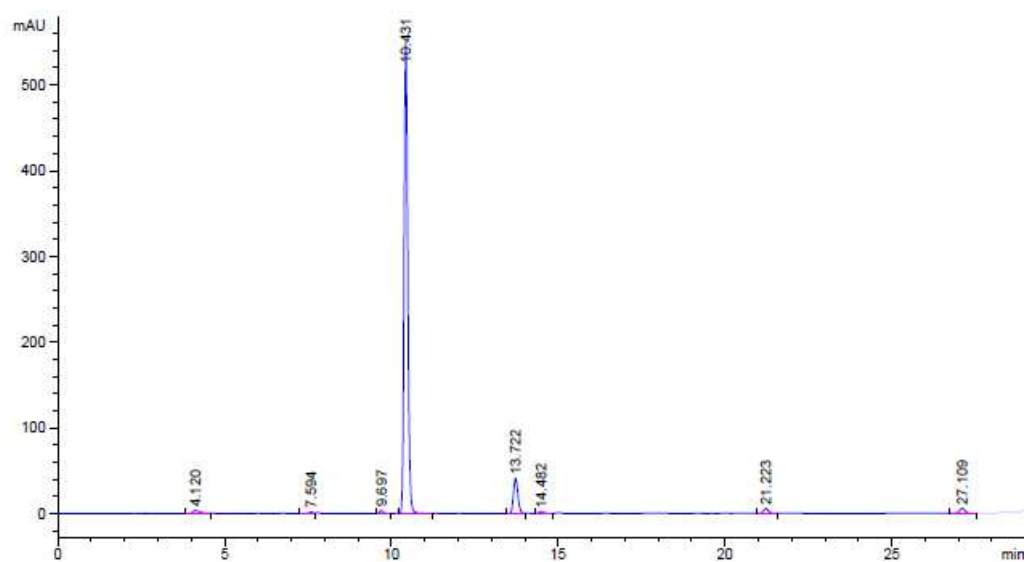
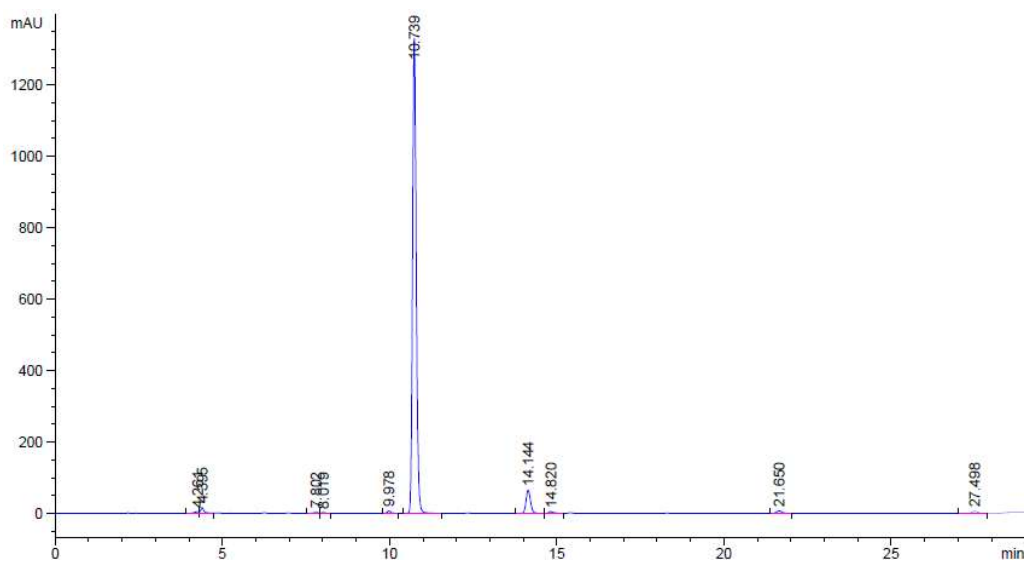


## Support information

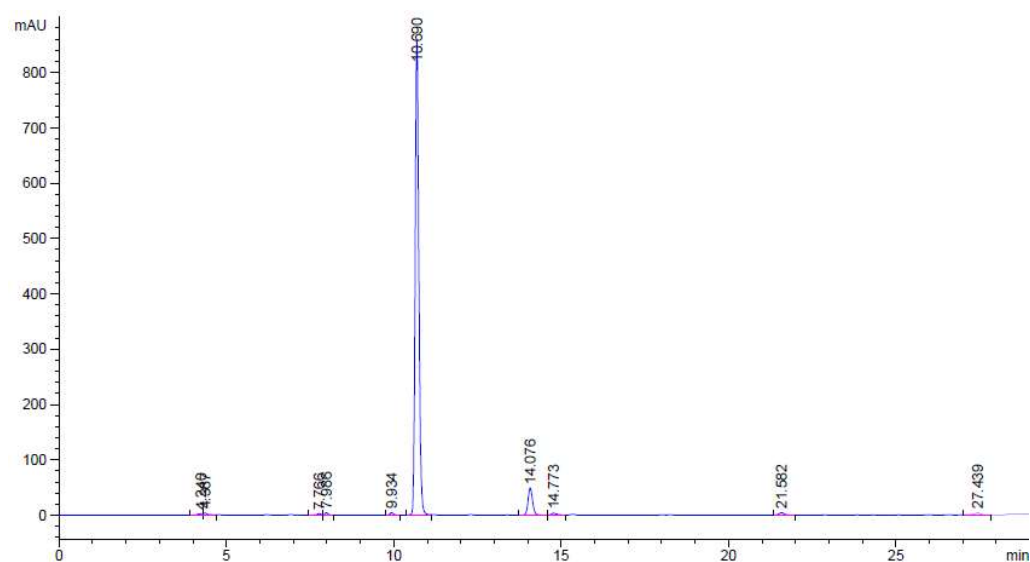
Figures S1-S7 below exhibit the chromatograms of the 7 crude extracts obtained with factorial design. Extraction conditions for each extract are shown in parentheses in the respective figure legend.



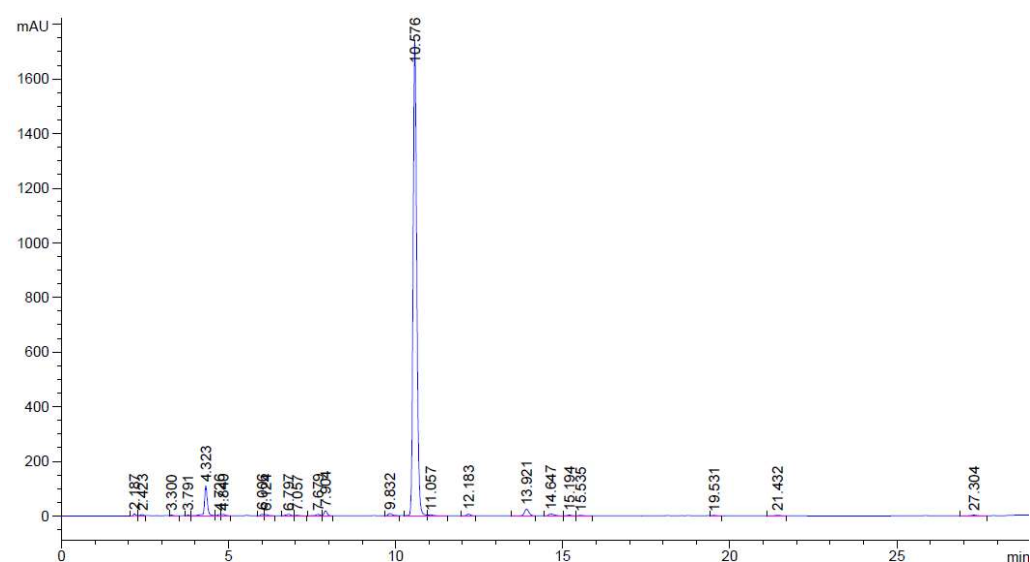
**Figure S1.** Chromatogram of extract 1 (P = 120 bar, cosolvent: water).



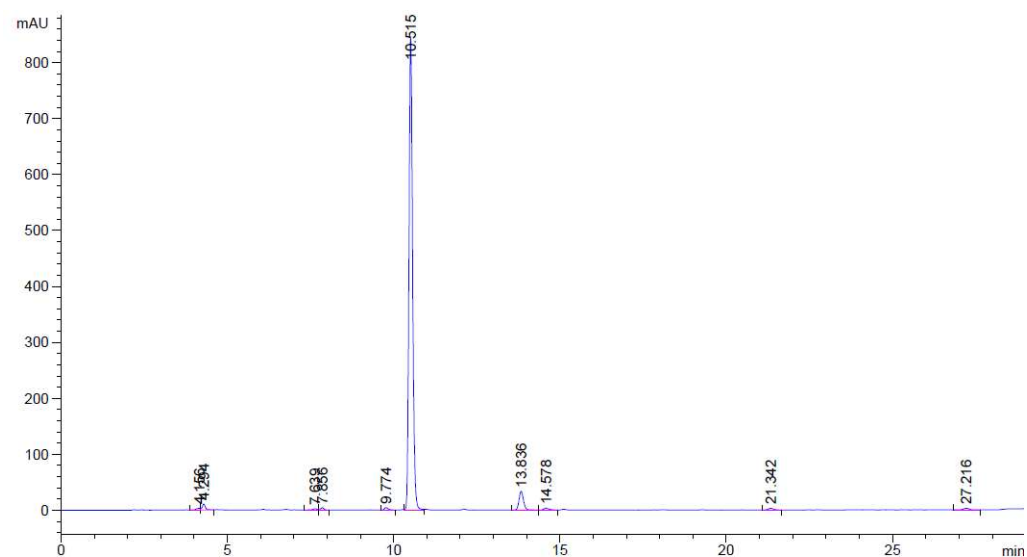
**Figure S2.** Chromatogram of extract 2 (P = 120 bar, cosolvent: ethanol).



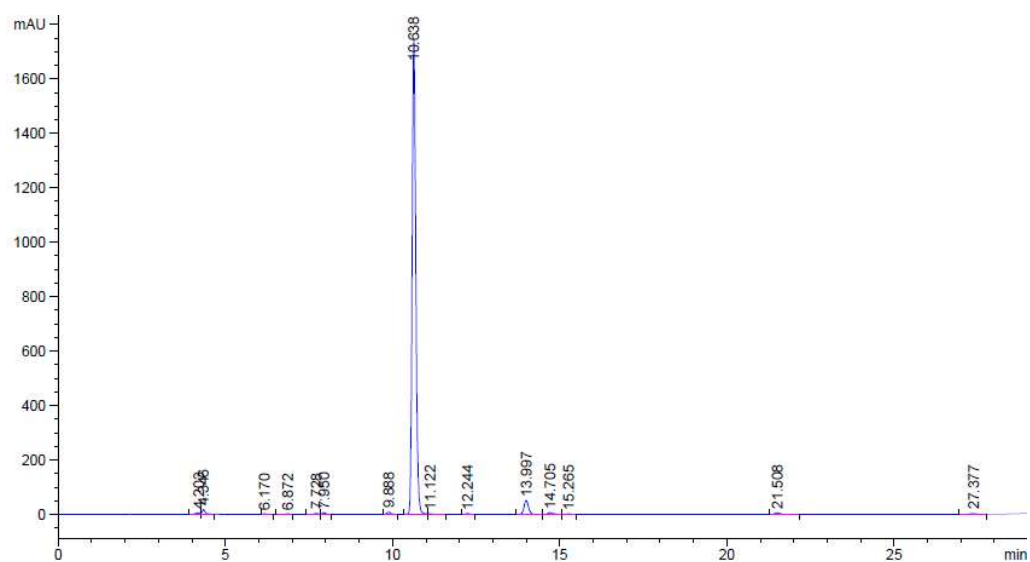
**Figure S3.** Chromatogram of extract 3 (P = 240 bar, cosolvent: water).



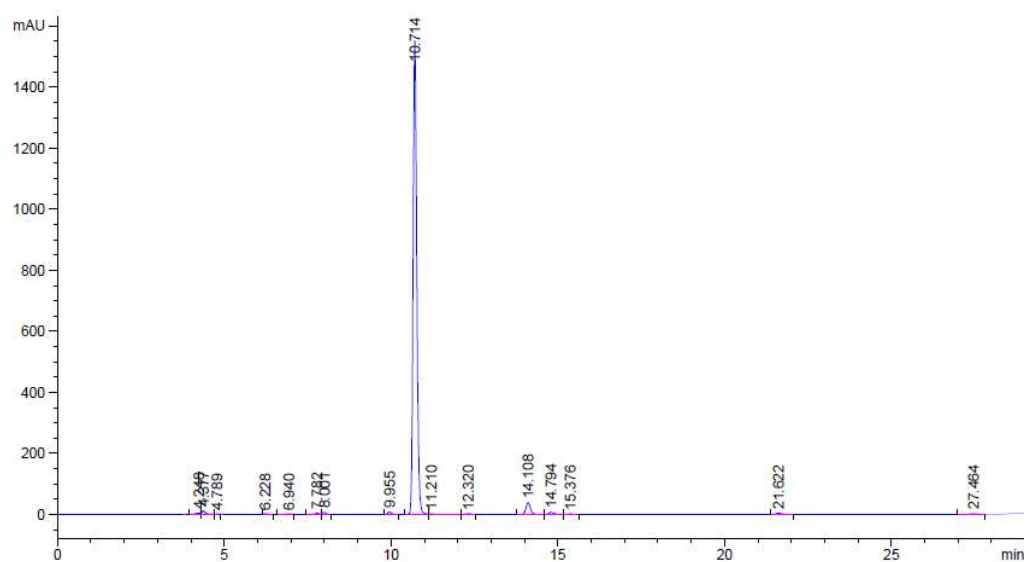
**Figure S4.** Chromatogram of extract 4 (P = 240 bar, cosolvent: ethanol).



**Figure S5.** Chromatogram of extract 5 (P = 180 bar, cosolvent: ethanol:water).

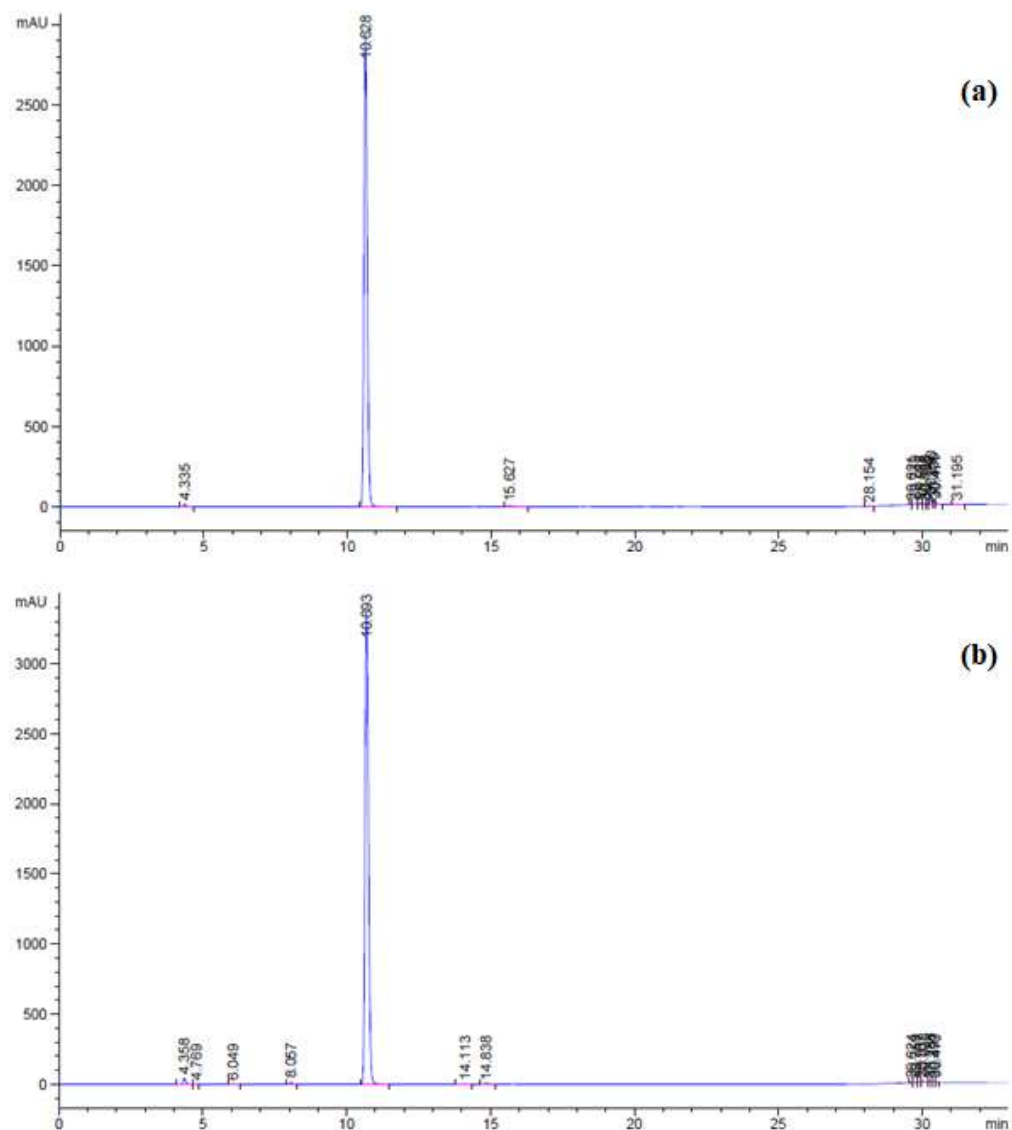


**Figure S6.** Chromatogram of extract 6 (P = 180 bar, cosolvent: ethanol:water).

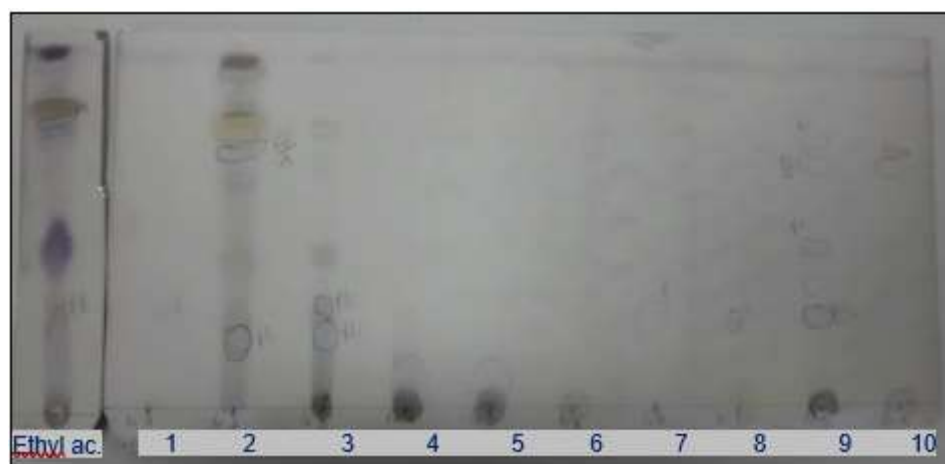


**Figure S7.** Chromatogram of extract 7 (P = 180 bar, cosolvent: ethanol:water).

Figures S8 and S9 present analysis results after extract purification. The compound *p*-anisic acid was concentrated at subfractions 4 and 5 that was evidenced by HPLC analysis (Figure S8).



**Figure S8.** Chromatograms of **(a)** subfraction 4 and **(b)** subfraction 5 were obtained by silica gel column chromatography separation.



**Figure S9.** TLC for the ethyl acetate fraction and subfractions 1–10 after application of sulfuric vanillin as color reagent (the marks drawn represent the spots visualized under U.V. light).

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Number of Words: 197 (approx.)  
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