

## Catalytic Ozonation of Reactive Black 5 in Aqueous Solution Using Iron-Loaded Dead Leaves Ash for Wastewater Remediation

Latif Hussain<sup>1</sup>, Farhan Javed<sup>1</sup>, Muhammad Wasim Tahir<sup>1</sup>, Hafiz Muhammad Shahzad Munir <sup>\*2</sup>,  
Amir Ikhlaz<sup>3</sup>, Anna Wołowicz<sup>\*4</sup>

<sup>1</sup> Department of Chemical Engineering, University of Engineering and Technology, Lahore, Pakistan; latifshifai@gmail.com (L.H.), farhan.javed@uet.edu.pk (F.J.), wasim.tahir@uet.edu.pk (M.W.T)

<sup>2</sup> Department of Chemical Engineering, Khawaja Fareed University of Engineering and Information Technology, Rahim Yar Khan, Pakistan

<sup>3</sup> Institute of Environmental Engineering and Research, University of Engineering and Technology, Lahore, Pakistan; aamirikhlaq@uet.edu.pk (A.I.)

<sup>4</sup> Department of Inorganic Chemistry, Faculty of Chemistry, Institute of Chemical Sciences, Maria Curie Skłodowska University, Maria Curie-Skłodowska Square 2, 20-031 Lublin, Poland

\* Correspondence: anna.wolowicz@mail.umcs.pl; Tel.: +48-81-537-57-27 (A.W.); shahzad.munir@kfueit.edu.pk (H.M.S.M.).

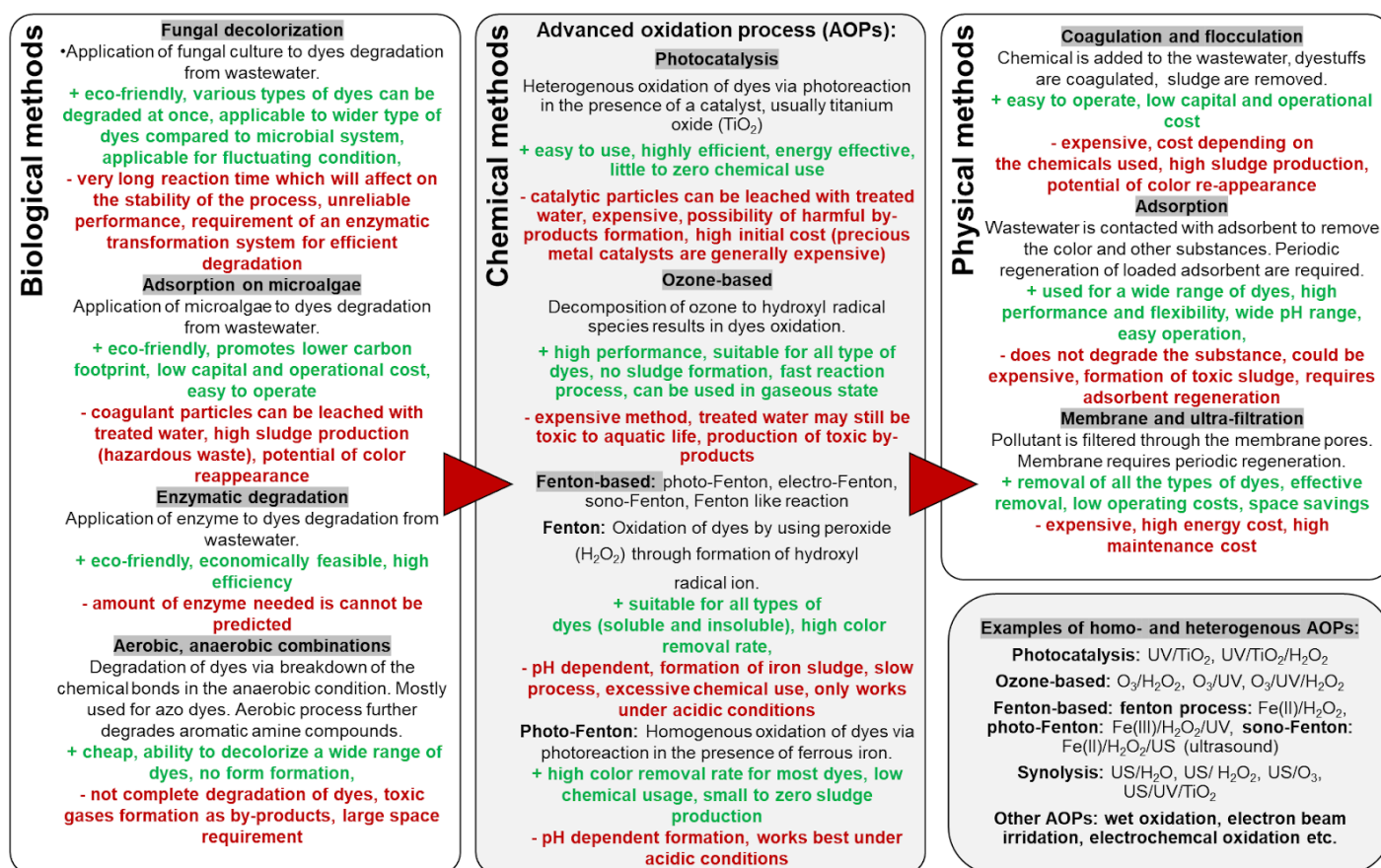
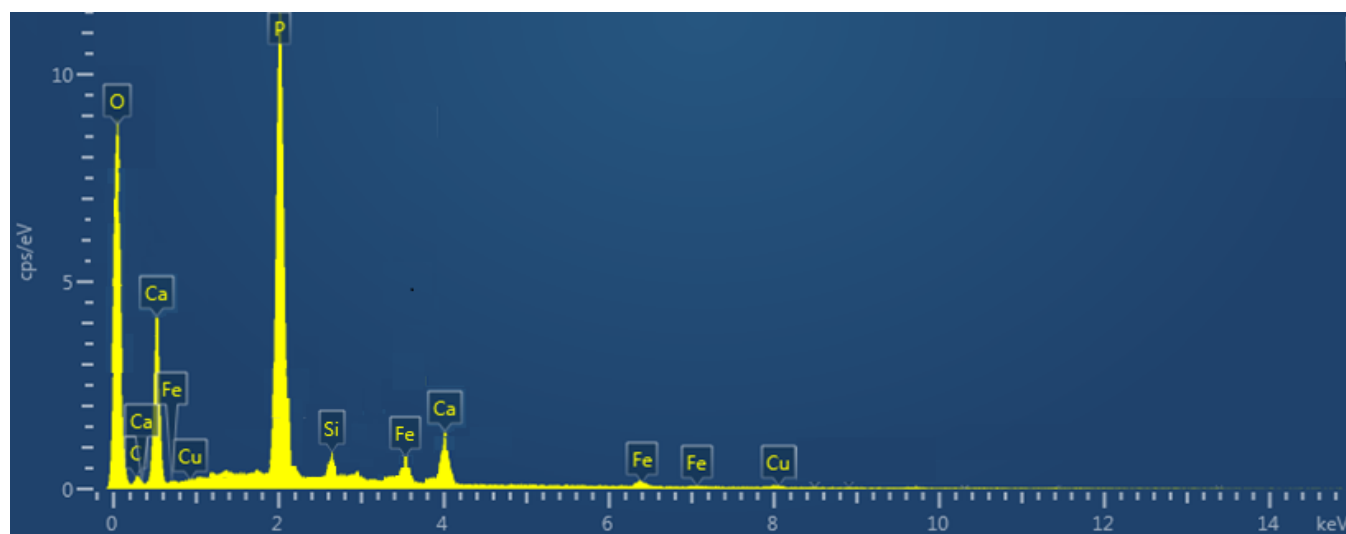


Figure S1. Comparison of advantageous and disadvantageous of dyes containing wastewater decolorization [14-15].

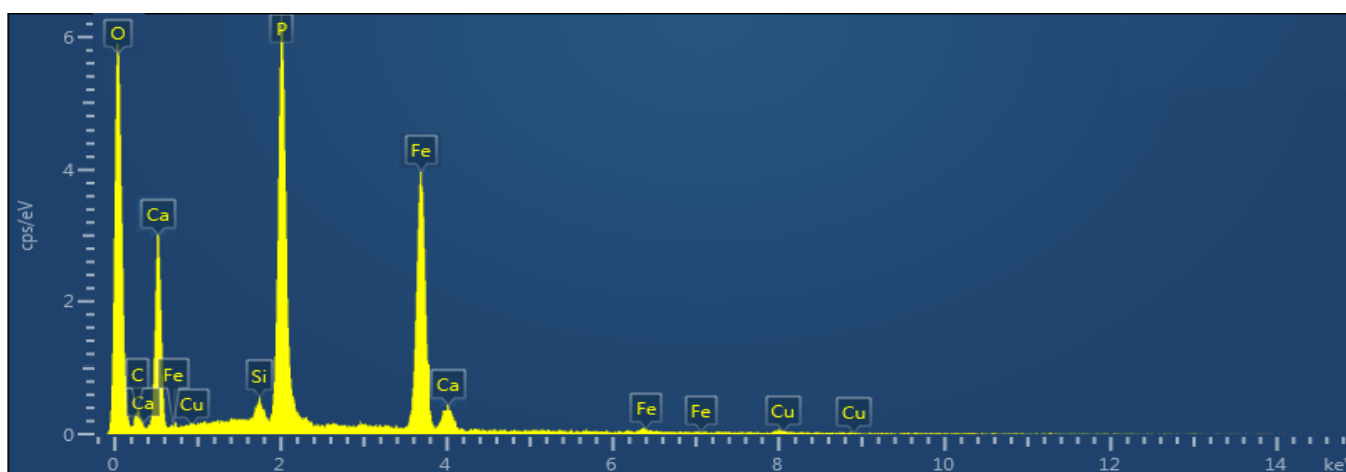
**Table S1.** The catalytic ozonation process for Reactive Black 5 dye containing wastewater decolorization.

Method	Dye	Results, removal efficiency and other parameters	Ref.
O <sub>3</sub> , O <sub>3</sub> /UV, O <sub>3</sub> /UV/persulfate (PS), and O <sub>3</sub> /Ce–Co–O catalyst	Reactive Black 5 C <sub>0</sub> =100–1000 mg/L	O <sub>3</sub> /Ce–Co–O: <sup>1</sup> %R <sub>TOC</sub> = 96% (60 LPH, pH 7, C <sub>0</sub> =100 mg/L, catalyst dose=1 g/L, t <sub>r</sub> =80 min)	
		O <sub>3</sub> /UV/persulfate: %R <sub>TOC</sub> = 90% (60 LPH, pH 12, C <sub>0</sub> =100 mg/L, UV intensity=66 W, TOC:PS=1:40, t <sub>r</sub> =80 min)	[29]
catalytic ozonation O <sub>3</sub> /Ag–Co–O catalyst: silver–cobalt composite oxide	Reactive Black 5 C <sub>0</sub> =100 mg/L	O <sub>3</sub> without catalyst: %R <sub>TOC</sub> =34% O <sub>3</sub> /Ag–Co–O: %R <sub>TOC</sub> =75% (ozone dose=30 mg/min, pH 7, t <sub>r</sub> =80 min, catalyst dose=0.5 g/L)	[30]
electro-flocculation, electro-flocculation with catalyst and catalytic ozonation in combination with electro-flocculation catalyst: zeolites modified with CuMn <sub>2</sub> O <sub>4</sub> /gC <sub>3</sub> N <sub>4</sub>	Reactive Black 5 C <sub>0</sub> =30–90 mg/L	electro-flocculation: 31.18%, electro-flocculation with catalyst: 39.27% catalytic ozonation with electro-flocculation: 90.31% (ozone dose=1 mg/min, pH 10, C <sub>0</sub> =30 mg/L, catalyst dose=1 mg, t <sub>r</sub> =30 min, voltage=10 V)	[31]
O <sub>3</sub> assisted electro-coagulation	Reactive Black 5 C <sub>0</sub> =100–400 mg/L	R <sub>TOC</sub> =94%, %R <sub>COD</sub> =60% (C <sub>0</sub> =100 mg/L, pH 5.5, current density=10 mA/cm <sup>2</sup> , salt concentration=5000 mg/L, T=20°C, ozone dose=20 mL/min, interelectrode distance of 1 cm)	[32]

%R<sub>TOC</sub> - the total organic carbon (TOC) removal efficiency, %R<sub>COD</sub> - the chemical oxygen demand (COD) removal efficiency, C<sub>0</sub> - the initial dye concentration, t<sub>r</sub> - the reaction time, T - temperature



(a)



(b)

**Figure S2.** EDX spectra of (a) DLA and (b) Fe-DLA.