

Supporting Information

Ferrocene-containing Pseudorotaxanes in Crystals: Aromatic Interactions with Hammett Correlation

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Experimental section

General

^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra were acquired on a MERCURY300 (Varian), EX-400 (JEOL) and a AV-400M (Bruker). The chemical shifts were referenced with respect to CHCl_3 (δ 7.26), CD_2HCN (δ 1.93) for ^1H , and CDCl_3 (δ 77.0), CD_3CN (δ 1.30) for ^{13}C as internal standards. Elemental analysis was carried out with a CHNS-932 (LECO) or MT-5 CHN (Yanaco) autorecorder. IR spectra were measured with a FTIR-8100A (Shimadzu) and FT/IR-4100 (JASCO). DSC was recorded by Seiko instrument DSC6200S. Thermogravimetric analysis (TGA) was recorded on Seiko TG/DTA6200R. $\text{H}_2\text{NCH}_2\text{Ar}$ ($\text{Ar} = -\text{C}_6\text{H}_4-4-\text{Br}$, $-\text{C}_6\text{H}_3-3-\text{F}-4-\text{Me}$) was prepared by reaction of LiAlH_4 and NCAr in THF under reflux condition. Other chemicals were commercially available and used without further purification.

Synthesis of $\text{FcCH}=\text{NCH}_2\text{C}_6\text{H}_4-3,4-\text{Cl}_2$ (**3a**) ($\text{Fc} = \text{Fe}(\text{C}_5\text{H}_4)(\text{C}_5\text{H}_5)$)

A solution of ferrocenecarboxaldehyde (1.07 g, 5.0 mmol) and $\text{H}_2\text{NCH}_2\text{C}_6\text{H}_4-3,4-\text{Cl}_2$ (0.89 g, 5.1 mmol) in EtOH (100 mL) was stirred for 14 h at 65 °C in the presence of MS4A (molecular sieves 4A, 1.0 g). The reaction mixture was filtered and then evaporated to yield the crude product which was purified by washing with hexane to yield $\text{FcCH}=\text{NCH}_2\text{C}_6\text{H}_4-3,4-\text{Cl}_2$ as a reddish-brown solid (1.69 g, 4.5 mmol, 90%). ^1H NMR (300 MHz, CDCl_3 , r.t.): δ = 4.19 (s, 5H, C_5H_5), 4.41 (m, 2H, C_5H_4), 4.60 (s, 2H, NCH_2), 4.66 (m, 2H, C_5H_4), 7.15 (d, 1H, C_6H_3), 7.39 (d, 1H, C_6H_3), 7.42 (d, 1H, C_6H_3), 8.24 (s, 1H, $\text{N}=\text{CH}$); $^{13}\text{C}\{^1\text{H}\}$ NMR (100 MHz, CDCl_3 , r.t.): δ = 63.6 (NCH_2), 68.4 (C_5H_5), 69.1 (C_5H_4), 70.6 (C_5H_4), 79.8 (C_5H_4), 126.9 (C_6H_3), 129.4 (C_6H_3), 130.2, (C_6H_3), 130.5 (C_6H_3), 132.1 (C_6H_3), 140.0 (C_6H_3), 163.2 ($\text{N}=\text{CH}$); IR (KBr disk): $\nu(\text{C}=\text{N}) = 1636\text{ cm}^{-1}$; elemental analysis: calcd (%) for $\text{C}_{18}\text{H}_{15}\text{NCl}_2\text{Fe}(\text{H}_2\text{O})_{0.5}$: C 56.73, H 4.23, N 3.68; found: C 56.91, H 3.98, N 3.70.

Synthesis of $\text{FcCH}=\text{NCH}_2\text{C}_6\text{H}_3-3,4-\text{F}_2$ (**3b**)

Compound **3b** was prepared in the same way as **3a**. The reaction of ferrocenecarboxaldehyde (1.07 g, 5.0 mmol) and $\text{H}_2\text{NCH}_2-\text{C}_6\text{H}_3-3,4-\text{F}_2$ (0.71 g, 5.0 mmol) in EtOH (100 mL) for 20 h at 65 °C in the presence of MS4A (1.0 g) yielded **3b** (1.66 g, 4.9 mmol, 98%) as a reddish-brown solid. ^1H NMR (300 MHz, CDCl_3 , r.t.): δ = 4.19 (s, 5H, C_5H_5), 4.40 (s, 2H, NCH_2), 4.59 (m, 2H, C_5H_4), 4.68 (m, 2H, C_5H_4), 6.90–7.19 (3H, C_6H_3), 8.24 (s, 1H, $\text{N}=\text{CH}$); IR (KBr disk): $\nu(\text{C}=\text{N}) = 1638\text{ cm}^{-1}$; elemental analysis: calcd (%) for $\text{C}_{18}\text{H}_{15}\text{NF}_2\text{Fe}(\text{H}_2\text{O})_{0.25}$: C 62.91, H 4.55, N 4.08, F 11.06; found: C 63.23, H 4.26, N 4.04, F 11.27.

Synthesis of FcCH=NCH₂C₆H₄-4-F (3c)

Compound **3c** was prepared in the same way as **3a**. The reaction of ferrocenecarboxaldehyde (1.07 g, 5.0 mmol) and H₂NCH₂C₆H₄-4-F (626 mg, 5.0 mmol) in toluene (100 mL) for 12 h at 75 °C in the presence of MS4A (1.0 g) yielded **3c** (1.29 g, 4.0 mmol, 80%) as a reddish-brown solid. ¹H NMR (300 MHz, CDCl₃, r.t.): δ = 4.17 (s, 5H, C₅H₅), 4.39 (m, 2H, C₅H₄), 4.68 (s, 2H, NCH₂), 4.62 (m, 2H, C₅H₄), 7.03 (d, 2H, C₆H₄, *J* = 8.7 Hz), 7.26 (d, 2H, C₆H₄, *J* = 8.7 Hz), 8.24 (s, 1H, N=CH); ¹³C{¹H} NMR (75.5 MHz, CDCl₃, r.t.): δ = 64.3 (NCH₂), 68.5 (C₅H₄), 69.0 (C₅H₅), 70.5 (C₅H₄), 80.2 (C₅H₄), 115.2 (C₆H₄, *J*(FH) = 22 Hz), 129.3 (C₅H₄, *J*(FH) = 8 Hz), 135.3 (C₆H₄), 161.8 (C₆H₄, *J*(FH) = 245 Hz), 162.3 (N=CH); IR (KBr disk): ν(C=N) = 1638 cm⁻¹; elemental analysis: calcd (%) for C₁₈H₁₆FFeN: C 67.31, H 5.02, N 4.36; found: C 67.25, H 4.83, N 4.49.

Synthesis of FcCH=NCH₂C₆H₄-4-Cl (3d)

Compound **3d** was prepared in the same way as **3a**. The reaction of ferrocenecarboxaldehyde (2.14 g, 10 mmol) and H₂NCH₂C₆H₄-4-Cl (1.48 g, 10 mmol) in EtOH (200 mL) for 6 h at 65 °C in the presence of MS4A (2.0 g) yielded **3d** (2.80 g, 8.3 mmol, 80%) as a reddish-brown solid. ¹H NMR (300 MHz, CDCl₃, r.t.): δ = 4.18 (s, 5H, C₅H₅), 4.39 (m, 2H, C₅H₄), 4.62 (s, 2H, NCH₂), 4.68 (m, 2H, C₅H₄), 7.24 (d, 2H, C₆H₄, *J* = 8.7 Hz), 7.31 (d, 2H, C₆H₄, *J* = 8.7 Hz), 8.24 (s, 1H, NCH); ¹³C{¹H} NMR (75.5 MHz, CDCl₃, r.t.): δ = 64.2 (NCH₂), 68.4 (C₅H₄), 68.9 (C₅H₅), 70.5 (C₅H₄), 80.1 (C₅H₄), 128.4 (C₆H₄), 129.0 (C₆H₄), 132.4 (C₆H₄), 138.1 (C₆H₄), 162.4 (NCH); IR (KBr disk): ν(C=N) = 1636 cm⁻¹; elemental analysis: calcd (%) for C₁₈H₁₆ClFeN: C 64.03, H 4.78, N 4.15; found: C 63.73, H 4.84, N 4.10.

Synthesis of FcCH=NCH₂C₆H₄-4-Br (3e)

Compound **3e** was prepared in the same way as **3a**. The reaction of ferrocenecarboxaldehyde (1.08 g, 5.0 mmol) and H₂NCH₂C₆H₄-4-Cl (939 mg, 5.0 mmol) in EtOH (80 mL) for 12 h at 65 °C in the presence of MS4A (1.0 g) yields **3e** (1.50 g, 3.9 mmol, 78%) as a reddish-brown solid. ¹H NMR (300 MHz, CDCl₃, r.t.): δ = 4.18 (s, 5H, C₅H₅), 4.39 (m, 2H, C₅H₄), 4.60 (s, 2H, NCH₂), 4.68 (m, 2H, C₅H₄), 7.18 (d, 2H, C₆H₄, *J* = 8.4 Hz), 7.46 (d, 2H, C₆H₄, *J* = 8.4 Hz), 8.23 (s, 1H, N=CH); ¹³C{¹H} NMR (75.5 MHz, CDCl₃, r.t.): δ = 64.2 (NCH₂), 68.5 (C₅H₄), 69.0 (C₅H₅), 70.5 (C₅H₄), 80.0 (C₅H₄), 120.6 (C₆H₄), 129.4 (C₆H₄), 131.4 (C₆H₄), 138.5 (C₆H₄), 162.6 (N=CH); IR (KBr disk): ν(C=N) 1636 cm⁻¹; elemental analysis: calcd (%) for C₁₈H₁₆BrFeN(H₂O)_{0.25}: C 55.92, H

4.30, N 3.62; found: C 55.98, H 4.12, N, 3.36.

Synthesis of FcCH=NCH₂C₆H₃-3-F-4-Me (3f)

Compound **3f** was prepared in the same way as **3a**. The reaction of ferrocene-carboxaldehyde (0.53 g, 2.5 mmol) and H₂NCH₂C₆H₃-3-F-4-Me (0.35 g, 2.5 mmol) in EtOH (100 mL) for 12 h at 65 °C in the presence of MS4A (1.0 g) yielded **3f** (0.76 g, 2.3 mmol, 92%) as a reddish-brown solid. ¹H NMR (300 MHz, CDCl₃, r.t.): δ = 2.54 (Me), 4.46 (s, 5H, C₅H₅), 4.67 (m, 2H, C₅H₄), 4.89 (s, 2H, NCH₂), 4.96 (ITI, 2H, C₅H₄), 7.24–7.54 (C₆H₃, 3H), 8.51 (s, 1H, N=CH); ¹³C {¹H} NMR (75.5 MHz, CDCl₃, r.t.): δ = 14.2 (Me), 64.2 (NCH₂), 68.6 (C₅H₄), 69.0 (C₅H₅), 70.5 (C₅H₄), 80.3 (C₅H₄), 114.1 (C₅H₃ *J*(FC) = 22 Hz), 123.0 (C₆H₃ *J*(FC) = 16 Hz), 131.3 (C₆H₃, *J*(FC) = 5 Hz), 139.3 (C₆H₃), 161.1 (C₆H₃), 162.4 (N=CH); IR (KBr disk): ν(C=N) = 1634 cm⁻¹.

Synthesis of FcCH=NCH₂C₆H₄-4-I (3g)

Compound **3g** was prepared in the same way as **3a**. The reaction of ferrocenecarboxaldehyde (420 mg, 2.0 mmol) and H₂NCH₂C₆H₄-4-I (477 mg, 2.0 mmol) in EtOH (20 mL) for 5 h at 65 °C in the presence of MS4A (1.0 g) yielded **3g** (711 mg, 1.7 mmol, 85%) as a reddish-brown solid. ¹H NMR (300 MHz, CDCl₃, r.t.): δ = 4.18 (s, 5H, C₅H₅), 4.39 (m, 2H, C₆H₄), 4.59 (s, 2H, NCH₂), 4.67 (m, 2H, C₅H₄), 7.06 (d, C₆H₄, *J* = 8.1 Hz), 7.66 (d, C₆H₄, *J* = 8.1 Hz), 8.23 (s, 1H, N=CH); ¹³C {¹H} NMR (75.5 MHz, CDCl₃, r.t.): δ = 64.2 (NCH₂), 68.4 (C₅H₄), 68.9 (C₅H₅), 70.4 (C₅H₄), 80.0 (C₅H₄), 92.0 (C₆H₄), 129.6 (C₆H₄), 137.2 (C₆H₄), 139.2 (C₆H₄), 162.4 (N=CH); IR (KBr disk): ν(C=N) = 1632 cm⁻¹; elemental analysis: calcd (%) for C₁₈H₁₆IFeN(H₂O)_{0.25} C 49.86, H 3.84, N 3.23, I 29.27; found: C 49.82, H 4.08, N 3.26, I 28.92.

Synthesis of [FcCH₂NH₂CH₂C₆H₃-3,4-Cl₂] PF₆ (2a)

FcCH = NCH₂C₆H₃-3,4-Cl₂ (**3a**) (1.4 g, 3.8 mmol) was dissolved in MeOH (50 mL) at room temperature. NaBH₄ (1.55 g, 41 mmol) was added to the solution and it was stirred for 12 h at room temperature and then mixed with 4 M HCl (150 mL) to cause the separation of a solid from the solution. The resulting solid was filtered and washed with water and Et₂O to yield [FcCH₂NH₂CH₂C₆H₃-3,4-Cl₂]Cl which was used without further purification. Acetone (100 mL) suspension of [FcCH₂NH₂CH₂C₆H₃-3,4-Cl₂]Cl and NH₄PF₆ (3.26 g, 20 mmol) was stirred for 2 h at room temperature. The resulting mixture was filtered and the solids were washed with water, CH₂Cl₂ and hexane to give [FcCH₂NH₂CH₂C₆H₃-3,4-Cl₂](PF₆) (**2a**) as a yellow solid (quant). ¹H NMR (300 MHz, CD₃CN, r.t.): δ = 4.03 (s, 2H, NCH₂), 4.07 (s, 2H, NCH₂), 4.20 (s, 5H, C₅H₅), 4.27

(m, 2H, C₅H₄), 4.36 (m, 2H, C₅H₄), 7.06 (s, 1H, C₆H₃), 7.34 (dd, 1H, C₆H₃, $J = 8.5, 2.2$ Hz), 7.59 (d, 1H, C₆H₃, $J = 8.0$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (75.5 MHz, CD₃CN, r.t.): $\delta = 48.5$ (NCH₂), 50.0 (NCH₂), 69.9 (C₅H₄), 70.5 (C₅H₅), 71.4 (C₅H₄), 76.5 (C₅H₄), 131.0 (C₆H₃), 132.0 (C₆H₃), 132.6 (C₆H₃), 133.0 (C₆H₃), 133.2 (C₆H₃), 134.0 (C₆H₃); elemental analysis: calcd (%) for C₁₈H₁₈NFeF₆Cl₂P: C 41.57, H 3.49, N 2.69, Cl 13.63; found: C 41.39, H 3.38, N 2.72, Cl 13.88.

Synthesis of [FcCH₂NH₂CH₂C₆H₃-3,4-F](PF₆) (2b)

Compound **2b** was prepared in the same way as **2a** (82% from **3b**). ^1H NMR (300 MHz, CD₃CN, r.t.): $\delta = 4.03$ (s, 2H, NCH₂), 4.07 (s, 2H, NCH₂), 4.21 (s, 5H, C₅H₅), 4.28 (m, 2H, C₅H₄), 4.36 (m, 2H, C₅H₄), 7.24–7.38 (3H, C₆H₃); $^{13}\text{C}\{^1\text{H}\}$ NMR (75.5 MHz, CD₃CN, r.t.): $\delta = 48.4$ (NCH₂), 50.2 (NCH₂), 69.9 (C₅H₄), 70.5 (C₅H₅), 71.4 (C₅H₄), 76.6 (C₅H₄), 118.8 (C₆H₃), 120.1 (C₆H₃, $J = 14, 5$ Hz), 128.0 (C₆H₃, $J(\text{CF}) = 7, 4$ Hz), 129.2 (C₆H₃, $J(\text{CF}) = 5, 5$ Hz), 150.0 (C₆H₃, $J(\text{CF}) = 73, 15$ Hz), 152.4 (C₆H₃, $J(\text{CF}) = 74, 14$ Hz); elemental analysis: calcd (%) for C₁₈H₁₉ClF₆FeNP: C 44.38, H 3.72, N 2.88; found: C 44.35, H 3.76, N 2.93.

Synthesis of [FcCH₂NH₂CH₂C₆H₄-4-F](PF₆) (2c)

Compound **2c** was prepared in the same way as **2a** (47% from **3c**). ^1H NMR (300 MHz, CD₃CN, r.t.): $\delta = 4.04$ (s, 2H, NCH₂), 4.09 (s, 2H, NCH₂), 4.21 (s, 5H, C₅H₅), 4.27 (m, 2H, C₅H₄), 4.37 (m, 2H, C₅H₄), 7.18 (d, 2H, C₆H₄, $J = 8.8$ Hz), 7.45 (d, 2H, C₆H₄, $J = 8.8$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (75.5 MHz, CD₃CN, r.t.): $\delta = 48.7$ (NCH₂), 50.9 (NCH₂), 69.6 (C₅H₅), 70.5 (C₅H₄), 71.4 (C₅H₄), 76.4 (C₅H₄), 116.9 (C₆H₄, $J(\text{FC}) = 22$ Hz), 127.8 (C₆H₄, $J(\text{FC}) = 3$ Hz), 133.5 (C₆H₄, $J(\text{FC}) = 9$ Hz), 164.2 (C₆H₄, $J(\text{FC}) = 247$ Hz); elemental analysis: calcd (%) for C₁₈H₁₉F₇FeNP: C 46.08, H 4.08, N 2.99; found: C 46.37, H 3.93, N 3.07.

Synthesis of [FcCH₂NH₂CH₂C₆H₄-4-Cl](PF₆) (2d)

Compound **2d** was prepared in the same way as **2a** (65% from **3d**). ^1H NMR (300 MHz, CD₃CN, r.t.): $\delta = 4.05$ (s, 2H, NCH₂), 4.09 (s, 2H, NCH₂), 4.21 (s, 5H, C₅H₅), 4.28 (m, 2H, C₅H₄), 4.37 (m, 2H, C₅H₄), 7.40 (d, 2H, C₆H₄, $J = 8.7$ Hz), 7.46 (d, 2H, C₆H₄, $J = 8.7$ Hz); $^{13}\text{C}\{^1\text{H}\}$ NMR (75.5 MHz, CD₃CN, r.t.): $\delta = 48.2$ (NCH₂), 50.4 (NCH₂), 69.6 (C₅H₅), 70.2 (C₅H₄), 71.1 (C₅H₄), 76.1 (C₅H₄), 129.7 (C₆H₄), 130.2 (C₆H₄), 132.4 (C₆H₄), 135.6 (C₆H₄); elemental analysis: calcd (%) for C₁₈H₁₉ClF₆FeNP: C 44.52, H 3.94, N 2.88; found: C 44.71, H 4.21, N 2.90.

Synthesis of [FcCH₂NH₂CH₂C₆H₄-4-Br](PF₆) (2e)

Compound **2e** was prepared in the same way as **2a** (38% from **3e**). ¹H NMR (300 MHz, CD₃CN, r.t.): δ = 4.04 (s, 2H, NCH₂), 4.07 (s, 2H, NCH₂), 4.21 (s, 5H, C₅H₅), 4.27 (m, 2H, C₅H₄), 4.37 (m, 2H, C₅H₄), 7.34 (d, 2H, C₆H₄, *J* = 8.4 Hz), 7.61 (d, 2H, C₆H₄, *J* = 8.4 Hz); ¹³C{¹H} NMR (75.5 MHz, CD₃CN, r.t.): δ = 48.8 (NCH₂), 51.0 (NCH₂), 69.9 (C₅H₅), 70.5 (C₅H₄), 71.5 (C₅H₄), 76.2 (C₅H₄), 124.3 (C₆H₄), 130.8 (C₆H₄), 133.0 (C₆H₄), 133.1 (C₆H₄); elemental analysis: calcd (%) for C₁₈H₁₉BrF₆FeNP: C 40.79, H 3.61, N 2.64; found: C 40.43, H 3.58, N, 2.67.

Synthesis of [FcCH₂NH₂CH₂C₆H₃-3-F-4-Me](PF₆) (2f)

Compound **2f** was prepared in the same way as **2a** (59% from **3f**). ¹H NMR (300 MHz, CD₃CN, r.t.): δ = 2.27 (Me), 4.06 (s, 2H, NCH₂), 4.07 (s, 2H, NCH₂), 4.21 (s, 5H, C₅H₅), 4.28 (m, 2H, C₅H₄), 4.37 (m, 2H, C₅H₄), 7.11-7.14 (2H, C₆H₃), 7.30 (1H, C₆H₃); ¹³C{¹H} NMR (75.5 MHz, CD₃CN, r.t.): δ = 48.9 (NCH₂), 51.0 (NCH₂), 70.1 (C₅H₅), 71.5 (C₅H₄), 76.3 (C₅H₄), 117.4 (C₆H₃, *J*(FC) = 23 Hz), 126.8 (C₆H₃, *J*(FC) = 4 Hz), 127.6 (C₆H₃, *J*(FC) = 17 Hz), 131.1 (C₆H₃, *J*(FC) = 8 Hz), 133.2 (C₆H₃, *J*(FC) = 5 Hz), 162.0 (C₆H₃, *J*(FC) = 254 Hz).

Synthesis of [FcCH₂NH₂CH₂C₆H₄-4-I]PF₆ (2g)

Compound **2g** was prepared in the same way as **2a** (50% from **3g**). ¹H NMR (300 MHz, CD₃CN, r.t.): δ = 4.03 (s, 2H, NCH₂), 4.05 (s, 2H, NCH₂), 4.20 (s, 5H, C₅H₅), 4.27 (m, 2H, C₅H₄), 4.36 (m, 2H, C₅H₄), 7.19 (d, 2H, C₆H₄, *J* = 8.7 Hz), 7.81 (d, 2H, C₆H₄, *J* = 8.7 Hz); ¹³C{¹H} NMR (75.5 MHz, CD₃CN, r.t.): δ = 48.8 (NCH₂), 51.1 (NCH₂), 69.9 (C₅H₅), 70.5 (C₅H₄), 71.4 (C₅H₄), 76.3 (C₅H₄), 131.4 (C₆H₄), 133.0 (C₆H₄), 139.1 (C₆H₄), 158.3 (C₆H₄); elemental analysis: calcd (%) for C₁₈H₁₉IF₆FeNP: C 37.46, H 3.32, N 2.43; found: C 37.65, H 3.33, N 2.40.

Crystal synthesis of [(FcCH₂NH₂CH₂C₆H₃-3,4-Cl₂)(DB24C8)](PF₆) (1a)

Yellow crystals of pseudorotaxane **1a** were obtained by slow evaporation of CH₂Cl₂/Et₂O solution of **2a** (52 mg, 0.10 mmol) and DB24C8 (46 mg, 0.10 mmol). **1a** was obtained in 10% yield. Elemental analysis: calcd (%) for C₄₂H₅₀NO₈FeF₆Cl₂P: C 52.08, H 5.20, N 1.45, Cl 7.32; found: C 51.58, H 4.93, N 1.45, Cl 7.38.

Crystal synthesis of [(FcCH₂NH₂CH₂C₆H₃-3,4-F₂)(DB24C8)](PF₆) (1b)

Yellow crystals of pseudorotaxane **1b** were obtained by slow evaporation of acetone/hexane solution of **2b** (46 mg, 0.093 mmol) and DB24C8 (49 mg, 0.11 mmol).

1b was obtained in 57% yield. Elemental analysis: calcd (%) for $C_{42}H_{50}NO_8FeF_8P$: C 53.91, H 5.39, N 1.50; found: C 53.81, H 5.39, N 1.52.

*Crystal synthesis of $[(FcCH_2NH_2CH_2C_6H_4-4-F)(DB24C8)](PF_6)$ (**1c**)*

Yellow crystals of pseudorotaxane **1c** were obtained by slow evaporation of acetone/hexane solution of **2c** (47 mg, 0.10 mmol) and DB24C8 (44 mg, 0.099 mmol). **1c** was obtained in 49% yield. Elemental analysis: calcd (%) for $C_{42}H_{51}NO_8FeF_7P$: C 54.97, H 5.60, N 1.53; found: C 54.92, H 5.65, N 1.89.

*Crystal synthesis of $[(FcCH_2NH_2CH_2C_6H_4-4-Cl)(DB24C8)](PF_6)$ (**1d**)*

Yellow crystals of pseudorotaxane **1d** were obtained by slow evaporation of acetone/hexane solution of **2d** (48 mg, 0.10 mmol) and DB24C8 (45 mg, 0.10 mmol). **1d** was obtained in 74% yield. Elemental analysis: calcd (%) for $C_{42}H_{51}NClF_6FeO_8P$: C 54.00, H 5.50, N 1.50; found: C 53.69, H 5.45, N 1.48.

*Crystal synthesis of $[(FcCH_2NH_2CH_2C_6H_4-4-Br)(DB24C8)](PF_6)$ (**1e**)*

Yellow crystals of pseudorotaxane **1e** were obtained by slow evaporation of $CHCl_3$ /acetone (2.0 mL/0.5 mL) solution of **2e** (53 mg, 0.10 mmol) and DB24C8 (45 mg, 0.10 mmol). **1e** was obtained in 85% yield. Elemental analysis: calcd (%) for $C_{42}H_{51}NBrF_6FeO_8P(H_2O)_{0.5}$: C 51.08, H 5.31, N 1.42; found: C 50.99, H 5.35, N 1.45.

*Crystal synthesis of $[(FcCH_2NH_2CH_2C_6H_3-3-F-4-Me)(DB24C8)](PF_6)$ (**1f**)*

Yellow crystals of pseudorotaxane **1f** were obtained by slow evaporation of CH_2Cl_2 /Et₂O solution of **2f** (47 mg, 0.094 mmol) and DB24C8 (45 mg, 0.10 mmol). **1f** was obtained in 41% yield. Elemental analysis: calcd (%) for $C_{43}H_{53}NF_7FeNO_8P$: C 55.43, H 5.73, N 1.50; found: C 55.45, H 5.29, N 1.55.

*Crystal synthesis of $[(FcCH_2NH_2CH_2C_6H_4-4-I)(DB24C8)](PF_6)$ (**1g**)*

Yellow crystals of pseudorotaxane **1g** were obtained by slow evaporation of CH_2Cl_2 /Et₂O solution of **2g** (57 mg, 0.10 mmol) and DB24C8 (45 mg, 0.10 mmol). **1g** was obtained in 69% yield. Elemental analysis: calcd (%) for $C_{42}H_{51}NI_6FeNO_8P$: C 49.19, H 5.01, N 1.37; found: C 49.29, H 4.98, N 1.37

Table S1. IR and TGA data of **1a-1g** and **2a-2g**.

compound	IR $\nu(\text{N-H})$ / cm^{-1}	$T_{\text{d}5^{\text{a}}}/^{\circ}\text{C}$
1a	3080, 3185	222 (29 ^{c)})
1b	3066, 3163	222 (23 ^{c)})
1c	3077, 3165	238 (50 ^{c)})
1d	3069, 3187	234 (35 ^{c)})
1e	3069, 3187	231 (36 ^{c)})
1f	3067, 3166	237 (20 ^{c)})
1g	3065, 3195	236 (34 ^{c)})
2a	3233, 3262	193
2b	3233, 3262	199
2c	3236, 3262	188
2d	3233, 3266	199
2e	3233, 3265	195
2f	3233, 3266	217
2g	3235, 3268	202

a) 5% weight loss temperature, b) Not determined, c) $T_{\text{d}5}(\text{rotaxane})$ -

$T_{\text{d}5}(\text{axle})$, rotaxane = **1a-1g**, axle = **2a-2g**

Table S2. Crystallographic results

compound	1a	1b	1c	1d
Formula	C ₄₂ H ₅₀ Cl ₂ F ₆ FeNO ₈ P	C ₄₂ H ₅₀ F ₈ FeNO ₈ P	C ₄₂ H ₅₁ F ₇ FeNO ₈ P	C ₄₂ H ₅₁ ClF ₆ FeNO ₈ P
ccdc	1032569	1032570	1032571	1032572
Temp/K	113	113	113	113
<i>a</i> /Å	10.639(8)	10.418(3)	10.326(2)	10.186(5)
<i>b</i> /Å	10.718(10)	11.204(3)	11.626(2)	11.028(6)
<i>c</i> /Å	19.744(14)	18.120(5)	17.311(3)	19.487(8)
α /°	87.30(3)	89.410(7)	88.365(3)	86.963(15)
β /°	76.15(2))	85.654(7)	86.948(4)	76.898(15)
γ /°	89.12(3)	88.006(6)	88.866(10)	89.125(18)
<i>V</i> /Å ³	2183	2107.5	2072.7	2128.8
crystal system	triclinic	triclinic	triclinic	triclinic
space group	P-1 (No. 2)	P-1 (No. 2)	P-1 (No. 2)	P-1 (No. 2)
<i>Z</i>	2	2	2	2
<i>Mr</i>	968.58	935.65	917.65	941.60
Dx/g cm ⁻³	1.473	1.474	1.444	1.453
μ /mm ⁻¹	0.582	0.485	0.488	0.927
F(000)	1004	972	956	972
Nref	9323	9146	8945	9151
R (reflections)	0.1032 (2608)	0.0361 (7353)	0.0346 (6359)	0.0667 (6426)
wR (reflections)	0.1338 (2608)	0.1056 (7353)	0.0791 (6359)	0.2109 (6426)
<i>GOF</i>	0.720	1.011	0.858	1.121
<i>N</i> _{par}	350	568	559	559

No	1e	1f	1g	1i-Ni
Formula by ICP	C ₄₂ H ₅₁ BrF ₆ FeNO ₈ P	C ₄₃ H ₅₃ F ₇ FeNO ₈ P	C ₄₃ H ₅₁ F ₆ FeINO ₈ P	C _{49.5} H ₅₄ ClFeNNiO ₈ S ₁₀
	1032573	1032574	1032575	2122024
<i>Temp</i> /K	113	113	113	113
<i>a</i> /Å	10.1826(3)	10.351(3)	10.335(2)	24.469(4)
<i>b</i> /Å	10.9946(5)	10.862(3)	10.991(3)	10.599(2)
<i>c</i> /Å	19.5223(6)	19.572(5)	19.677(4)	21.608(4)
α /°	87.183(5)	87.190(9)	87.097(6)	90
β /°	76.844(5)	76.858(9)	77.503(5)	97.371(2)
γ /°	88.816(6)	89.111(11)	88.510(7)	90
<i>V</i> /Å ³	2125.6	2140.4	2144.8	5558.0
crystal system	triclinic	triclinic	triclinic	monoclinic
space group	P-1 (No. 2)	P-1 (No. 2)	P-1 (No. 2)	P2 ₁ /c (No. 14)
<i>Z</i>	2	2	2	2
<i>Mr</i>	978.56	931.68	1025.55	1261.54
Dx/g cm ⁻³	1.529	1.446	1.563	1.508
μ /mm ⁻¹	1.412	0.474	1.169	1.196
F(000)	1008	972	1044	968
Nref	8877	9280	9346	12361
R (reflections)	0.0375 (7491)	0.0653 (6406)	0.0343 (8084)	0.0596 (10925)
wR (reflections)	0.0798 (7491)	0.1815 (6406)	0.0752 (8084)	0.1123 (10925)
<i>GOF</i>	1.048	1.003	1.017	1.196
<i>N</i> _{par}	549	569	606	667