

Experimental and Theoretical Studies of the Optical Properties of the Schiff Bases and Their Materials Obtained from o-Phenylenediamine

Magdalena Barwiolek ^{1,*}, Dominika Jankowska ^{1,*}, Anna Kaczmarek-Kędziera ¹, Sławomir Wojtulewski ², Łukasz Skowroński ³, Tomasz Rerek ³, Paweł Popielarski ⁴ and Tadeusz M. Muziol ¹

- 1 Faculty of Chemistry, Nicolaus Copernicus University in Torun, Gagarina 7, 87-100 Torun, Poland
- 2 Faculty of Chemistry, University of Białystok, Ciolkowskiego 1K, Białystok, 15-245, Poland
- 3 Faculty of Chemical Technology and Engineering, Bydgoszcz University of Science and Technology, Kaliskiego 7, Poland, 85-796 Bydgoszcz, Poland
- 4 Institute of Physics, Kazimierz Wielki University, Chodkiewicza 30, 85-064 Bydgoszcz, Poland

* Correspondence: mbarwiolek@umk.pl; Tel.: +48-56-611-4516;

Table of contents:

Figure S1. ¹H NMR spectrum of **L1** (500 MHz, CDCl₃-d₃).

Figure S2. ¹³C NMR spectrum of **L1** (500 MHz, CDCl₃-d₃).

Figure S3. ¹H NMR spectrum of **L2** (700 MHz, CDCl₃-d₃).

Figure S4. ¹³C NMR spectrum of **L2** (700 MHz, CDCl₃-d₃).

Figure S5. IR spectrum of **L1**, KBr.

Figure S6. IR spectrum of **L2**, KBr.

Figure S7. Structure of **L1** with two crystallographically independent molecules resulting from two halves of the macrocyclic compound found in the asymmetric unit. There is given a numbering scheme and thermal ellipsoids at 30% probability

Figure S8. Hirshfeld surfaces and fingerprints of selected interactions created in the crystal network of **L1**: a. Hirshfeld surface for H...H (red markers correspond to the shortest H...H interactions at 1.15, 1.1), b. fingerprint for H...H (51.1%), c. Hirshfeld surface for C...H, d. fingerprint for C...H (19.3%), e. Hirshfeld surface for H...C (red markers correspond mainly to the spike at 0.95; 1.5), f. fingerprint for H...C (15.2%) for

O34 molecule. In brackets, there is given surface area included as a percentage of the total surface area.

Figure S9. Structure of **L2** with a numbering scheme and thermal ellipsoids at 30% probability.

Figure S10. Hirshfeld surfaces and fingerprints of selected interactions created in the crystal network of **L2**: a. Hirshfeld surface for H...H (red markers correspond to the large spike at 1.1; 1.15), b. fingerprint for H...H (65.4%), c. Hirshfeld surface for C...H, d. fingerprint for C...H (12.4%), e. Hirshfeld surface for H...C, f. fingerprint for H...C (10.5%) for O34 molecule. In brackets, there is given surface area included as a percentage of the total surface area.

Figure S11. Frontier molecular orbitals of **L1** for the most intensive transitions (PBE0/6-311++G(d,p)/PCM(acetonitrile))

Figure S12. Absorption spectrum of **L1**, **L2** calculated with PBE0/6-311++G(d,p) approach in vacuum and in solvents.

Figure S13. Frontier molecular orbitals of **L2** for the most intensive transitions (PBE0/6-311++G(d,p)/PCM(acetonitrile)).

Figure S14. SEM images of **L1**/Si mapping scanning size 100 μm .

Figure S15. a), b) and c) DRIFT spectrum of **L1** and **L2** samples and **L1**/Si and **L2**/Si.

Table S1. Crystal data and structure refinement for **L1** and **L2**.

Table S2. Selected bond length [\AA] and valence angles [$^\circ$] for the **L1**.

Table S3. Relevant photophysical data of studied compounds, (λ_{em} , λ_{ex} nm, $\lambda[\text{nm}]$ (ϵ [$\text{dm}^3 \text{mol}^{-1} \text{cm}^{-1}$]), bp=8

Table S4. Theoretical PBE0/6-311++G(d,p)/PCM(ACN) vertical excitation wavelengths λ [nm] for most intensive transitions together with the corresponding oscillator strengths f and the orbital contributions for investigated species.

Table S5. Relevant fluorescent data of studied compounds in the solid state (λ_{em} , λ_{ex}) [nm].

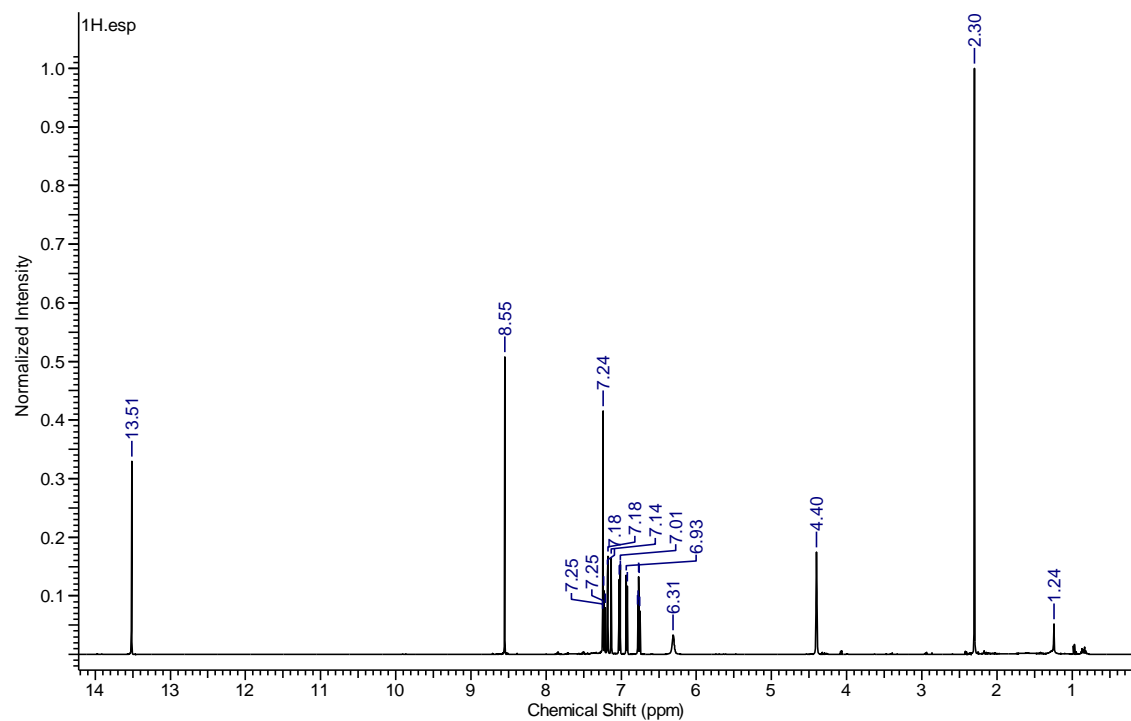


Figure S1. ¹H NMR spectrum of L1 (500 MHz, CDCl₃-d₃).

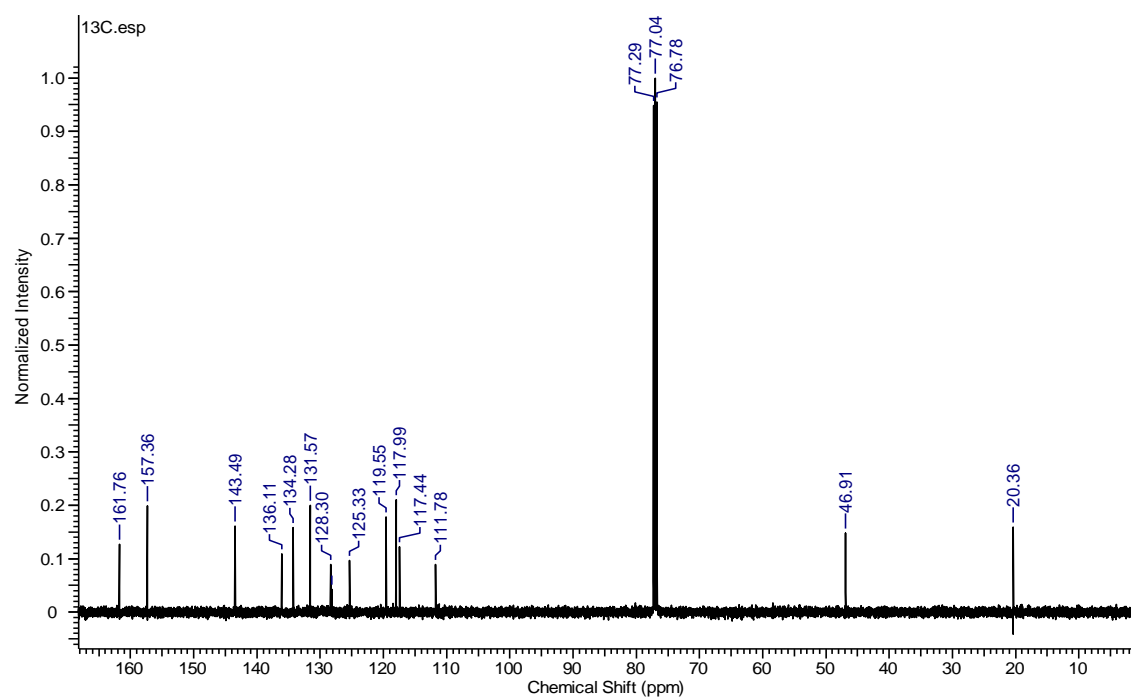


Figure S2. ¹³C NMR spectrum of L1 (500 MHz, CDCl₃-d₃).

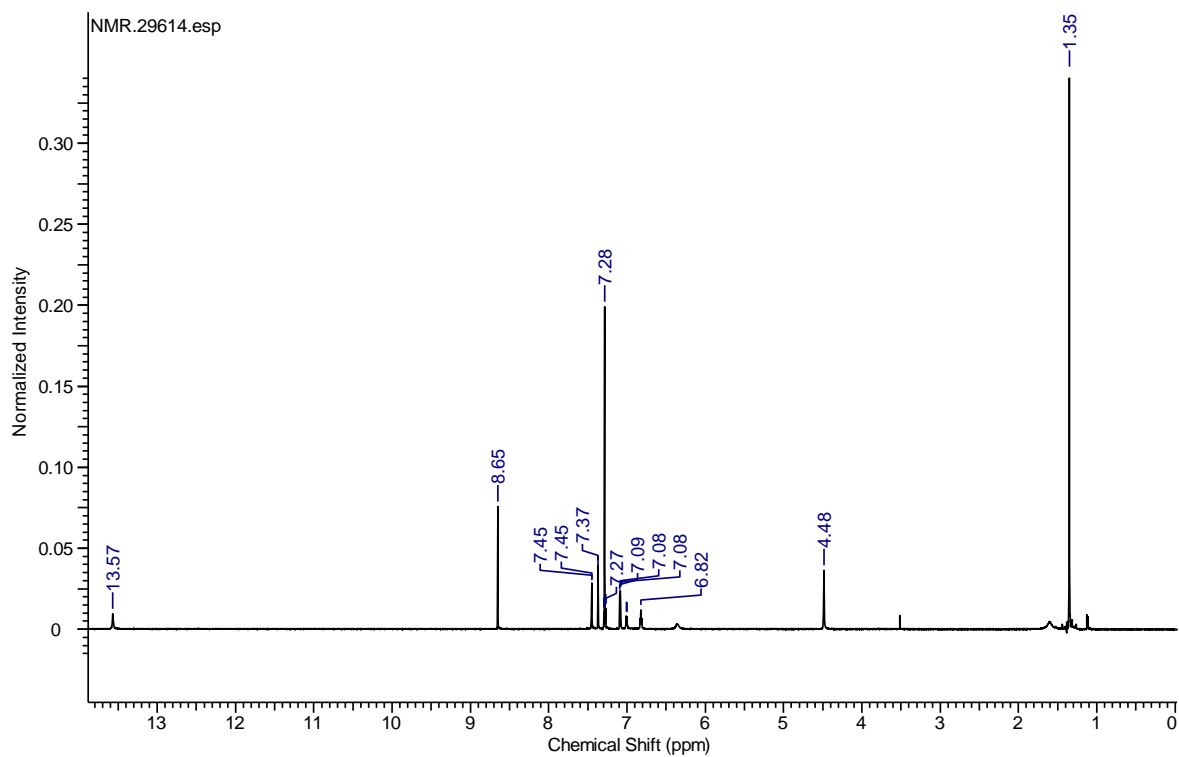


Figure S3. ^1H NMR spectrum of L2 (700 MHz, $\text{CDCl}_3\text{-d}_3$).

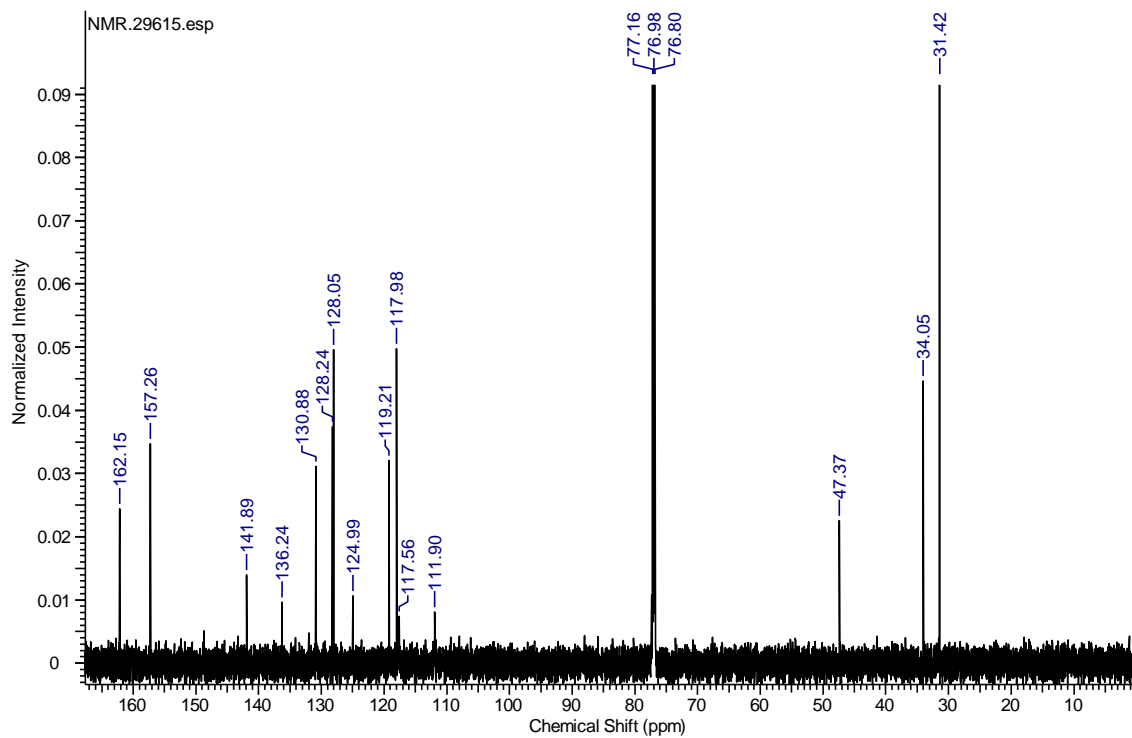


Figure S4. ^{13}C NMR spectrum of L2 (700 MHz, $\text{CDCl}_3\text{-d}_3$).

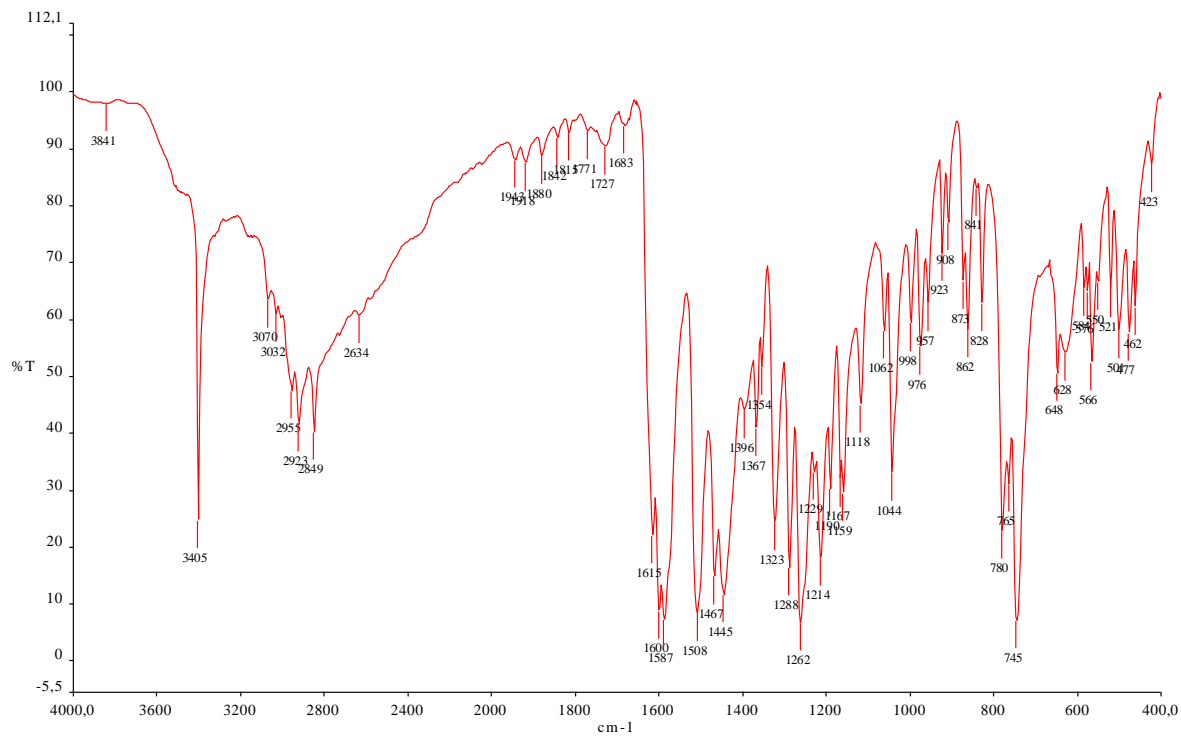
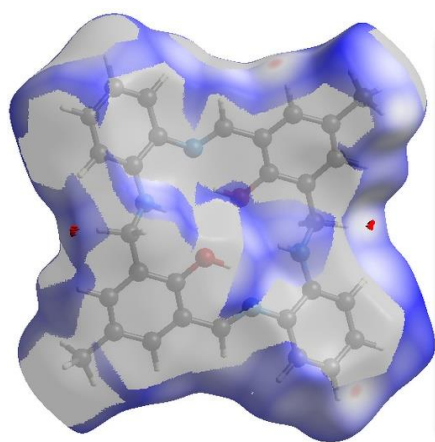
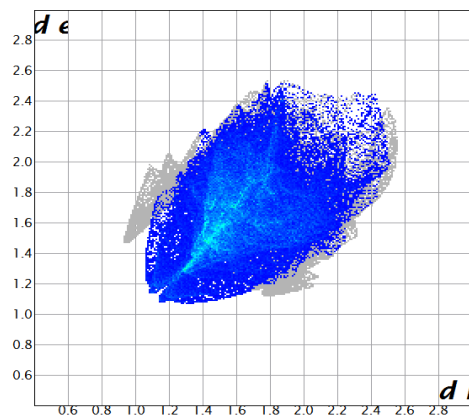


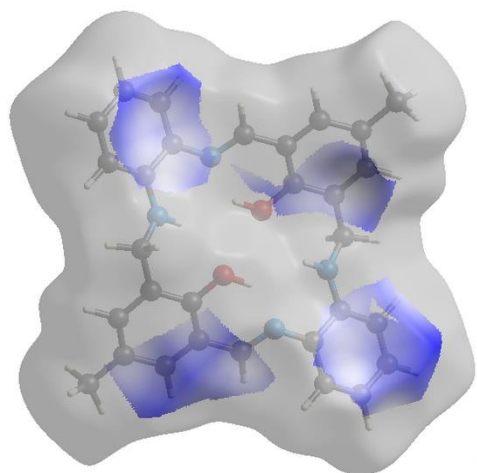
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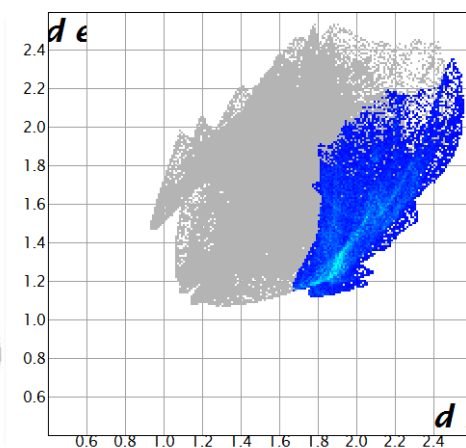
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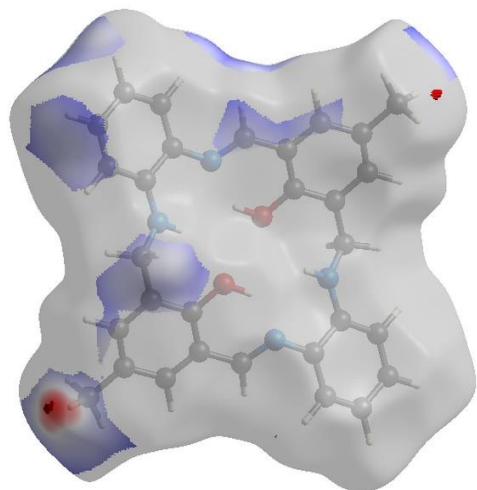
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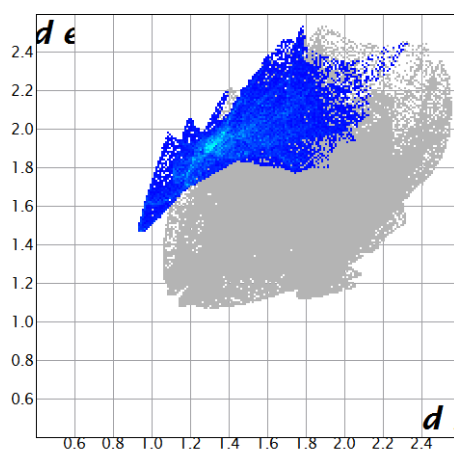
c



d



e



f

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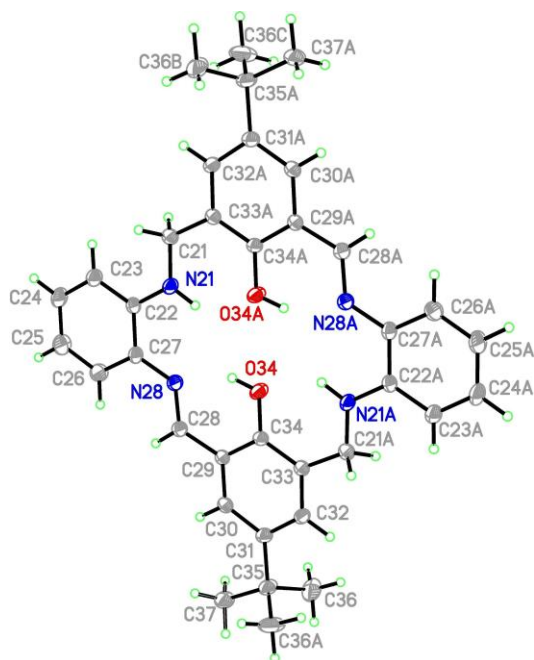
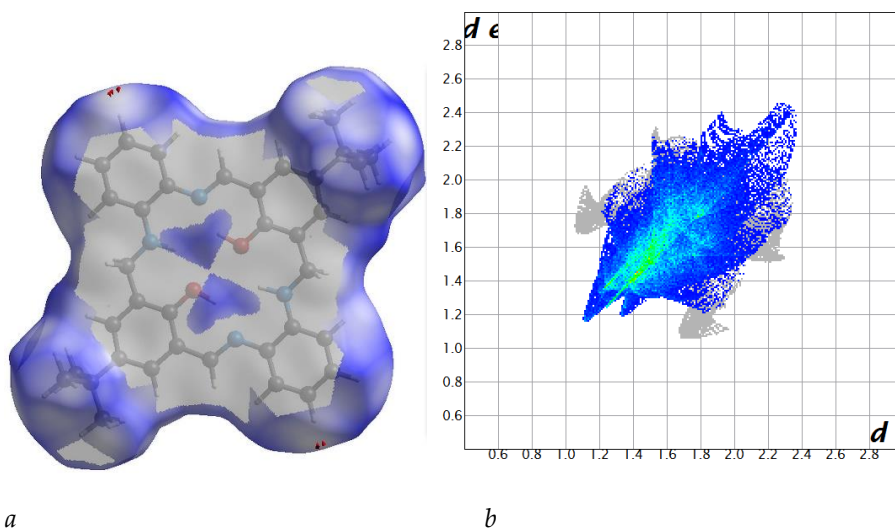


Figure S9. Structure of L2 with a numbering scheme and thermal ellipsoids at 30% probability.



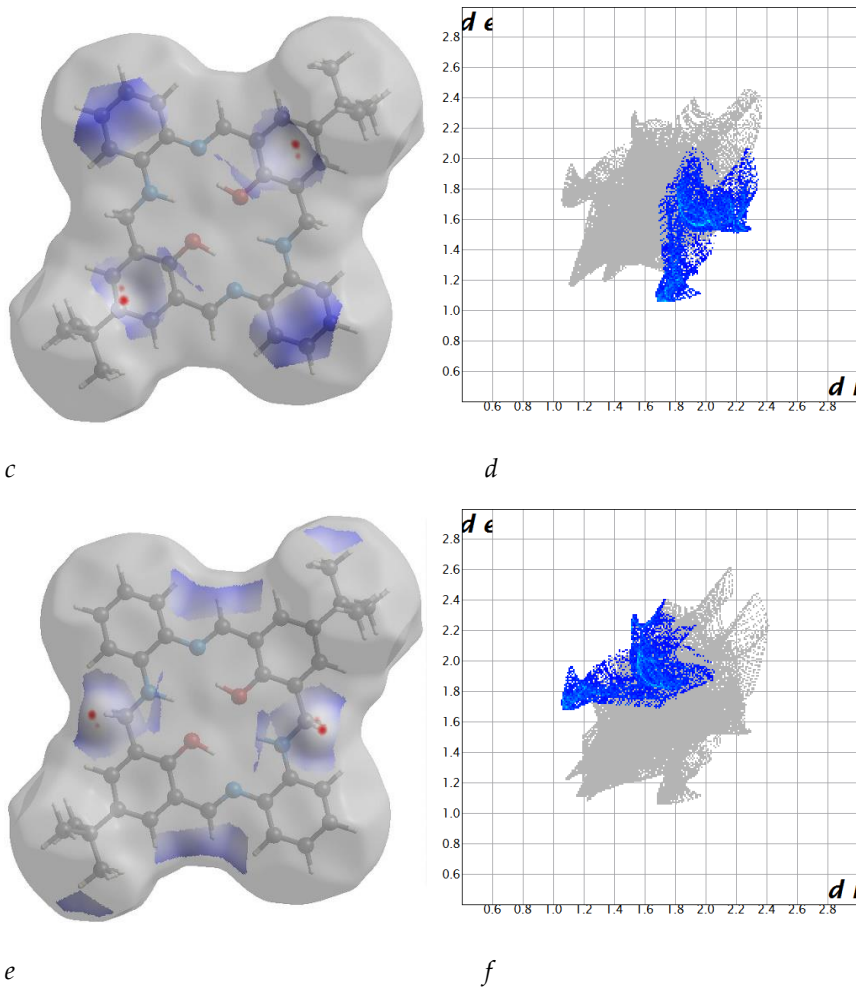


Figure S10. Hirshfeld surfaces and fingerprints of selected interactions created in the crystal network of L2: a. Hirshfeld surface for H...H (red markers correspond to the large spike at 1.1; 1.15), b. fingerprint for H...H (65.4%), c. Hirshfeld surface for C...H, d. fingerprint for C...H (12.4%), e. Hirshfeld surface for H...C, f. fingerprint for H...C (10.5%) for O34 molecule. In brackets, there is given surface area included as a percentage of the total surface area.

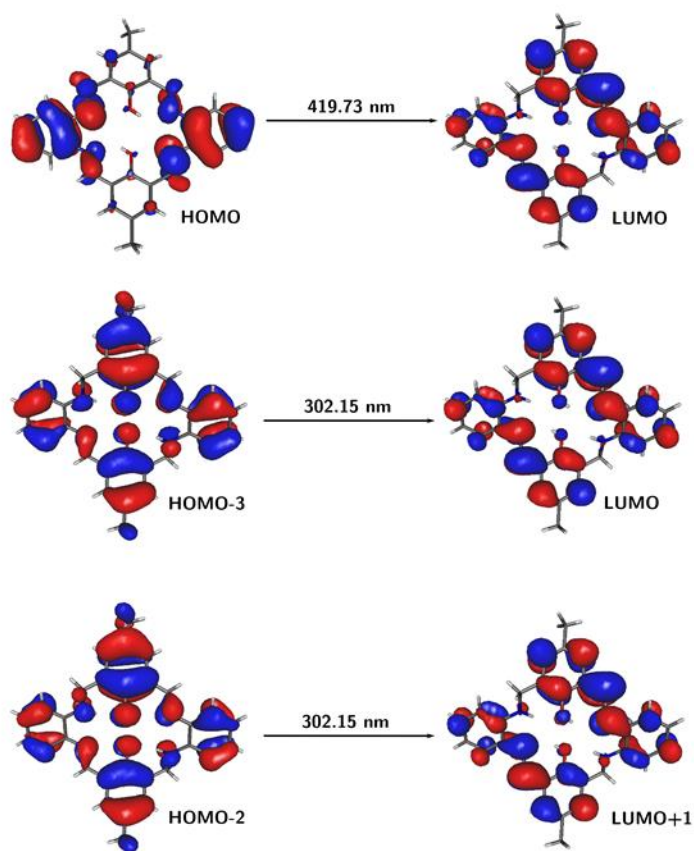


Figure S11. Frontier molecular orbitals of **L1** for the most intensive transitions (PBE0/6-311++G(d,p)/PCM(acetonitrile)).

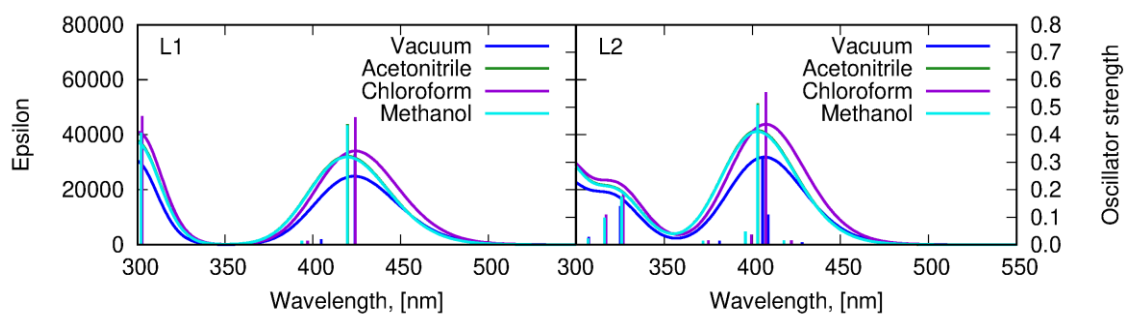


Figure S12. Absorption spectrum of **L1**, **L2** calculated with PBE0/6-311++G(d,p) approach in vacuum and in solvents

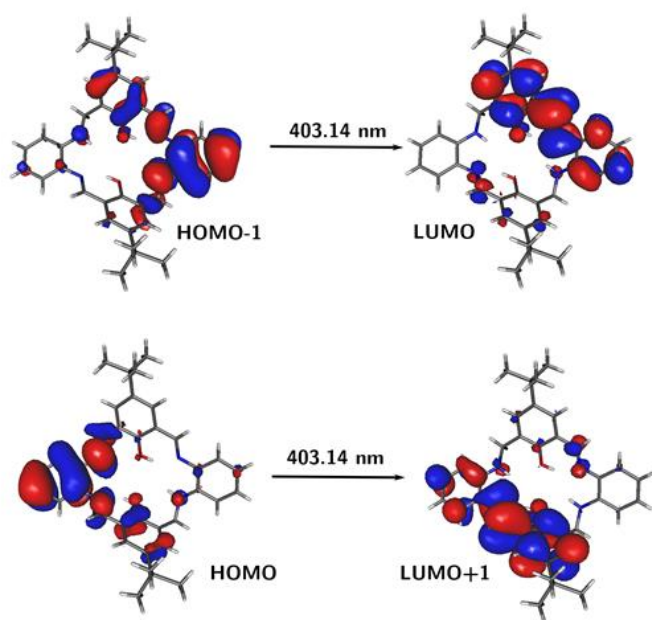


Figure S13. Frontier molecular orbitals of **L2** for the most intensive transitions (PBE0/6-311++G(d,p)/PCM(acetonitrile)).

Table S1. Crystal data and structure refinement for **L1** and **L2**.

Identification code	L1	L2
Empirical formula	C ₃₀ H ₂₈ N ₄ O ₂	C ₃₆ H ₄₀ N ₄ O ₂
Formula weight	476.56	560.72
Temperature [K]	293(2)	100(2)
Wavelength [Å]	0.71073	1.54184
Crystal system, space group	Monoclinic, P2 ₁ /c	Orthorhombic, Pbam

Unit cell dimensions [\AA] and [$^{\circ}$]	$a = 11.911(2)$ $\alpha = 90$ $b = 8.9920(15)$ $\beta = 93.153(19)$ $c = 22.463(5)$ $\gamma = 90$	$a = 28.2607(7)$ $\alpha = 90$ $b = 15.2760(3)$ $\beta = 90$ $c = 6.7643(2)$ $\gamma = 90$
Volume [\AA^3]	2402.3(8)	2920.22(13)
Z, Calculated density [$\text{Mg}\cdot\text{m}^{-3}$]	4, 1.318 Mg/m^3	4, 1.275
Absorption coefficient [mm^{-1}]	0.084	0.625
F(000)	1008	1200
Crystal size [mm]	0.630 x 0.210 x 0.080	0.160 x 0.070 x 0.040
Theta range for data collection [$^{\circ}$]	2.427 to 26.364	3.289 to 74.463
Limiting indices	$-14 \leq h \leq 14$ $-11 \leq k \leq 11$ $-28 \leq l \leq 28$	$-35 \leq h \leq 34$ $-19 \leq k \leq 19$ $-8 \leq l \leq 7$
Reflections collected/unique	5698 / 5698 [R(int) = 0.1042]	16898 / 3256 [R(int) = 0.0416]
Completeness to theta = 29.732° [%]	99.9	100.0
Max. and min. transmission	0.993 and 0.949	0.975 and 0.907
Refinement method	Full-matrix least-squares on F^2	Full-matrix least-squares on F^2
Data/restraints/parameters	5698 / 2 / 335	3256 / 2 / 251
Goodness-of-fit on F^2	0.887	0.987
Final R Indices [$I > 2\sigma(I)$]	$R_1^a = 0.0746$, $wR_2^b = 0.1856$	$R_1^a = 0.0538$, $wR_2^b = 0.1507$

R indices (all data)	$R_1^a = 0.1788$, $wR_2^b = 0.2247$	$R_1^a = 0.0657$, $wR_2^b = 0.1622$
Largest diff. peak and hole [eÅ ⁻³]	0.318 and -0.241	0.366 and -0.361

$$^a R_1 = \Sigma ||F_o| - |F_c|| / \Sigma |F_o|$$

$$^b wR_2 = [\Sigma w(F_o^2 - F_c^2)^2 / \Sigma (w(F_o^2)^2)]^{1/2}$$

Table S2. Selected bond length [Å] and valence angles [°] for the **L1**.

	L1		
C1-N1	1.282(7)	N8-C8	1.455(7)
N1-C2	1.436(8)	C7-N8	1.362(8)
C1-C13 ⁱ	1.460(9)	C8-C9	1.501(9)
C21-N21	1.295(7)	N28-C28	1.442(7)
N21-C22	1.416(7)	C27-N28	1.371(7)
C21-C33 ⁱⁱ	1.464(9)	C28-C29	1.508(9)
	Valence angles [°]		
C1-N1-C2	118.4(6)	C21-N21-C22	119.3(6)
N1-C1-C13 ⁱ	124.2(7)	N21-C21-C33 ⁱⁱ	124.1(7)

(**L1**) ⁱ -x+1,-y+1,-z ⁱⁱ -x+2,-y+2,-z

Table S3. Relevant photophysical data of studied compounds, (λ_{em} , λ_{ex} nm, λ [nm] (ϵ [dm³ mol⁻¹ cm⁻¹])). Bp=8

Compound	Solvent	λ_{ex} [nm]	λ_{em} [nm]	Fluorescence Intensity a. u.	λ [nm] (ϵ [dm ³ mol ⁻¹ cm ⁻¹])	A	C
L1	acetonitrile	295	454	1491630	338(16176)	0,275	1,7*10 ⁻⁵
		350	459	1349115	346(16294)	0,277	
					382(16353)	0,278	
	chloroform	295	498	1939430	338(16765)R	0,285	1,7*10 ⁻⁵
		350	500	2957245	350(17117)	0,291	
					382(15823)	0,269	

L2	methanol	295	497	2792725	342(16800)	0,336	2*10 ⁻⁵
		350	498	4940850			
	benzene	295	516	968195	336(14529)	0,247	1,7*10 ⁻⁵
		350	514	2100280	350(14765)	0,251	
					390(14412)	0,245	
	acetonitrile	295	452	3270265	340(19529)	0,332	1,7*10 ⁻⁵
		350	453	2621965	382(20882)	0,355	
	chloroform	295	492	5620800	338(27882)	0,474	1,7*10 ⁻⁵
		350	494	6858510	346(28471)	0,484	
					380(27176)	0,462	
	methanol	295	493	2646015	342(22214)	0,311	1,4*10 ⁻⁵
		350	495	3994280			
	benzene	295	517	3085945	342(16941)	0,288	1,7*10 ⁻⁵
		350	514	5024360	390(16176)	0,275	

Table S4. Theoretical PBE0/6-311++G(d,p)/PCM(ACN) vertical excitation wavelengths λ [nm] for most intensive transitions together with the corresponding oscillator strengths f and the orbital contributions for investigated species.

L1		
λ [nm]	f	Orbitals (contribution)
419.73	0.4374	HOMO->LUMO (63%)
302.15	0.4114	HOMO-3->LUMO (47%) HOMO-2->LUMO+1 (40%)
L2		
403.14	0.5140	HOMO-1->LUMO (63%)
325.87	0.1753	HOMO-2->LUMO (67%)

Table S5. Relevant fluorescent data of studied compounds in the solid state (λ_{em} , λ_{ex}) [nm].

Compound	λ_{ex} [nm]	λ_{em} [nm]
L1	295	548
	350	552
L2	295	549
	350	561

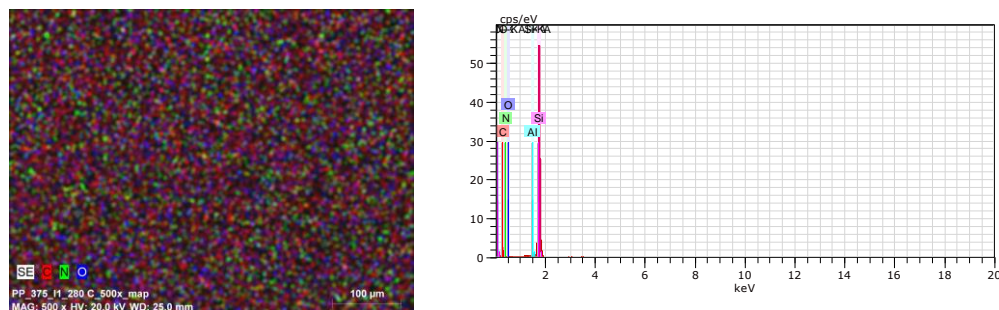
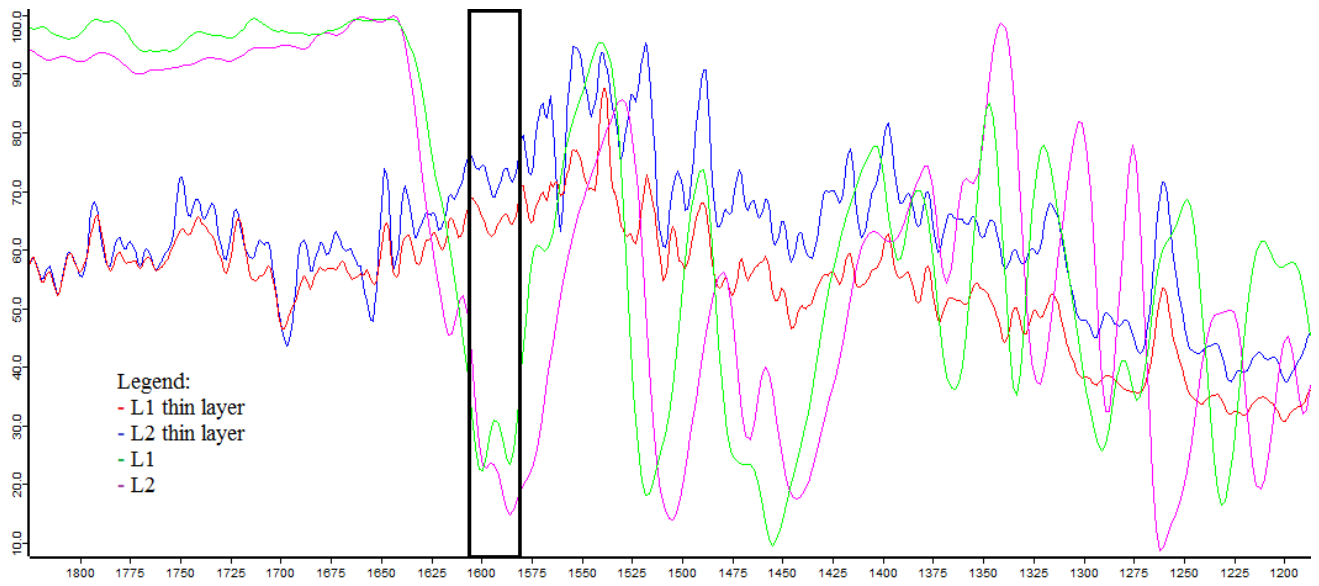
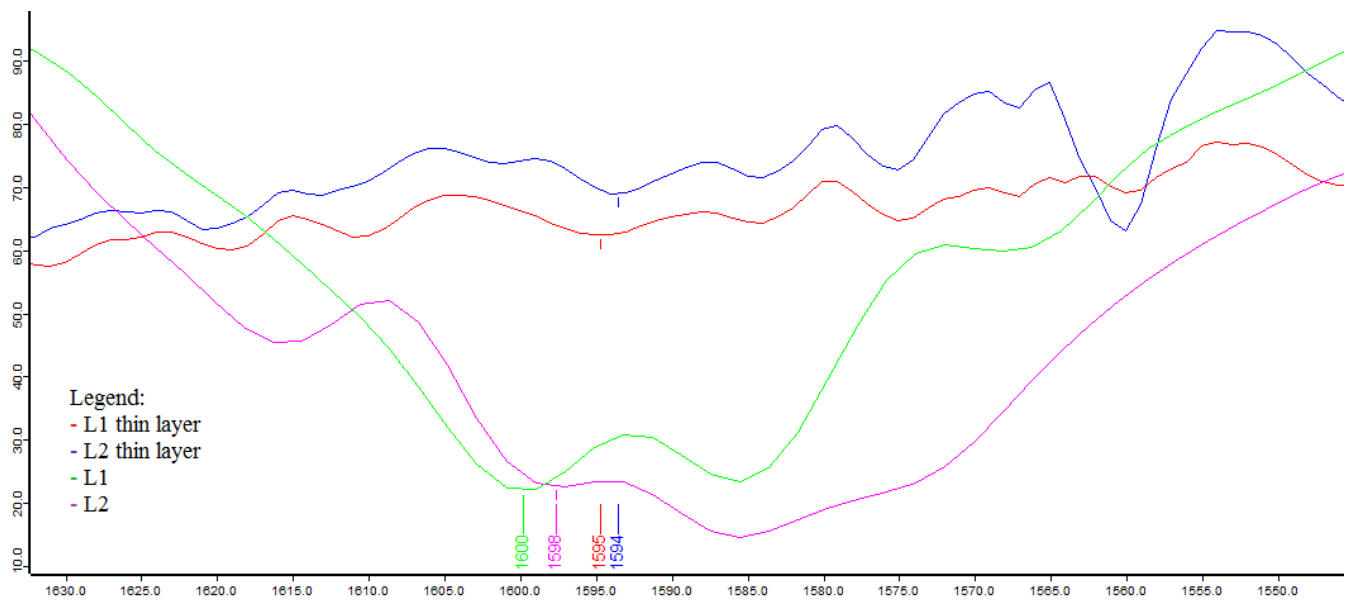


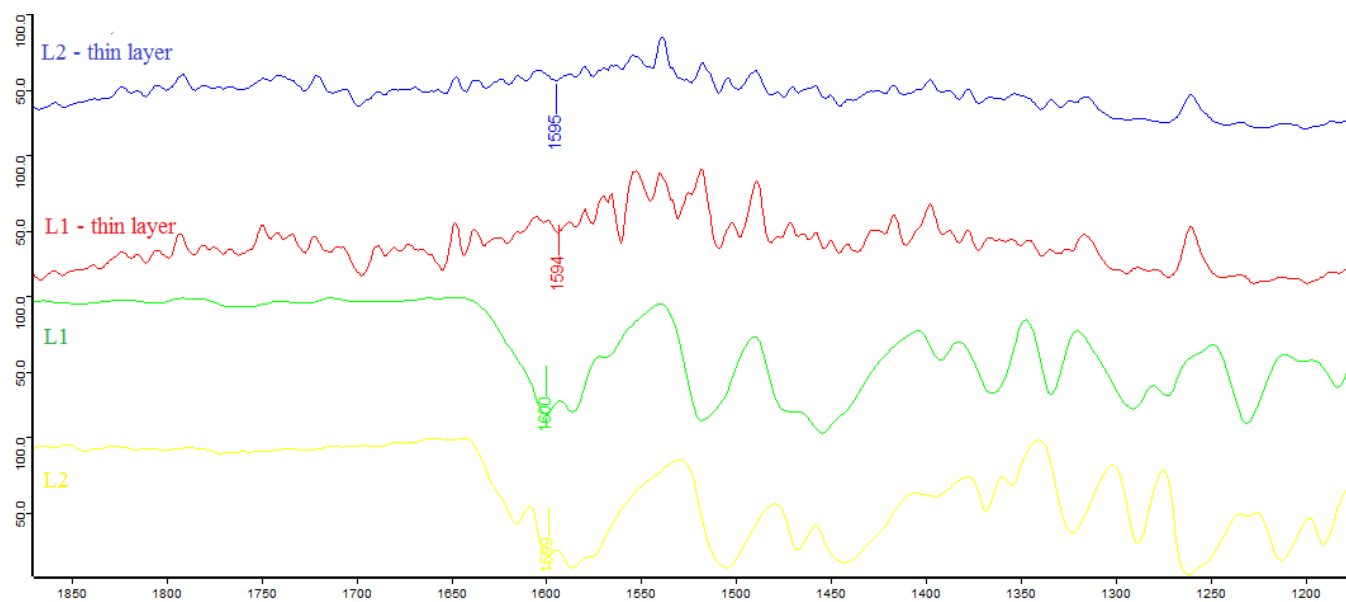
Figure S14. SEM images of **L1**/Si mapping scanning size 100 μm .



a)



b)



c)

Figure S15. (a), (b) and (c) DRIFT spectra of **L1** and **L2** samples and **L1/Si** and **L2/Si**.