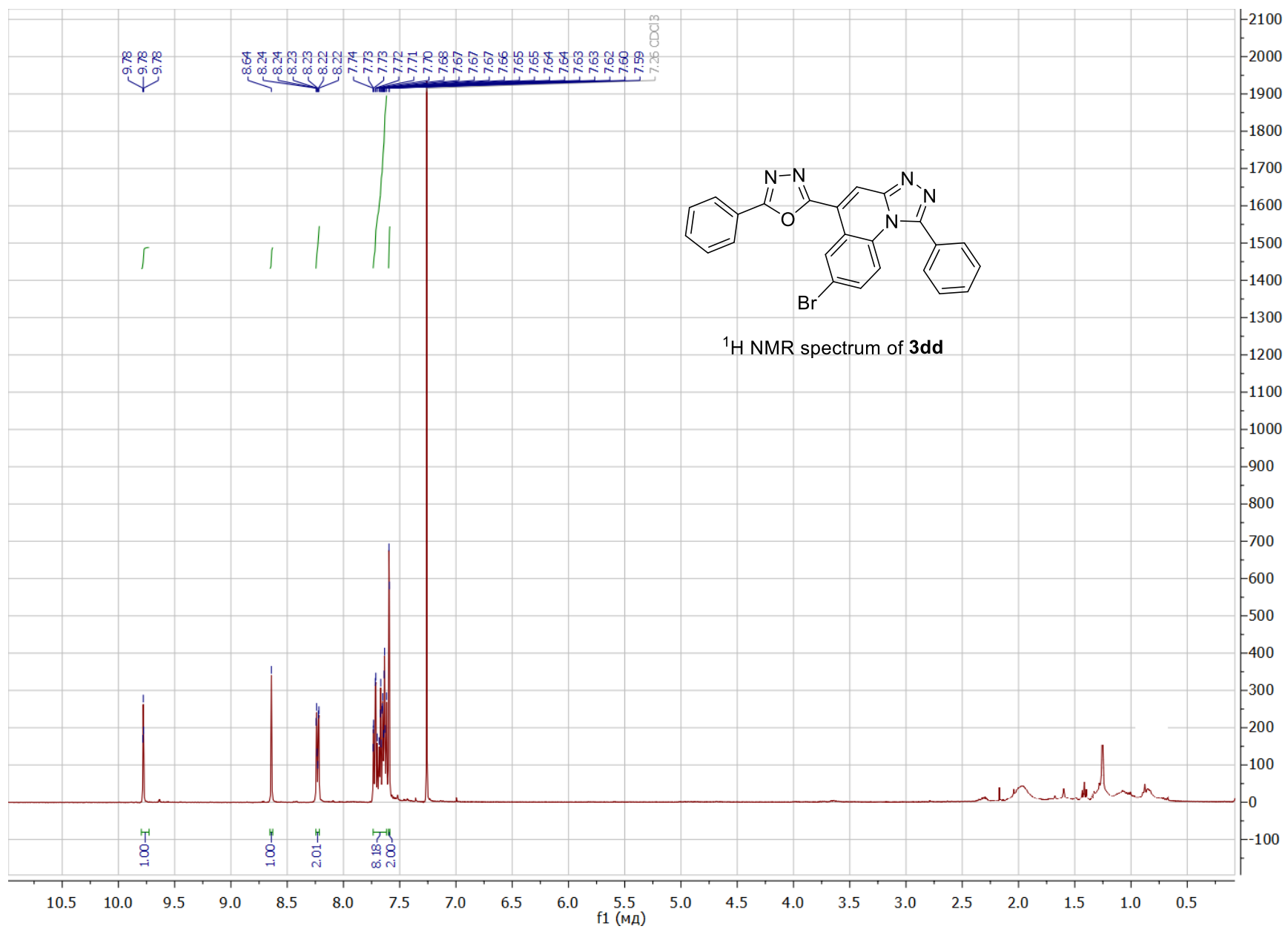


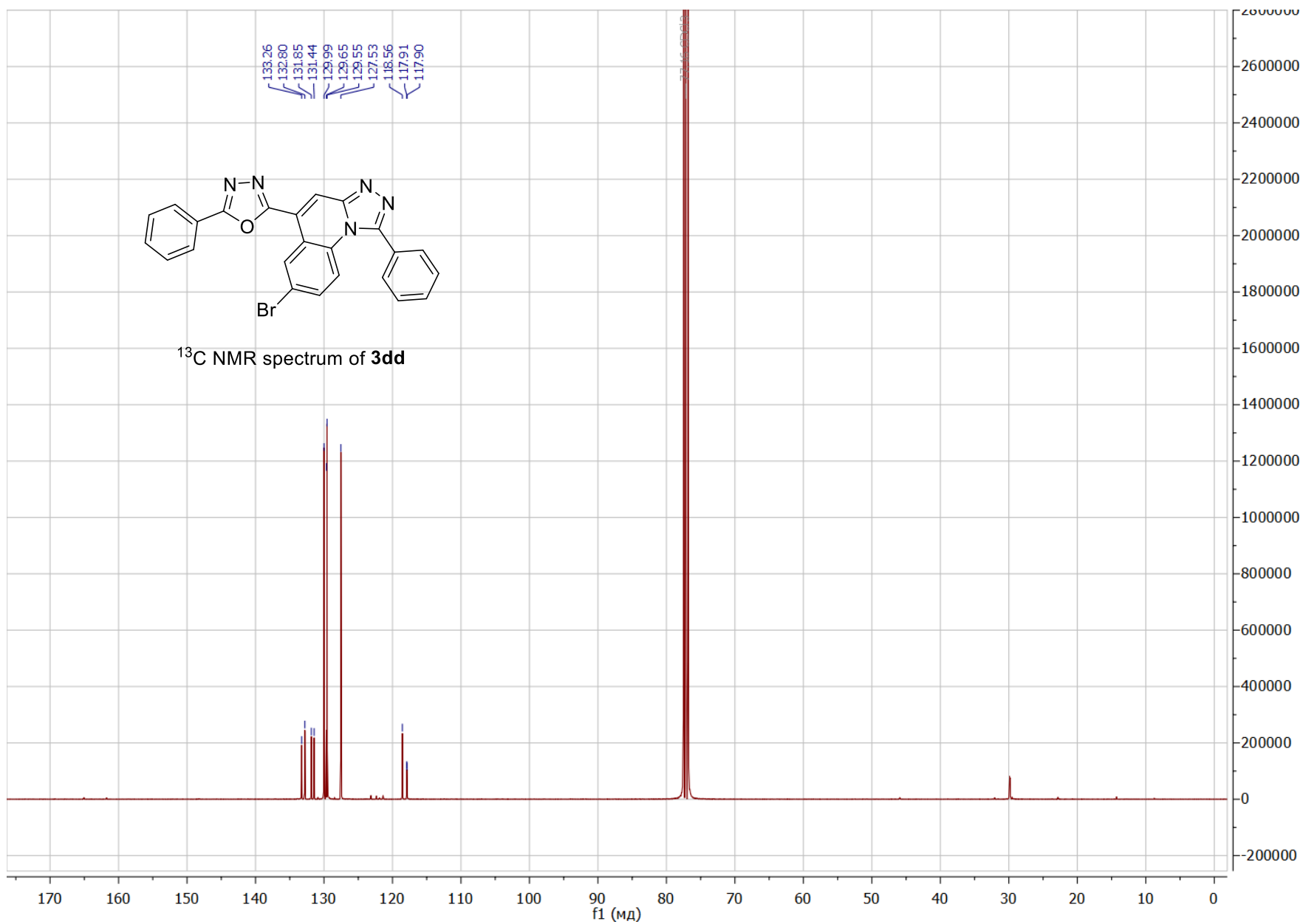


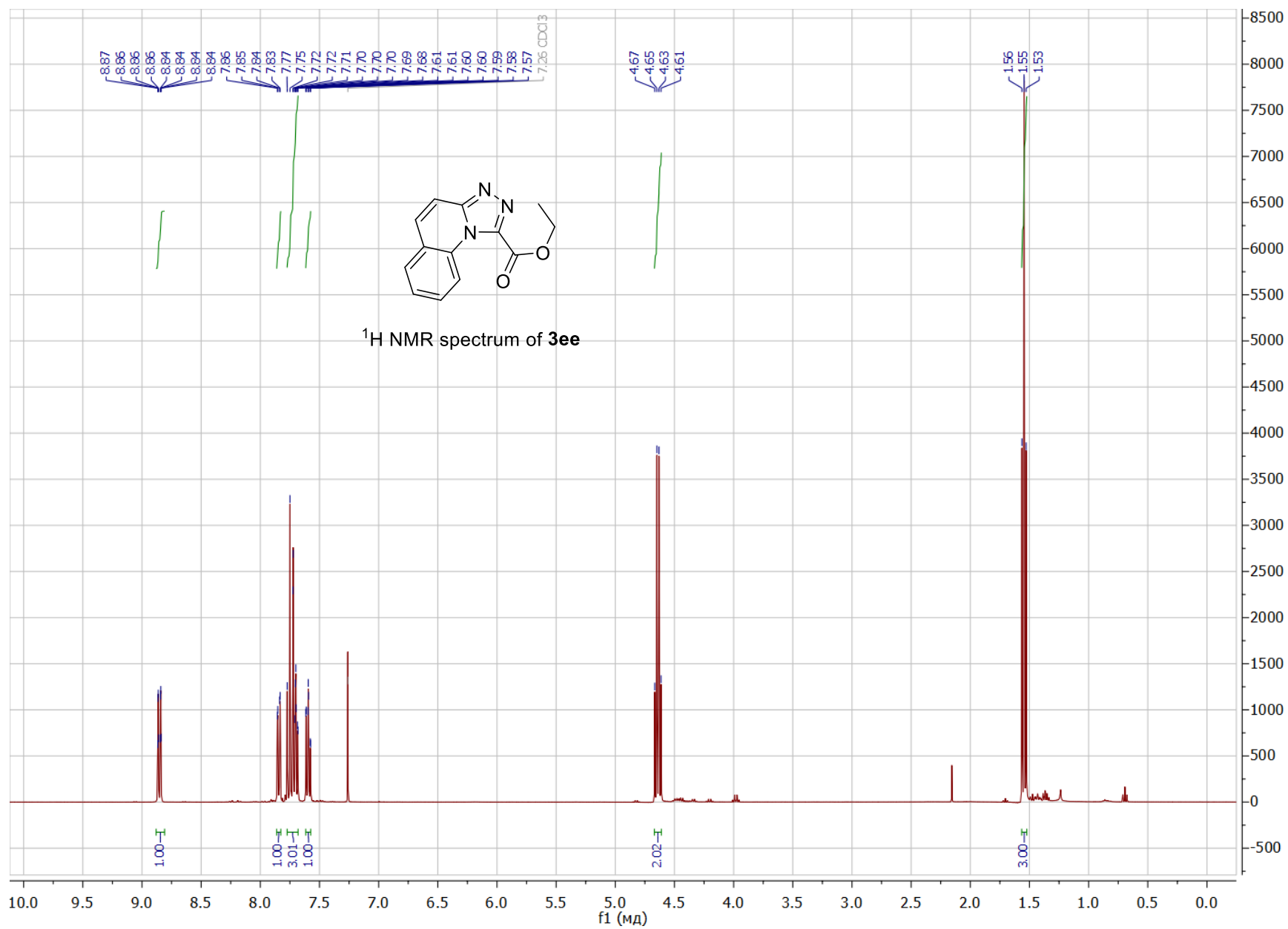


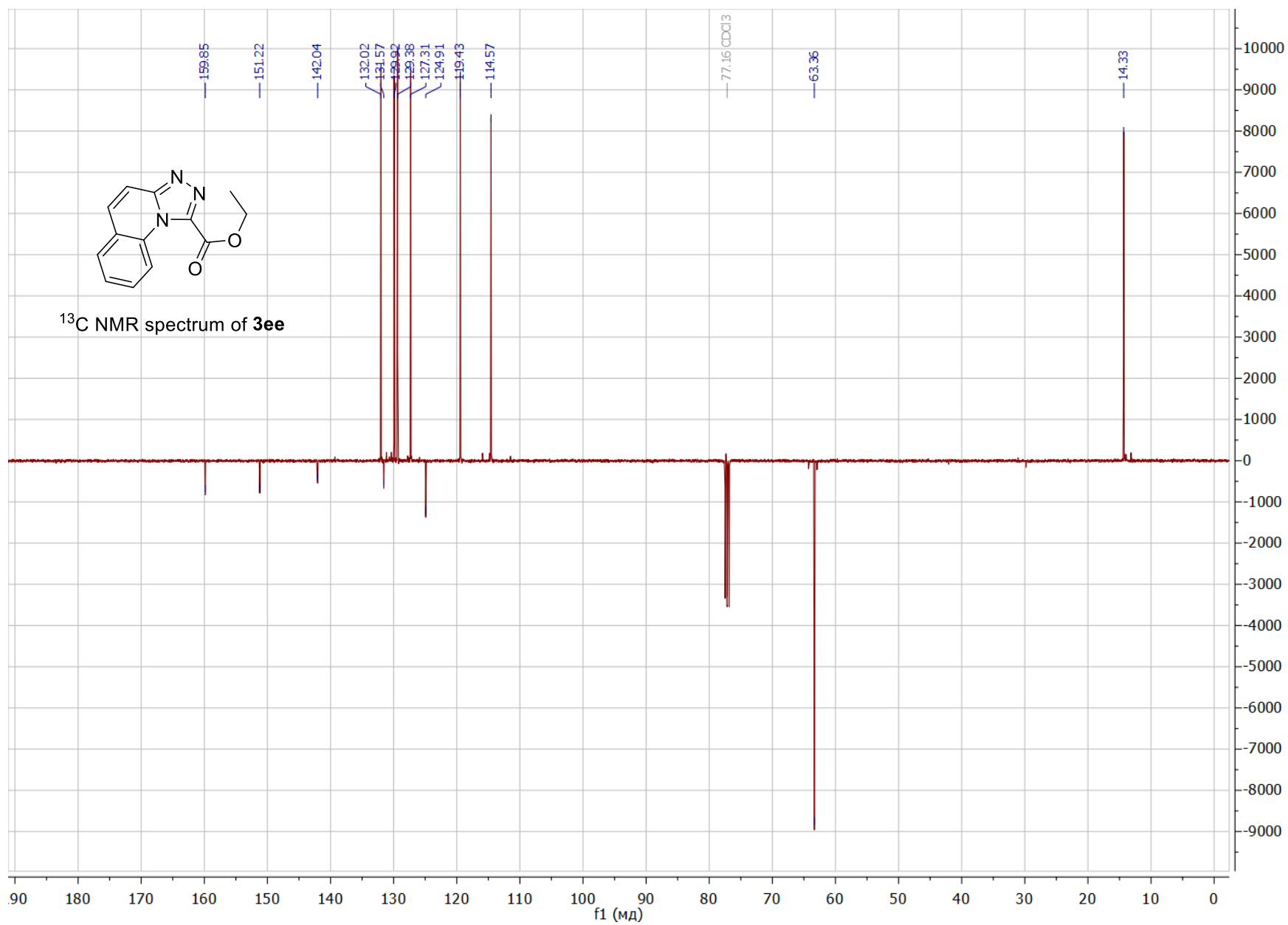


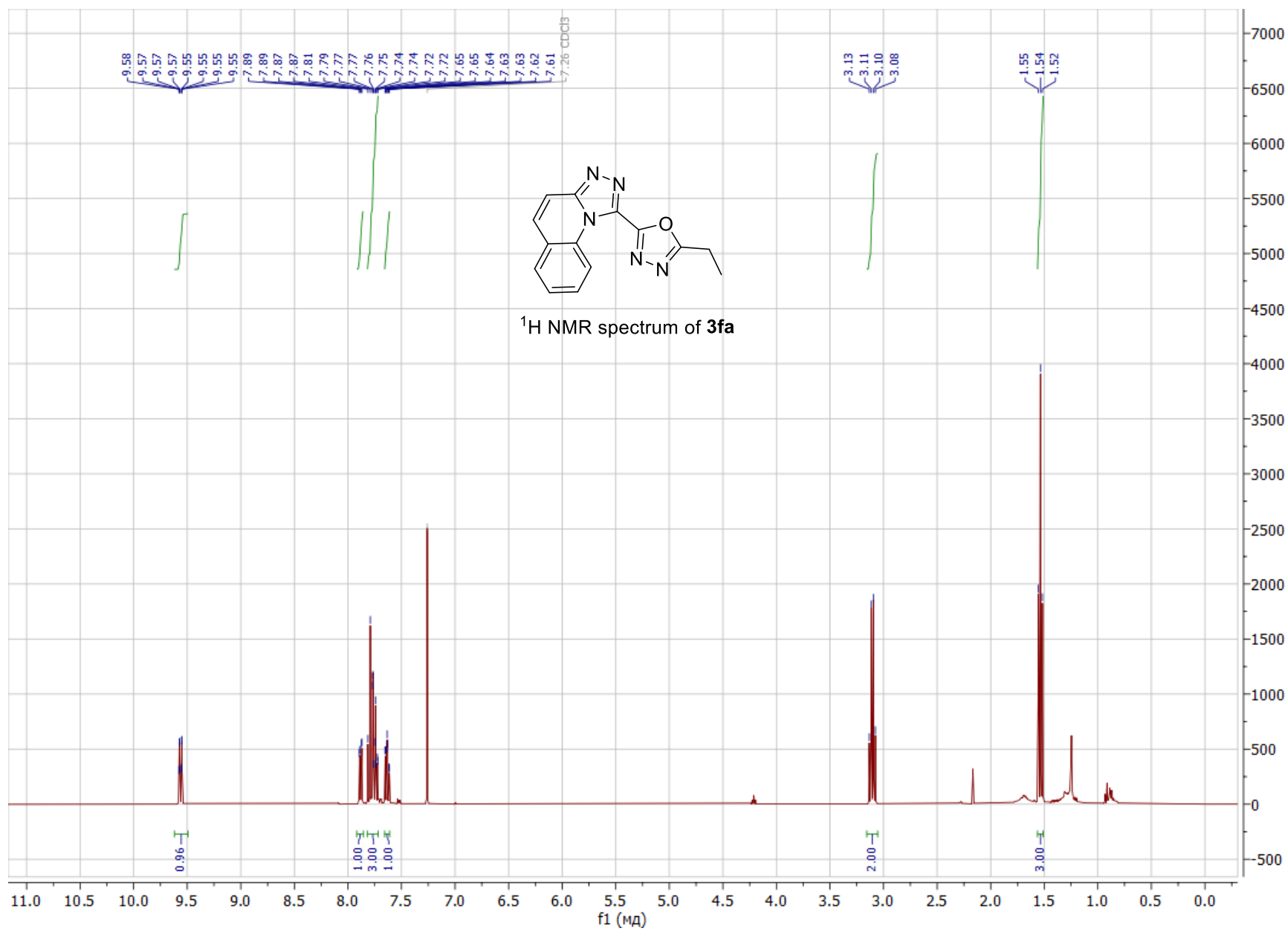


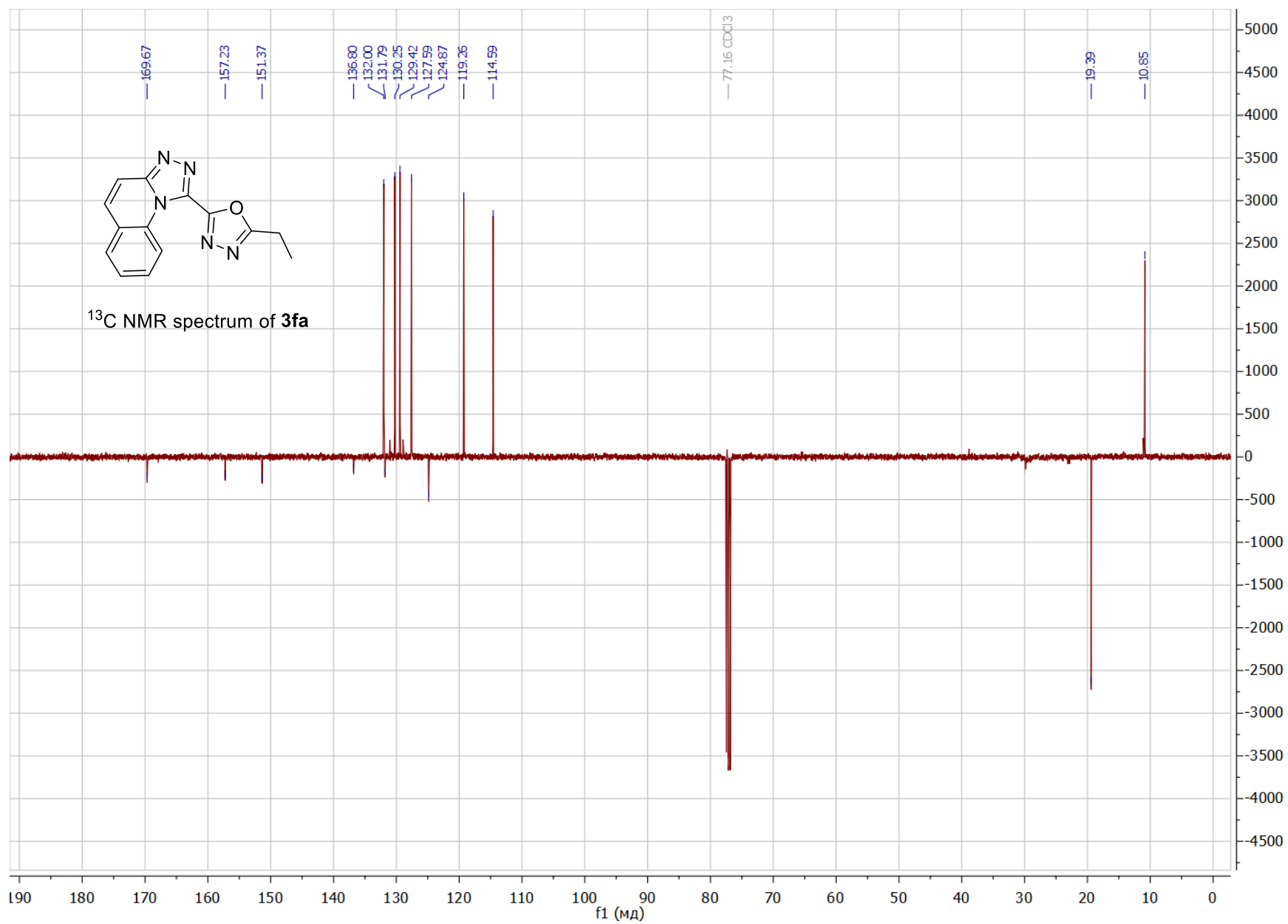


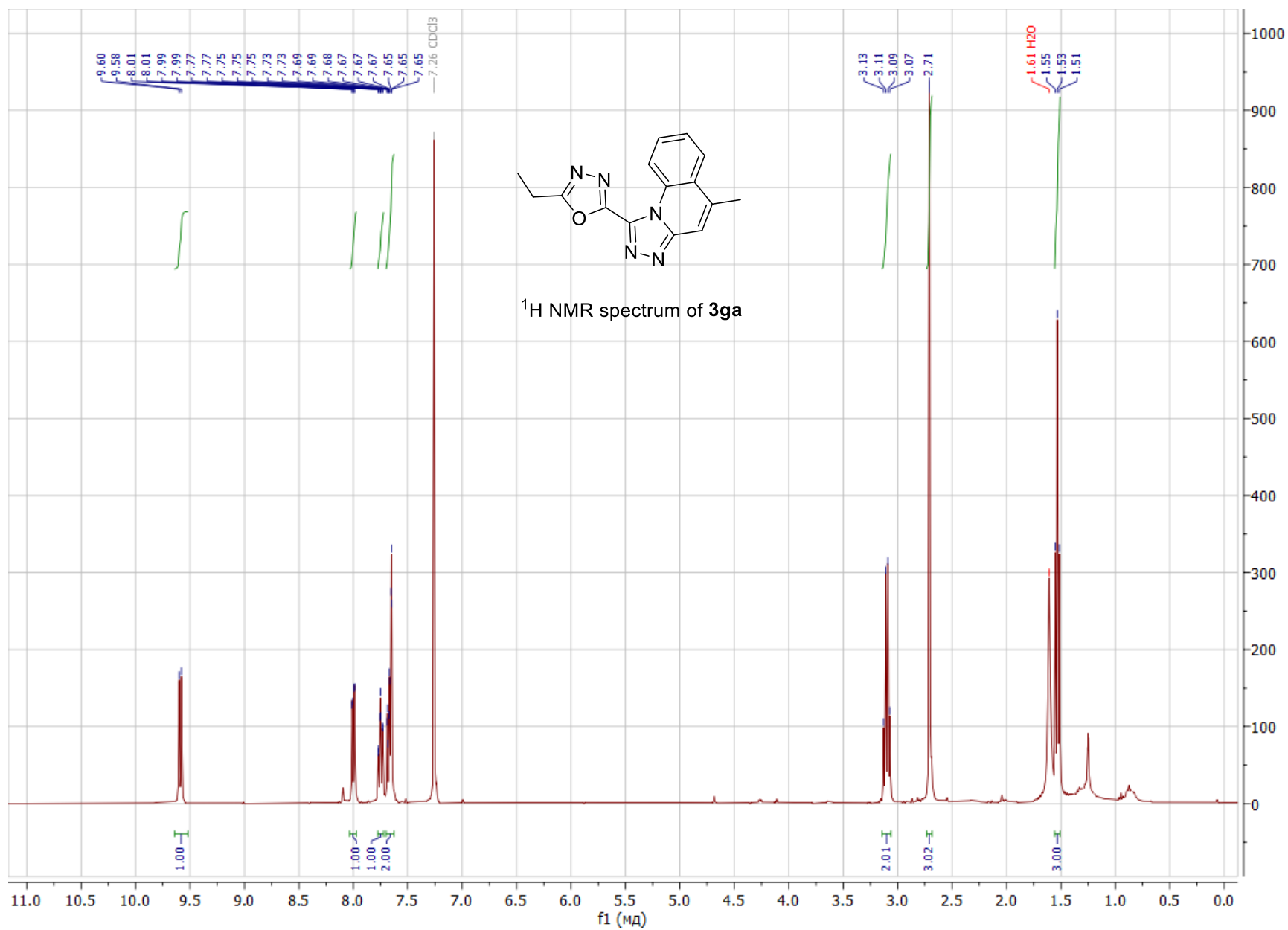


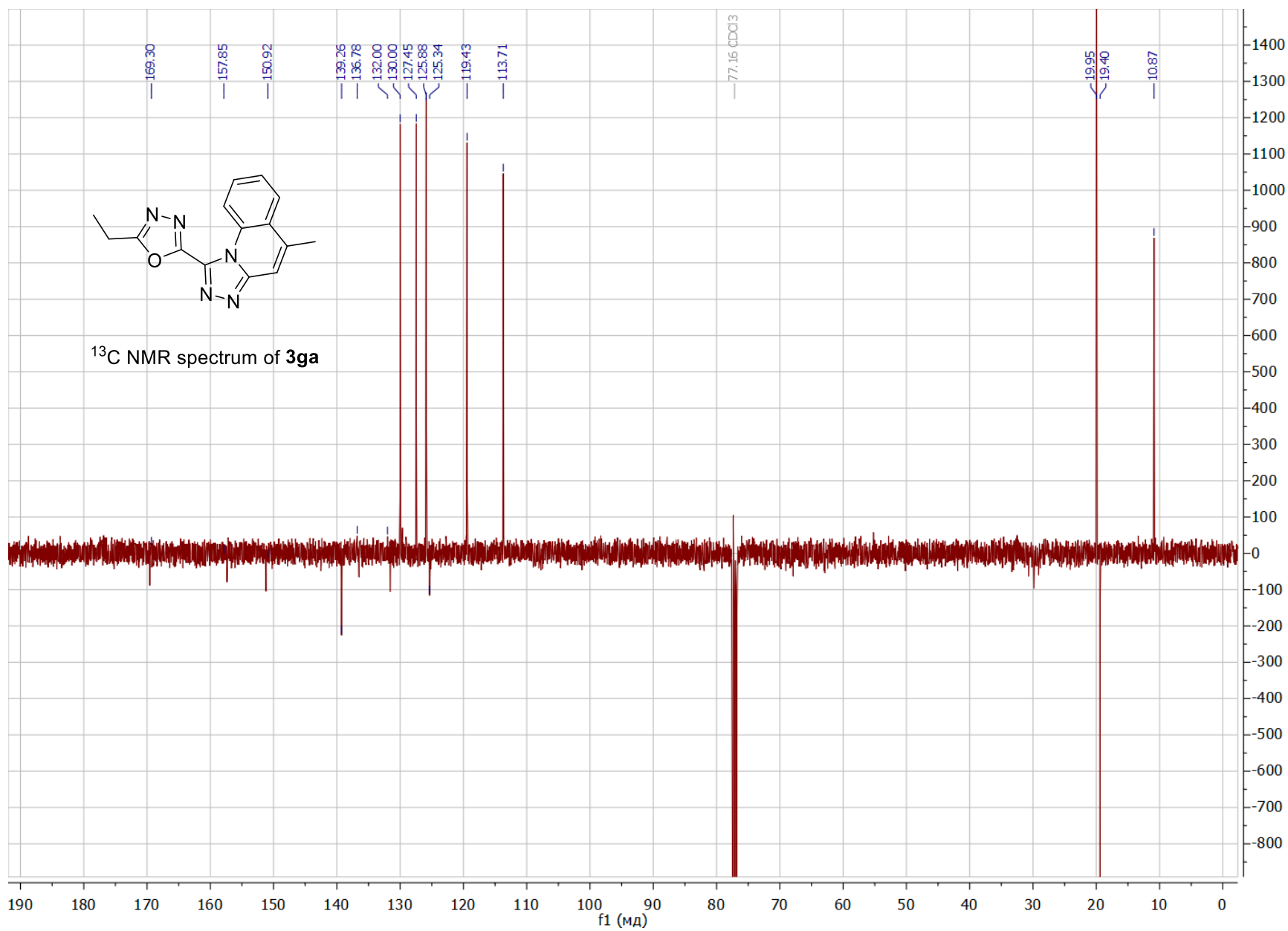




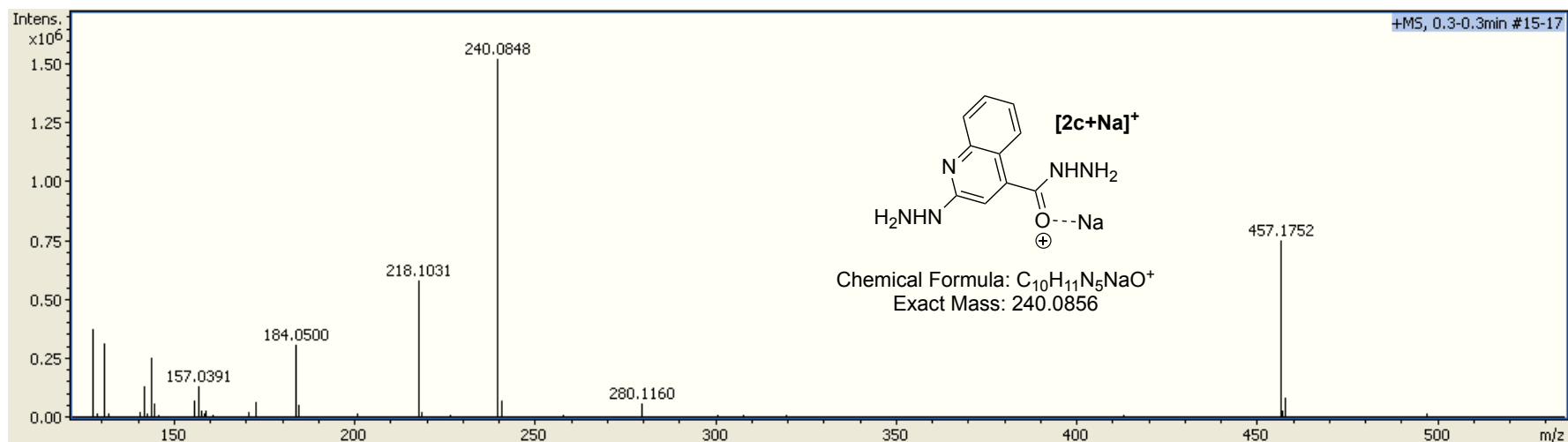
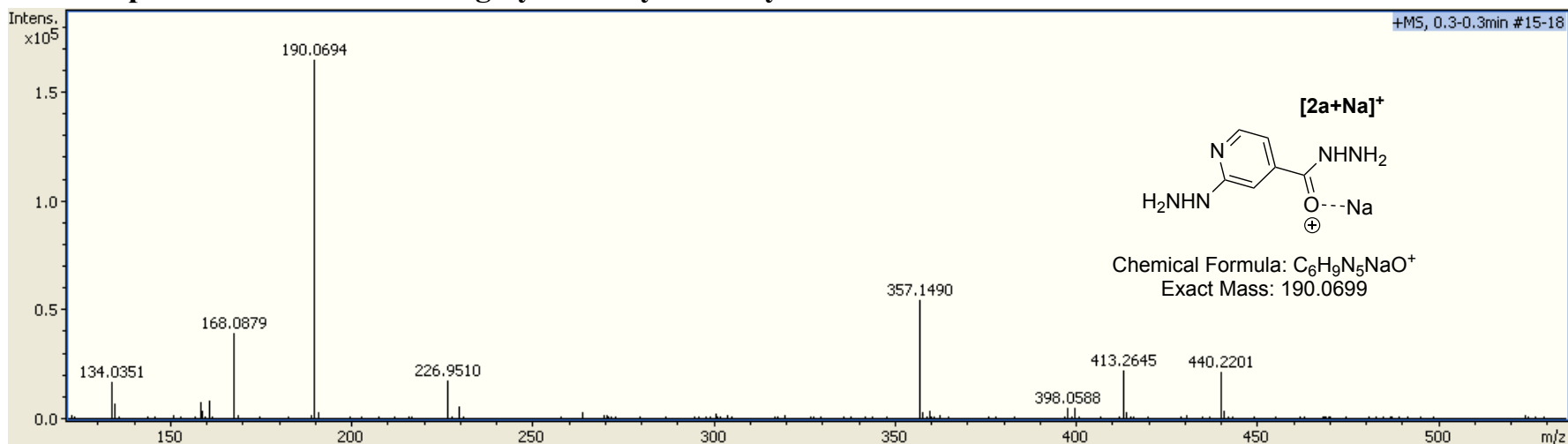


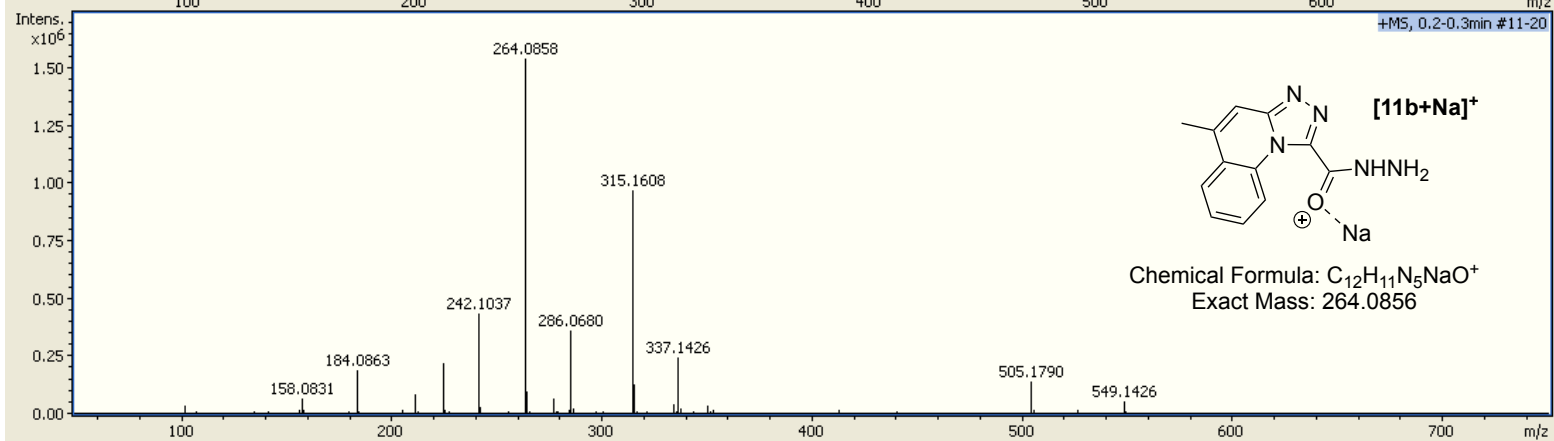
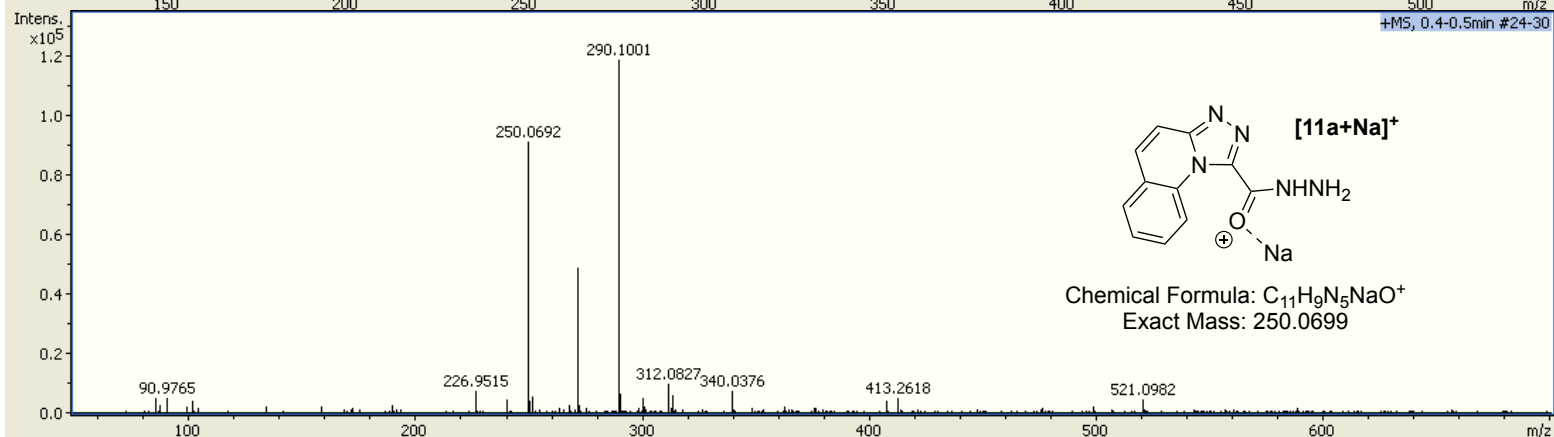
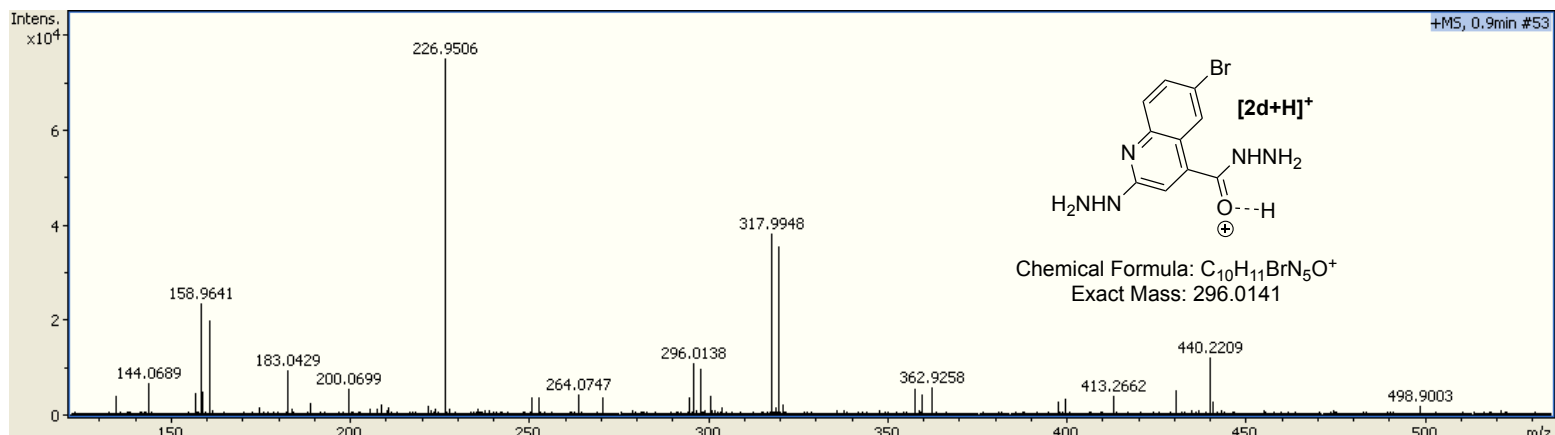




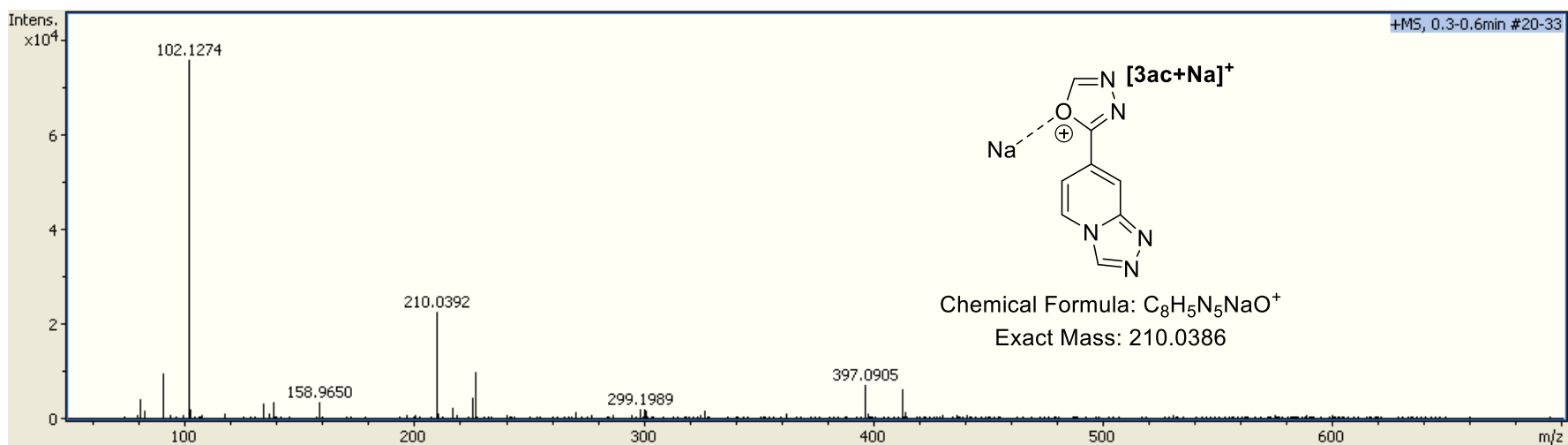
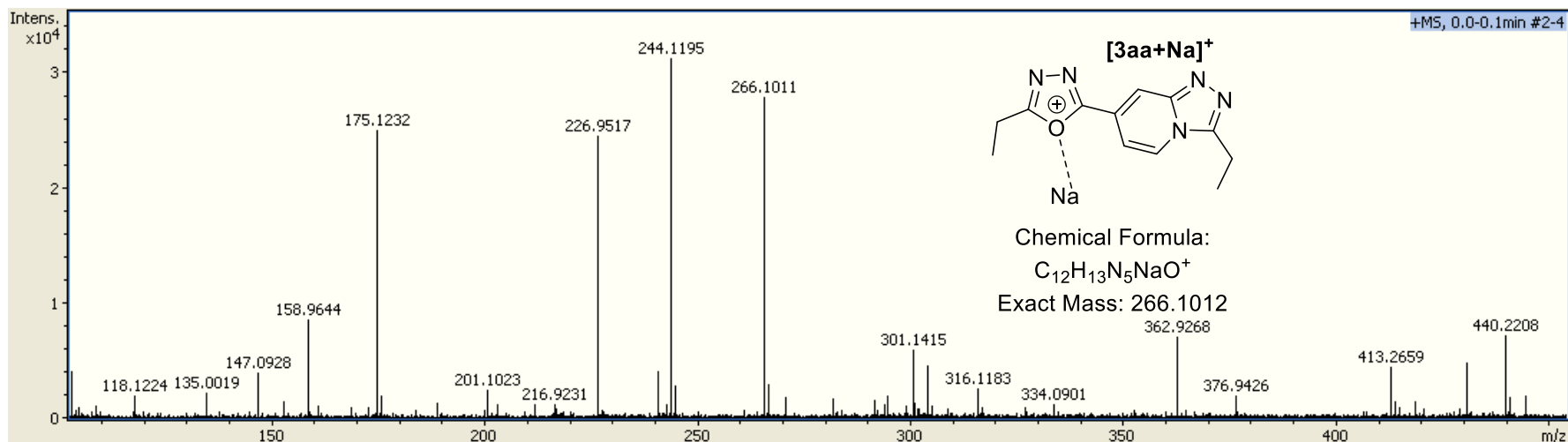


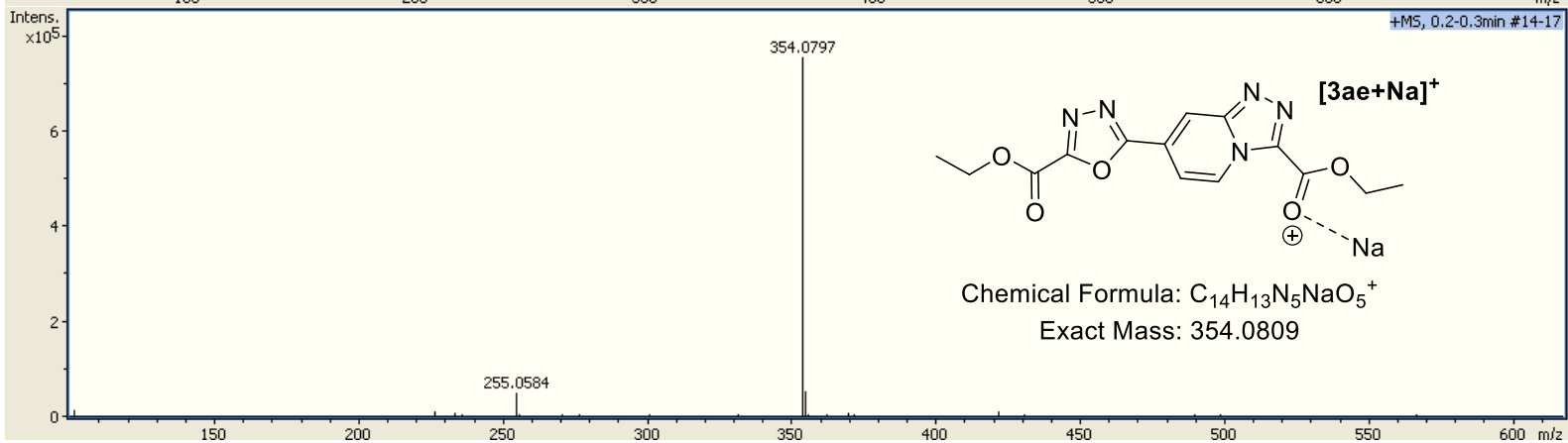
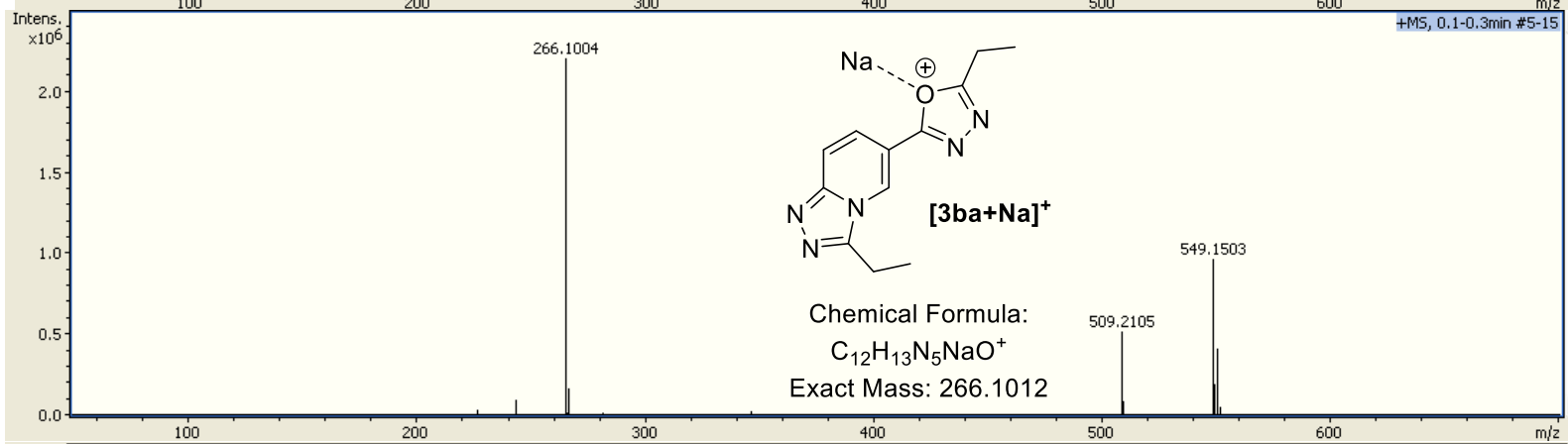
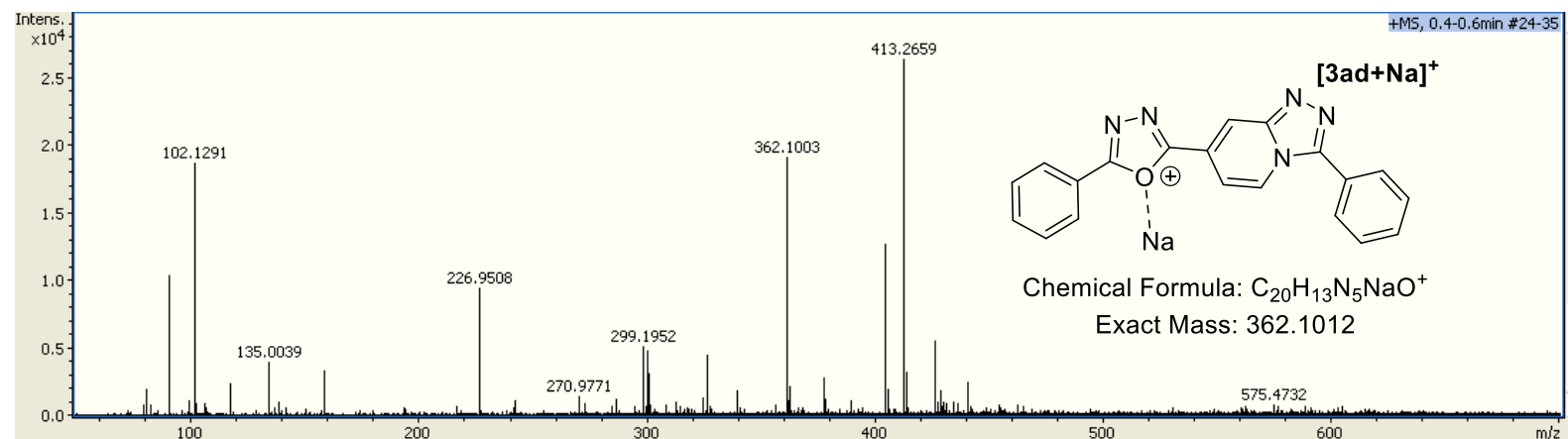
HRMS spectral charts for starting hydrazineylcarbohydrazides



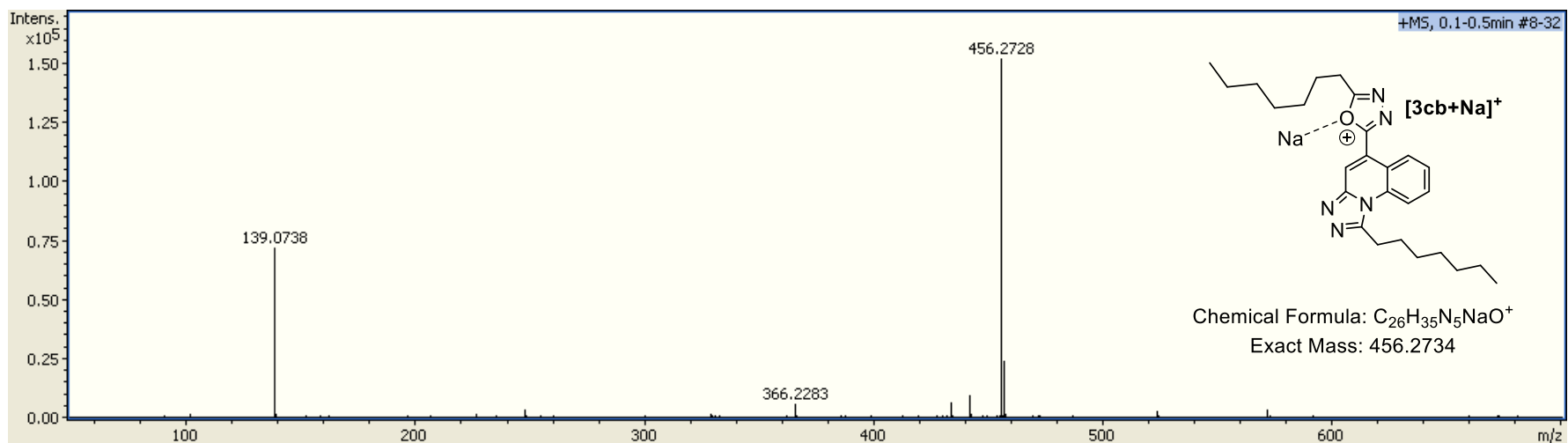
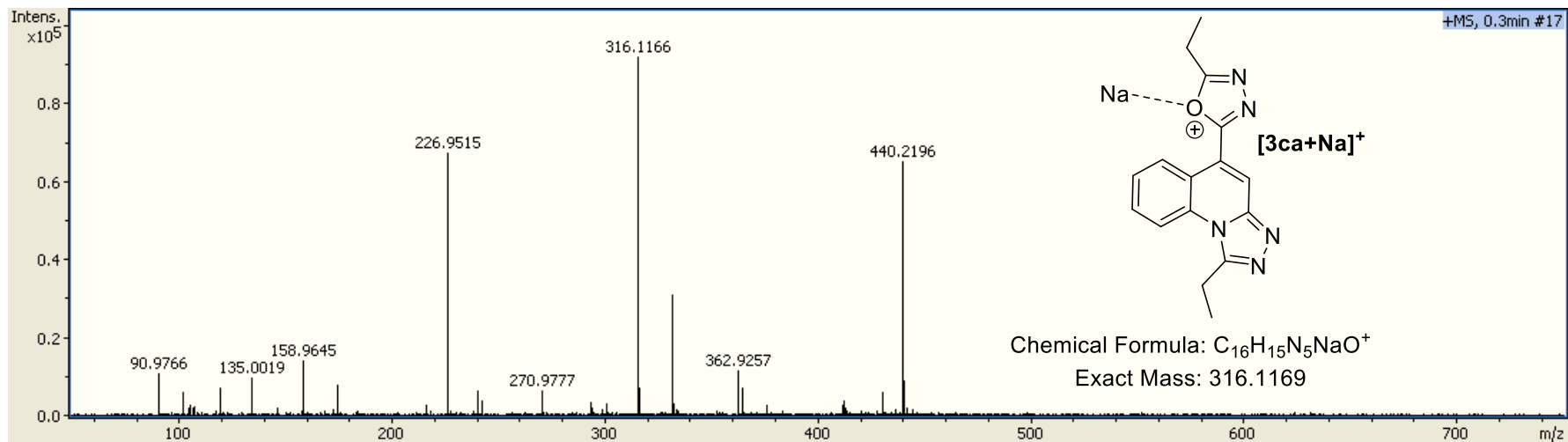


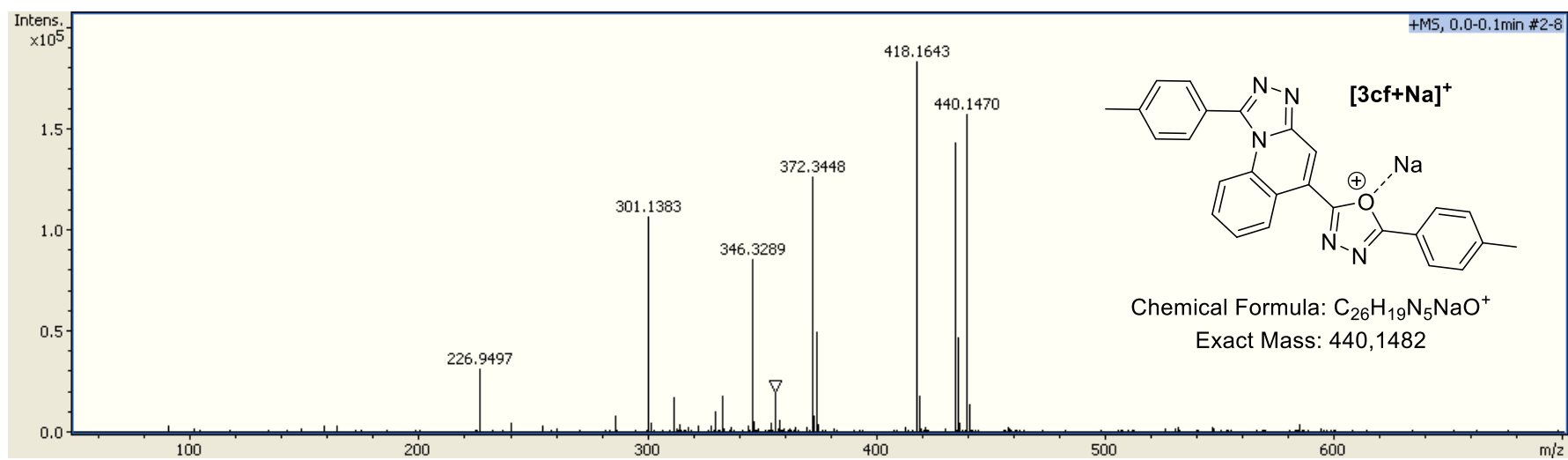
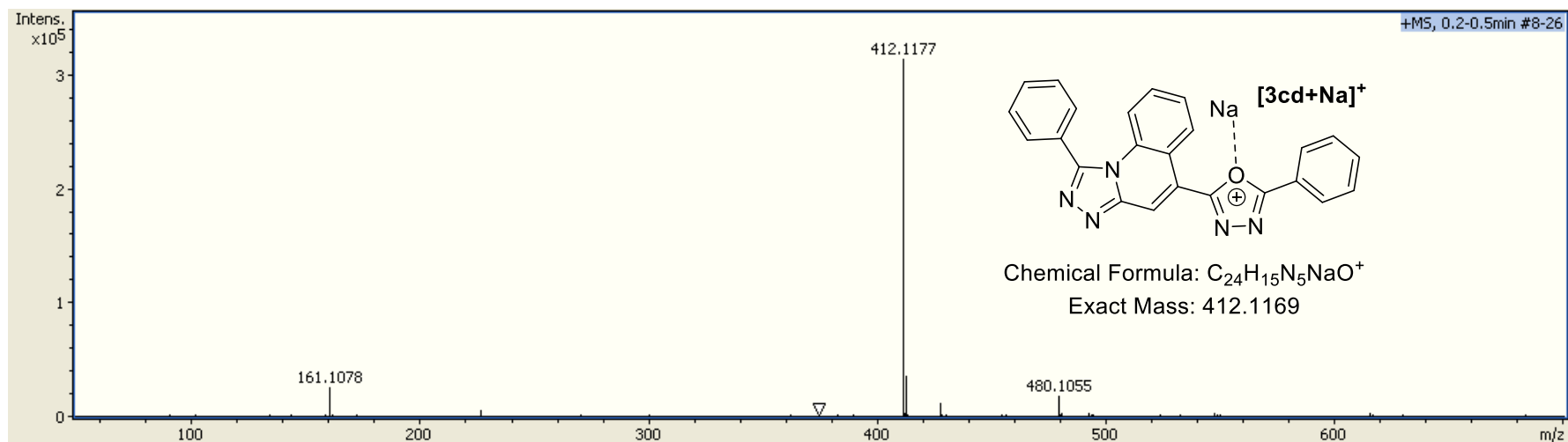
HRMS spectral charts for [1,2,4]triazolo[4,3-a]pyridin-(6)7-yl)-1,3,4-oxadiazoles

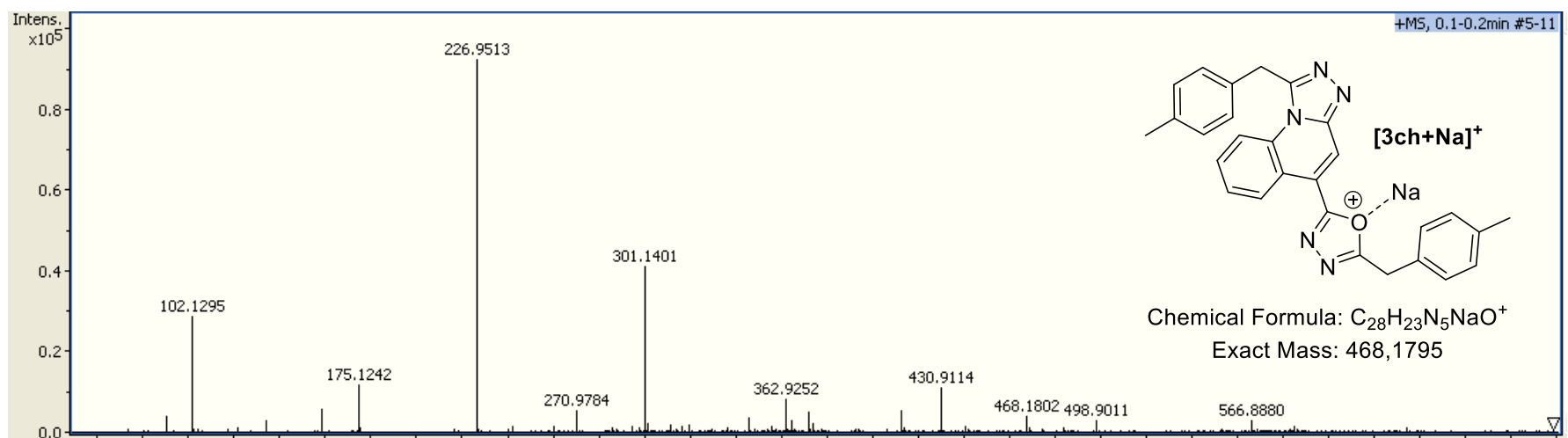
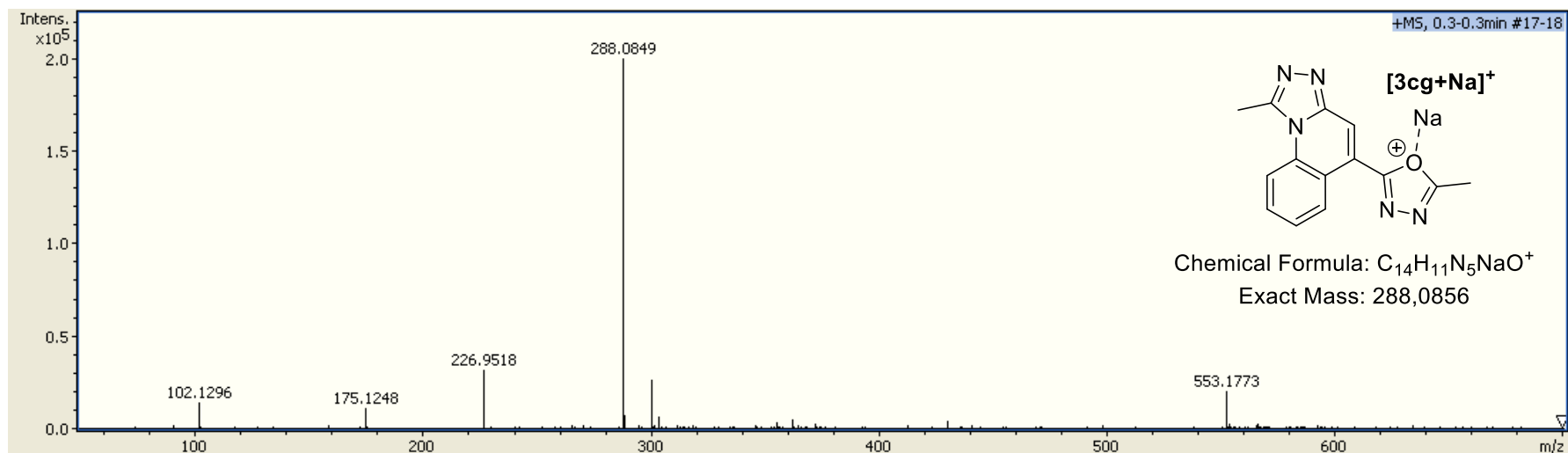


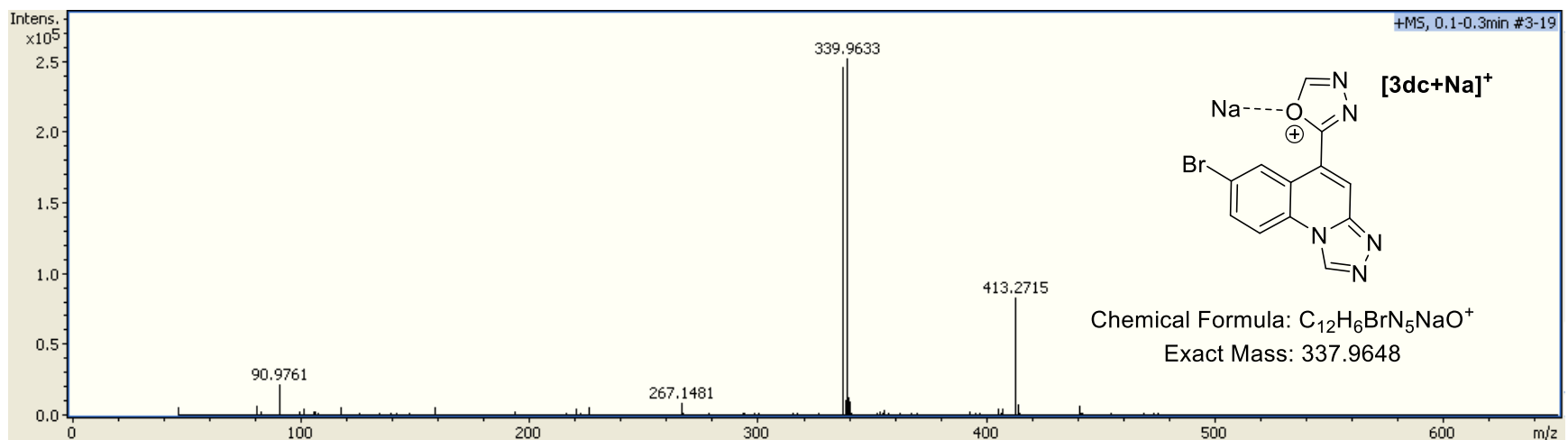
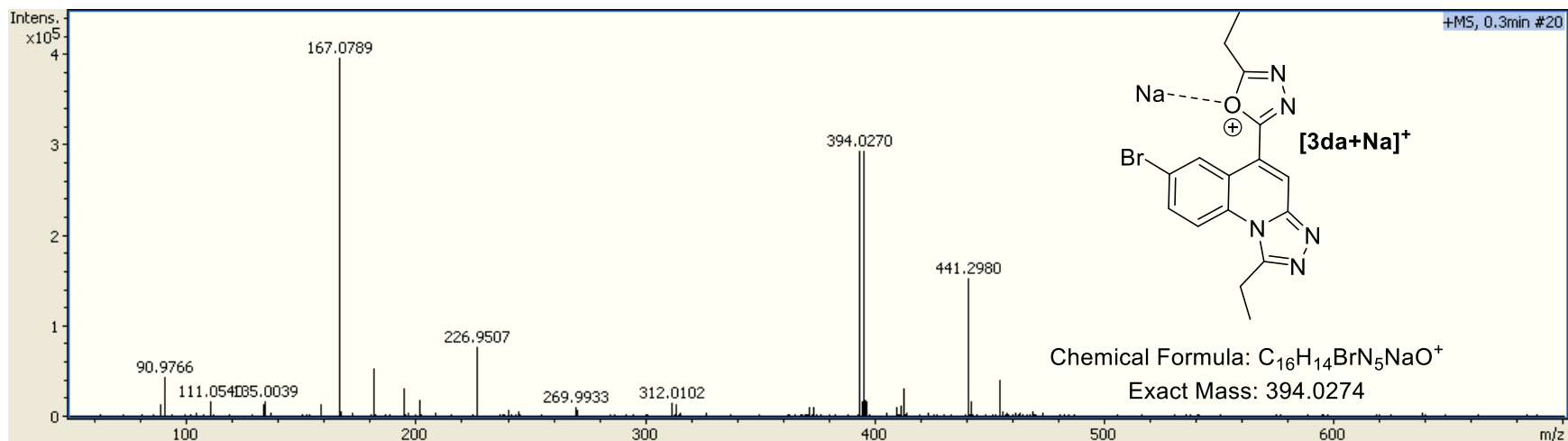


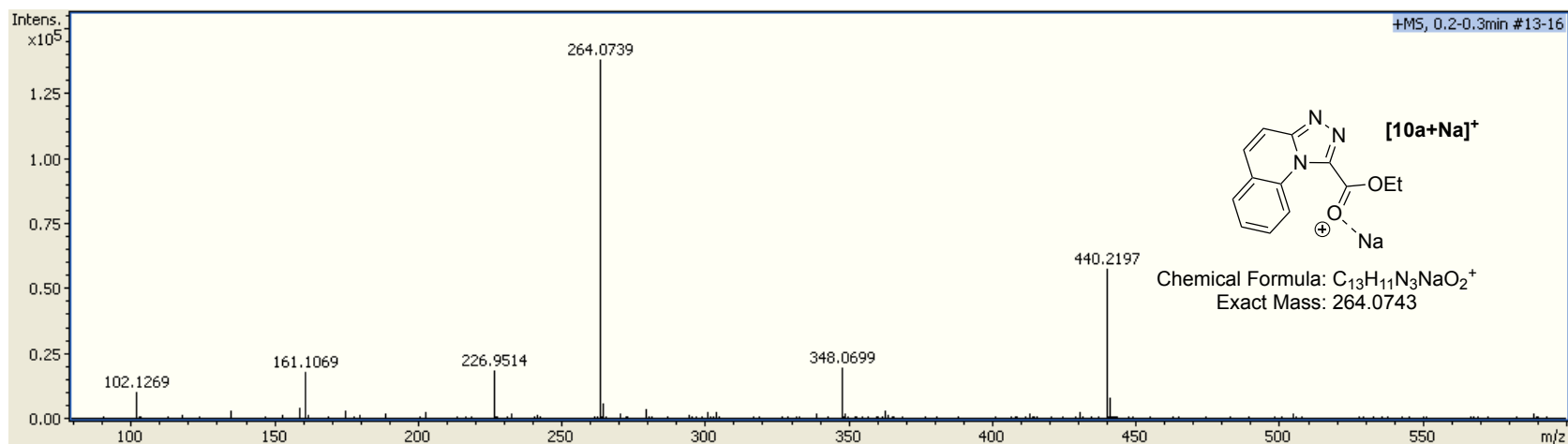
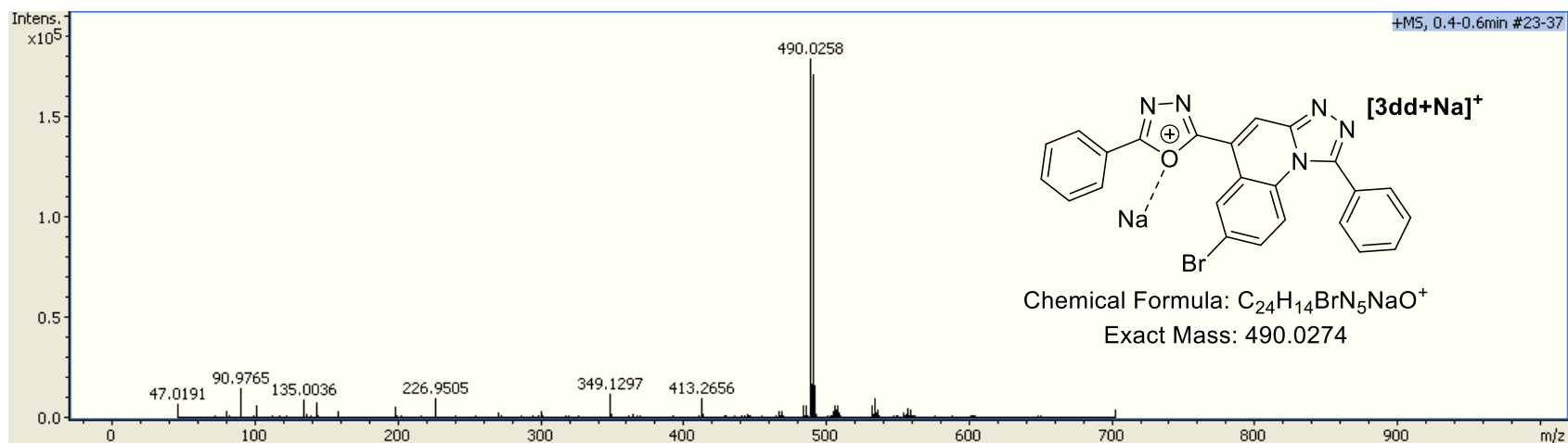
HRMS spectral charts for (1,3,4-oxadiazol-2-yl)[1,2,4]triazolo[4,3-a]quinolines

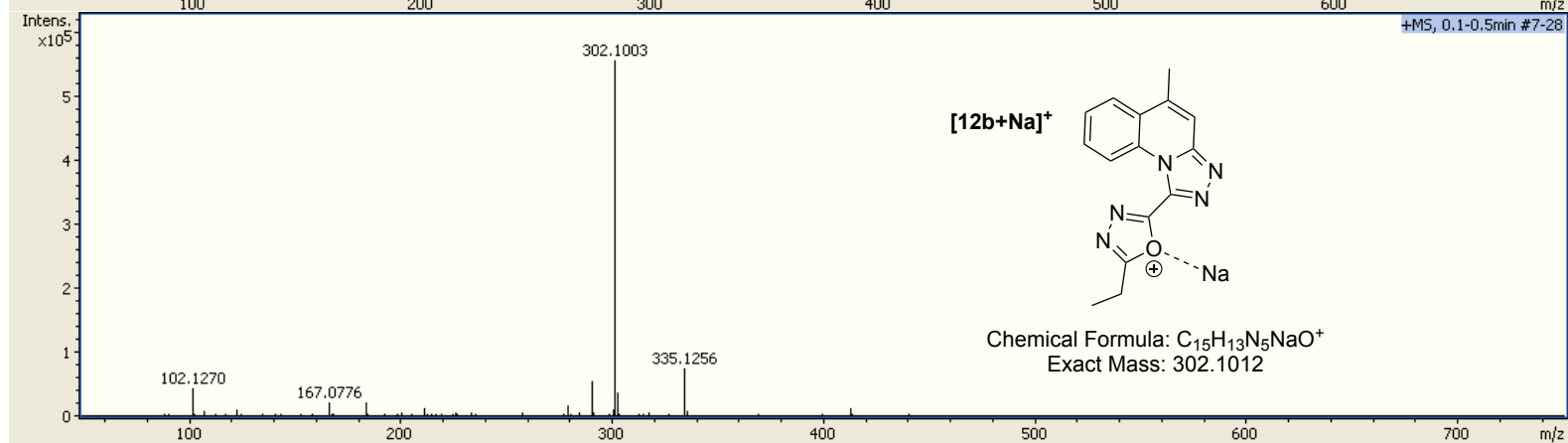
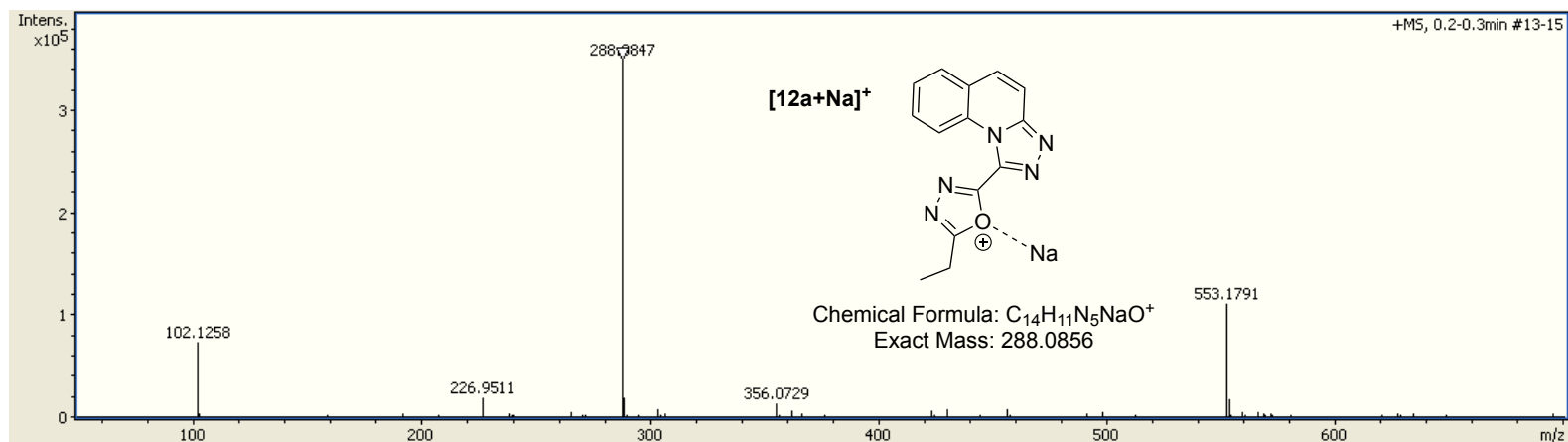












X-Ray crystallography data

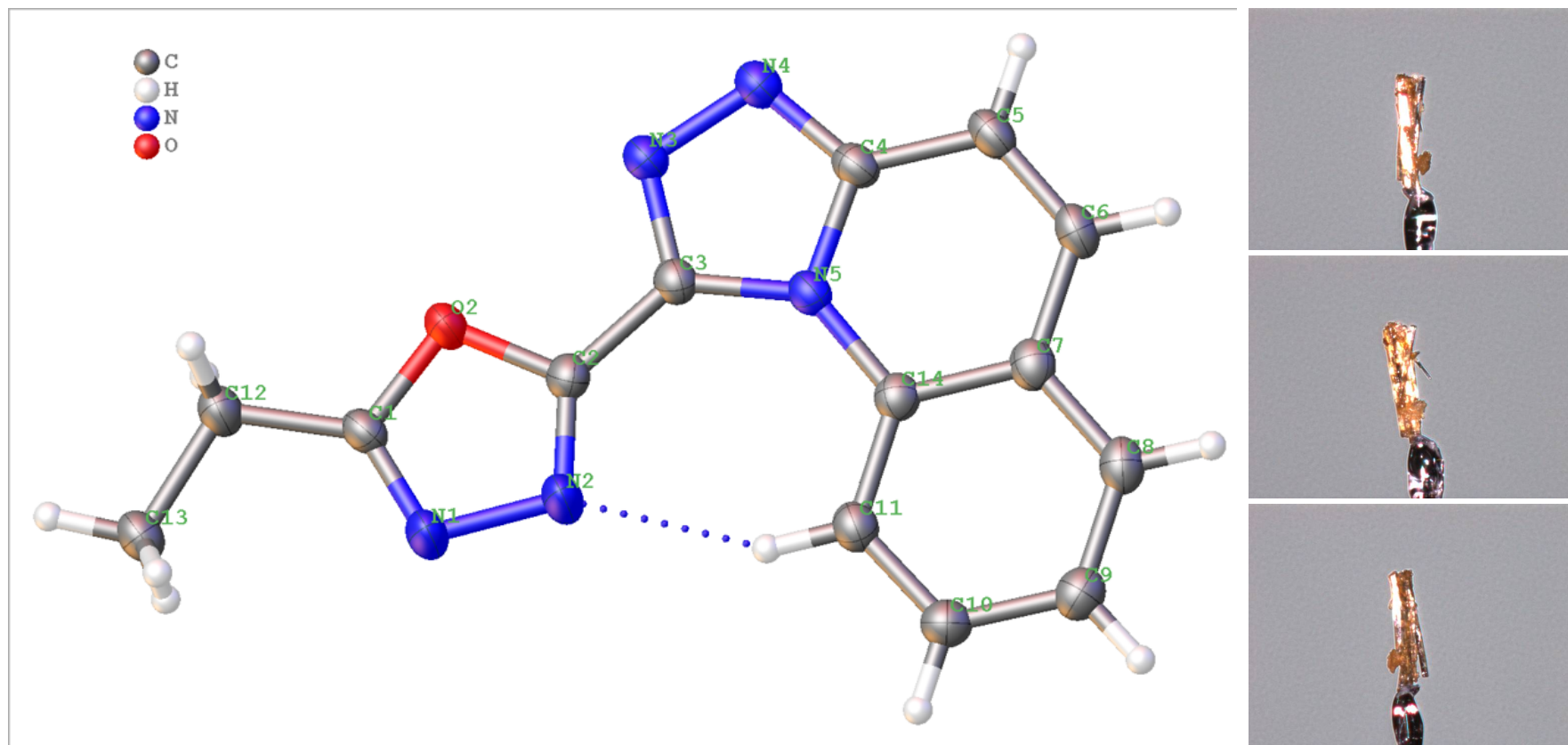


Figure S1. ORTEP drawing of the crystal structure (left) and microphotography of the single crystal of compound **12a** used for X-Ray diffraction analysis (right)

Table S1 Crystal data and structure refinement for 12a.

Identification code	ANNA_NIK424_2
Empirical formula	C ₁₄ H ₁₁ N ₅ O
Formula weight	265.28
Temperature/K	100.01(10)
Crystal system	monoclinic
Space group	P2 ₁ /c
a/Å	4.14345(14)
b/Å	11.4862(4)
c/Å	24.9424(8)
α /°	90
β /°	90.307(3)
γ /°	90
Volume/Å ³	1187.06(7)
Z	4
ρ_{calc} /cm ³	1.484
μ /mm ⁻¹	0.820
F(000)	552.0
Crystal size/mm ³	0.399 × 0.115 × 0.084
Radiation	Cu K α (λ = 1.54184)
2 Θ range for data collection/°	7.088 to 152.766
Index ranges	-5 ≤ h ≤ 5, -14 ≤ k ≤ 14, -30 ≤ l ≤ 31
Reflections collected	27813
Independent reflections	2489 [R_{int} = 0.1099, R_{sigma} = 0.0345]
Data/restraints/parameters	2489/0/183
Goodness-of-fit on F ²	1.058
Final R indexes [$I \geq 2\sigma(I)$]	R_1 = 0.0516, wR_2 = 0.1256
Final R indexes [all data]	R_1 = 0.0535, wR_2 = 0.1269
Largest diff. peak/hole / e Å ⁻³	0.22/-0.28

Table S2 Fractional Atomic Coordinates ($\times 10^4$) and Equivalent Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for 12a. U_{eq} is defined as 1/3 of the trace of the orthogonalised U_{IJ} tensor.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	$U(\text{eq})$
O2	2526(4)	1398.2(13)	5178.6(6)	30.9(4)
N4	2246(5)	1623.3(16)	6777.8(7)	29.1(4)
N5	5273(4)	2998.7(15)	6394.7(7)	23.8(4)
N3	2087(5)	1483.0(15)	6235.4(7)	27.5(4)
N2	5764(6)	2890.5(19)	5090.2(8)	46.5(6)
C4	4160(5)	2525.7(18)	6869.7(8)	25.6(4)
C14	7285(5)	3997.3(17)	6395.0(8)	24.5(4)
C7	8097(5)	4478.3(18)	6900.1(8)	26.7(4)
N1	5171(6)	2396.7(18)	4577.8(8)	42.8(6)
C3	3865(5)	2293.9(17)	6006.7(8)	24.3(4)
C11	8406(5)	4514.2(19)	5926.6(9)	27.8(5)
C2	4176(5)	2273.8(17)	5424.4(8)	25.5(4)
C1	3289(5)	1539.5(18)	4651.9(8)	25.7(4)
C8	10053(5)	5474.5(19)	6915.7(9)	29.3(5)
C6	6925(5)	3961.7(19)	7386.8(8)	28.6(5)
C5	5036(6)	3012.0(19)	7375.5(8)	28.4(5)
C10	10342(6)	5488.6(19)	5957.2(9)	29.7(5)
C9	11185(6)	5977.1(19)	6450.9(9)	30.7(5)
C13	2839(6)	982(2)	3684.0(8)	30.8(5)
C12	1955(6)	699.7(19)	4258.7(8)	31.3(5)

Table S3 Anisotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for 12a. The Anisotropic displacement factor exponent takes the form: - $2\pi^2[h^2a^{*2}U_{11}+2hka^*b^*U_{12}+\dots]$.

Atom	U_{11}	U_{22}	U_{33}	U_{23}	U_{13}	U_{12}
O2	40.4(9)	34.2(8)	18.0(7)	-1.9(6)	2.3(6)	-10.4(7)
N4	36.1(10)	30.0(9)	21.1(8)	0.2(7)	1.2(7)	-2.8(8)
N5	28.2(9)	24.1(8)	19.2(8)	-0.6(6)	1.9(7)	0.6(7)
N3	34.9(10)	27.1(9)	20.6(8)	-0.4(6)	1.5(7)	-3.1(8)
N2	73.2(16)	46.9(12)	19.6(9)	-6.9(8)	6.5(10)	-30.1(12)
C4	29.9(10)	27.1(10)	19.7(10)	1.7(8)	2.3(8)	1.7(9)
C14	25.1(9)	23.8(9)	24.7(10)	-0.9(8)	-0.8(8)	1.4(8)
C7	28.3(10)	25.7(10)	26.0(10)	-2.8(8)	-1.1(9)	3.5(8)
N1	66.8(15)	42.1(11)	19.7(9)	-6.0(8)	6.3(10)	-23.6(11)
C3	28.2(10)	23.4(9)	21.4(10)	-1.4(7)	0.5(8)	0.3(8)
C11	31.5(11)	28.4(10)	23.5(10)	-1.8(8)	-0.7(9)	0.1(9)
C2	31.0(11)	22.5(9)	23.0(10)	-1.6(8)	0.3(8)	-1.0(8)
C1	31.5(11)	27.4(10)	18.3(9)	0.9(8)	2.2(8)	0.1(9)
C8	30.3(11)	30.0(11)	27.7(11)	-5.9(8)	-3.5(9)	2.4(9)
C6	30.9(11)	34.6(11)	20.1(9)	-4.2(8)	-1.4(8)	2.3(9)
C5	34.0(11)	32.9(11)	18.4(9)	-0.4(8)	1.8(9)	3.1(9)
C10	32.7(11)	28.5(11)	27.9(10)	1.8(8)	1.4(9)	-2.4(9)
C9	31.2(11)	27.7(10)	33.1(11)	-1.9(9)	-1.0(9)	-1.7(9)
C13	37.1(12)	34.4(11)	20.9(10)	-1.3(8)	0.6(9)	-1.6(10)
C12	37.3(12)	34.9(11)	21.8(10)	-2.5(9)	-0.1(9)	-6.4(9)

Table S4 Bond Lengths for 12a.

Atom	Atom	Length/Å	Atom	Atom	Length/Å
O2	C2	1.360(3)	C14	C11	1.392(3)
O2	C1	1.363(2)	C7	C8	1.403(3)
N4	N3	1.364(2)	C7	C6	1.438(3)
N4	C4	1.324(3)	N1	C1	1.270(3)
N5	C4	1.385(3)	C3	C2	1.459(3)
N5	C14	1.418(3)	C11	C10	1.379(3)
N5	C3	1.388(3)	C1	C12	1.480(3)
N3	C3	1.319(3)	C8	C9	1.380(3)
N2	N1	1.419(3)	C6	C5	1.343(3)
N2	C2	1.279(3)	C10	C9	1.396(3)
C4	C5	1.425(3)	C13	C12	1.516(3)
C14	C7	1.415(3)			

Table S5 Bond Angles for 12a.

Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
C2	O2	C1	103.17(16)	N5	C3	C2	131.64(19)
C4	N4	N3	106.85(17)	N3	C3	N5	110.13(17)
C4	N5	C14	121.02(17)	N3	C3	C2	118.14(19)
C4	N5	C3	103.13(17)	C10	C11	C14	119.8(2)
C3	N5	C14	135.82(18)	O2	C2	C3	114.43(18)
C3	N3	N4	108.74(17)	N2	C2	O2	112.02(19)
C2	N2	N1	106.17(19)	N2	C2	C3	133.5(2)
N4	C4	N5	111.15(18)	O2	C1	C12	118.25(18)
N4	C4	C5	127.56(19)	N1	C1	O2	112.24(19)
N5	C4	C5	121.28(19)	N1	C1	C12	129.49(19)
C7	C14	N5	116.97(18)	C9	C8	C7	121.1(2)
C11	C14	N5	122.90(19)	C5	C6	C7	121.1(2)
C11	C14	C7	120.12(19)	C6	C5	C4	118.86(19)
C14	C7	C6	120.7(2)	C11	C10	C9	121.2(2)
C8	C7	C14	118.55(19)	C8	C9	C10	119.3(2)
C8	C7	C6	120.72(19)	C1	C12	C13	113.29(18)
C1	N1	N2	106.40(18)				

Table S6 Torsion Angles for 12a.

A	B	C	D	Angle/°	A	B	C	D	Angle/°
O2	C1	C12	C13	177.17(19)	C14	N5	C3	C2	5.8(4)
N4	N3	C3	N5	0.2(2)	C14	C7	C8	C9	-0.3(3)
N4	N3	C3	C2	177.01(19)	C14	C7	C6	C5	0.0(3)
N4	C4	C5	C6	-177.5(2)	C14	C11	C10	C9	-0.6(3)
N5	C4	C5	C6	1.4(3)	C7	C14	C11	C10	0.8(3)
N5	C14	C7	C8	-179.28(19)	C7	C8	C9	C10	0.5(3)
N5	C14	C7	C6	0.3(3)	C7	C6	C5	C4	-0.8(3)
N5	C14	C11	C10	179.6(2)	N1	N2	C2	O2	-0.3(3)
N5	C3	C2	O2	176.4(2)	N1	N2	C2	C3	177.3(2)
N5	C3	C2	N2	-1.1(4)	N1	C1	C12	C13	-4.6(4)
N3	N4	C4	N5	0.4(2)	C3	N5	C4	N4	-0.3(2)
N3	N4	C4	C5	179.4(2)	C3	N5	C4	C5	-179.4(2)
N3	C3	C2	O2	0.4(3)	C3	N5	C14	C7	177.9(2)
N3	C3	C2	N2	-177.1(3)	C3	N5	C14	C11	-1.1(4)
N2	N1	C1	O2	0.0(3)	C11	C14	C7	C8	-0.3(3)
N2	N1	C1	C12	-178.4(2)	C11	C14	C7	C6	179.3(2)
C4	N4	N3	C3	-0.3(2)	C11	C10	C9	C8	-0.1(3)
C4	N5	C14	C7	0.2(3)	C2	O2	C1	N1	-0.1(3)
C4	N5	C14	C11	-178.7(2)	C2	O2	C1	C12	178.4(2)
C4	N5	C3	N3	0.1(2)	C2	N2	N1	C1	0.2(3)
C4	N5	C3	C2	-176.2(2)	C1	O2	C2	N2	0.3(3)
C14	N5	C4	N4	178.03(18)	C1	O2	C2	C3	-177.78(19)
C14	N5	C4	C5	-1.0(3)	C8	C7	C6	C5	179.6(2)
C14	N5	C3	N3	-177.9(2)	C6	C7	C8	C9	-179.9(2)

Table S7 Hydrogen Atom Coordinates ($\text{\AA} \times 10^4$) and Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for 12a.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	U(eq)
H11	7850.91	4203.36	5594.43	33
H8	10597.08	5801.78	7245.1	35
H6	7485.73	4290.87	7714.98	34
H5	4311.97	2677.44	7692.41	34
H10	11099.79	5826.41	5643.3	36
H9	12494.24	6633.94	6466.25	37
H13A	1982.93	393.66	3450.08	46
H13B	5145.53	1007.21	3650.87	46
H13C	1949.04	1725.45	3587.39	46
H12A	-376.84	688.37	4289.8	38
H12B	2744.66	-73.06	4345.53	38

Experimental

Single crystals of $\text{C}_{14}\text{H}_{11}\text{N}_5\text{O}$ (**3fa**) were crystallized by slow evaporation of EtOAc solution. A suitable crystal was selected and mounted by acrylic glue on the glass stick on a **SuperNova, Dual, Cu at home/near, AtlasS2** diffractometer. The crystal was kept at 100.01(10) K during data collection. Using Olex2 [S1], the structure was solved with the SHELXT [S2] structure solution program using Intrinsic Phasing and refined with the SHELXL [S3] refinement package using Least Squares minimization.

Crystal structure determination of 3fa

Crystal Data for $\text{C}_{14}\text{H}_{11}\text{N}_5\text{O}$ ($M = 265.28$ g/mol): monoclinic, space group $\text{P2}_1/\text{c}$ (no. 14), $a = 4.14345(14)$ \AA , $b = 11.4862(4)$ \AA , $c = 24.9424(8)$ \AA , $\beta = 90.307(3)^\circ$, $V = 1187.06(7)$ \AA^3 , $Z = 4$, $T = 100.01(10)$ K, $\mu(\text{Cu K}\alpha) = 0.820$ mm^{-1} , $D_{\text{calc}} = 1.484$ g/cm^3 , 27813 reflections measured ($7.088^\circ \leq 2\theta \leq 152.766^\circ$), 2489 unique ($R_{\text{int}} = 0.1099$, $R_{\text{sigma}} = 0.0345$) which were used in all calculations. The final R_1 was 0.0516 ($I > 2\sigma(I)$) and wR_2 was 0.1269 (all data).

Refinement model description of 3fa

Number of restraints - 0, number of constraints - unknown.

Details:

1. Twinned data refinement

Scales: 0.8428(18)

0.1572(18)

2. Fixed Uiso

At 1.2 times of:

All C(H) groups, All C(H,H) groups

At 1.5 times of:

All C(H,H,H) groups

3.a Secondary CH₂ refined with riding coordinates:

C12(H12A,H12B)

3.b Aromatic/amide H refined with riding coordinates:

C11(H11), C8(H8), C6(H6), C5(H5), C10(H10), C9(H9)

3.c Idealised Me refined as rotating group:

C13(H13A,H13B,H13C)

X-Ray crystallography data

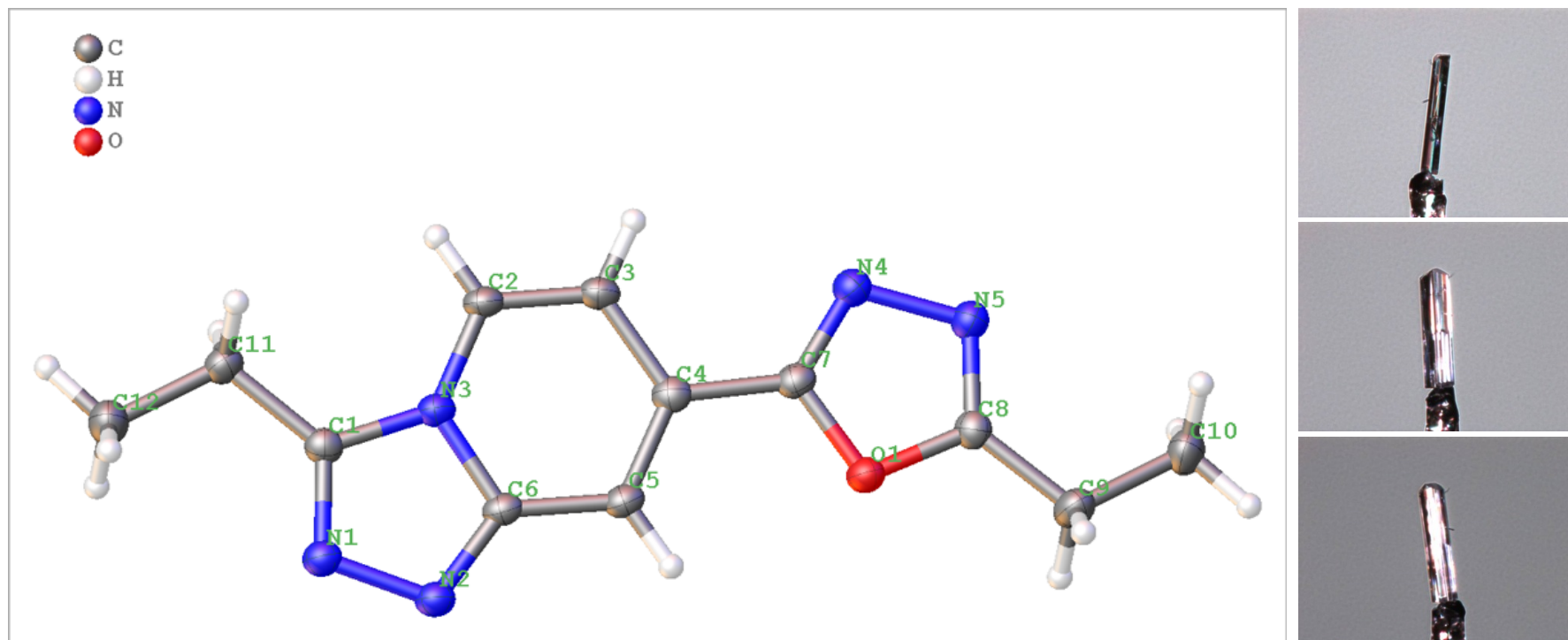


Figure S2. ORTEP drawing of the crystal structure (left) and microphotography of the single crystal of compound **3ba** used for X-Ray diffraction analysis (right)

Table S9 Crystal data and structure refinement for 3ba.

Identification code	ANNA_SOK3_2
Empirical formula	C ₁₂ H ₁₃ N ₅ O
Formula weight	243.27
Temperature/K	100.00(10)
Crystal system	monoclinic
Space group	P2 ₁ /c
a/Å	14.4997(5)
b/Å	12.4867(4)
c/Å	6.7627(2)
α/°	90
β/°	103.384(4)
γ/°	90
Volume/Å ³	1191.16(7)
Z	4
ρ _{calc} /cm ³	1.357
μ/mm ⁻¹	0.758
F(000)	512.0
Crystal size/mm ³	0.424 × 0.1 × 0.081
Radiation	Cu Kα (λ = 1.54184)
2Θ range for data collection/°	6.266 to 152.694
Index ranges	-18 ≤ h ≤ 17, -15 ≤ k ≤ 15, -5 ≤ l ≤ 8
Reflections collected	12747
Independent reflections	2479 [R _{int} = 0.0256, R _{sigma} = 0.0135]
Data/restraints/parameters	2479/0/166
Goodness-of-fit on F ²	1.079
Final R indexes [I >= 2σ (I)]	R ₁ = 0.0391, wR ₂ = 0.1102
Final R indexes [all data]	R ₁ = 0.0405, wR ₂ = 0.1129
Largest diff. peak/hole / e Å ⁻³	0.14/-0.30

Table S10 Fractional Atomic Coordinates ($\times 10^4$) and Equivalent Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for 3ba. U_{eq} is defined as 1/3 of the trace of the orthogonalised U_{IJ} tensor.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	$U(\text{eq})$
O1	8533.2(7)	8287.5(8)	6133.4(15)	21.8(2)
N3	5316.1(8)	6834.5(10)	2803.2(18)	19.0(3)
N2	4965.6(9)	8558.7(10)	2377.6(18)	21.7(3)
N4	8757.6(9)	6544.1(10)	6558.1(19)	23.3(3)
N5	9609.5(9)	7075.5(10)	7477.4(19)	23.4(3)
N1	4164.5(9)	7958.4(10)	1553.6(19)	21.8(3)
C6	5652.2(10)	7873.4(11)	3124(2)	19.0(3)
C1	4379.1(10)	6936.9(11)	1824(2)	20.0(3)
C3	6797.6(10)	6098.6(11)	4336(2)	20.8(3)
C5	6613.2(10)	8039.9(11)	4123(2)	20.2(3)
C8	9443.6(10)	8084.6(12)	7174(2)	21.4(3)
C2	5879.5(10)	5944.8(11)	3400(2)	20.6(3)
C7	8160.2(10)	7284.8(11)	5801(2)	20.2(3)
C11	3742.8(10)	5994.4(11)	1246(2)	23.9(3)
C4	7173.2(10)	7164.3(11)	4728(2)	20.1(3)
C9	10095.4(11)	9009.5(13)	7719(3)	29.9(4)
C12	2753.5(11)	6294.7(13)	65(3)	32.2(4)
C10	11038.8(11)	8698.3(13)	9100(3)	32.0(4)

Table S11 Anisotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for 3ba. The Anisotropic displacement factor exponent takes the form: $-2\pi^2[h^2a^{*2}U_{11}+2hka^*b^*U_{12}+\dots]$.

Atom	U_{11}	U_{22}	U_{33}	U_{23}	U_{13}	U_{12}
O1	22.7(5)	18.0(5)	23.3(5)	0.1(4)	2.4(4)	-0.2(4)
N3	21.7(5)	15.9(6)	19.6(5)	0.3(5)	5.1(5)	0.4(4)
N2	24.3(6)	17.5(6)	22.6(6)	0.3(5)	4.1(5)	1.0(4)
N4	22.8(6)	20.9(6)	24.8(6)	-0.7(5)	2.7(5)	0.0(4)
N5	22.2(6)	21.8(6)	24.7(6)	-0.6(5)	2.8(5)	-0.9(5)
N1	23.3(6)	18.8(6)	22.6(6)	0.1(5)	4.1(5)	0.1(4)
C6	24.6(6)	15.0(6)	17.6(6)	-0.7(5)	5.4(5)	-0.2(5)
C1	21.7(6)	19.4(7)	19.2(6)	1.2(5)	5.4(5)	0.4(5)
C3	24.3(7)	16.2(6)	21.9(6)	1.6(5)	5.6(5)	2.5(5)
C5	24.6(7)	16.1(6)	20.3(6)	-1.4(5)	5.9(5)	-1.1(5)
C8	21.5(6)	22.6(7)	19.2(6)	-0.8(5)	3.0(5)	-0.4(5)
C2	25.2(6)	14.2(6)	22.3(6)	-0.3(5)	5.3(5)	1.4(5)
C7	24.0(7)	18.1(7)	18.6(6)	-1.8(5)	5.2(5)	-0.6(5)
C11	23.7(7)	19.1(6)	28.1(7)	0.8(6)	4.5(6)	-2.2(5)
C4	24.0(7)	19.8(6)	16.8(6)	-0.8(5)	5.1(5)	0.4(5)
C9	28.8(7)	20.7(7)	36.4(8)	1.9(6)	0.0(7)	-3.2(6)
C12	24.2(7)	28.7(8)	41.1(9)	2.3(7)	2.2(6)	-1.8(6)
C10	23.3(7)	28.4(8)	41.1(9)	2.0(7)	1.3(7)	-5.2(6)

Table S12 Bond Lengths for 3ba.

Atom	Atom	Length/Å	Atom	Atom	Length/Å
O1	C8	1.3681(17)	N1	C1	1.3156(18)
O1	C7	1.3618(16)	C6	C5	1.416(2)
N3	C6	1.3852(18)	C1	C11	1.4906(19)
N3	C1	1.3727(18)	C3	C2	1.349(2)
N3	C2	1.3827(18)	C3	C4	1.4391(19)
N2	N1	1.3863(17)	C5	C4	1.3670(19)
N2	C6	1.3211(19)	C8	C9	1.484(2)
N4	N5	1.4132(17)	C7	C4	1.4548(19)
N4	C7	1.2894(19)	C11	C12	1.518(2)
N5	C8	1.2903(19)	C9	C10	1.518(2)

Table S13 Bond Angles for 3ba.

Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
C7	O1	C8	102.34(11)	C2	C3	C4	120.56(13)
C1	N3	C6	105.15(11)	C4	C5	C6	118.44(13)
C1	N3	C2	131.88(12)	O1	C8	C9	117.84(13)
C2	N3	C6	122.96(12)	N5	C8	O1	112.64(13)
C6	N2	N1	106.89(12)	N5	C8	C9	129.49(15)
C7	N4	N5	106.04(12)	C3	C2	N3	118.35(13)
C8	N5	N4	106.06(13)	O1	C7	C4	118.99(12)
C1	N1	N2	108.61(12)	N4	C7	O1	112.91(12)
N3	C6	C5	118.94(12)	N4	C7	C4	128.08(13)
N2	C6	N3	109.88(12)	C1	C11	C12	113.27(12)
N2	C6	C5	131.18(13)	C3	C4	C7	118.31(13)
N3	C1	C11	122.44(12)	C5	C4	C3	120.74(13)
N1	C1	N3	109.46(12)	C5	C4	C7	120.95(13)
N1	C1	C11	128.09(13)	C8	C9	C10	112.66(13)

Table S14 Torsion Angles for 3ba.

A	B	C	D	Angle/°	A	B	C	D	Angle/°
O1	C8	C9	C10	172.46(13)	C6	N3	C2	C3	-0.1(2)
O1	C7	C4	C3	178.18(12)	C6	N2	N1	C1	0.28(15)
O1	C7	C4	C5	-2.4(2)	C6	C5	C4	C3	0.8(2)
N3	C6	C5	C4	0.1(2)	C6	C5	C4	C7	-178.61(12)
N3	C1	C11	C12	-176.51(13)	C1	N3	C6	N2	-0.36(16)
N2	N1	C1	N3	-0.51(16)	C1	N3	C6	C5	179.96(12)
N2	N1	C1	C11	178.17(13)	C1	N3	C2	C3	179.31(14)
N2	C6	C5	C4	-179.48(14)	C8	O1	C7	N4	0.69(15)
N4	N5	C8	O1	0.83(17)	C8	O1	C7	C4	179.25(12)
N4	N5	C8	C9	-176.79(15)	C2	N3	C6	N2	179.16(12)
N4	C7	C4	C3	-3.5(2)	C2	N3	C6	C5	-0.5(2)
N4	C7	C4	C5	175.94(14)	C2	N3	C1	N1	-178.92(13)
N5	N4	C7	O1	-0.23(16)	C2	N3	C1	C11	2.3(2)
N5	N4	C7	C4	-178.63(13)	C2	C3	C4	C5	-1.4(2)
N5	C8	C9	C10	-10.0(2)	C2	C3	C4	C7	178.01(13)
N1	N2	C6	N3	0.06(15)	C7	O1	C8	N5	-0.94(16)
N1	N2	C6	C5	179.69(14)	C7	O1	C8	C9	176.98(13)
N1	C1	C11	C12	5.0(2)	C7	N4	N5	C8	-0.36(16)
C6	N3	C1	N1	0.53(16)	C4	C3	C2	N3	1.0(2)
C6	N3	C1	C11	-178.24(12)					

Table S15 Hydrogen Atom Coordinates ($\text{\AA} \times 10^4$) and Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for 3ba.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	U(eq)
H3	7191.18	5510.91	4730.31	25
H5	6857.58	8728.72	4360.51	24
H2	5631.55	5257.42	3163.07	25
H11A	3695.2	5614.66	2469.67	29
H11B	4023.95	5511.14	427.87	29
H9A	9795.15	9547.41	8394.18	36
H9B	10207.67	9325.31	6485.79	36
H12A	2790.15	6623.94	-1197.07	48
H12B	2476.12	6787.93	850	48
H12C	2368.69	5661.94	-203.84	48
H10A	11340.94	8167.93	8438.6	48
H10B	10934.44	8410.94	10347.17	48
H10C	11437.81	9319.18	9389.11	48

Experimental

Single crystals of $\text{C}_{12}\text{H}_{13}\text{N}_5\text{O}$ (**3ba**) were crystallized by slow evaporation of saturated solution in EtOAc. A suitable crystal was selected and mounted by acrylic glue on the glass stick on a **SuperNova, Dual, Cu at home/near, AtlasS2** diffractometer. The crystal was kept at 100.00(10) K during data collection. Using Olex2 [S1], the structure was solved with the SHELXT [S2] structure solution program using Intrinsic Phasing and refined with the SHELXL [S3] refinement package using Least Squares minimization.

Crystal structure determination of 3ba

Crystal Data for $\text{C}_{12}\text{H}_{13}\text{N}_5\text{O}$ ($M = 243.27$ g/mol): monoclinic, space group $\text{P2}_1/\text{c}$ (no. 14), $a = 14.4997(5)$ \AA , $b = 12.4867(4)$ \AA , $c = 6.7627(2)$ \AA , $\beta = 103.384(4)^\circ$, $V = 1191.16(7)$ \AA^3 , $Z = 4$, $T = 100.00(10)$ K, $\mu(\text{Cu K}\alpha) = 0.758$ mm^{-1} , $D_{\text{calc}} = 1.357$ g/cm^3 , 12747 reflections measured ($6.266^\circ \leq 2\theta \leq 152.694^\circ$), 2479 unique ($R_{\text{int}} = 0.0256$, $R_{\text{sigma}} = 0.0135$) which were used in all calculations. The final R_1 was 0.0391 ($I > 2\sigma(I)$) and wR_2 was 0.1129 (all data).

Refinement model description

Number of restraints - 0, number of constraints - unknown.

Details:

1. Twinned data refinement

Scales: 0.7827(13)

0.2173(13)

2. Fixed Uiso

At 1.2 times of:

All C(H) groups, All C(H,H) groups

At 1.5 times of:

All C(H,H,H) groups

3.a Secondary CH₂ refined with riding coordinates:

C11(H11A,H11B), C9(H9A,H9B)

3.b Aromatic/amide H refined with riding coordinates:

C3(H3), C5(H5), C2(H2)

3.c Idealised Me refined as rotating group:

C12(H12A,H12B,H12C), C10(H10A,H10B,H10C)

This report has been created with Olex2, compiled on 2020.11.12 svn.r5f609507 for OlexSys. Please [let us know](#) if there are any errors or if you would like to have additional features.

References

(S1) Dolomanov, O. V.; Bourhis, L. J.; Gildea, R. J.; Howard, J. A. K.; Puschmann, H. OLEX 2: A Complete Structure Solution, Refinement and Analysis Program. *J. Appl. Cryst.* **2009**, 42, 339-341.

(S2) Sheldrick, G. M. SHELXT – Integrated Space-Group and Crystal-Structure Determination. *Acta Cryst.* **2015**, A71, 3-8.

(S3) Sheldrick, G. M. Crystal Structure Refinement with SHELXL. *Acta Cryst.* **2015**, C71, 3-8