

Article

Synthesis of Cyano-Benzylidene Xanthene Synthons Using a Diprotic Brønsted Acid Catalyst, and Their Application as Efficient Inhibitors of Aluminum Corrosion in Alkaline Solutions

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S1. General procedure of the synthesis of the studied compounds

S1.1. Synthesis of 3-(3,3,6,6-tetramethyl-1,8-dioxo-2,3,4,5,6,7,8,9-octahydro-1H-xanthen-9-yl)benzaldehyde (3)

The synthesis of compound **3** was performed by the reaction of 5,5-dimethylcyclohexan-1,3-dione (**1**; 20 mmol, 2.80 g) with isophthaldehyde (**2**; 10 mmol, 1.34 g) and oxalic acid solution as a catalyst (10 mmol, 0.99 g) at 80 °C in a water bath under constant stirring for 40 min. The reaction mixture was then poured onto ice-cold water. The solid product was collected, filtered off, dried and recrystallized from ethanol as buff crystals in 67% yield; M.p. 156-158 °C; ¹H-NMR (DMSO-*d*₆) (δ/ppm): 2.36 (s, 12H, 3Me), 2.49 (s, 4H, Xanthene_{(C4,5)-H}), 3.35 (s, 4H, Xanthene_{(C2,7)-H}), 7.72-7.91(m, 4H, Ar-H), 9.88 (s, 1H, CHO). ¹³C-NMR (300 MHz): 27.5 (4 Me), 32.2 (C-3,6-xanth), 38.7, 51.6 (4CH₂), 39.1 (C-9-xanth), 113.8, 126.8, 129.1, 130.1, 133.4, 139.3, 142.5, 155.1 (5C=C), 191.1 (CHO), 198.8 (2C=O); MS (m/z, %): 379.19 (M+1, 26.4). Anal. Calcd. for C₂₄H₂₆O₄ (378.46) C, 76.17; H, 6.92; O, 16.91%. Found: C, 76.01; H, 6.75; O, 16.81%.

S1.2. General procedure for the synthesis of 2-(4-(12H-dibenzo[c]xanthen-7-yl)benzylidene) derivatives (compounds 4-6).

The xanthene derivative (**3**; 10 mmol, 3.86 g) was allowed to react with a molar ratio of the active nitrile compounds, malononitrile (0.66 g), cyanoacetic acid (0.85 g) or ethylcyanoacetate (1.13 g). The reaction mixture was kept under magnetic stirring at ambient temperature for 2 h in 30 mL ethanol and piperidine (2 mL) as basic catalyst. The mixture was poured onto ice-cold water and then acidified with dilute hydrochloric acid (2 M). The product was filtered off, dried and recrystallized from ethanol, yielding the cyanobenzylidene xanthenes (compounds **4-6**), respectively.

2-(3-(3,3,6,6-Tetramethyl-1,8-dioxo-2,3,4,5,6,7,8,9-octahydro-1H-xanthen-9-yl)benzylidene)malononitrile (4).

Faint yellow crystals; Yield: 73% (3.10 g); M.p. >250 °C. IR: ν_{max}=2871(=CH Ar), 2190, (C≡N_{str.}), 1666 (C=O_{str.} aldehydic), 1476 (CH₂), 1365 (CH₃) and 1213 (C-O xanthene ring) cm⁻¹. ¹H-NMR (300 MHz, DMSO-*d*₆) (δ/ppm): 2.38 (s, 12H, 4 Me); 2.51 (s, 4H, Xanthene_{(C4,5)-H}), 3.37 (s, 4H, Xanthene_{(C2,7)-H}); 7.35-7.84 (m, 4H, Ar-H); 7.93 (s, 1H, olefinic proton). ¹³C-NMR (75 MHz) δ: 199.12, 159.11, 147.12, 136.13, 128.10, 124.95, 108.31. MS (m/z, %): 427.20 (M+1, 29.6). Anal. calc. for C₂₇H₂₅N₂O₃ (426.51) C, 76.03; H, 6.14; N, 6.57 (Found: C, 76.45; H, 6.56; N, 6.87).

2-Cyano-3-(3-(3,3,6,6-tetramethyl-1,8-dioxo-2,3,4,5,6,7,8,9-octahydro-1H-xanthen-9-yl)phenyl)acrylic acid

(5). Faint yellow crystals; Yield: 68% (3.02 g); M.p. 120-121 °C. IR: $\nu_{\text{max}}=3392$ (OH_{str.}), 2226, (C≡N_{str.}), 1666 (C=O_{str.}carboxylic), 1470 (CH₂), 1365 (CH₃) and 1213 (C-O xanthene ring) cm⁻¹. ¹H-NMR (300 MHz, DMSO-*d*₆) (δ /ppm): 2.38 (s, 12H, 4 CH₃); 2.51 (s, 4H, Xanthene_(C4,5)-H), 3.35 (s, 4H, Xanthene_(C2,7)-H); 7.74-7.95 (m, 4H, Ar-H); 8.53 (s, 1H, olefinic proton); 12.45 (s, 1H, COOH). ¹³C-NMR (75 MHz) δ : 199.12, 159.11, 147.12, 136.13, 128.10, 124.95, 108.31. MS (*m/z*, %): 445.19 (M⁺, 4.31). Anal. calc. for C₂₇H₂₇NO₅ (445.51) C, 72.79; H, 6.11; N, 3.14 (Found: C, 72.65; H, 6.33; N, 3.46).

Ethyl-2-cyano-3-(3-(3,3,6,6-tetramethyl-1,8-dioxo-2,3,4,5,6,7,8,9-octahydro-1H-xanthen-9-yl)phenyl)acrylate

(6). Yellow crystals; Yield: 81% (4.73 g); M.p. 134-136 °C; IR: $\nu_{\text{max}}=2957$ (=CH Ar), 2232, (C≡N_{str.}), 1666 (C=O_{str.} aldehydic), 1447 (CH₂), 1365 (CH₃) and 1213 (C-O xanthene ring) cm⁻¹. ¹H-NMR (300 MHz, DMSO-*d*₆) (δ /ppm): 1.28 (s, 12H, 4 CH₃); 1.31 (t, 3H, CH₃); 2.51 (s, 4H, Xanthene_(C4,5)-H), 3.44 (s, 4H, Xanthene_(C2,7)-H); 4.25 (q, 2H, CH₂); 7.36-7.95 (m, 4H, Ar-H); 8.51 (s, 1H, olifinic proton). ¹³C-NMR (75 MHz) δ : 199.12, 159.11, 147.12, 136.13, 128.10, 124.95, 108.31. MS (*m/z*, %): 474.22 (M+1, 31.9). Anal. calc. for C₂₉H₃₁NO₅ (473.56) C, 73.55; H, 6.60; N, 2.96 (Found: C, 73.45; H, 6.54; N, 3.26).

S2. ^1H -NMR and FTIR spectra of the studied xanthene derivatives

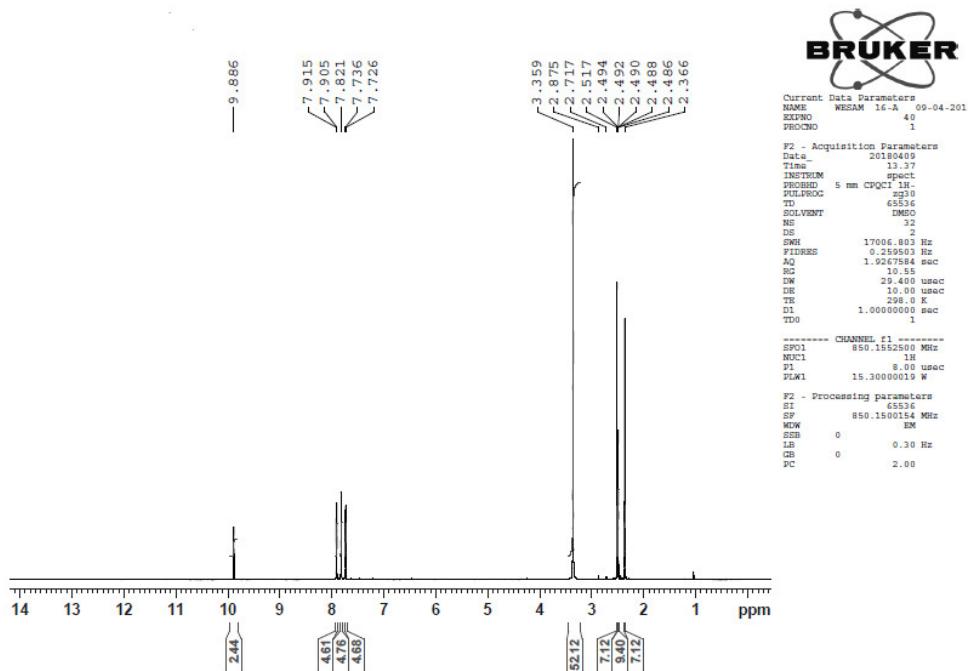


Figure S1. ^1H -NMR spectrum of benzaldehyde xanthene derivative (3).

The xanthene-benzaldehyde (3) is hence incorporated in a set of reactions with different active nitriles, malononitrile, cyanoacetic acid or ethylcyanoacetate in presence of basic catalyst (piperidine), to synthesize derivatives (4-6), as shown in **Scheme 2**.

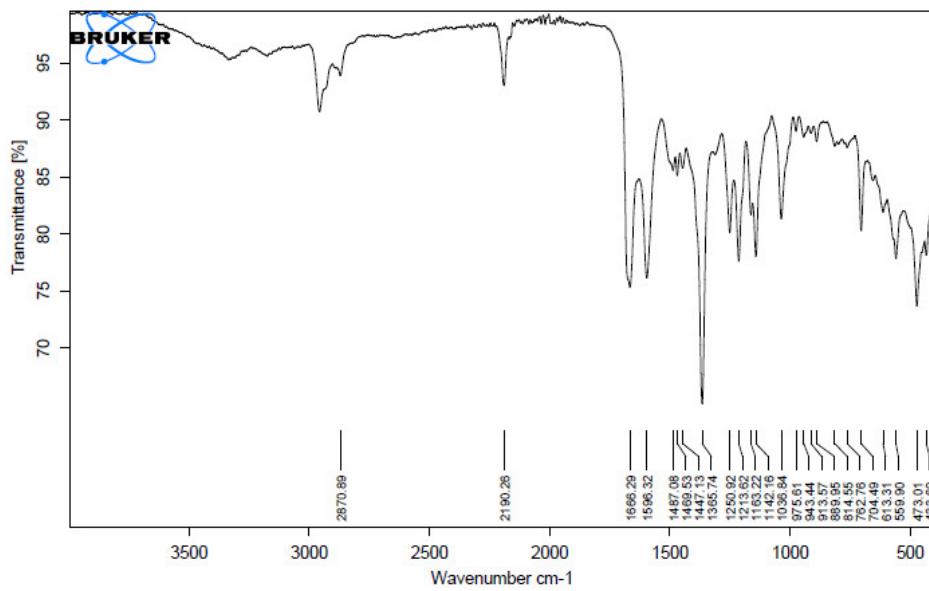


Figure S2. FTIR spectrum of malononitrile xanthene derivative (4).

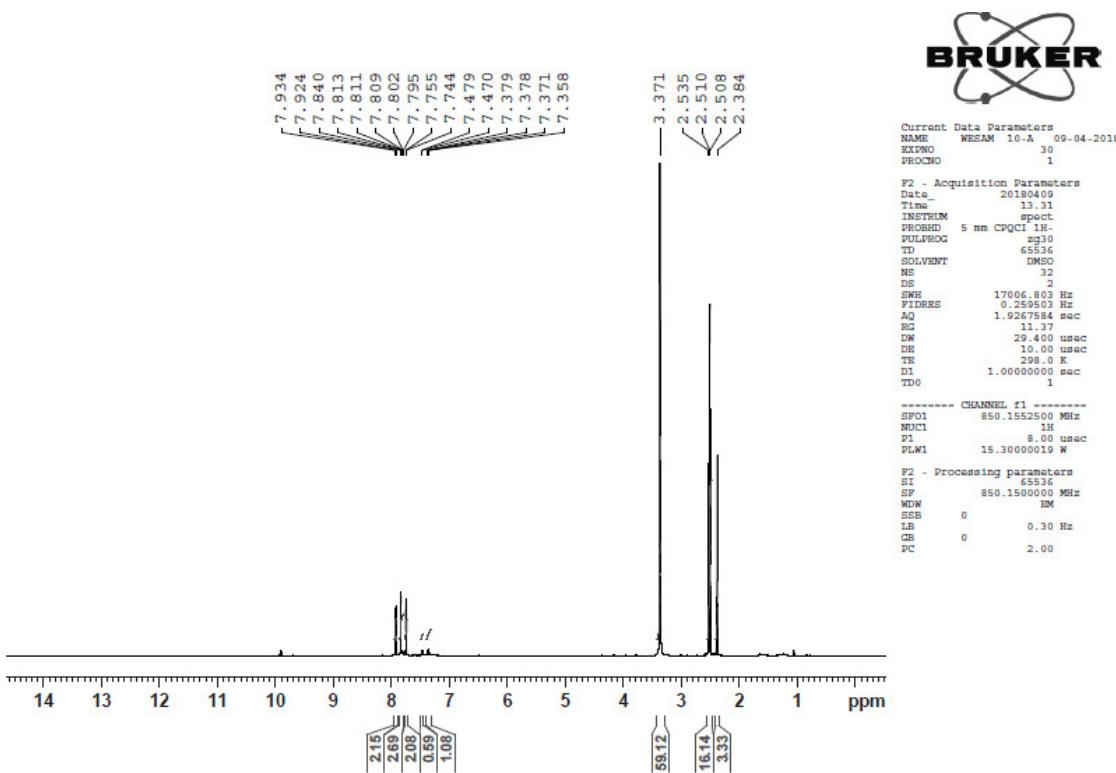


Figure S3. ¹H-NMR spectrum of malononitrile xanthene derivative (**4**).

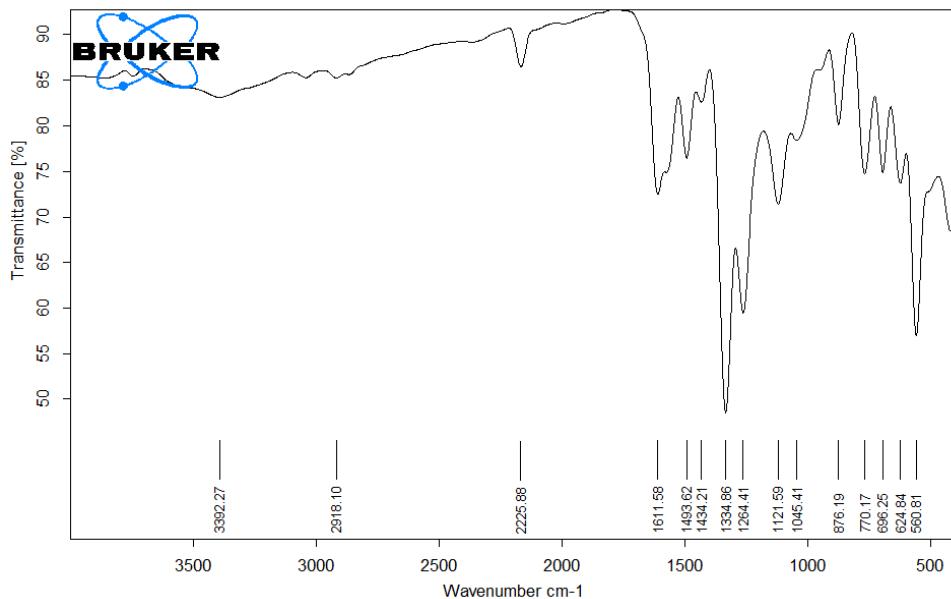


Figure S4. FTIR spectrum of acrylic acid xanthene derivative (**5**).

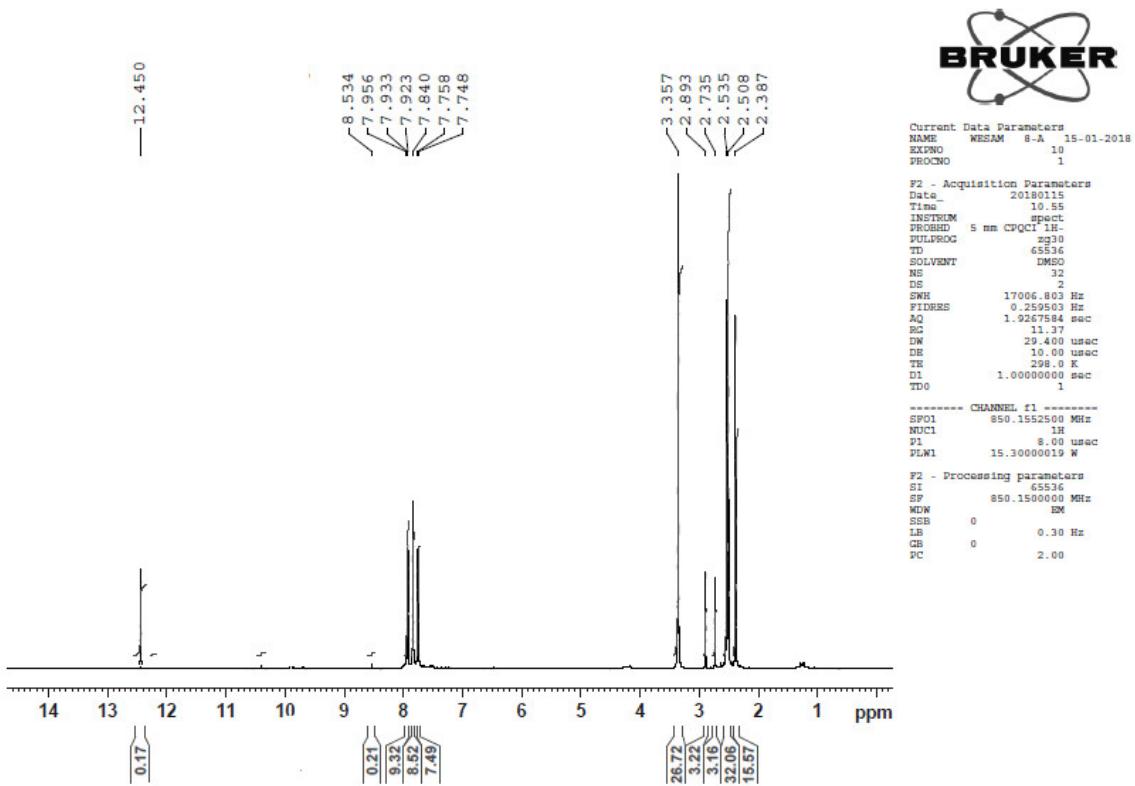


Figure S5. ¹H-NMR plot of acrylic acid xanthene derivative (**5**).

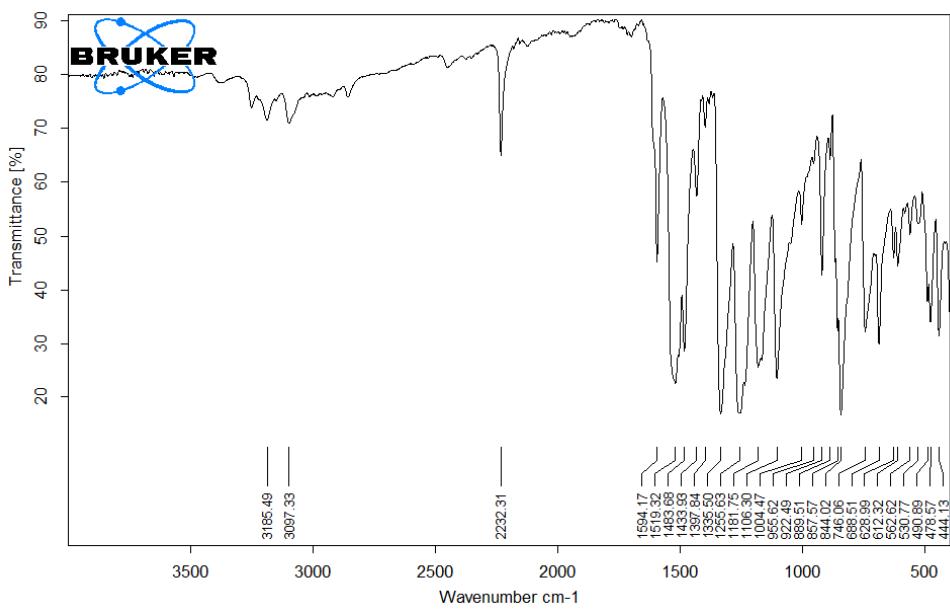


Figure S6. FTIR spectrum of acrylate xanthene derivative (**6**).

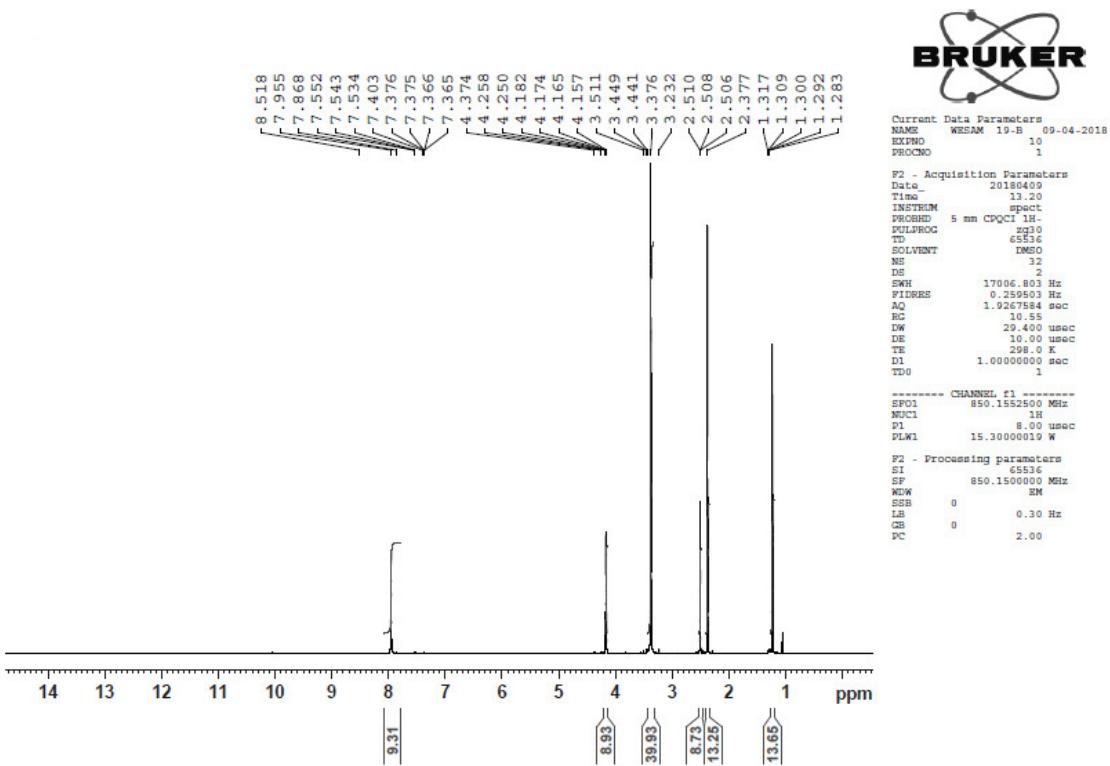


Figure S7. ^1H -NMR spectrum of acrylate xanthene derivative (**6**).

S3. Electrochemical kinetic parameters based on polarization studies

Table S1 – Various electrochemical kinetic parameters recorded for compound 3, derived from polarization measurements employing the Tafel extrapolation and LPR methods.

[Inhibitor] / mM	Tafel extrapolation method					LPR method				
	E_{corr} / V vs. Ag/AgCl	j_{corr} / A cm ⁻²	β_c / mV dec ⁻¹	β_a / mV dec ⁻¹	v / mm yr ⁻¹	I(%)	j_{corr} / A cm ⁻²	R_p / Ω cm ²	v / mm yr ⁻¹	I(%)
0.0	-1.520	1.26×10^{-3}	-0.249	0.241	13.73	----	1.17×10^{-3}	45.45	12.75	----
0.1	-1.482	7.7×10^{-4}	-0.245	0.290	8.39	38.9	7.04×10^{-4}	81.91	7.67	39.8
0.2	-1.469	6.4×10^{-4}	-0.259	0.219	6.98	49.2	5.66×10^{-4}	91.03	6.17	51.6
0.5	-1.452	2.78×10^{-4}	-0.256	0.260	3.03	77.9	2.38×10^{-4}	235.34	2.59	79.7
1.0	-1.427	1.46×10^{-4}	-0.273	0.250	1.59	88.4	1.19×10^{-4}	476.17	1.30	89.8
2.0	-1.399	1.03×10^{-4}	-0.261	0.196	1.12	91.8	8.78×10^{-5}	553.6	0.96	92.5

Table S2 – Various electrochemical kinetic parameters recorded for compound 5, derived from polarization measurements employing the Tafel extrapolation and LPR methods.

[Inhibitor] / mM	Tafel extrapolation method					LPR method				
	E_{corr} / V vs. Ag/AgCl	j_{corr} / A cm ⁻²	β_c / mV dec ⁻¹	β_a / mV dec ⁻¹	v / mm yr ⁻¹	I(%)	j_{corr} / A cm ⁻²	R_p / Ω cm ²	v / mm yr ⁻¹	I(%)
0.0	-1.520	1.26×10^{-3}	-0.249	0.241	13.72	----	1.17×10^{-3}	45.45	12.74	----
0.1	-1.489	6.72×10^{-4}	-0.243	0.288	7.32	46.7	6.44×10^{-4}	88.86	7.015	44.9
0.2	-1.464	6.12×10^{-4}	-0.257	0.221	6.67	51.4	5.53×10^{-4}	93.3	6.02	52.7
0.5	-1.446	2.2×10^{-4}	-0.255	0.262	2.39	82.6	2.23×10^{-4}	251.6	2.43	81.0
1.0	-1.422	9.07×10^{-5}	-0.270	0.255	0.99	92.8	9.83×10^{-5}	579.3	1.07	91.6
2.0	-1.383	6.43×10^{-5}	-0.263	0.199	0.70	94.9	7.72×10^{-5}	637.2	0.84	93.4

Table S3 – Various electrochemical kinetic parameters recorded for compound 6, derived from polarization measurements employing the Tafel extrapolation and LPR methods.

[Inhibitor] / mM	Tafel extrapolation method					LPR method				
	E_{corr} / V vs. Ag/AgCl	j_{corr} / A cm ⁻²	β_c / mV dec ⁻¹	β_a / mV dec ⁻¹	v / mm yr ⁻¹	I(%)	j_{corr} / A cm ⁻²	R_p / Ω cm ²	v / mm yr ⁻¹	I(%)
0.0	-1.520	1.26×10^{-3}	-0.249	0.241	13.73	----	1.17×10^{-3}	45.45	12.75	----

0.0	-1.520	1.26×10^{-3}	-0.249	0.241	13.72	----	1.17×10^{-3}	45.45	12.74	----
0.1	-1.499	6.44×10^{-4}	-0.245	0.290	7.02	48.9	5.9×10^{-4}	97.74	6.43	49.6
0.2	-1.482	5.57×10^{-4}	-0.255	0.218	6.07	55.8	5.00×10^{-4}	102.06	5.45	57.2
0.5	-1.463	1.86×10^{-4}	-0.256	0.260	2.03	85.2	1.97×10^{-4}	284.3	2.15	83.2
1.0	-1.449	7.18×10^{-5}	-0.273	0.257	0.78	94.3	7.61×10^{-5}	755.3	0.83	93.5
2.0	-1.435	4.03×10^{-5}	-0.265	0.202	0.44	96.8	4.80×10^{-5}	1036.9	0.52	95.9

Table S4 – Various electrochemical kinetic parameters recorded for compound **4 (the best one)**, derived from polarization measurements employing the Tafel extrapolation and LPR methods.

[Inhibitor] / mM	Tafel extrapolation method					LPR method				
	E_{corr} / V vs. Ag/AgCl	j_{corr} / $A\ cm^{-2}$	β_c / $mV\ dec^{-1}$	β_a / $mV\ dec^{-1}$	v / $mm\ yr^{-1}$	$I(\%)$	j_{corr} / $A\ cm^{-2}$	R_p / $\Omega\ cm^2$	v / $mm\ yr^{-1}$	$I(\%)$
0.0	-1.520	1.26×10^{-3}	-0.249	0.241	13.72	----	1.17×10^{-3}	45.45	12.74	----
0.1	-1.531	5.97×10^{-4}	-0.247	0.289	6.50	52.6	5.62×10^{-4}	102.90	6.12	52.0
0.2	-1.581	5.20×10^{-4}	-0.258	0.220	5.66	58.7	4.75×10^{-4}	108.55	5.17	59.4
0.5	-1.590	1.32×10^{-4}	-0.255	0.258	1.44	89.5	1.16×10^{-4}	480.05	1.26	90.1
1.0	-1.586	3.02×10^{-5}	-0.275	0.256	0.33	97.6	3.74×10^{-5}	1539.26	0.41	96.8
2.0	-1.567	1.39×10^{-5}	-0.268	0.199	0.15	98.9	1.87×10^{-5}	2651.77	0.20	98.4

S4. Cartesian coordinates for the optimized geometry of the studied xanthene:

(3)

C 0.62052000 -3.76771300 -0.21785500
C 2.14134700 -3.58836400 -0.00710400
C 2.39235900 -2.34100100 0.89289400
C 1.55127100 -1.19113900 0.46303700
C 0.41405300 -1.23807900 -0.25667300
C -0.06722700 -2.53859700 -0.78507900
O 2.11984900 0.00090800 0.91975900
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C 0.61727200 3.76823700 -0.21785100
C -0.06941000 2.53853000 -0.78508200
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C 2.83943700 -3.39720100 -1.36135200
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C 2.83651000 3.39964400 -1.36134600
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O -5.54622600 -0.00235600 -1.51025800
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H -2.98556500 -0.00126800 -1.47071300
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H -2.74525200 -0.00119800 3.50914600
H -0.65277400 -0.00030200 2.17122000
H -6.30188100 -0.00265400 0.36062500

(4)

C 0.38550500 3.55345200 -0.25201800
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C -0.64380200 1.23099800 -0.29933600
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O -2.64688400 0.91185800 1.01064100
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C -3.25551700 -3.04416900 -0.30379300
C -2.09453600 -2.27466700 -0.90679300
C -5.52901500 -3.00728100 0.77235900
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C 0.54338900 -1.45977900 1.27161000
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(5)

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C -5.74111400 -3.16108600 0.71499600
C -5.27867900 -1.76578800 -1.32332800
C 0.36965200 -0.78840900 0.09880800
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(6)

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