

## Supplementary Information

# Accelerated Design for Perovskite-Oxide-Based Photocatalysts Using Machine Learning Techniques

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**Table S1** Overview of ML applications in ABO<sub>3</sub>-type perovskites

No.	Target properties	Algorithms	Ref
1	Thermodynamic phase stability	ET and KRR	[1]
2	Stable and metastable	GBDT	[2]
3	Formability and cubic structure stability	RF and Gradient boosting tree	[3]
4	Formation energy and bandgap	GBR	[4]
5	Curie temperature	SVR	[5]
6	Curie temperature	RR, SVM and ERT	[6]
7	Néel temperature	SVM	[7]
8	Maximum magnetic entropy change	GPR	[8]
9	Ionic conductivity	SVM	[9]
10	Specific surface area	SVM	[10]
11	The nature of band gap	RF	[11]
12	Thermodynamic stable	PSO-SVR and SVR	[12]
13	Crystal structure	Boosting algorithms, KNN and XGBoost	[13]
14	Perovskite structures	Quantum Machine Learning	[14]
15	Formability of perovskite structures	GBDT	[15]
16	Formation energy	DTR, GBRT, RFR and ETR	[16]
17	Defect formation energy	RFR	[17]
18	Thermal expansion coefficient	SVM	[18]
19	Oxygen ionic conductivities	GPR	[19]
20	Formability and stability	RF	[20]
21	Experimental design	ANN, Gaussian process and SVR	[21]
22	Stability	PCA and LFA	[22]
23	Band gaps and Band-edge positions	GBDT, LRR and KRR	[23]
24	Hydrogen production rate	GBR, BPANN and SVR	[24]

**Table S2** The dataset including 102 ABO<sub>3</sub>-type perovskite samples

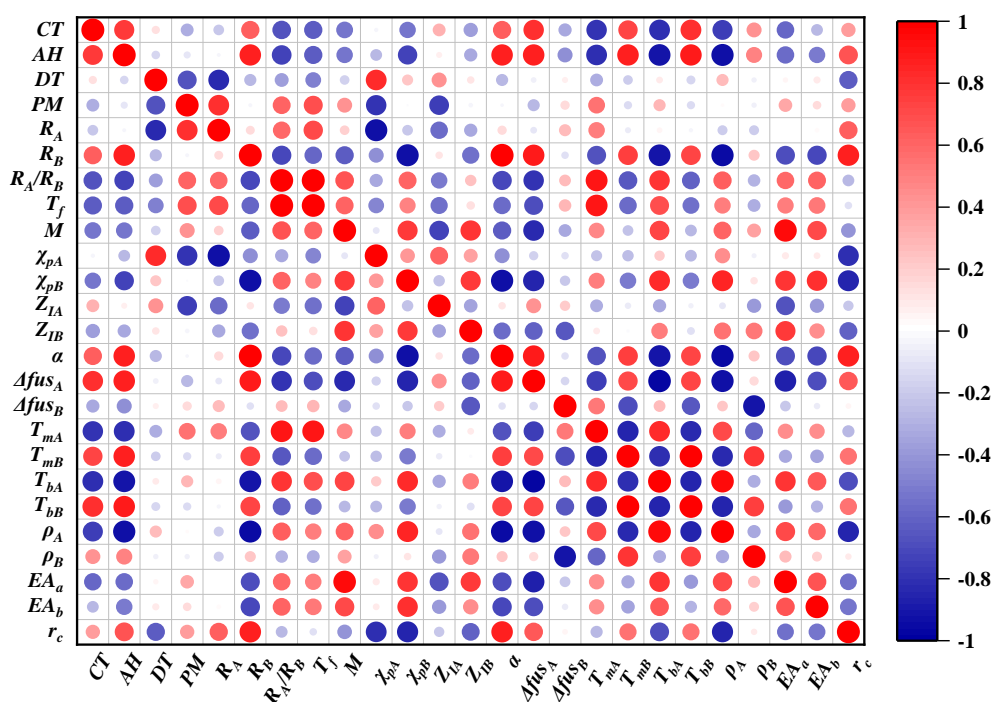
No.	Molecular formulas	<i>SSA</i> (m <sup>2</sup> g <sup>-1</sup> )	<i>CT</i> (°C)	<i>AH</i> (h)	<i>DT</i> (°C)	PM	Ref.
1	ZnTiO <sub>3</sub>	1.05	900	2	120	1	[25]
2	LaFeO <sub>3</sub>	1.08	900	4	150	3	[26]
3	LaCo <sub>0.94</sub> Mg <sub>0.06</sub> O <sub>3</sub>	19	750	4	110	3	[27]
4	LaCo <sub>0.90</sub> Mg <sub>0.10</sub> O <sub>3</sub>	21	750	4	110	3	[27]
5	LaCo <sub>0.80</sub> Mg <sub>0.20</sub> O <sub>3</sub>	22	750	4	110	3	[27]
6	La <sub>0.5</sub> Bi <sub>0.2</sub> Ba <sub>0.2</sub> Mn <sub>0.1</sub> FeO <sub>3</sub>	20.63	600	4	120	3	[28]
7	La <sub>0.5</sub> Bi <sub>0.2</sub> Ba <sub>0.2</sub> Mn <sub>0.1</sub> FeO <sub>3</sub>	12.46	700	4	120	3	[28]
8	La <sub>0.5</sub> Bi <sub>0.2</sub> Ba <sub>0.2</sub> Mn <sub>0.1</sub> FeO <sub>3</sub>	5.91	800	4	120	3	[28]
9	La <sub>0.5</sub> Bi <sub>0.2</sub> Ba <sub>0.2</sub> Mn <sub>0.1</sub> FeO <sub>3</sub>	4.19	900	4	120	3	[28]
10	LaCrO <sub>3</sub>	3.95	600	5	100	3	[29]
11	LaMg <sub>0.2</sub> Cr <sub>0.8</sub> O <sub>3</sub>	8.42	600	5	100	3	[29]
12	LaMg <sub>0.4</sub> Cr <sub>0.6</sub> O <sub>3</sub>	29.71	600	5	100	3	[29]
13	LaMg <sub>0.6</sub> Cr <sub>0.4</sub> O <sub>3</sub>	18.41	600	5	100	3	[29]
14	LaMg <sub>0.8</sub> Cr <sub>0.2</sub> O <sub>3</sub>	14.46	600	5	100	3	[29]
15	PrFeO <sub>3</sub>	10.88	700	5	90	3	[30]
16	LaFe <sub>0.9</sub> Co <sub>0.1</sub> O <sub>3</sub>	51.2	750	10	110	3	[31]
17	LaFe <sub>0.1</sub> Co <sub>0.9</sub> O <sub>3</sub>	42.8	750	10	110	3	[31]
18	SrTiO <sub>3</sub>	16.4	650	10	110	3	[32]
19	La <sub>0.002</sub> Sr <sub>0.998</sub> TiO <sub>3</sub>	19.7	650	10	110	3	[32]
20	La <sub>0.01</sub> Sr <sub>0.99</sub> TiO <sub>3</sub>	24.1	650	10	110	3	[32]
21	La <sub>0.02</sub> Sr <sub>0.98</sub> TiO <sub>3</sub>	23.2	650	10	110	3	[32]
22	LaNiO <sub>3</sub>	14.1	600	2	130	2	[33]
23	LaNiO <sub>3</sub>	12.7	700	2	130	2	[33]
24	LaNiO <sub>3</sub>	11.8	800	2	130	2	[33]
25	LaNiO <sub>3</sub>	6.5	900	2	130	2	[33]
26	LaNiO <sub>3</sub>	12.2	600	6	130	2	[33]
27	LaFeO <sub>3</sub>	21.9	500	4	120	3	[34]
28	LaFeO <sub>3</sub>	15.37	600	4	120	3	[34]
29	LaFeO <sub>3</sub>	10.07	700	4	120	3	[34]
30	LaFeO <sub>3</sub>	5.24	800	4	120	3	[34]
31	SrTiO <sub>3</sub>	12.24	700	4	80	1	[35]
32	SrTiO <sub>3</sub>	10	700	4	80	1	[36]
33	LaCuO <sub>3</sub>	5	800	15	100	3	[37]
34	LaCu <sub>0.53</sub> Ni <sub>0.47</sub> O <sub>3</sub>	8.2	800	15	100	3	[37]
35	La <sub>0.8</sub> Sr <sub>0.2</sub> Ni <sub>0.8</sub> Cu <sub>0.2</sub> O <sub>3</sub>	9	700	5	100	3	[38]
36	La <sub>0.8</sub> Sr <sub>0.2</sub> Ni <sub>0.8</sub> Fe <sub>0.2</sub> O <sub>3</sub>	13.9	700	5	100	3	[38]
37	La <sub>0.8</sub> Sr <sub>0.2</sub> Ni <sub>0.8</sub> Bi <sub>0.2</sub> O <sub>3</sub>	12.1	700	5	100	3	[38]
38	LaNiO <sub>3</sub>	7.5	750	5	100	3	[39]
39	LaNi <sub>0.8</sub> Fe <sub>0.2</sub> O <sub>3</sub>	7.9	750	5	100	3	[39]
40	LaNi <sub>0.6</sub> Fe <sub>0.4</sub> O <sub>3</sub>	9.6	750	5	100	3	[39]
41	LaNi <sub>0.3</sub> Fe <sub>0.7</sub> O <sub>3</sub>	6.8	750	5	100	3	[39]
42	LaFeO <sub>3</sub>	15.1	750	5	100	3	[39]
43	LaFeO <sub>3</sub>	25.8	500	2	130	2	[40]
44	LaFeO <sub>3</sub>	22.55	600	2	130	2	[40]

No.	Molecular formulas	<i>SSA</i> (m <sup>2</sup> g <sup>-1</sup> )	<i>CT</i> (°C)	<i>AH</i> (h)	<i>DT</i> (°C)	PM	Ref.
45	LaFeO <sub>3</sub>	20.04	700	2	130	2	[40]
46	LaFeO <sub>3</sub>	8.5	800	2	130	2	[40]
47	GaFeO <sub>3</sub>	56	600	8	300	1	[41]
48	GaFeO <sub>3</sub>	38	700	8	300	1	[41]
49	GaFeO <sub>3</sub>	10	900	8	300	1	[41]
50	NaTaO <sub>3</sub>	2.25	900	12	120	3	[42]
51	Na <sub>0.995</sub> La <sub>0.005</sub> TaO <sub>3</sub>	3.83	900	12	120	3	[42]
52	Na <sub>0.985</sub> La <sub>0.015</sub> TaO <sub>3</sub>	4.55	900	12	120	3	[42]
53	Na <sub>0.98</sub> La <sub>0.02</sub> TaO <sub>3</sub>	4.54	900	12	120	3	[42]
54	Na <sub>0.95</sub> La <sub>0.05</sub> TaO <sub>3</sub>	4.36	900	12	120	3	[42]
55	SrTiO <sub>3</sub>	16.2	900	4	80	3	[43]
56	Ca <sub>0.98</sub> Ag <sub>0.01</sub> La <sub>0.01</sub> TiO <sub>3</sub>	10.61	850	10	120	3	[44]
57	Ca <sub>0.96</sub> Ag <sub>0.02</sub> La <sub>0.02</sub> TiO <sub>3</sub>	11.39	850	10	120	3	[44]
58	Ca <sub>0.94</sub> Ag <sub>0.03</sub> La <sub>0.03</sub> TiO <sub>3</sub>	12.17	850	10	120	3	[44]
59	Ca <sub>0.92</sub> Ag <sub>0.04</sub> La <sub>0.04</sub> TiO <sub>3</sub>	11.44	850	10	120	3	[44]
60	LaFeO <sub>3</sub>	9.5	700	4	90	3	[45]
61	LaFeO <sub>3</sub>	11.39	600	5	150	3	[46]
62	LaMg <sub>0.2</sub> Fe <sub>0.8</sub> O <sub>3</sub>	15.07	600	5	150	3	[46]
63	LaMg <sub>0.4</sub> Fe <sub>0.6</sub> O <sub>3</sub>	17.63	600	5	150	3	[46]
64	LaMg <sub>0.6</sub> Fe <sub>0.4</sub> O <sub>3</sub>	24.41	600	5	150	3	[46]
65	LaMgO <sub>3</sub>	10.17	600	5	150	3	[46]
66	PrCo <sub>0.7</sub> Fe <sub>0.3</sub> O <sub>3</sub>	7.9	700	2	90	3	[47]
67	PrCo <sub>0.6</sub> Fe <sub>0.4</sub> O <sub>3</sub>	8.7	700	2	90	3	[47]
68	PrCo <sub>0.5</sub> Fe <sub>0.5</sub> O <sub>3</sub>	7.5	700	2	90	3	[47]
69	PrCo <sub>0.3</sub> Fe <sub>0.7</sub> O <sub>3</sub>	9.9	700	2	90	3	[47]
70	PrCo <sub>0.1</sub> Fe <sub>0.9</sub> O <sub>3</sub>	9.7	400	2	90	3	[47]
71	PrCo <sub>0.1</sub> Fe <sub>0.9</sub> O <sub>3</sub>	15.2	600	2	90	3	[47]
72	PrCo <sub>0.1</sub> Fe <sub>0.9</sub> O <sub>3</sub>	15	700	2	90	3	[47]
73	PrCo <sub>0.1</sub> Fe <sub>0.9</sub> O <sub>3</sub>	9.5	800	2	90	3	[47]
74	LaMnO <sub>3</sub>	26	600	4	120	3	[48]
75	LaFeO <sub>3</sub>	19	600	4	120	3	[48]
76	LaCoO <sub>3</sub>	16	600	4	120	3	[48]
77	LaCoO <sub>3</sub>	1.1	750	4	300	3	[49]
78	LaCo <sub>0.8</sub> Ru <sub>0.2</sub> O <sub>3</sub>	2.9	750	4	300	3	[49]
79	LaCo <sub>0.6</sub> Ru <sub>0.4</sub> O <sub>3</sub>	5	750	4	300	3	[49]
80	LaAlO <sub>3</sub>	17.1	800	2	200	3	[50]
81	LaAl <sub>0.9</sub> Ni <sub>0.1</sub> O <sub>3</sub>	6	800	2	200	3	[50]
82	LaAl <sub>0.8</sub> Ni <sub>0.2</sub> O <sub>3</sub>	7	800	2	200	3	[50]
83	LaAl <sub>0.7</sub> Ni <sub>0.3</sub> O <sub>3</sub>	6	800	2	200	3	[50]
84	La <sub>0.9</sub> Ca <sub>0.1</sub> Al <sub>0.9</sub> Ni <sub>0.1</sub> O <sub>3</sub>	17	800	2	200	3	[50]
85	La <sub>0.9</sub> Ca <sub>0.1</sub> Al <sub>0.8</sub> Ni <sub>0.2</sub> O <sub>3</sub>	15	800	2	200	3	[50]
86*	LaCoO <sub>3</sub>	17	750	4	110	1	[27]
87*	La <sub>0.5</sub> Bi <sub>0.2</sub> Ba <sub>0.2</sub> Mn <sub>0.1</sub> FeO <sub>3</sub>	27.75	500	4	120	1	[28]
88*	LaFeO <sub>3</sub>	8.5	700	3	110	3	[51]
89*	La <sub>0.005</sub> Sr <sub>0.995</sub> TiO <sub>3</sub>	22.3	650	10	110	3	[32]
90*	LaNiO <sub>3</sub>	15.1	600	4	130	2	[33]

No.	Molecular formulas	<i>SSA</i> (m <sup>2</sup> g <sup>-1</sup> )	<i>CT</i> (°C)	<i>AH</i> (h)	<i>DT</i> (°C)	PM	Ref.
91*	LaFeO <sub>3</sub>	1.09	1000	4	120	1	[34]
92*	La <sub>0.8</sub> Sr <sub>0.2</sub> Ni <sub>0.8</sub> Co <sub>0.2</sub> O <sub>3</sub>	10.3	700	5	100	3	[38]
93*	LaFeO <sub>3</sub>	5.8	900	2	130	2	[40]
94*	Na <sub>0.92</sub> La <sub>0.08</sub> TaO <sub>3</sub>	4.21	900	12	120	1	[42]
95*	CaTiO <sub>3</sub>	9.64	850	10	120	1	[44]
96*	LaMg <sub>0.8</sub> Fe <sub>0.2</sub> O <sub>3</sub>	13.32	600	5	150	3	[46]
97*	PrCo <sub>0.9</sub> Fe <sub>0.1</sub> O <sub>3</sub>	7.2	700	2	90	3	[47]
98*	PrCo <sub>0.1</sub> Fe <sub>0.9</sub> O <sub>3</sub>	19.3	500	2	90	1	[47]
99*	La <sub>0.9</sub> Ca <sub>0.1</sub> Al <sub>0.7</sub> Ni <sub>0.3</sub> O <sub>3</sub>	12	800	2	200	3	[50]
100 <sup>#</sup>	LaFeO <sub>3</sub>	7	850	4	110	3	[52]
101 <sup>#</sup>	GdCoO <sub>3</sub>	8.69	750	4	110	3	[53]
102 <sup>#</sup>	LaMnO <sub>3</sub>	25	700	3	110	3	[54]

**Table S3** The meanings of the twenty-five candidate features

No.	Meanings	Features	No.	Meanings	Features
1	Calcination Temperature (°C)	$CT$	14	Unit Cell Lattice Edge	$a$
2	Calcination Time (h)	$AH$	15	Enthalpy of fusion at the melting point of A-site (kJ/mol)	$\Delta fus_A$
3	Drying Temperature (°C)	$DT$	16	Enthalpy of fusion at the melting point of B-site (kJ/mol)	$\Delta fus_B$
4	Synthetic Mode	$PM$	17	Melting Point of A-site (°C)	$T_{mA}$
5	Ionic Radius of A-site (pm)	$R_A$	18	Melting Point of B-site (°C)	$T_{mB}$
6	Ionic Radius of B-site (pm)	$R_B$	19	Boiling Point of A-site (°C)	$T_{bA}$
7	Ratio of Ionic Radius	$R_A/R_B$	20	Boiling Point of B-site (°C)	$T_{bB}$
8	Tolerance Factor	$T_f$	21	Density of A-site (g/cm <sup>3</sup> )	$\rho_A$
9	Molecular Mass (g/mol)	$M$	22	Density of B-site (g/cm <sup>3</sup> )	$\rho_B$
10	Pauling Electronegativity of A-site	$\chi_{pA}$	23	Electron Affinity of A-site (kJ/mol)	$EA_a$
11	Pauling Electronegativity of B-site	$\chi_{pB}$	24	Electron Affinity of B-site (kJ/mol)	$EA_b$
12	Ionization Energy of A-site (eV)	$Z_{IA}$	25	Critical Radius	$r_c$
13	Ionization Energy of B-site (eV)	$Z_{IB}$			

**Figure S1** Pearson correlation matrix diagram of the twenty-five features

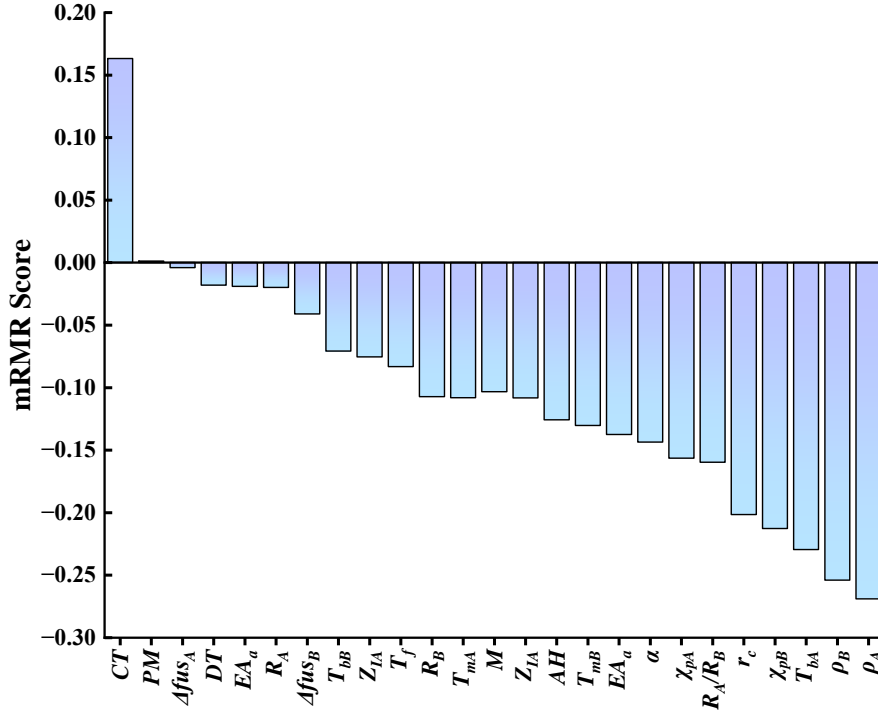


Figure S2 The mRMR scores of the twenty-five features

$$P_{CA1} = -0.259Z_{IA} + 0.012EA_a + 0.00365EA_b + 0.00134T_{mA} - 0.000692T_{mB} + 0.000495T_{bA} - 0.0105Afus_A + 0.2355\rho_A - 0.00216CT - 0.0934AH + 5.448E-05DT + 0.1657PM + 0.3918 \quad (S1)$$

$$P_{CA2} = 1.252Z_{IA} - 0.0176EA_a - 0.00437EA_b + 0.000174T_{mA} - 0.000698T_{mB} - 3.958E-05T_{bA} + 0.00197Afus_A + 0.1162\rho_A - 0.0004163CT - 0.0859AH + 0.003415DT - 0.782PM - 3.542 \quad (S2)$$

$$P_{CA3} = -0.1473Z_{IA} + 0.01277EA_a + 0.00852EA_b - 0.001796T_{mA} + 0.000290T_{mB} + 3.97E-05T_{bA} - 0.005132Afus_A + 0.07203\rho_A + 0.00192CT + 0.00353AH + 0.01098DT - 0.6485PM - 0.1763 \quad (S3)$$

$$P_{CA4} = 0.3796Z_{IA} - 0.006746EA_a + 0.01845EA_b + 0.001076T_{mA} - 0.000336T_{mB} + 6.914E-05T_{bA} + 0.00348Afus_A + 0.003075\rho_A + 0.00531CT + 0.01077AH - 0.001409DT + 0.3009PM - 7.915 \quad (S4)$$

$$P_{CA5} = 0.1268Z_{IA} - 0.00104EA_a + 0.0104EA_b - 0.000318T_{mA} + 0.0008359T_{mB} - 7.796E-05T_{bA} - 0.005807Afus_A + 0.01999\rho_A - 0.001758CT - 0.03819AH - 0.01129DT - 0.819PM + 3.624 \quad (S5)$$

$$P_{CA6} = 0.591Z_{IA} + 0.01055EA_a - 0.01066EA_b - 0.0006895T_{mA} + 0.000652T_{mB} + 1.383E-05T_{bA} - 0.002944Afus_A + 0.1368\rho_A + 0.004784CT - 0.1168AH - 0.00406DT + 0.379PM - 7.712 \quad (S6)$$

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