

## Supporting Information

### Facile synthesis of P-doped ZnIn<sub>2</sub>S<sub>4</sub> with enhanced visible-light-driven photocatalytic hydrogen production

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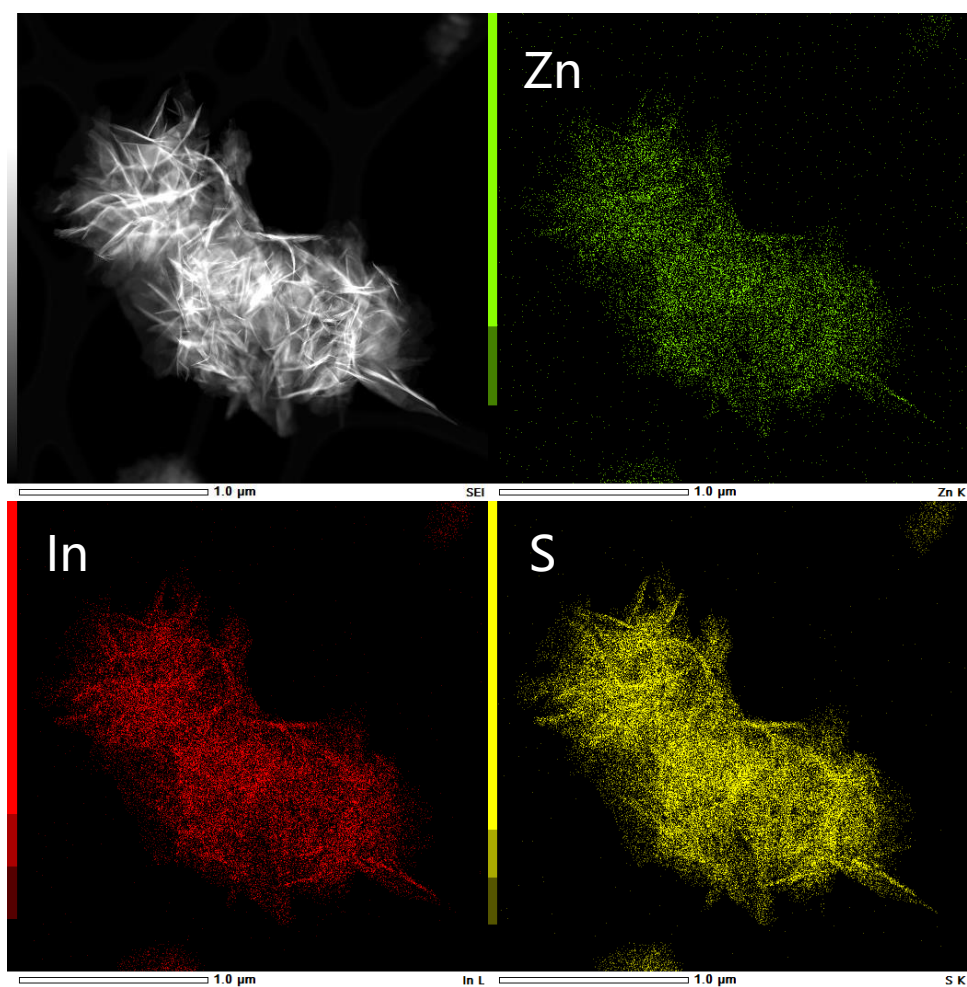


Figure S1. Energy dispersive spectrum elemental mapping of the pure  $\text{ZnIn}_2\text{S}_4$ .

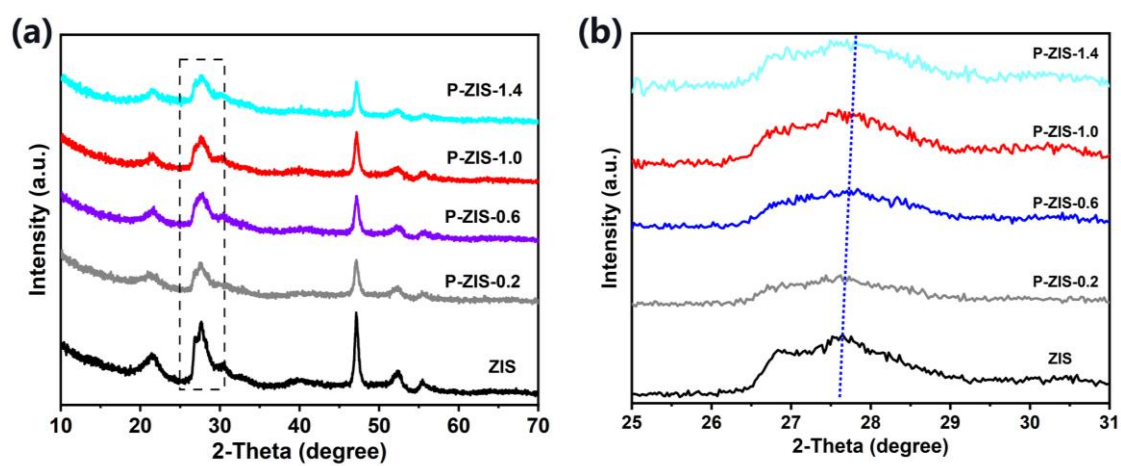


Figure S2. XRD patterns of the as-prepared photocatalysts.

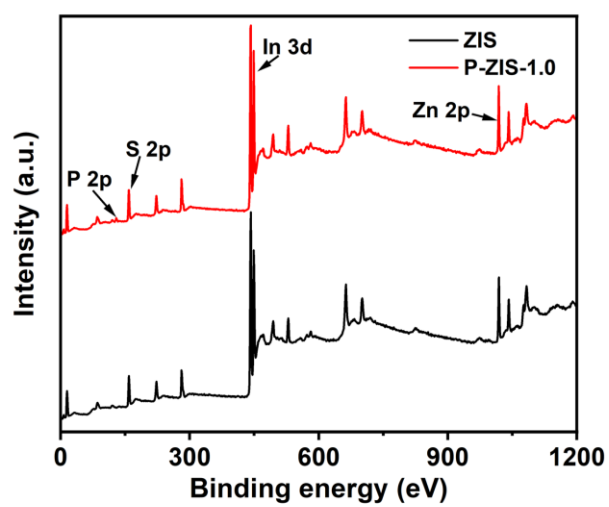


Figure S3. High-resolution XPS survey spectra of ZIS and P-ZIS-1.0.

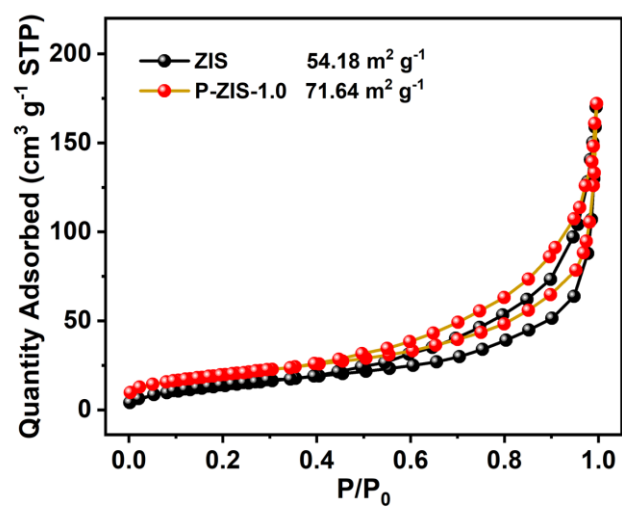


Figure S4. BET spectra of ZIS and P-ZIS-1.0.

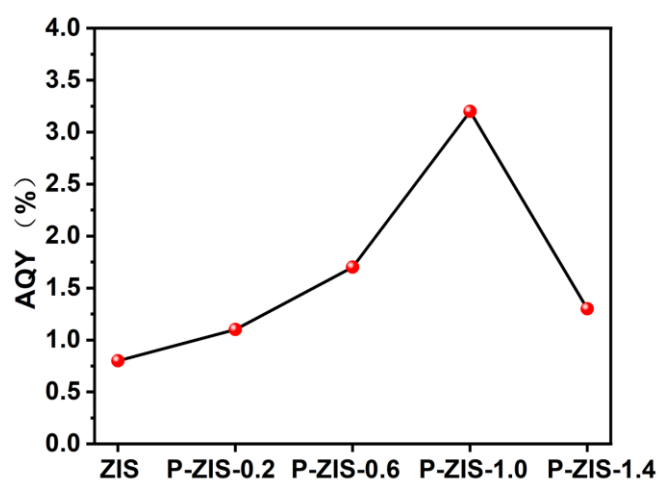


Figure S5. AQY (%) of different samples. Reaction conditions: same as photocatalytic activity test.

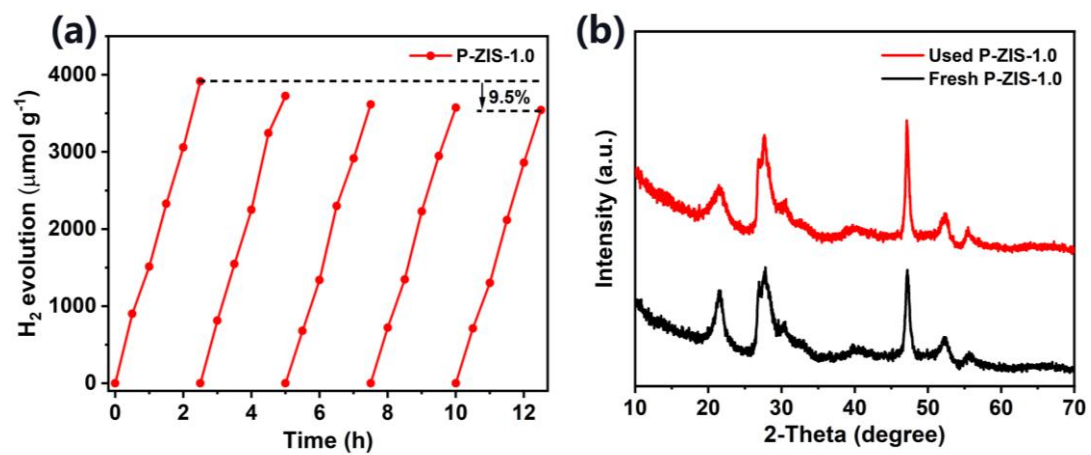


Figure S6. (a) Recycling photocatalytic H<sub>2</sub> generation tests over P-ZIS-1.0. (b) XRD of the fresh and used P-ZIS-1.0 samples.

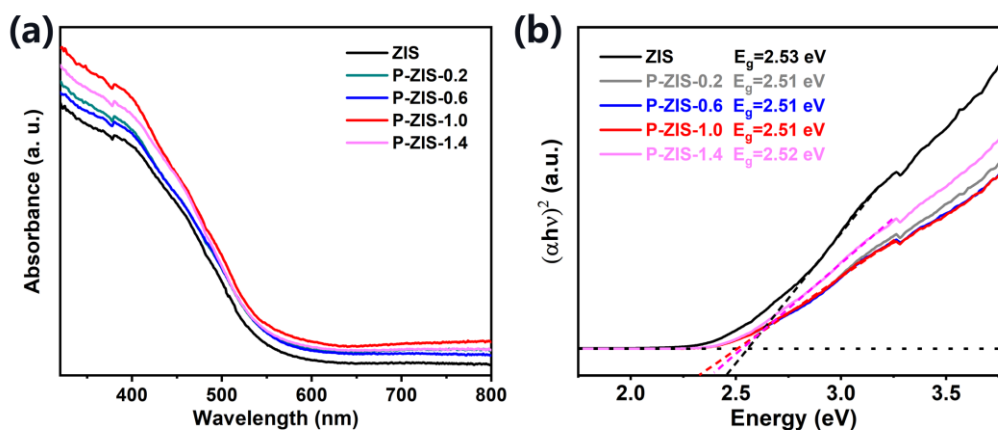


Figure S7. (a) UV-vis DRS of the as-prepared P-ZIS; (b) The band gap value curves of the as-prepared photocatalysts obtained from UV-vis DRS spectra.

Table S1. Comparison of the performance of P-ZIS-1.0 with other recently reported improved materials based on ZIS.

Catalysts	Condition	H <sub>2</sub> generation ( $\mu\text{mol g}^{-1} \text{h}^{-1}$ )	Year	Ref.
P-ZIS-1.0	Na <sub>2</sub> S/Na <sub>2</sub> SO <sub>3</sub>	1556.6	2023	This work
N-ZIS	Na <sub>2</sub> S/Na <sub>2</sub> SO <sub>3</sub>	1575.7	2023	[1]
ZIS/Ti <sub>3</sub> C <sub>2</sub>	TEOA (10 vol%)	978.7	2022	[2]
ZIS/Ti <sub>3</sub> C <sub>2</sub> O <sub>x</sub>	TEOA (10 vol%)	363.0	2023	[3]
20%ZIS/CN	TEOA (10 vol%)	75.2	2023	[4]
ZIS/Ce-U66	Na <sub>2</sub> S/Na <sub>2</sub> SO <sub>3</sub>	500	2023	[5]
ZIS/NiMoO <sub>4</sub>	TEOA (10 vol%)	140.6	2023	[6]
BiOCl@ZIS	TEOA (10 vol%)	674	2023	[7]
Pt@UiO-66-NH <sub>2</sub> @ZIS	TEOA (10 vol%)	850	2023	[8]
NiCo <sub>2</sub> S <sub>4</sub> /ZIS	TEOA (10 vol%)	221.8	2023	[9]
Bi/ZIS	TEOA (10 vol%)	3658.8	2023	[10]
Co <sub>9</sub> S <sub>8</sub> /N, S-CNTs-ZIS	K <sub>3</sub> [Fe(CN) <sub>6</sub> ]/ K <sub>4</sub> [Fe(CN) <sub>6</sub> ] /KCl	2409.2	2023	[11]
W <sub>18</sub> O <sub>49</sub> /ZIS	C <sub>3</sub> H <sub>6</sub> O <sub>3</sub>	902.57	2022	[12]
ReS <sub>2</sub> /ZIS	TEOA(10 vol%)	3092.9	2023	[13]

Calculation method for Fermi level, valence band, and conduction band <sup>[14]</sup>

UPS was used to determine the ionization energy value, the cutoff ( $E_{\text{cutoff}}$ ), and the edge ( $E_{\text{edge}}$ ) energy of samples. The ionization energy values (equivalent to the work function ( $\phi$ ) of metallic materials) of samples were calculated by the following equation. According to the relationship between the vacuum energy ( $E_{\text{ve}}$ ) and the normal electrode potential ( $E_{\text{n}}$  vs NHE), the valence band energy ( $E_{\text{VB}}$  vs NHE) and conduction band energy ( $E_{\text{CB}}$  vs NHE) of samples were determined as follows:

$$(h\nu = \phi + E_{\text{cutoff}})$$

$$(E_{\text{VB}} = E_{\text{edge}} + \phi)$$

$$(E_{\text{CB}} = E_{\text{VB}} - E_{\text{g}})$$

$$(-E_{\text{ve}} = E_{\text{n}} + 4.4 \text{ eV})$$

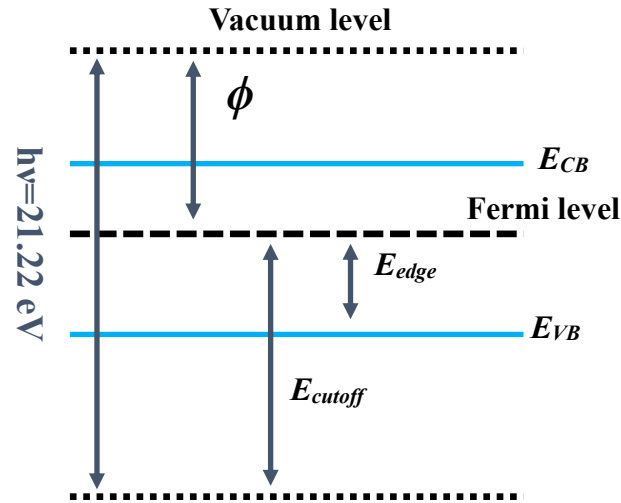


Figure S8. Schematic illustration of the energy level calculation from UPS <sup>[15]</sup>.

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