

Combination NIPS/TIPS Synthesis of α -Fe₂O₃ and α/γ -Fe₂O₃ doped PVDF Composite for Efficient Piezocatalytic Degradation of Rhodamine B

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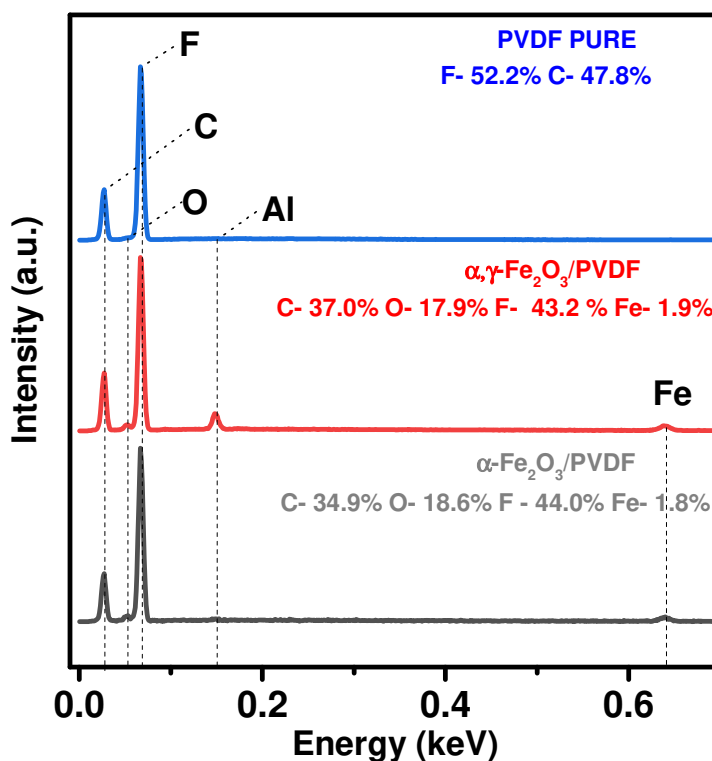


Figure S1. EDX spectra of samples

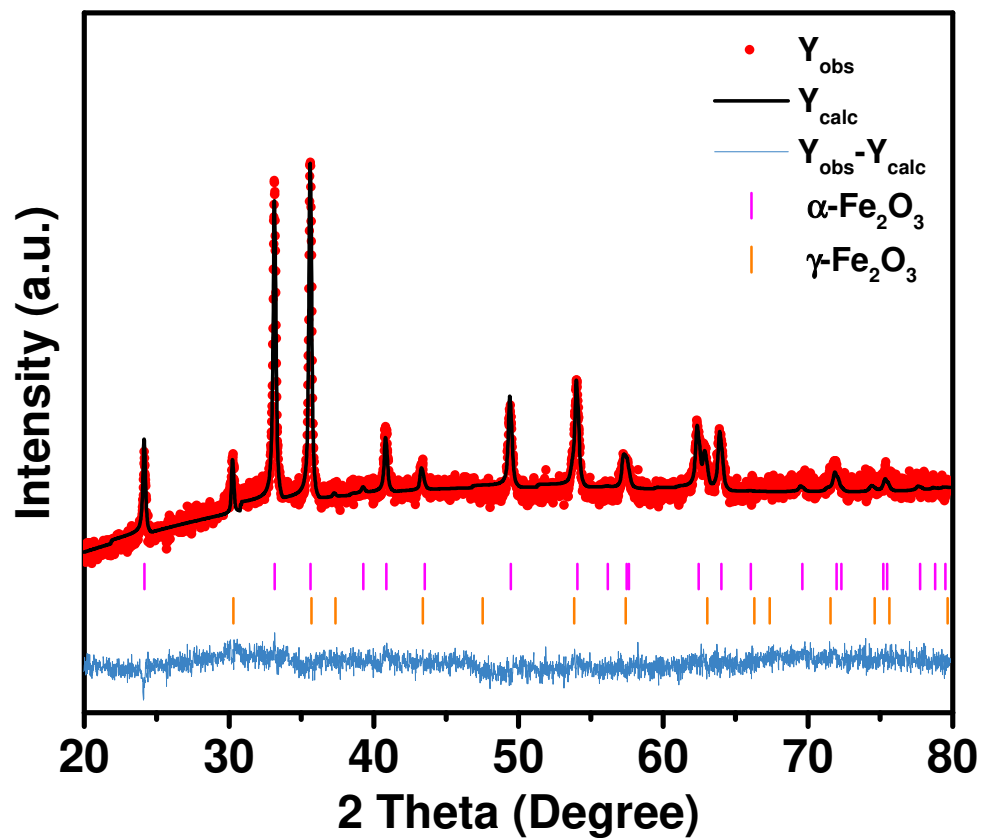


Figure S2. XRD patterns of $\alpha\text{-Fe}_2\text{O}_3$ and $\alpha,\gamma\text{-Fe}_2\text{O}_3$ with standards.

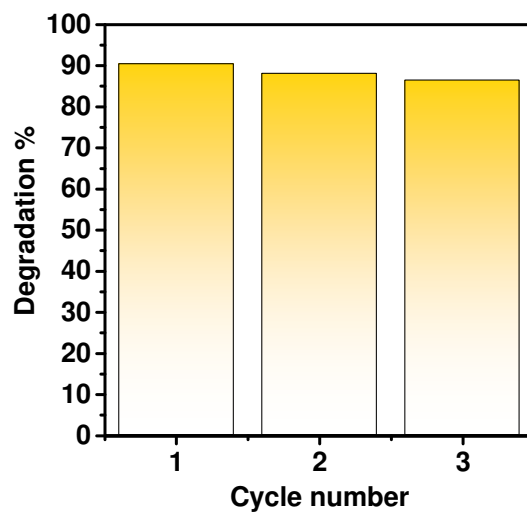


Figure S3. Cycling tests of the $\alpha\text{-Fe}_2\text{O}_3/\text{PVDF}$ membrane in the degradation of RhB.

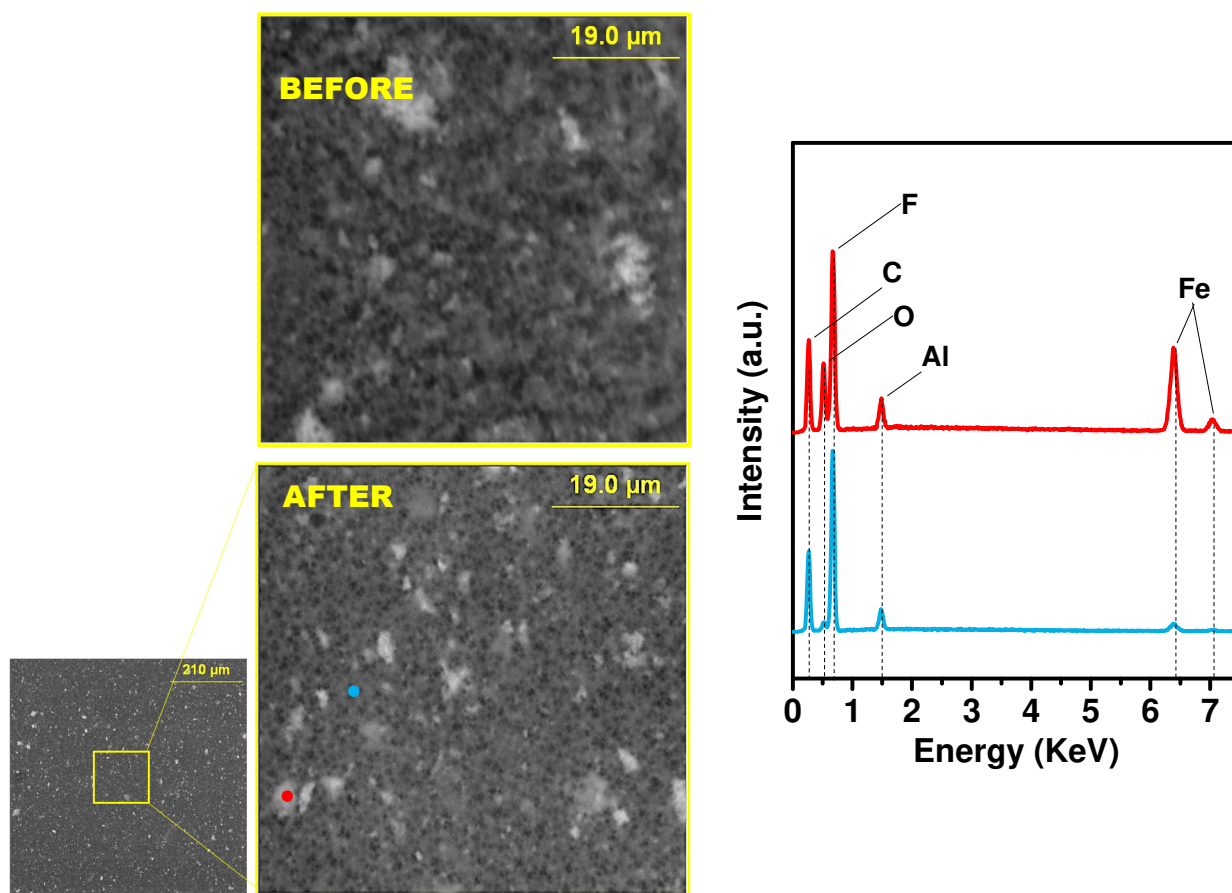


Figure S4. SEM images of α -Fe₂O₃/PVDF before and after 3 cycles piezocatalytic experiment and selected area EDX spectra

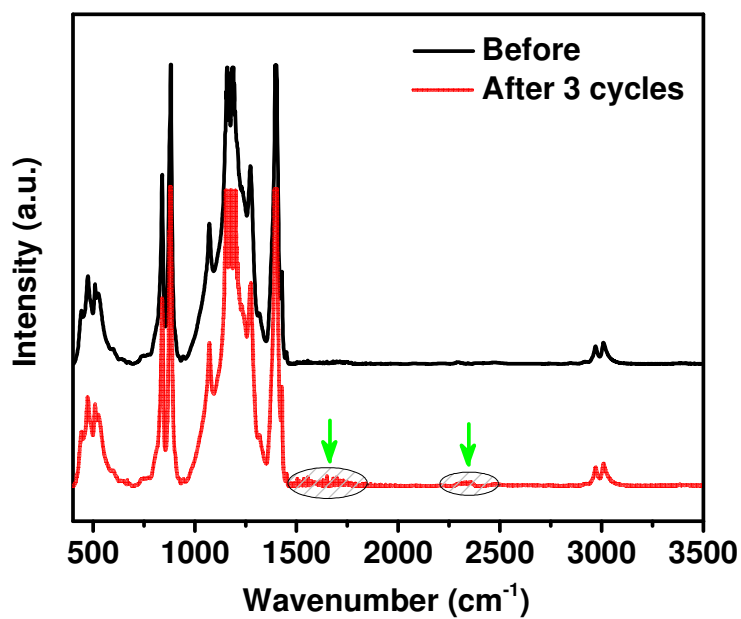


Figure S5. FTIR spectra of α -Fe₂O₃/PVDF before and after 3 cycles piezocatalytic experiment.

Table. S1. Previously reported work and its comparison with our present work in field Piezocatalytic properties of PVDF based composites.

Materials	Pollutants	Time (min)	Mechanical source	Degradation, %	Rate constant, min ⁻¹	References
E-MoS ₂ /PVD F EFMs 10 wt % 100 mg	OTC C= 20 mg/L V= 100 mL	24	US v= 20 kHz	93.05	0.09124	1
Ag@LiNbO ₃ /PVDF 5 wt % 2.5 cm Diameter	Rhodamine B C= 5 mg/L V= 10 mL	120	Ultrasonicator W= 70 W v= 40 kHz	~80	-	2
CBO/PVDF (1:1 mass) ?	Rhodamine B C=10 mg·L ⁻¹ V=?	10	Ultrasonic instrument W = 100 W	99.9	-	3
MoS ₂ - PVDF 10 wt% 2×2 cm ² thickness 50 μm	Rhodamine B C= 10 ppm V= 10 mL	20	RS Pro Ultrasonic Cleaner W=100 W	>90	0.21	4
PVDF/ZnS nO ₃ /MoS ₂ 20 wt % ?	MB C= 5 mg/L V=?	4	Ultrasonicator W= 50 W v= 40 kHz	100	-	5
BTO-PDMS (25 wt %) ?	Rhodamine B C= 5 mg/L V= 40 mL	120	Ultrasonic machine W= 400W v= 40 kHz	~94	0.02254	6
CNT/PVDF (SCP, 0.015 g)	Rhodamine B C= 5 mg/L V= 15 mL	120	US W= 240 W	≥ 95	-	7
Bi ₂ ZnB ₂ O ₇ - Polyacrylo nitrile (BBZO- PAN)	MB C= 5 mg/L V= 10 mL	180	Ultrasonicator (Labman) W=150 W v= 40 kHz	37	2.1*10 ⁻³	8

1.5×1.5 cm ²						
α- Fe ₂ O ₃ /PVD F (2wt%) 3×1 cm ² thickness ~110 μm	Rhodamine B (RhB) C= 8 mg/L V= 20 mL	60	US bath W=120 W 40 KHz	90	0.036	This work

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