

Supporting Information

Electrocatalytic Properties of Quasi-2D Oxides $\text{LaSrMn}_{0.5}\text{M}_{0.5}\text{O}_4$ ($\text{M} = \text{Co}, \text{Ni}, \text{Cu}, \text{and Zn}$) for Hydrogen and Oxygen Evolution Reactions

Kinithi M. K. Wickramaratne^a, Farshid Ramezanipour^{a,*}

^aDepartment of Chemistry, University of Louisville, Louisville, Kentucky 40292, USA

*Corresponding author. Email: farshid.ramezanipour@louisville.edu, Phone: (502) 852-7061

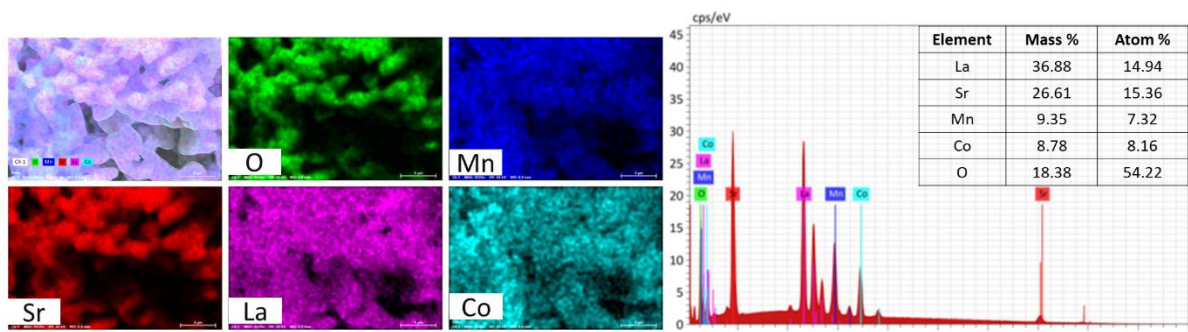


Figure S1. Scanning electron microscopy (SEM) images and the corresponding EDS mapping for $\text{LaSrMn}_{0.5}\text{Co}_{0.5}\text{O}_4$.

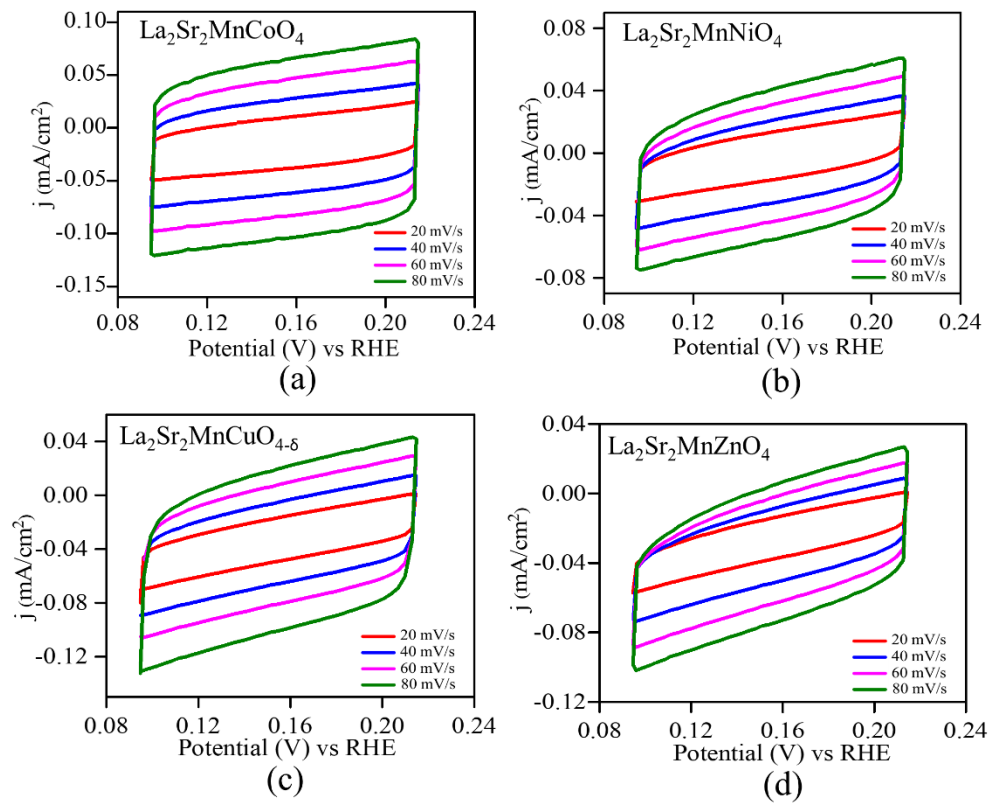


Figure S2. Cyclic voltammetry data in the non-faradaic region in 0.5 M H₂SO₄.

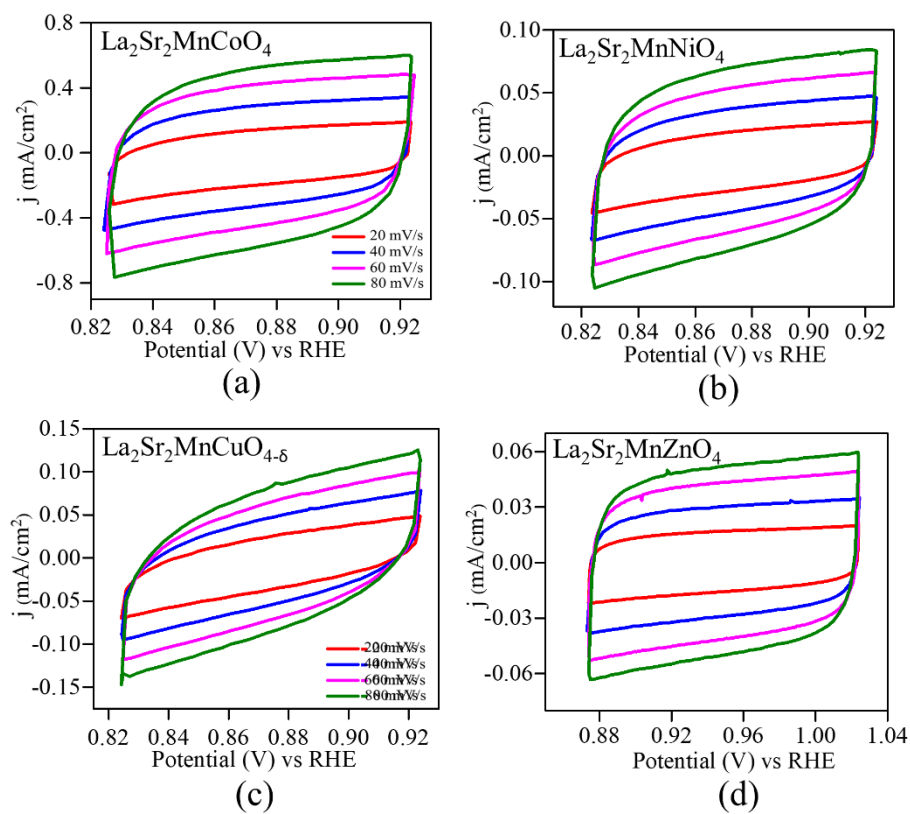


Figure S3. Cyclic voltammetry data in the non-faradaic region in 1 M KOH.

Table S1. Refined structural parameters for $\text{LaSrMn}_{0.5}\text{Ni}_{0.5}\text{O}_4$ at room temperature using powder X-ray diffraction data. Space group: $I4/mmm$, $a = 3.8409(1) \text{ \AA}$, $c = 12.5694(3) \text{ \AA}$, $R_p = 0.0529$, $wR_p = 0.0714$.

Atom	x	y	z	Occupancy	Multiplicity	$U_{\text{iso}} (\text{\AA}^2)$
La	0	0	0.3597(1)	0.5	4	0.0283(5)
Sr	0	0	0.3597(1)	0.5	4	0.0283(5)
Mn	0	0	0	0.5	2	0.0224(9)
Ni	0	0	0	0.5	2	0.0224(9)
O1	0.5	0	0	1	4	0.028(2)
O2	0	0	0.1604(6)	1	4	0.032(2)

Table S2. Refined structural parameters for $\text{LaSrMn}_{0.5}\text{Cu}_{0.5}\text{O}_{4.8}$ at room temperature using powder X-ray diffraction data. Space group: $I4/mmm$, $a = 3.8139(1) \text{ \AA}$, $c = 12.8459(3) \text{ \AA}$, $R_p = 0.0555$, $wR_p = 0.0721$.

Atom	x	y	z	Occupancy	Multiplicity	$U_{\text{iso}} (\text{\AA}^2)$
La	0	0	0.35906(7)	0.5	4	0.0128(5)
Sr	0	0	0.35906(7)	0.5	4	0.0128(5)
Mn	0	0	0	0.5	2	0.0101(9)
Cu	0	0	0	0.5	2	0.0101(9)
O1	0.5	0	0	0.98125	4	0.017(2)
O2	0	0	0.1659(5)	0.98125	4	0.036(2)

Table S3. Refined structural parameters for $\text{LaSrMn}_{0.5}\text{Zn}_{0.5}\text{O}_4$ at room temperature using powder X-ray diffraction data. Space group: $I4/mmm$, $a = 3.8628(1) \text{ \AA}$, $c = 12.6301(1) \text{ \AA}$, $R_p = 0.0549$, $wR_p = 0.0706$.

Atom	x	y	z	Occupancy	Multiplicity	$U_{\text{iso}} (\text{\AA}^2)$
La	0	0	0.35982(7)	0.5	4	0.0170(5)
Sr	0	0	0.35982(7)	0.5	4	0.0170(5)
Mn	0	0	0	0.5	2	0.0200(8)
Zn	0	0	0	0.5	2	0.0200(8)
O1	0.5	0	0	1	4	0.036(2)
O2	0	0	0.1662(5)	1	4	0.034(2)